

# A coupled geochemical-geodynamical approach for predicting mantle melting in space and time

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## Key Points:

- We present the BDD21 peridotite-melting parameterization that can be coupled with geodynamical simulations to calculate melt composition
- BDD21 is applied to simulations of mid-oceanic ridges to replicate observed patterns of crustal thickness and melt chemistry
- Geodynamical simulations that incorporate BDD21 can be used to study melting adjacent to lithospheric steps and in other geologic settings

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**Abstract**

Geodynamical simulations underpin our understanding of upper-mantle processes, but their predictions require validation against observational data. Widely used geophysical datasets provide limited constraints on dynamical processes into the geological past, whereas under-exploited geochemical observations from volcanic lavas at Earth’s surface constitute a valuable record of mantle processes back in time. Here, we describe a new peridotite-melting parameterization, BDD21, that can predict the incompatible-element concentrations of melts within geodynamical simulations, thereby providing a means to validate these simulations against geochemical datasets. Here, BDD21’s functionality is illustrated using the Fluidity computational modelling framework, although it is designed so that it can be integrated with other geodynamical software. To validate our melting parameterization and coupled geochemical-geodynamical approach, we develop 2-D single-phase flow simulations of melting associated with passive upwelling beneath mid-oceanic ridges and edge-driven convection adjacent to lithospheric steps. We find that melt volumes and compositions calculated for mid-oceanic ridges at a range of mantle temperatures and plate-spreading rates closely match those observed at present-day ridges. Our lithospheric-step simulations predict spatial and temporal melting trends that are consistent with those recorded at intra-plate volcanic provinces in similar geologic settings. Taken together, these results suggest that our coupled geochemical-geodynamical approach can accurately predict a suite of present-day geochemical observations. Since our results are sensitive to small changes in upper-mantle thermal and compositional structure, this novel approach provides a means to improve our understanding of the mantle’s thermo-chemical structure and flow regime into the geological past.

**Plain Language Summary**

Earth’s mantle is a  $\sim 3000$  km thick layer of hot rock that lies between Earth’s crust and core. Its slow, creeping, convective motion over billions of years has been integral to Earth’s thermal, chemical, tectonic and geological evolution. However, an inability to reproduce observational constraints derived from the composition of volcanic lavas at Earth’s surface limits our capacity to validate models of mantle convection back in time. Here, we present a new framework that can predict the volume and composition of melts generated within the mantle. These predictions compare favourably with those recorded by igneous rocks at Earth’s surface in two geologic settings: mid-oceanic ridges, where plates move apart to drive decompression melting, and lithospheric steps, where instabilities associated with changes in the thickness of Earth’s rigid outermost shell generate volcanism far from plate boundaries. The approach and tools presented here will allow scientists to better understand the mantle’s past structure, dynamics, evolution and impact on Earth’s surface.

**1 Introduction**

Geodynamical models are important tools for investigating the spatio-temporal evolution of the upper mantle across a wide variety of geologic settings (e.g., Houseman et al., 1981; Turcotte & Emerman, 1983; Bercovici et al., 1989; Tackley et al., 1993; Steinberger, 2000; Schellart et al., 2007). To assess how well geodynamical model predictions replicate uppermost mantle conditions, it is necessary to validate simulations against a diverse range of observations. To date, such comparisons have focused primarily on geophysical constraints, such as gravimetric, seismic tomographic, seismic anisotropic and dynamic topographic datasets (e.g., Becker et al., 2006; Forte et al., 2010; Garel et al., 2014; Davies et al., 2019; Cooper et al., 2021; Ghelichkhan et al., 2021). These geophysical observations are confined to the present day and, hence, offer limited information on the thermo-chemical structure of the uppermost mantle into the geological past.

62 To address this shortcoming, a growing number of studies incorporate geochem-  
63 ical tracers, such as Nd isotopes, when modeling the mantle’s evolutionary pathway over  
64 billions of years (e.g., Nakagawa et al., 2010; van Heck et al., 2016; Barry et al., 2017;  
65 R. E. Jones et al., 2019; T. D. Jones et al., 2021). This approach has been instrumen-  
66 tal to our understanding of the development of present-day mantle structure, particu-  
67 larly formation of large-low-velocity zones at the base of the mantle. However, simula-  
68 tions of this nature can only investigate whole-mantle processes over Earth history. This  
69 starting point limits their ability to generate high-resolution simulations of short-lived  
70 geologic events. Moreover, like studies that rely on geophysical constraints, these isotopic  
71 simulations are validated using present-day observations of mantle structure. To deter-  
72 mine whether geodynamical simulations accurately reproduce past mantle processes, they  
73 must compute predictions for a wider variety of geologic observations.

74 The composition of igneous rocks at Earth’s surface acts as a permanent, albeit in-  
75 complete, record of melting within the mantle and, hence, can be used to inform simu-  
76 lations into the geologic past. Melting in the mantle is rare, and so simulations that  
77 predict the distribution and extent of melting, and compare these predictions to the spread  
78 of igneous rocks at the surface, can offer important insights into upper-mantle structure  
79 at mid-oceanic ridges (e.g., Keller et al., 2017; Sim et al., 2020), subduction zones (e.g.,  
80 Perrin et al., 2018; Rees Jones et al., 2018), and intraplate settings (e.g., Lees et al., 2020;  
81 Duvernay et al., 2021). However, comparisons between calculated melt estimates and  
82 observed igneous-rock distributions are often imperfect since in many locations the ma-  
83 jority of melt is trapped within the lithosphere (Crisp, 1984). The ability to predict melt  
84 chemistry within geodynamical simulations unlocks the full potential of igneous rocks  
85 to constrain upper-mantle processes through space and time (e.g., Watson & M<sup>c</sup>Kenzie,  
86 1991; Plank & Langmuir, 1992; Bown & White, 1994; Asimow & Langmuir, 2003; Gregg  
87 et al., 2009; Behn & Grove, 2015; Brown et al., 2020; Krein et al., 2020). To first order,  
88 incompatible-element compositions of mafic igneous rocks are analogous to those of prim-  
89 itive mantle melts, i.e., melts that have not undergone fractional crystallisation (Gast,  
90 1968; Frey et al., 1974). Primitive melts commonly represent an accumulation of melts  
91 generated at a wide range of thermal, chemical, mineralogical and dynamical conditions  
92 (Langmuir et al., 1977). The volume and composition of each melt is sensitive to these  
93 conditions and, hence, aggregate primitive melts, and corollary igneous rocks observed  
94 at the surface, largely reflect the thermo-chemical and dynamical structure of the up-  
95 per mantle (e.g., Klein & Langmuir, 1987; McKenzie & O’Nions, 1991; Niu et al., 2011;  
96 Dalton et al., 2014; Ball et al., 2021). By comparing computed melt compositions with  
97 observed igneous rock compositions, it is possible to test the veracity of geodynamical  
98 simulations through geologic time. Furthermore, integrating geochemical calculations within  
99 a geodynamical modeling framework provides a means to investigate how changes to upper-  
100 mantle thermo-chemical structure affect melt compositions.

101 In this study, we present a peridotite melting parameterization, BDD21, that cal-  
102 culates melt incompatible-element concentrations as a function of pressure, temperature,  
103 melt fraction and mantle composition. BDD21 is designed so that it can be integrated  
104 with a range of geodynamical codes that incorporate a tracer-particle functionality. It  
105 can therefore be used within geodynamical simulations over a wide range of geologic set-  
106 tings to predict melt incompatible-element compositions through space and time. Here,  
107 we demonstrate BDD21’s functionality using the Fluidity computational modelling frame-  
108 work (e.g., Davies et al., 2011; Mathews, 2021).

109 To test the versatility of our approach, we apply our coupled Fluidity-BDD21 frame-  
110 work to two contrasting geologic settings: melting at mid-oceanic ridges and adjacent  
111 to step-changes in lithospheric thickness. At mid-oceanic ridges, extensive melt produc-  
112 tion occurs in response to passive upwelling generated by divergent plate velocities. If  
113 prescribed plate velocities, underlying mantle temperature, and mantle composition re-  
114 main constant through time, upwelling rates and melting diagnostics will approach a steady-

115 state that persists for millions of years. As such, we are able to test the sensitivity of melt  
 116 volume and melt composition to variables such as plate velocity, mantle temperature and  
 117 mantle depletion. These predictions can subsequently be compared to global datasets  
 118 from mid-oceanic ridges (e.g., Gale et al., 2013; Hoggard et al., 2017; Richards et al., 2020).  
 119 At lithospheric steps, lateral density variations that accompany steep gradients in litho-  
 120 spheric thickness trigger small-scale convective flow (King & Anderson, 1998). This flow  
 121 field changes over time, as the instabilities and overlying thermal boundary layer evolve.  
 122 In contrast to mid-oceanic ridges, melting at lithospheric steps is deeper, less volumi-  
 123 nous and ephemeral, as melting is inhibited by thickening of the overlying thermal-boundary  
 124 layer (Duvernay et al., 2021). Since melt composition is strongly sensitive to the extent  
 125 and depth of melting, applying BDD21 to these contrasting geologic settings allows us  
 126 to demonstrate the sensitivity of melt composition to temperature, lithospheric thick-  
 127 ness and mantle flow across different melting regimes. Therefore, integrating BDD21 into  
 128 geodynamical simulations will help to reveal how the uppermost mantle’s thermo-chemical  
 129 structure and flow regime influence melt compositions observed at the surface through  
 130 space and time.

## 131 2 Geodynamical Modeling Framework

132 In the upper-mantle convection simulations considered below, we solve the incom-  
 133 pressible Stokes and energy equations relevant to mantle convection using *Fluidity*, a finite-  
 134 element, control-volume computational modeling framework based upon adaptive, un-  
 135 structured discretisations (Davies et al., 2011; Kramer et al., 2012, 2021). Our simula-  
 136 tions are 2-D, and we focus upon melting at mid-oceanic ridges and lithospheric steps.  
 137 Melting is tracked using the particle-in-cell method, with particles advected using a 4<sup>th</sup>-  
 138 order Runge-Kutta scheme. Further details on *Fluidity* and the implementation of its particle-  
 139 in-cell scheme can be found in Davies et al. (2011) and Mathews (2021), respectively. Our  
 140 melting and geochemical parameterizations are described below. Note that our simula-  
 141 tions assume that the mantle is incompressible and, hence, temperature does not decrease  
 142 adiabatically as particles ascend to shallower depths. Accordingly, when calculating melt  
 143 productivity within this study, we assume that simulation mantle temperatures repre-  
 144 sent potential temperatures ( $T_p$ ).

### 145 2.1 Melting Parameterization

146 Within our simulations, melt generation is governed by an adapted version of the  
 147 hydrous lherzolite melting parameterization of Katz et al. (2003). Here, some of the ma-  
 148 terial constants within this parameterization have been updated to honor experimental  
 149 constraints that have subsequently been obtained (Supplementary Materials; Shorttle  
 150 et al., 2014). Assuming that clinopyroxene remains present, or has already been exhausted  
 151 during melting, melt productivity as the mantle decompresses is defined as

$$152 \left. \frac{dX}{dP} \right|_S = \frac{-\left. \frac{C_P dT}{T dP} \right|_X + \frac{\alpha_s}{\rho_s} + X \left( \frac{\alpha_f}{\rho_f} - \frac{\alpha_s}{\rho_s} \right)}{\Delta S + \left. \frac{C_P dT}{T dX} \right|_P}, \quad (1)$$

153 where  $P$ ,  $T$ ,  $X$ ,  $S$ ,  $\Delta S$ ,  $C_P$  represent pressure, temperature, melt fraction, entropy, en-  
 154 tropy of fusion and specific heat capacity, respectively; the coefficient of thermal expan-  
 155 sion and density are denoted by  $\alpha$  and  $\rho$ , respectively; and subscripts  $s$  and  $f$  refer to  
 156 the solid and fluid phases, respectively (McKenzie, 1984). This equation describes adi-  
 157 abatic decompression melting, and changes in temperature are tracked during melting

158 by integrating

$$159 \quad \left. \frac{dT}{dP} \right|_S = T \left[ \frac{\frac{\alpha_s}{\rho_s} + X \left( \frac{\alpha_f}{\rho_f} - \frac{\alpha_s}{\rho_s} \right) - \Delta S \left. \frac{dX}{dP} \right|_S}{C_P} \right], \quad (2)$$

160 simultaneously with Equation 1. Here, we adapt these equations to allow each particle  
 161 to decompress along its respective temperature gradient calculated within each Fluidity  
 162 simulation. We assume that

$$163 \quad \frac{\alpha_s}{\rho_s} + X \left( \frac{\alpha_f}{\rho_f} - \frac{\alpha_s}{\rho_s} \right) \sim \frac{C_P}{T} \left. \frac{dT}{dP} \right|_{\text{Fluidity}}, \quad (3)$$

164 where the derivative super-script signals that this temperature gradient is sourced from  
 165 Fluidity (for derivation of this relationship, see Supplementary Equation 3). Accordingly,  
 166 by substituting Equation 3 into Equations 1 and 2 and rearranging, we obtain the fol-  
 167 lowing system of coupled equations:

$$168 \quad \left. \frac{dX}{dP} \right|_S \sim \frac{\left. \frac{dT}{dP} \right|_{\text{Fluidity}} - \left. \frac{dT}{dP} \right|_X^{\text{Katz}}}{\frac{T \Delta S}{C_P} + \left. \frac{dT}{dX} \right|_P^{\text{Katz}}}, \quad (4)$$

$$169 \quad \left. \frac{dT}{dP} \right|_S \sim \left. \frac{dT}{dP} \right|_{\text{Fluidity}} - \frac{T \Delta S}{C_P} \left. \frac{dX}{dP} \right|_S. \quad (5)$$

171 When a particle exceeds the solidus, Equations 4 and 5 are jointly integrated to calcu-  
 172 late the evolution of melt fraction and temperature as a function of pressure. Each La-  
 173 grangian particle thus records a value of  $X$  at each time-step and stores the maximum  
 174 value encountered throughout the simulation ( $X_{\text{max}}$ ). Melting occurs when the newly  
 175 obtained  $X$  is higher than  $X_{\text{max}}$  and a melting rate ( $M$ ) is subsequently calculated from  
 176 the value of the time-step ( $\delta t$ ) at this stage:

$$M = \max\left(0, \frac{X - X_{\text{max}}}{\delta t}\right). \quad (6)$$

177 During melting, latent heat is consumed and, hence, temperatures decrease. These tem-  
 178 perature variations are fed back into Fluidity through a source-term in the energy equa-  
 179 tion. For simplicity, we do not attempt to simulate melt extraction or melt refreezing.

## 180 2.2 Mid-Oceanic Ridge Simulation Set-Up

181 We simulate mantle flow beneath a mid-oceanic ridge in an idealised 2-D compu-  
 182 tational geometry (Figure 1a; Table A1). Our domain extends horizontally ( $x$ ) for 990  
 183 km either side of the ridge axis and vertically ( $z$ ) from the surface to 660 km depth. The  
 184 upper boundary has an imposed horizontal velocity ( $v_x$ ) which represents the motion of  
 185 the oceanic plate; no vertical flow through this boundary is permitted. Plate velocity acts  
 186 in opposite directions either side of the ridge, and so plate speed is equivalent to the half  
 187 spreading rate ( $R_s$ ).  $v_x$  is also prescribed at side boundaries and varies as a function of  
 188  $1/z^2$  from  $v_x = R_s$  at the surface to  $v_x = 0$  at 660 km depth. The bottom boundary  
 189 is open. Temperature boundary conditions are fixed to 0 °C at the surface of the simu-  
 190 lation, a constant user-defined temperature ( $T$ ) at the base, and zero heat flux bound-  
 191 ary conditions on both sides. Initial conditions comprise a plate with an age increasing  
 192 from the ridge (0 Myr at  $x=990$  km) to the sides, following a half-space cooling model.  
 193 Mesh spacing varies from  $\sim 0.5$  km near the top boundaries to 70 km at the domain's  
 194 base.

195 We consider a mantle undergoing deformation through a composite diffusion and  
 196 dislocation creep (i.e. mixed) rheology. Viscosity ( $\mu$ ) is calculated at each time step from  
 197 the general power law

$$198 \quad \mu = A^{-\frac{1}{n}} \exp\left(\frac{E + PV}{nRT}\right) \dot{\epsilon}_{II}^{\frac{1-n}{n}}, \quad (7)$$

199 where  $\dot{\epsilon}_{II}$  is the second invariant of the strain-rate tensor,  $n$  is the stress exponent that  
 200 depends on the deformation mechanism,  $E$  and  $V$  are activation energy and volume,  $A$   
 201 is a pre-factor,  $P = \rho g z$  is the lithostatic pressure,  $R$  is the gas constant and  $T$  is tem-  
 202 perature. Note that the rheological parameters, detailed in Table A1, are chosen to yield  
 203 realistic upper-mantle viscosity values that are consistent with geophysical constraints,  
 204 following the methodology of Garel et al. (2014). These parameters are also compati-  
 205 ble with experimental data (Hirth & Kohlstedt, 2003; Korenaga & Karato, 2008). The  
 206 effective viscosity is calculated via a harmonic mean as  $(\mu_{\text{diff}}^{-1} + \mu_{\text{disl}}^{-1})^{-1}$  with  $\mu_{\text{diff}}$  and  
 207  $\mu_{\text{disl}}$  the viscosity associated to diffusion and dislocation creep, respectively.

208 Melting commences when mantle material reaches the peridotite solidus and ceases  
 209 when this material no longer advects upwards or cools down to sub-solidus conditions.  
 210 The occurrence of melting, and consequently the geometry of the melt region, is tracked  
 211 by tracer particles. For our mid-oceanic ridge simulations, particles are randomly ini-  
 212 tialized at depths below the solidus over a wide region so that all possible melting paths  
 213 are captured throughout each simulation. Diagnostics are analysed once each simulation  
 214 has reached steady state.

### 215 **2.3 Lithospheric Step Simulation Set-Up**

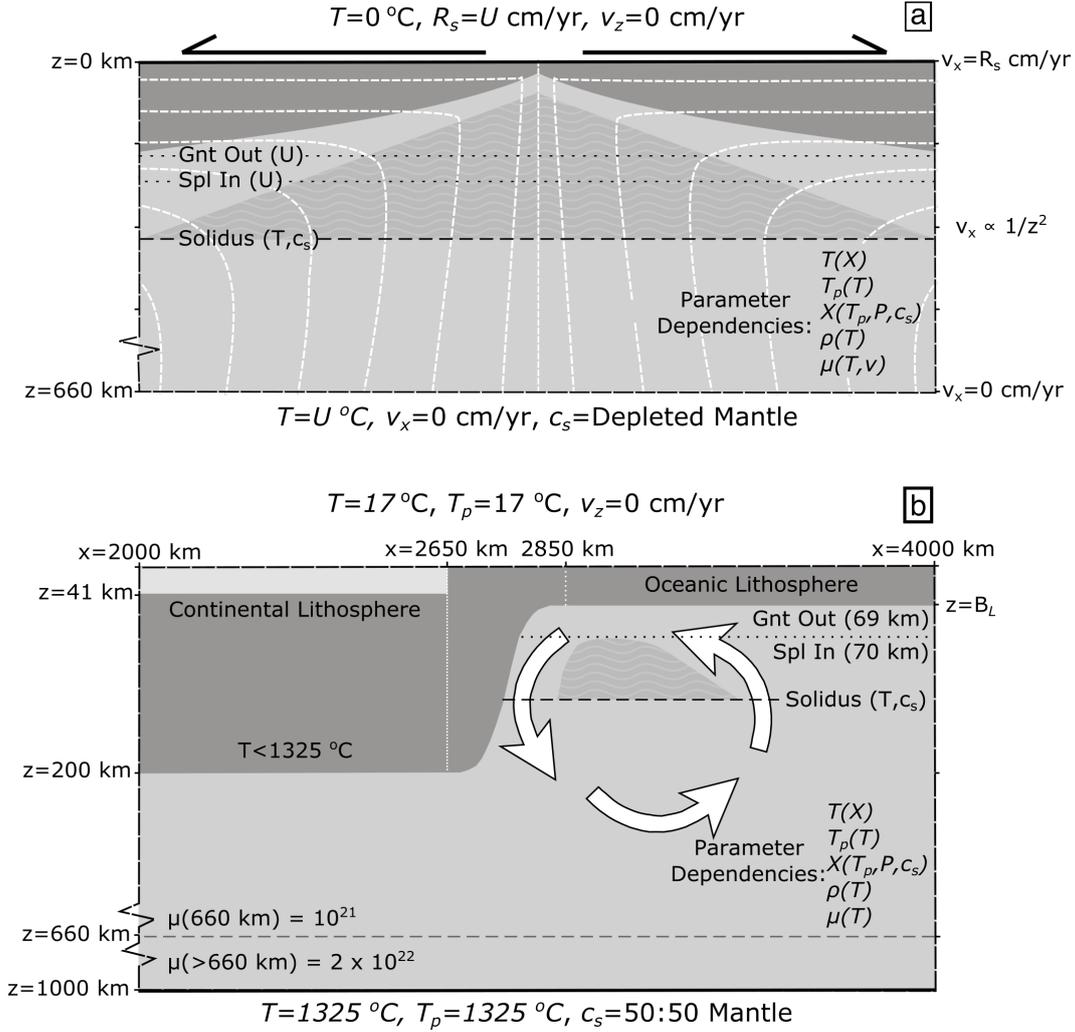
216 The geodynamical set-up of our lithospheric step simulations is similar to the 2-  
 217 D cases described in Duvernay et al. (2021). Key simulation parameters are summarised  
 218 in Table A2. The modeling domain extends in the vertical direction ( $z$ ) from the sur-  
 219 face to 1000 km depth and horizontally ( $x$ ) over 4000 km (Figure 1b). No external forc-  
 220 ing is included within this simulation. Simulation boundaries are free-slip apart from the  
 221 bottom boundary which is no-slip. The asthenospheric interior of these simulations have  
 222 initial mantle temperatures of 1325 °C, while the surface temperature is set to 17 °C.  
 223 No heat transfer is allowed through the sidewalls of the simulation.

224 Steps in lithospheric thickness are imposed over 200 km wide zones between a cen-  
 225 tral region of thick continental lithosphere and surrounding thinner oceanic lithosphere.  
 226 The continent lies between  $x = 1350$ – $2650$  km and has a thickness of 200 km, with the  
 227 upper 41 km representing the crust. The initial temperature distribution within the con-  
 228 tinental interior follows a conductive geotherm obtained by solving the 1-D heat equa-  
 229 tion. For this calculation, we use a thermal conductivity of  $3 \text{ W m}^{-1} \text{ K}^{-1}$  and internal  
 230 heat generation with an exponential decrease of characteristic length scale 9 km (Lachenbruch,  
 231 1970; Schatz & Simmons, 1972; Pollack & Chapman, 1977). Either side of this continen-  
 232 tal core, the oceanic lithosphere is treated as a thermal-boundary layer with a temper-  
 233 ature profile that adheres to a half-space cooling model.

234 We define viscosity ( $\mu$ ) through a diffusion creep rheology law

$$235 \quad \mu = A \times \exp\left(\frac{E^* + \rho_0 g z V^*}{R(T + \psi z)}\right). \quad (8)$$

236 Viscosity is temperature- and depth-dependent. At 660 km depth,  $\mu$  reaches a value of  
 237  $10^{21}$  Pa·s in the upper mantle and jumps to a constant value of  $2 \times 10^{22}$  Pa·s in the un-  
 238 derlying lower mantle. Within the continent, to ensure the stability and longevity of the  
 239 lithospheric block, viscosity is increased by a factor of 100.



**Figure 1.** Schematic diagrams of geodynamical simulation domains with boundary conditions and parameter dependencies. a) Mid-oceanic ridge simulation illustrated with simplified representation of the thermal boundary layer (darkest grey), melting region (dark grey with wavy white lines) and mantle streamlines (dashed white lines). Bottom and side boundaries are open for normal flow. Text above and below simulation domain describe boundary conditions at  $z = 0$  km and  $z = 660$  km, respectively.  $U =$  User defined value. b) Same as panel a but for the lithospheric step case. Additional features include: continental crust (lightest grey); stylised mantle flow lines (white arrows);  $B_L =$  oceanic thermal boundary layer thickness at time = 0 Myr. Dashed boundaries = free-slip, solid boundary = no-slip. Simulation domain mirrored along left-hand side.

240 Our mid-oceanic-ridge simulations are analysed once they achieve steady state and  
 241 particles are present throughout the melt region. In the lithospheric step case, we are  
 242 interested in the spatio-temporal melting trends as these simulations do not approach  
 243 a steady-state melting regime. Therefore, our lithospheric step simulations are initial-  
 244 ized with particles distributed across the entire computational domain. Particles initial-  
 245 ized at supra-solidus conditions are assigned initial melt fractions assuming that they  
 246 underwent instantaneous melting at that depth.

### 3 Geochemical Melting Parameterization

Our geodynamical modeling framework provides a means to compute the spatio-temporal evolution of the upper mantle’s structure, flow regime and associated melt distribution. To generate melt compositions that can be compared to the composition of mafic igneous rocks erupted at Earth’s surface, a geochemical melting parameterization must be integrated into our geodynamical framework. Our geochemical melting parameterization is similar to those of Kimura and Kawabata (2014) and Brown and Leshner (2016), which exploit peridotite melting experiments to parameterize mineralogy and melt stoichiometry throughout the mantle (Baker & Stolper, 1994; Walter, 1998; Falloon et al., 1999). This approach represents an improvement on earlier parameterizations that were developed prior to these experiments being conducted (e.g., McKenzie & O’Nions, 1991; Langmuir et al., 1992). It is less computationally expensive than sophisticated thermodynamically-consistent parameterizations (e.g., Ghiorso et al., 2002; Connolly, 2005; Holland et al., 2018) and, thus, is very-well suited for integration within geodynamical simulations. Our approach is different to some previous coupled geochemical-geodynamical melting schemes (e.g., Behn & Grove, 2015; Krein et al., 2020). These schemes primarily focused on major element systems, and so they parameterize the equilibrated composition of the melt rather than the mineralogy of the source (Till et al., 2012). Here, we describe BDD21, a Python package for calculating incompatible element concentrations within melts as a function of pressure, temperature, melt fraction and mantle composition.

#### 3.1 Fractional Melting Functions

The concentration of an incompatible element within an instantaneous melt ( $c_l$ ) and the concentration of that element within the residue ( $c_s$ ) are related to each other by two equations that must be simultaneously solved (White et al., 1992). These equations are

$$\frac{dc_s}{dX} = \frac{c_s - c_l}{1 - X} \quad \text{and} \quad c_l = \frac{c_s(1 - X)}{\bar{D} - \bar{P}X}, \quad (9)$$

where  $X$  is the melt fraction, and where  $\bar{D}$  and  $\bar{P}$  are the bulk distribution coefficients for a given element within the solid assemblage and the melting assemblage, respectively (Shaw, 1970).

To calculate  $c_l$  and  $c_s$  for a given particle, these equations are numerically integrated using the LSODA algorithm (Hindmarsh, 1983; Petzold, 1983) from the solidus, where  $X = 0$ , to the particle’s present location, where  $X = X'$ . The path between these limits for each particle,  $X(P, T)$ , is dictated by the flow field of our geodynamical simulations (Section 2).

Together,  $\bar{D}$  and  $\bar{P}$  determine how readily an incompatible element will partition into the melt.  $\bar{D}$  and  $\bar{P}$  are the sum of the distribution coefficients for each mineral phase,  $D_{mnl}$ , scaled by their modal abundance in the solid and liquid phase, respectively. Therefore,

$$\bar{D} = \sum_{n=1}^N F_{mnl} D_{mnl} \quad \text{and} \quad \bar{P} = \sum_{n=1}^N p_{mnl} D_{mnl}, \quad (10)$$

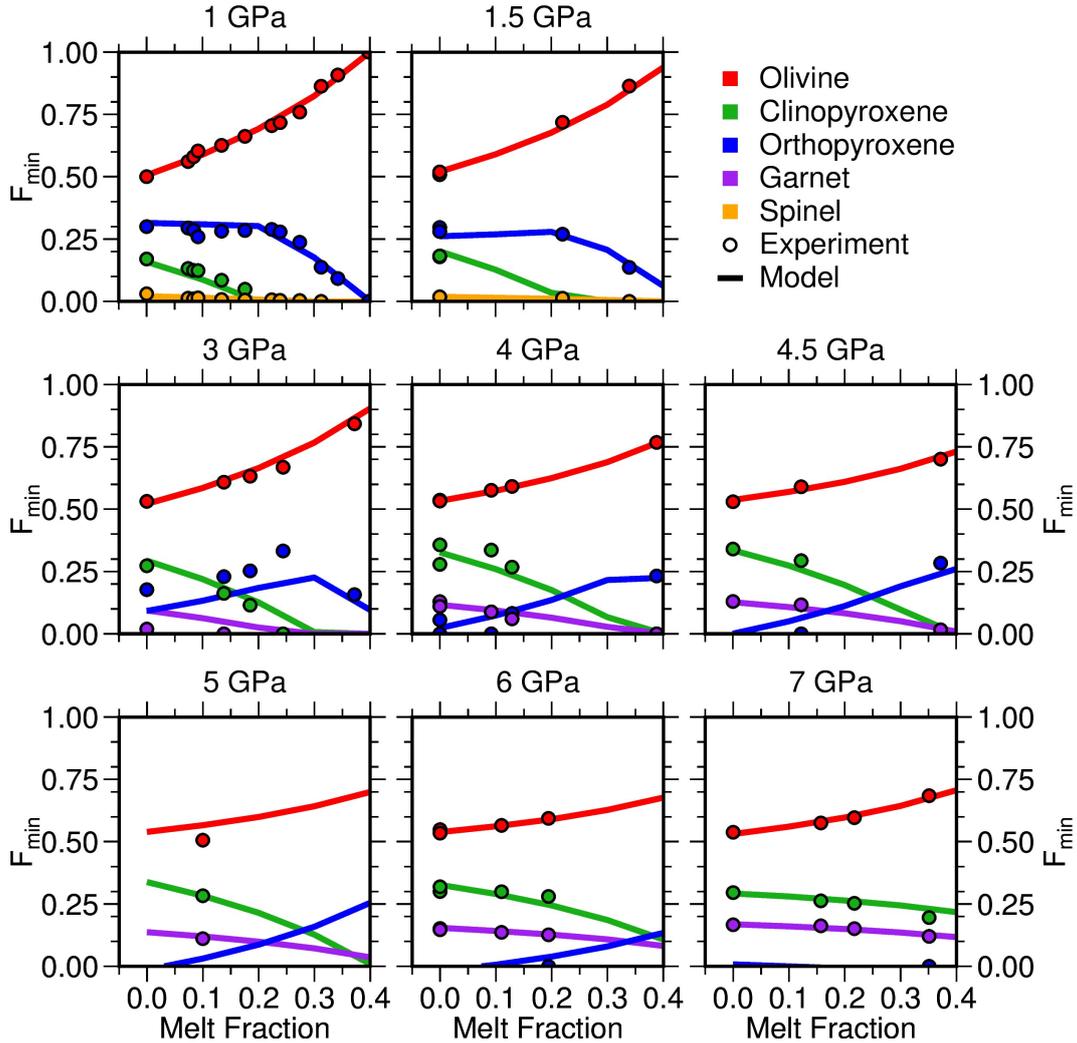
where  $N$  is the total number mineral phases present,  $F_{mnl}$  is the modal abundance of a mineral, and  $p_{mnl}$  is linked to the change in  $F_{mnl}$  at each melting step, as follows:

$$\sum_{n=1}^N F_{mnl} = 1 \quad \text{and} \quad p_{mnl} = F_{mnl} - (1 - X) \frac{dF_{mnl}}{dX} \quad (11)$$

289 (Shaw, 1979).  $\bar{D}$  and  $\bar{P}$  vary as a function of depth since the mineral assemblage, melt  
 290 stoichiometry, and compatibility of an element within each mineral are pressure and melt-  
 291 fraction dependent. In the following sections, we parameterize each of these variables within  
 292 our geochemical framework.

### 293 3.2 Mantle Modal Mineralogy and Melting Reaction Stoichiometry

294 Peridotite melting experiments can be utilised to develop a simple parameteriza-  
 295 tion of mantle mineralogy as a function of pressure and melt fraction. Along similar lines  
 296 to recent geochemical parameterizations (e.g., Kimura & Kawabata, 2014; Brown & Leshner,  
 297 2016), we exploit experimental studies that record mineral abundances as a function of



**Figure 2.** Modal mineralogy as a function of melt fraction at a range of pressures. Red, blue, green, purple and orange circles represent experimental modal proportions of olivine, orthopyroxene, clinopyroxene, garnet and spinel, respectively (Baker & Stolper, 1994; Walter, 1998; Falloon et al., 1999). Colored lines = parameterized proportions of each mineral as a function of melt fraction using Equations 12, 13 and 15. Note that experiments conducted at 3 and 5 GPa were not used to predict modal mineralogy.

298 melt fraction for the MM3 and KR4003 lherzolites at 1–1.5 and 3–7 GPa, respectively  
 299 (Baker & Stolper, 1994; Walter, 1998; Falloon et al., 1999). We assume that these ex-  
 300 perimental results can be combined since these lherzolites have similar major element  
 301 compositions (Kimura & Kawabata, 2014). Four mineral phases exist within these lher-  
 302 zolites: olivine (ol), orthopyroxene (opx), clinopyroxene (cpx) and an aluminous phase.  
 303 The aluminous phase present varies as a function of depth between plagioclase, spinel  
 304 (spl) and garnet (gnt). Plagioclase is stable within the mantle at pressures  $\lesssim 1$  GPa (Borghini  
 305 et al., 2010). At ambient mantle temperatures, once the mantle has decompressed to  $\sim$   
 306 1 GPa, melt fractions are sufficiently high such that plagioclase is commonly exhausted.  
 307 Accordingly, and for simplicity, at this stage BDD21 does not include a parameteriza-  
 308 tion for melting within the plagioclase-stability field: any experiment where plagioclase  
 309 exists as a stable phase is excised from our experimental database. We also remove ex-  
 310 periments with  $> 20$  % melting at depths  $\geq 6$  GPa since these conditions are unlikely  
 311 to occur within the mantle.

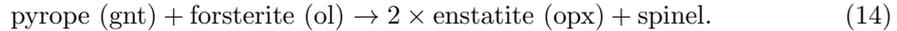
312 At constant pressure ( $P$ ), modal abundances for most minerals within these exper-  
 313 iments vary approximately linearly as a function of  $X$  (if  $X$  is included as a modal phase,  
 314 i.e., if all mineral phases sum to  $1 - X$ ; Figure 2). Here, to parameterize  $F_{mnl}(P, X)$   
 315 for olivine, clinopyroxene, garnet and spinel, we perform linear regressions through these  
 316 data at constant pressure:

$$317 \quad F_{mnl}(1 - X) = a_{mnl}X + b_{mnl}. \quad (12)$$

318 The empirically determined constants,  $a_{mnl}$  and  $b_{mnl}$ , from each linear regression at each  
 319 pressure are then combined to parameterize  $F_{mnl}(P, X)$  via a second-order polynomial  
 320 regression:

$$321 \quad a_{mnl} = a_0 + a_1P + a_2P^2 \quad \text{and} \quad b_{mnl} = b_0 + b_1P + b_2P^2. \quad (13)$$

322 At depths between  $\sim 60$ – $90$  km, the stable aluminous phase changes from spinel  
 323 to garnet through the following reaction:



324 When performing polynomial regressions from 1–7 GPa, the proportion of garnet present  
 325 at the solidus (i.e.,  $F_{gnt}$  when  $X = 0$ ), approaches zero at  $\sim 2.7$  GPa. The pressure  
 326 range over which the spinel-garnet transition occurs is controversial: thermodynamical  
 327 models typically place this transition at lower pressures than it occurs at within peri-  
 328 dotite melting experiments (Green et al., 2012). Since partitioning of trace elements, es-  
 329 pecially heavy rare earth elements, is strongly dependent on the presence of garnet, we  
 330 allow the depth of the spinel-garnet-transition zone to be a user-defined variable within  
 331 our parameterization. We therefore calculate the modal mineralogy constants ( $a_{0-2}$  and  
 332  $b_{0-2}$ ) for spinel- and garnet-bearing peridotite separately using experiments between 1–  
 333 1.5 GPa and 4–7 GPa, respectively. All experiments from 1–7 GPa are used to calibrate  
 334  $F_{cpx}(P, X)$  since clinopyroxene is not precipitated or consumed at this phase boundary.  
 335 We assume that mantle modal mineralogy varies linearly between the mineral assemblages  
 336 estimated to be present at the top and bottom of the spinel-garnet-transition zone.

337 The proportion of orthopyroxene present in the mantle ( $F_{opx}$ ) does not vary lin-  
 338 early as a function of melt fraction, since it strongly depends on whether clinopyroxene  
 339 is a stable phase (Figure 2). For simplicity, we assume that  $F_{opx}$  accounts for the total  
 340 missing fraction when all other mineral contributions are combined:

$$341 \quad F_{opx} = 1 - F_{ol} - F_{cpx} - F_{spl} - F_{gnt}. \quad (15)$$

Our empirically determined modal mineralogy constants are listed in Table 1. The fit between these parameterizations and the experimental data can be seen in Figure 2. Note that, with the exception of clinopyroxene, experiments performed at 3 GPa are not exploited to calibrate  $F_{mnl}(P, X)$  since these experiments occur close to the spinel-garnet transition zone. The single experiment conducted at 5 GPa is excluded since a linear regression cannot be performed on one data point.

### 3.3 Mineral Distribution Coefficients

To calculate bulk distribution coefficients, the partition coefficients between each mineral and the melt phase ( $D_{mnl}$ ) must be parameterized. A notable difference between BDD21 and those of similar melting parameterizations (e.g., Kimura & Kawabata, 2014; Brown & Leshner, 2016) is that, where possible, we estimate partition coefficients as a function of pressure, temperature and composition using a lattice strain equation (Brice, 1975). The compatibility of an element within a mineral is principally controlled by the valency of the element and the size difference between the element and the mineral host site. Isovalent cations in minerals can be described using the same lattice strain equation, where  $D_{mnl}$  depends on three material properties: the radius of the site ( $r_o$ ), the elastic response of the site ( $E_M$ ) and the partition coefficient for an ideal element with a radius of  $r_o$ , ( $D_o$ ; Brice, 1975). These material properties vary as a function of pressure ( $P$ ), temperature ( $T$ ) and mineral major-element chemistry ( $\chi$ ) since changes in these variables cause sites within mineral lattices to expand and contract (Brice, 1975). As such, to use this lattice strain equation to define  $D_{mnl}(P, T, \chi)$  for each cation site within each mineral, we must parameterize  $r_o(P, T, \chi)$ ,  $E_M(P, T, \chi)$  and  $D_i(P, T, \chi)$ . Mineral-melt partitioning experiments at a range of thermochemical conditions can be exploited to empirically estimate  $r_o$ ,  $E_M$  and  $D_i$  as a function of  $P$ ,  $T$  and  $\chi$  (Wood & Blundy, 1997). Chosen lattice strain parameterizations used within BDD21 are shown in Table 2. For some cation valency-mineral pairs,  $D_{mnl}(P, T, \chi)$  is less well constrained and so in these cases we assume  $D_{mnl}$  is invariant (Supplementary Table 1; McKenzie & O’Nions, 1995). For a full description of how we parameterize  $D_{mnl}(P, T, \chi)$  for each element, see Supplementary Information (Shannon, 1976; Wood & Banno, 1973; Hazen & Finger, 1979; Landwehr et al., 2001; Blundy & Wood, 2003; Hill et al., 2011).

### 3.4 Mantle Composition

The experiments used to parameterize mantle mineralogy during melting were conducted using lherzolites that are assumed to represent primitive mantle (PM). However, more refractory sources, such as depleted MORB mantle (DMM), may also be present during melting. For melting of DMM, we adopt the same approach as Kimura and Kawa-

**Table 1.** Constants for calculating modal mineralogy as a function of melt fraction in Equations 12 and 13.

Spinel Peridotite	$a_0$	$a_1$	$a_2$	$b_0$	$b_1$	$b_2$
$F_{ol}$	0.318	0.314	-0.115	0.419	0.126	-0.039
$F_{cpx}$	-0.606	-0.229	0.037	0.058	0.112	-0.011
$F_{spl}$	-0.087	-0.013	0.026	0.020	0.004	-0.004
Garnet Peridotite	$a_0$	$a_1$	$a_2$	$b_0$	$b_1$	$b_2$
$F_{ol}$	1.298	-0.558	0.048	0.445	0.035	-0.003
$F_{cpx}$	-0.606	-0.229	0.037	0.058	0.112	-0.011
$F_{gnt}$	-0.557	0.078	-0.005	0.008	0.033	-0.001

377 bata (2014), increasing  $F_{ol}$  by 0.04 at the expense of  $F_{cpx}$ . Note that when clinopyroxene  
 378 is exhausted, or close to exhaustion, this reduction in  $F_{cpx}$  is accommodated via a  
 379 complementary reduction in  $F_{opx}$ , ensuring that Equation 15 remains satisfied. When  
 380 a combination of PM and DMM is used, this change to modal mineralogy is scaled lin-  
 381 early according to the proportions of PM and DMM in the source (e.g., for a mantle of  
 382 50% PM and 50% DMM,  $F_{ol}$  and  $F_{cpx}$  are increased and reduced by 0.02, respectively).  
 383 Concentrations of incompatible elements within the source region prior to melt initia-  
 384 tion are also linearly varied between the primitive and depleted compositions of McDonough  
 385 and Sun (1995) and Salters and Stracke (2004), respectively. In the primitive case,  $H_2O$   
 386 concentration is set to 280 ppm based on the assumption that the concentration of  $H_2O$   
 387 is  $200\times$  greater than that of Ce (Michael, 1995). The concentration of water in depleted  
 388 mantle is set to 100 ppm (Salters & Stracke, 2004).

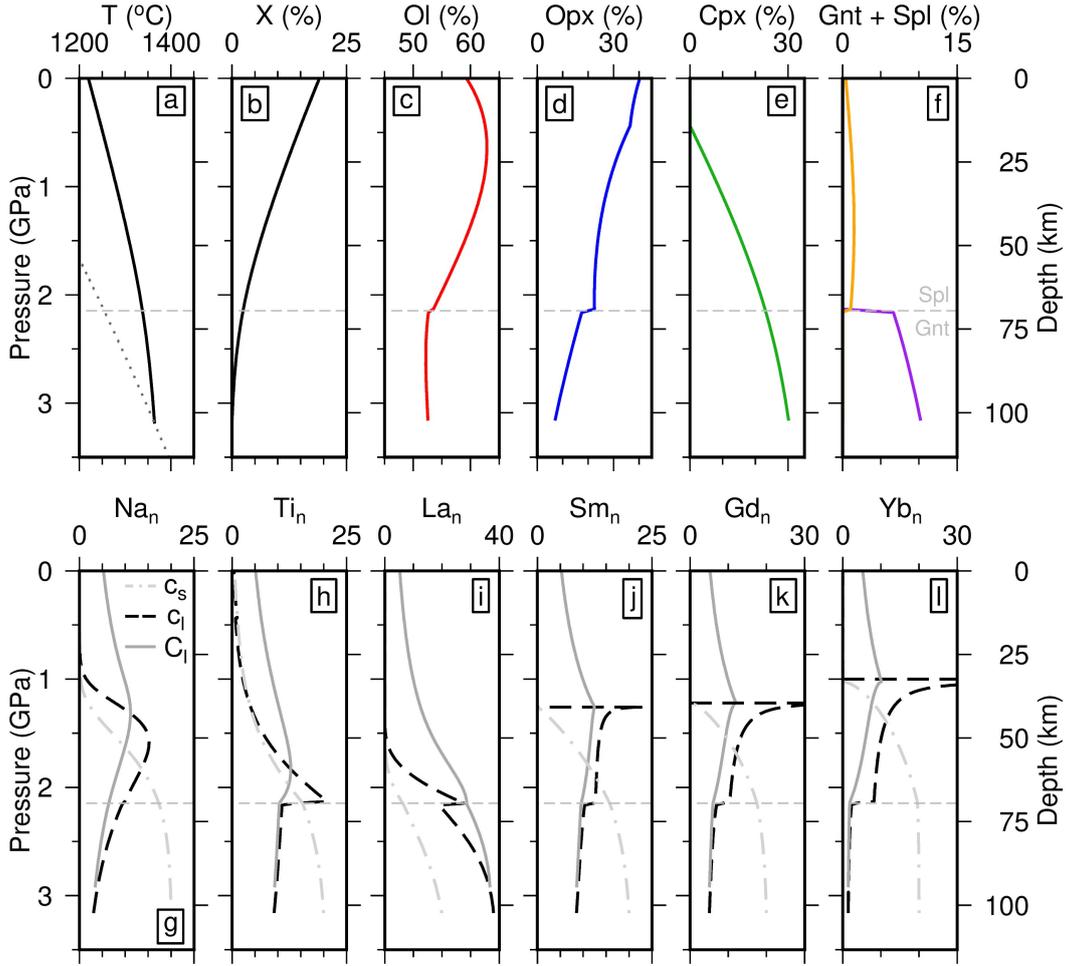
### 389 3.5 Application to a One-Dimensional Melting Column

390 Before applying BDD21 to 2-D geodynamical simulations, we first describe how man-  
 391 tle mineralogy and melt composition respond to isentropic melting in a simple 1-D case.  
 392 A particle of primitive mantle with  $T_p = 1325$  °C ascends vertically from 3.5 GPa to  
 393 the surface. As this particle rises, it intersects the solidus at  $\sim 100$  km depth and be-  
 394 gins to melt (Figure 3a). Melt productivity increases as  $X$  increases, and so the first few  
 395 degrees of melting happen over a larger depth interval (Figure 3b; Katz et al., 2003). These  
 396 initial melts are generated in the garnet stability field through the consumption of clinopy-  
 397 roxene, garnet and small amounts of olivine, and the precipitation of orthopyroxene (Fig-  
 398 ure 3c-f). For this illustrative example, we have placed the spinel-garnet transition at  
 399 69–70 km depth (Jennings & Holland, 2015). Both the mantle mineralogy and melting  
 400 reaction stoichiometry are radically altered as the spinel-garnet-transition zone is crossed.  
 401 At the transition zone, garnet and olivine react to form orthopyroxene and spinel (Equa-  
 402 tion 14). Beyond this boundary, clinopyroxene is consumed and olivine is precipitated  
 403 during melting. At shallow depths olivine begins to be consumed and orthopyroxene is  
 404 precipitated, with the importance of this reaction increasing once clinopyroxene has been  
 405 exhausted.

406 As elements vary in abundance and compatibility, each element behaves differently  
 407 during melting. Of the major mineral phases present in the mantle, Na is most compat-  
 408 ible within clinopyroxene and the compatibility of Na within clinopyroxene increases as  
 409 a function of pressure (Blundy et al., 1995). As the particle decompresses and melts, clinopy-  
 410 roxene is consumed and Na becomes increasingly incompatible within the solid phase.  
 411 Therefore, an increase in instantaneous melt Na concentrations is observed between the

**Table 2.** Incompatible element partitioning parameterizations for each mineral as a function of element valency. All other mineral-elements partition coefficients treated as constant values (Supplementary Table 1; McKenzie & O’Nions, 1995).

Mineral	Parameterisation	Mineral	Parameterisation
<b>Olivine</b>		<b>Orthopyroxene</b>	
3+	Sun and Liang (2013)	2+	Wood and Blundy (2014)
		3+	Yao et al. (2012)
<b>Clinopyroxene</b>		<b>Garnet</b>	
1+	Wood and Blundy (2014)	3+	Sun and Liang (2013)
2+	Wood and Blundy (2014)	4+	Mallmann and O’Neill (2007)
3+	Sun and Liang (2012)		
4+	Wood and Blundy (2014)		



**Figure 3.** 1-D isentropic decompression melting of a particle of primitive mantle at  $T_p = 1325$  °C. a) Temperature as a function of pressure. Solid line = prediction from melt parameterization, dotted grey line = solidus, dashed gray lines = spinel-garnet-transition zone. b) melt fraction ( $X$ ). c-f) percentage abundance of each mineral; purple and orange lines represent garnet and spinel abundances, respectively. g-l) Concentration of each element normalised by initial source concentration (McDonough & Sun, 1995). Solid / dashed / dot-dashed lines =  $C_i / c_i / c_s$  normalised by source composition. Note that, for illustrative purposes,  $c_s$  values have been increased by a factor of 20.

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solidus and  $\sim 1.5$  GPa (Figure 3g). Beyond  $\sim 1.5$  GPa, the concentration of Na in the instantaneous melt decreases.

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The compatibility of many elements changes considerably at the spinel-garnet transition zone. Ti is less compatible within spinel peridotite than within garnet peridotite, and so once this threshold is exceeded the concentration of Ti within the instantaneous liquid sharply increases (Figure 3h). The concentration of highly incompatible elements entering the melt decreases and melting continues as these elements become less abundant within the source. Light rare earth elements, such as La, are highly incompatible in garnet peridotite and, accordingly, concentrations of these elements within the liquid rapidly decrease as a function of melt fraction (Figure 3i). Note that, for un-normalised

422 concentrations, the gradient of this decrease is a function of both the incompatibility of  
 423 the element and its concentration in the residue. Heavier rare earth elements, such as  
 424 Sm, Gd and Yb, are more compatible in garnet peridotite than spinel peridotite. There-  
 425 fore, the concentrations of these elements in the instantaneous melts increase when the  
 426 spinel-garnet transition zone is exceeded (Figure 3j-1). Between the spinel-garnet tran-  
 427 sition zone and  $\sim 25$  km depth, clinopyroxene and olivine are consumed and precipi-  
 428 tated, respectively. Therefore, heavy rare earth elements become increasingly incompat-  
 429 ible in the solid phase and preferentially enter the melt phase until they are exhausted.  
 430 Since the integrated melt composition ( $C_l$ ) is a weighted sum of all instantaneous ( $c_l$ )  
 431 generated thus-far,

$$432 \quad C_l = \frac{1}{X_{\max}} \int_0^{X_{\max}} c_l(X) dX, \quad (16)$$

433  $C_l$  changes at a slower rate than  $c_l$ . After each element is exhausted, they no longer en-  
 434 ter the instantaneous melt phase and so their concentrations in the integrated liquid com-  
 435 position decrease as melting continues.

#### 436 **4 Application to Mid-Oceanic Ridges**

437 Using BDD21 and Fluidity, we have constructed a coupled geochemical-geodynamical  
 438 computational modeling framework of steady-state mid-oceanic spreading centres, at a  
 439 range of different mantle potential temperatures, plate spreading rates and source com-  
 440 positions. To validate these mid-oceanic ridge simulations and, therefore, our overall ap-  
 441 proach, we next compare the predictions from our simulations with a suite of geologi-  
 442 cal observations.

443 To compare our simulation predictions to observations, crustal thickness and melt  
 444 composition must be calculated for each simulation. Melt production beneath mid-oceanic  
 445 ridges can extend for  $> 100$  km either side of the ridge axis (Forsyth et al., 1998). To  
 446 form oceanic crust, melts must travel laterally towards the ridge axis. Generally, melts  
 447 that migrate greater lateral distances are more likely to refreeze within the lithosphere  
 448 and are, therefore, less likely to contribute to oceanic crust formation (Plank & Lang-  
 449 muir, 1992; Katz, 2008; Keller et al., 2017; Sim et al., 2020). We do not attempt to model  
 450 complex melt extraction and refreezing processes. Instead, we rely on simple relation-  
 451 ships observed by Keller et al. (2017) within their mid-oceanic ridge melt extraction sim-  
 452 ulations. Keller et al. (2017) suggest that the passive focusing distance ( $x_f$ ), i.e., the fur-  
 453 thest lateral distance that melts can travel to ridge axis without freezing, is a function  
 454 of upwelling velocity ( $v_y$ ) and temperature, finding that  $x_f$  approximately coincides with  
 455 the intersection between the isopleth that defines  $v_z = R_s/3$  and the anhydrous peri-  
 456 dotite solidus. We assume that all melts generated beyond these limits freeze within the  
 457 lithosphere. Thus, the thickness of crust produced at the ridge axis ( $t_c$ ) is calculated by

$$458 \quad t_c = \frac{1}{2R_s} \frac{\rho_s}{\rho_c} \int_A M dA, \quad (17)$$

459 where  $M$  and  $A$  correspond to the the melting rate and melting region over which melt  
 460 reaches the ridge axis, respectively (Forsyth, 1993). We assume that the density of oceanic  
 461 crust ( $\rho_c$ ) is  $2900 \text{ kg m}^{-3}$ .

462 Melt compositions are calculated once melting diagnostics (i.e.,  $F$  and  $M$ ) within  
 463 the simulation reach a steady state. For the majority of these calculations, we assume  
 464 that the mantle source is depleted (see Section 3.4; Salters & Stracke, 2004). We impose  
 465 the spinel-garnet transition zone between 69–70 km depth (Jennings & Holland, 2015).  
 466 Observations of radiometric isotopic systems at mid-oceanic ridges demonstrate that melts  
 467 ascend through the mantle at velocities of order  $10\text{--}100 \text{ s m yr}^{-1}$  (Stracke et al., 2006).

At these speeds, melts reach the ridge  $10^3$ – $10^5$  yrs after inception. Given that these timescales are much shorter than those associated with mantle flow, we can assume that all melts generated within the magmatic focussing area ( $A$ ), regardless of depth, are instantaneously extracted at the ridge axis. More productive regions of the simulation will contribute higher volumes of melt to the ridge axis. To account for these variations in productivity, when calculating the average composition of this pooled melt ( $C_l$ ), we weight each instantaneous melt generated within the melt zone by its corresponding melt rate:

$$C_l = \frac{\int_A c_l M dA}{\int_A M dA}. \quad (18)$$

Crustal thicknesses and melt compositions calculated by our mid-oceanic ridge simulation are a function of mantle  $T_p$ ,  $R_s$ , mantle composition, and the depth of the spinel-garnet transition zone. We systematically vary these parameters and compare our crustal thickness and melt composition predictions to global databases of mid-oceanic ridge observations.

#### 4.1 Effect of Potential Temperature on Melting

In Figure 4,  $T_p$  is varied in 50 °C increments between 1275–1475 °C and  $R_s$  is held constant at 2.1 cm yr<sup>-1</sup>. Increasing  $T_p$  displaces the solidus to greater depth, which causes the melt region to expand and higher melt fractions to be attained (McKenzie & Bickle, 1988, Figure 4a–c). Additionally, since the effective viscosity is inversely proportional to temperature, increasing  $T_p$  from 1275 °C to 1475 °C results in concomitant increases in upwelling and melting rates (1.96–2.34 cm yr<sup>-1</sup> and 0.034–0.073 Myr<sup>-1</sup> at 1.5 GPa beneath the ridge axis, respectively). The combined effects of a deeper solidus and increased upwelling rates widen the magmatic focusing distance ( $x_f$ ) and supply a greater volume of melt to the ridge axis. As a result, we find that calculated crustal thicknesses increase from 3.19 – 14.94 km as  $T_p$  increases from 1275 – 1475 °C (Figure 4g).

We compare our results to a compilation of oceanic plate crustal thickness estimates derived from seismic velocity profiles (Hoggard et al., 2017). Where these estimates occur on crust < 3 Ma, we assign these crustal thickness values to their corresponding ridge segment (78 locations world-wide; Figure 5a). Potential temperatures for each ridge segment are estimated using a global parameterization of mantle  $T_p$  at 150±25 km depth calculated by converting the SL2013sv shear-wave tomographic model to temperature by fitting upper mantle seismic observations beneath the oceans to a plate model (Figure 5a; Schaeffer & Lebedev, 2013; Richards et al., 2020). Corresponding spreading rates for each ridge segment are taken from a global database (Gale et al., 2013). These rates are based upon established estimates of present-day absolute plate motion (NUVEL-1A and NNR-MORVEL56; Figure 5b; DeMets et al., 1994; Argus et al., 2011). Since our simulations are generated with a constant half-spreading rate of 2.1 cm yr<sup>-1</sup>, results are only compared to crustal thickness values from ridges where  $R_s = 2.1 \pm 1$  cm yr<sup>-1</sup>. We find that crustal thickness estimates from our simulations are more sensitive to variations in  $T_p$  than crustal thicknesses measured at present-day mid-oceanic ridges (Figure 4g). For example, our simulations predict a crustal thickness of 8.68 km at a  $T_p$  of 1375 °C, whereas potential temperatures are estimated to be ~ 100 °C hotter under modern-day ridges with similar crustal thicknesses (Figure 4g).

Na<sub>2</sub>O concentration is among the most-commonly used geochemical metrics at mid-oceanic ridges (e.g., Klein & Langmuir, 1987). As demonstrated in Section 3.5, Na becomes more compatible within clinopyroxene as depth increases (Figure 3; Blundy et al., 1995). Therefore, as  $T_p$  increases and the solidus deepens, Na<sub>2</sub>O concentrations within the first melts decrease (Figure 4d–f). We find that simulations with elevated potential temperatures generate greater melt fractions over deeper melting columns, and consequently supply melts with lower Na<sub>2</sub>O concentrations to the ridge axis. In Figure 4h, our Na<sub>2</sub>O results are compared to the average Na<sub>90</sub> value (i.e., Na<sub>2</sub>O concentrations

518 corrected for the effects of fractionation so that the melt is in equilibrium with 0.9 Mg#  
 519 olivine) recorded for basalts within each ridge segment (Figure 5b; Gale et al., 2014). As-  
 520 suming that this correction is appropriate, our predictions of Na<sub>2</sub>O concentrations should  
 521 therefore be comparable to Na<sub>90</sub> values. Again, we filter this database to only show Na<sub>90</sub>  
 522 values for ridges where  $R_s = 2.1 \pm 1$  cm yr<sup>-1</sup> and compare these values to  $T_p$  estimates  
 523 at  $150 \pm 25$  km depth (Richards et al., 2020). Observed Na<sub>90</sub> concentrations decrease  
 524 as potential temperatures increase up to  $\sim 1400$  °C. At  $\sim 1400$  °C, there is a break in  
 525 slope and Na<sub>90</sub> values decrease at a slower rate as  $T_p$  approaches 1500 °C. Our results  
 526 closely match observed Na<sub>90</sub> concentrations at potential temperatures  $< 1400$  °C, al-  
 527 though they do not fully-replicate the observed break in slope between Na<sub>90</sub> and  $T_p$ . In-  
 528 stead, our simulations with  $T_p > 1400$  °C slightly underestimate observed Na<sub>90</sub> values.

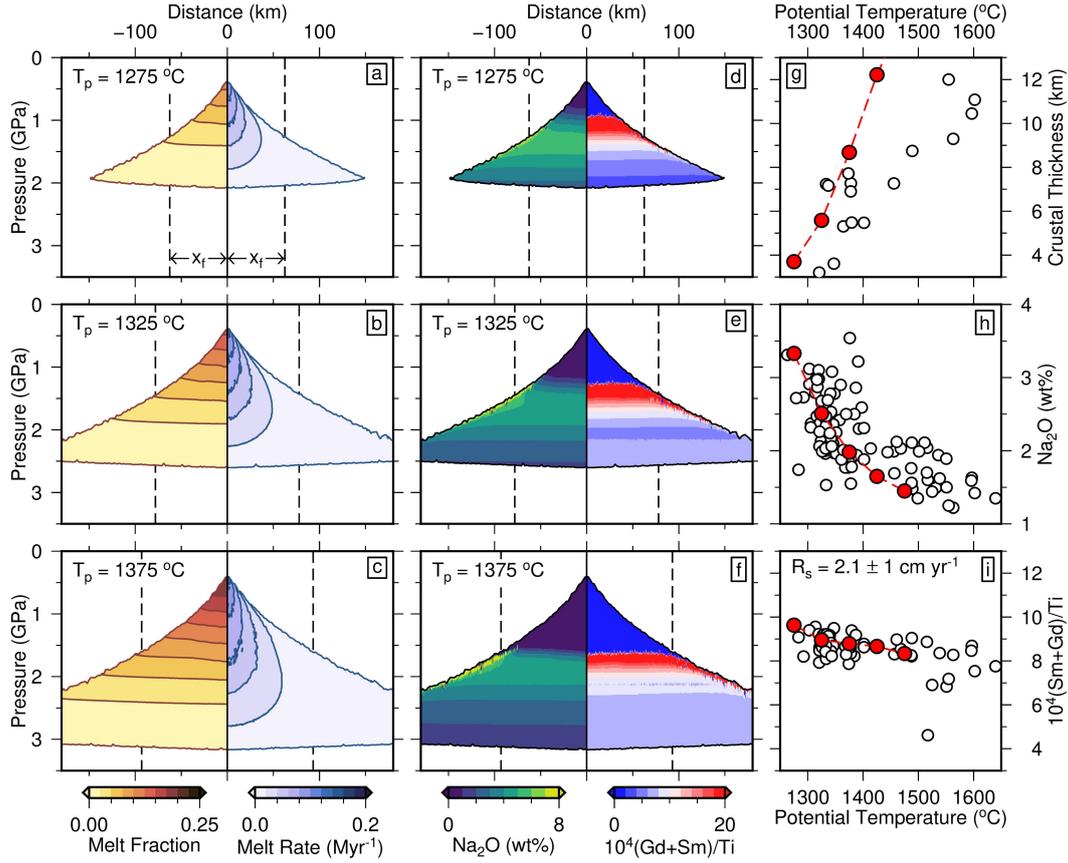
529 The hypothesis that Na<sub>2</sub>O concentrations in mid-oceanic ridge basalts are inversely  
 530 proportional to mantle temperature variations is well established (e.g., Klein & Lang-  
 531 muir, 1987; Langmuir et al., 1992; Plank & Langmuir, 1992; Gale et al., 2013; Dalton  
 532 et al., 2014). However, some authors argue that shallow magma chamber processes and  
 533 mantle compositional heterogeneity, rather than  $T_p$ , are the primary drivers of geochem-  
 534 ical variability in mid-oceanic ridge basalts (e.g., O’Hara, 1977, 1985; Niu & O’Hara, 2008;  
 535 O’Neill & Jenner, 2012; Niu, 2016; O’Neill & Jenner, 2016). These studies highlight that  
 536 correcting Na<sub>2</sub>O concentrations back to Na<sub>90</sub> values is difficult since Na is compatible  
 537 in plagioclase and magma chambers are constantly replenished by primitive melts.

538 One way to avoid the pitfalls of conducting a fractional crystallisation correction  
 539 is to employ incompatible element ratios. The geochemical ratio  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  is thought  
 540 to be largely unaffected by fractional crystallisation and is observed to be almost con-  
 541 stant throughout the mid-oceanic ridge system ( $8.61 \pm 0.47$ ; O’Neill & Jenner, 2016).  
 542 Sm and Gd are far more compatible in clinopyroxene than in orthopyroxene, whereas  
 543 the compatibility of Ti is more evenly spread between the two (Supplementary Table 1).  
 544 Given this assumption, as clinopyroxene is consumed during partial melting, O’Neill and  
 545 Jenner (2016) predicts that the  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  value of the melt should rapidly decrease.  
 546 Given that this trend is not observed, O’Neill and Jenner (2016) infer that melt fractions  
 547 do not vary significantly beneath mid-oceanic ridges (Figure 4i).

548 Within our simulations, however, we find that  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  increases as a func-  
 549 tion of melt fraction until Gd and Sm are exhausted, at which point the ratio becomes  
 550 zero (Figure 4d-f). This difference between our predictions and those of O’Neill and Jen-  
 551 ner (2016) arise since the relative abundances of clinopyroxene and orthopyroxene in the  
 552 residue and entering the instantaneous melt are also a function of depth (Figure 2). By  
 553 the time melt fractions reach  $\sim 15\%$ , Sm, Gd, and Ti are almost exhausted in the man-  
 554 tle and so the  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  value of the integrated melt at the ridge axis is approx-  
 555 imately equal to that of the original source (8.33; Figure 4i; Salters & Stracke, 2004). Con-  
 556 sequently, we recover  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values within error of the global average at a range  
 557 of  $T_p$  values (Figure 4h). Our results suggest that the limited spread of  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$   
 558 values cannot be used as evidence that primitive melt compositions, and therefore cumu-  
 559 lative melt fractions, are approximately uniform beneath the global mid-oceanic ridge  
 560 system. We note that the upper flanks of our melting region record high Na<sub>2</sub>O and  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$   
 561 values. These high values arise from low melt productivity at the margins of the simu-  
 562 lation and, as such, have a negligible influence on our predicted melt compositions, since  
 563 they are down-weighted in our calculations (Equation 18).

## 564 4.2 The Effect of Spreading Rate on Melting

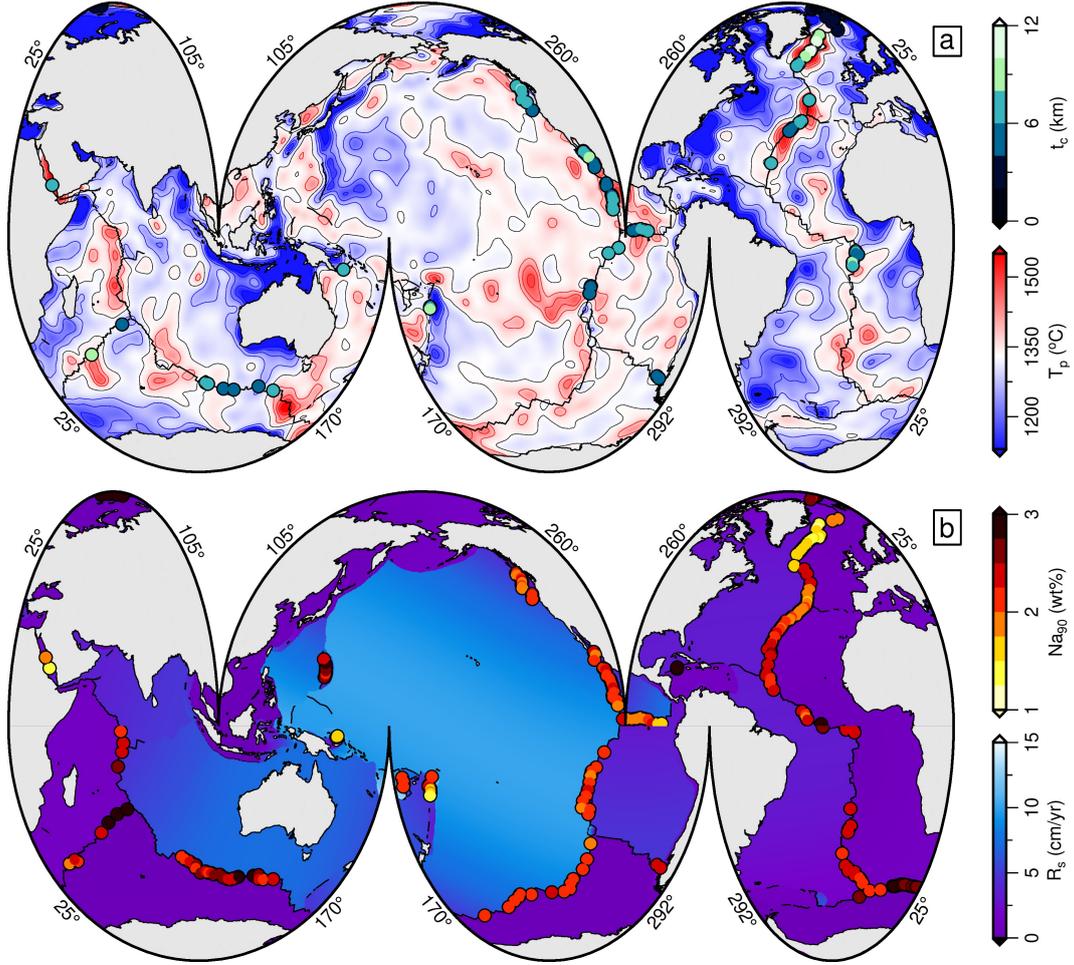
565 To generate the mid-oceanic ridge cases shown in Figure 6, we keep  $T_p$  fixed at 1325 °C  
 566 and vary half-spreading rate ( $R_s$ ) between 0.5–10 cm yr<sup>-1</sup>. At slow spreading rates, pas-  
 567 sive upwelling speeds, and therefore melting rates, are substantially reduced. Mantle as-  
 568 cending at slower speeds is subjected to more extensive conductive cooling from the sur-



**Figure 4.** Melt region geometry at mid-oceanic ridges as a function of mantle temperature. a) Ridge-centered simulation with left- and right-hand sides coloured by melt fraction and melting rate, respectively. Simulation run with a potential temperature,  $T_p$ , of 1275 °C and a half-spreading rate,  $R_s$ , of 2.1 cm yr<sup>-1</sup>. b-c) Same as panel a but with different  $T_p$  as indicated in top left corner of each panel. d) Same as panel a but left- and right-hand sides coloured by Na<sub>2</sub>O concentration and  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  within the instantaneous melt phase, respectively. e-f) Same as panel d but with different  $T_p$  as indicated in top left corner of each panel. g) Crustal thickness as function of  $T_p$ . Red circles and dashed lines = calculated crustal thicknesses from cases shown in panels a-c and Supplementary Figure 7b,e; white circles = observed crustal thicknesses at mid-oceanic ridge segments with  $R_s = 2.1 \pm 1$  cm yr<sup>-1</sup> (Gale et al., 2013; Hoggard et al., 2017; Richards et al., 2020). h) Same as panel g with predicted Na<sub>2</sub>O concentrations compared to estimated Na<sub>90</sub> values at each ridge segment (Gale et al., 2013, 2014; Richards et al., 2020). i) Same as panel g with predictions compared to average  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values at each ridge calculated using samples with MgO > 8.5 wt% (Gale et al., 2013; Richards et al., 2020).

569 face, which increases effective viscosity, further decreasing upwelling rates and, thus, caus-  
 570 ing melting to cease at greater depths.

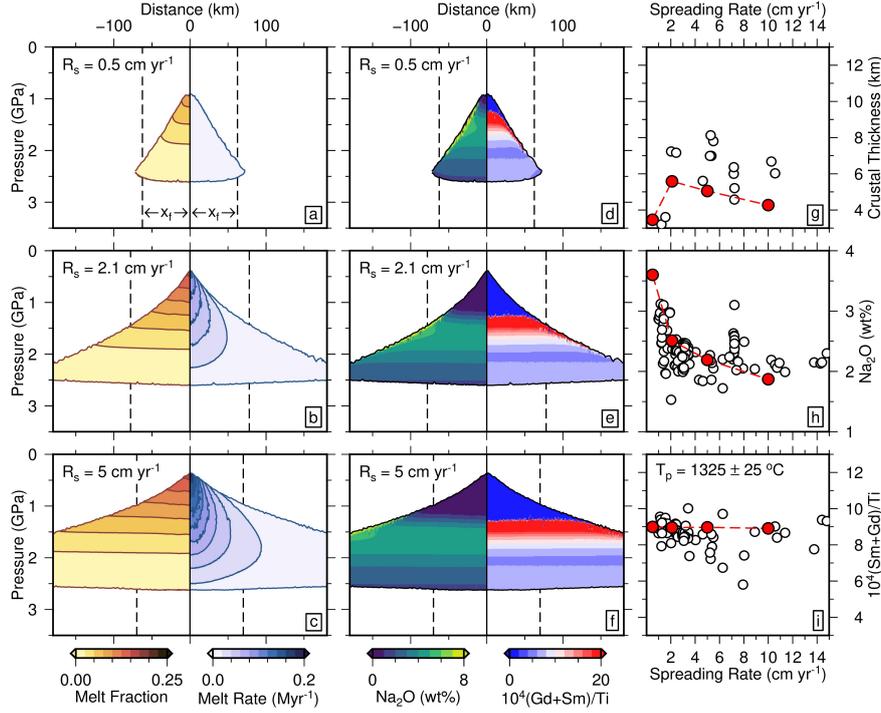
571 In our simulations, these effects are particularly pronounced at  $R_s < 2.1$  cm yr<sup>-1</sup>.  
 572 When  $R_s \geq 2.1$  cm yr<sup>-1</sup>, conductive cooling is reduced and maximum melt fraction  
 573 and melting rates, in addition to crustal thickness, increase significantly (Figures 6a-c).  
 574 However, as  $R_s$  increases beyond 2.1 cm yr<sup>-1</sup> we observe a narrowing in the melt focus-  
 575 ing distance,  $x_f$ , and a gradual reduction in crustal thickness estimates. This pattern



**Figure 5.** Geographic distribution of mid-oceanic ridge observations. a) Segmented Mollweide projection of globe showing estimate of potential temperature ( $T_p$ ) at  $150 \pm 25$  km depth (Richards et al., 2020). Circles = loci of oceanic crustal thickness estimates (Hoggard et al., 2017). b) Grid of present-day plate velocity magnitude (Argus et al., 2011). Circles = loci of mid-oceanic ridge segments with basaltic rock trace element data coloured by average  $Na_{90}$  value (Gale et al., 2014).

576 of low crustal thicknesses at  $R_s < 2.1$  cm yr<sup>-1</sup>, a sharp increase in crustal thicknesses  
 577 when  $R_s \geq 2.1$  cm yr<sup>-1</sup>, and a subsequent slow decrease in crustal thickness as a func-  
 578 tion of  $R_s$ , is observed, to a certain degree, in our global crustal-thickness database (af-  
 579 ter filtering to only include ridges overlying ambient mantle i.e.,  $T_p = 1325 \pm 25$  °C;  
 580 Figure 6g). However, it is unclear to what extent observed crustal thicknesses decrease  
 581 at  $R_s \geq 2.1$  cm yr<sup>-1</sup> and at these half-spreading rates simulated crustal thicknesses are  
 582  $\sim 1$  km less than observed crustal thicknesses (Figure 6g).

583 Since the melting region beneath mid-oceanic ridges tapers towards the ridge axis,  
 584 narrowing the magmatic focusing distance decreases the relative contribution of deep,  
 585 incompatible-element rich melts (Figure 6d–f). Therefore, our simulations predict a de-  
 586 crease in  $Na_{2O}$  concentrations as spreading rates increase. This predicted decrease is some-  
 587 what matched by the distribution of observed  $Na_{90}$  values. We find that the highest  $Na_{90}$   
 588 values decrease as a function of  $R_s$ , but that the lowest  $Na_{90}$  values remain at  $\sim 2$  wt%

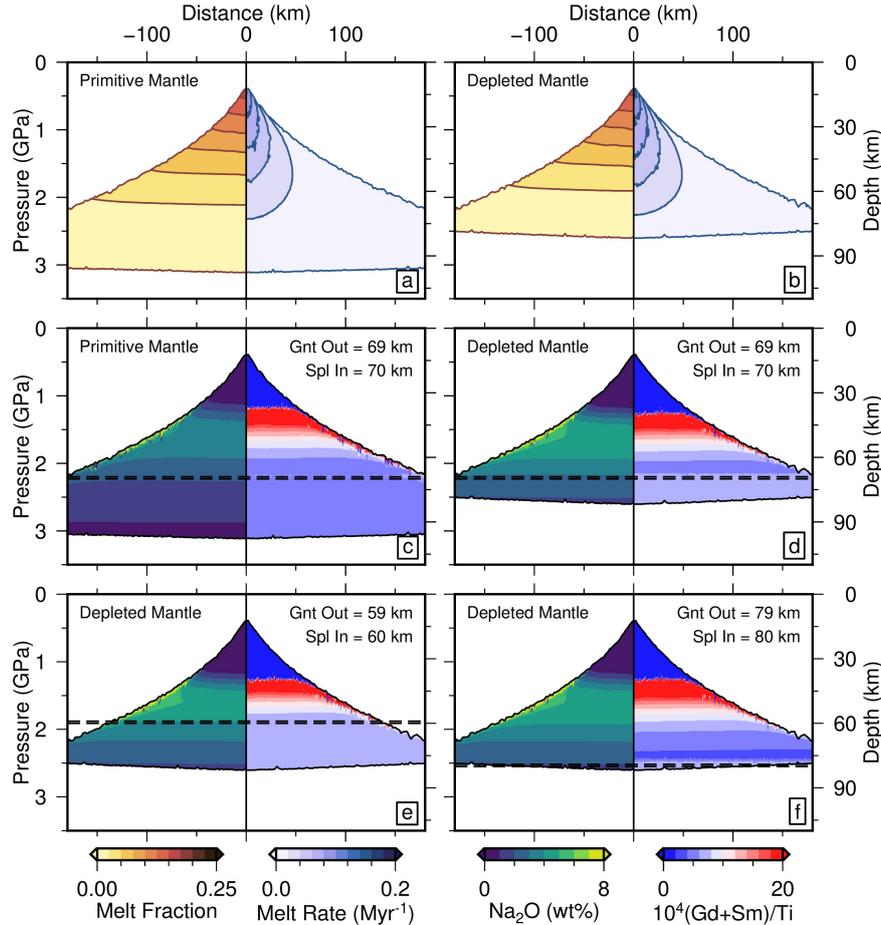


**Figure 6.** Melt region geometry at mid-oceanic ridges as a function of half-spreading rate ( $R_s$ ). a) Ridge-centered simulation with left- and right-hand sides coloured by melt fraction and melting rate, respectively. Simulation run with a potential temperature ( $T_p$ ) of  $1325\text{ }^\circ\text{C}$  and a  $R_s$  of  $0.5\text{ cm yr}^{-1}$ . b–c) Same as panel a but with different  $R_s$  as indicated in top left corner of each panel. d) Same as panel a but left- and right-hand sides coloured by  $\text{Na}_2\text{O}$  concentration and  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  within the instantaneous melt phase, respectively. e–f) Same as panel d but with different  $R_s$  as indicated in top left corner of each panel. g) Crustal thickness as function of  $R_s$ . Red circles and dashed lines = calculated crustal thicknesses within melt focusing region from cases shown in panels a–c and Supplementary Figure 7g; white circles = observed crustal thicknesses at mid-oceanic ridge segments with  $T_p = 1325 \pm 25\text{ }^\circ\text{C}$  (Gale et al., 2013; Hoggard et al., 2017; Richards et al., 2020). h) Same as panel g with predicted  $\text{Na}_2\text{O}$  concentrations compared to  $\text{Na}_{90}$  values at each ridge segment (Gale et al., 2013, 2014; Richards et al., 2020). i) Same as panel g with predictions compared to average  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values at each ridge calculated using samples with  $\text{MgO} > 8.5\text{ wt}\%$ .

589 at all spreading rates (Figure 6h; Gale et al., 2013). Varying spreading rate or  $x_f$  does  
 590 not significantly affect  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  values since, in all cases, melting is sufficiently  
 591 extensive that this ratio remains close to that of the original mantle source (Figure  
 592 6i).

### 593 4.3 Effect of Mantle Composition and Mineralogy on Melting

594 Mantle composition and mineralogy exert a strong influence on both melt produc-  
 595 tivity and composition (Section 3). Fertile mantle has greater concentrations of incom-  
 596 compatible elements and volatiles, and a greater abundance of clinopyroxene than depleted  
 597 mantle (Gast, 1968; Morgan, 1968; Walter, 1998). Consequently, if  $T_p$  and  $R_s$  are held  
 598 constant, the melting region beneath a ridge underlain by more fertile mantle will be deeper,



**Figure 7.** Melt region geometry at mid-oceanic ridges as a function of mantle mineralogy and composition. a) Ridge-centered simulation with left- and right-hand sides coloured by melt fraction and melting rate, respectively. Simulation run with a potential temperature of 1325 °C and a half-spreading rate of 2.1 cm yr<sup>-1</sup>. Simulation has primitive mantle composition and the spinel garnet transition zone is between 69–70 km depth. b) Same as panel a but with depleted mantle composition. c/d) Same as panels a/b but left- and right-hand sides coloured by Na<sub>2</sub>O concentration and 10<sup>4</sup>(Sm+Gd)/Ti within the instantaneous melt phase, respectively. Dashed lines denote spinel-garnet-transition zone depths e–f) Same as panel d but with different spinel-garnet-transition-zone depths as indicated in top right corner of each panel.

599 since increasing volatile content of the mantle deepens the solidus (Katz et al., 2003).  
 600 Mid-oceanic ridge crustal thicknesses and basaltic compositions may therefore be more  
 601 sensitive to mantle source heterogeneity than to temperature or spreading rate (e.g., Niu  
 602 & O’Hara, 2008; Niu, 2016). Here, we assess the sensitivity of our modelling results to  
 603 mantle composition.

604 To ascertain the extent to which source fertility and spinel-garnet-transition-zone  
 605 depth affect the cumulative melt compositions, we perform simulations where these param-  
 606 eters are varied between extrema (Figure 7). In these cases,  $T_p$  and  $R_s$  are 1325 °C  
 607 and 2.1 cm yr<sup>-1</sup>, respectively. Our primitive source has 180 ppm more H<sub>2</sub>O than our  
 608 depleted source and, so, increasing mantle fertility from depleted to primitive mantle in-  
 609 creases the pressure of the solidus from  $\sim 2.6 - 3.2$  GPa (Figure 7a,b). The more ex-

610 tensive melting region of the primitive mantle case generates crustal thicknesses 0.35 km  
 611 greater than those calculated for the comparable depleted mantle case. Clinopyroxene  
 612 is more abundant in fertile mantle and, as noted above, Na is more compatible within  
 613 clinopyroxene-rich mantle. As a result, we calculate lower Na concentrations in the cu-  
 614 mulative melts delivered to the ridge axis in the fertile mantle case (2.02 wt% compared  
 615 to 2.51 wt% for the depleted mantle case; Figure 7c,d). As mantle temperatures increase,  
 616 we expect that the difference between Na concentrations in fertile and depleted mantle  
 617 simulations to reduce as clinopyroxene is progressively consumed. At the highest poten-  
 618 tial temperatures examined, where clinopyroxene is exhausted during melting, we would  
 619 expect primitive mantle to generate crust with higher Na concentrations than depleted  
 620 mantle since fertile mantle contains a higher initial concentration of incompatible ele-  
 621 ments.  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  is lower in our primitive mantle source than in our depleted man-  
 622 tle source. As a result, the  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  value recorded for our primitive mantle case  
 623 is lower than for our depleted case (8.08 and 8.96, respectively). However, both values  
 624 are close to the observed global average along mid-oceanic ridges ( $8.61 \pm 0.47$  O'Neill  
 625 & Jenner, 2016).

626 At the spinel-garnet-transition zone, garnet and olivine are replaced by a combi-  
 627 nation of orthopyroxene and spinel. Increasing the spinel-garnet-transition-zone depth  
 628 does not greatly affect  $\text{Na}_2\text{O}$  concentrations since  $\text{Na}_2\text{O}$  is most sensitive to the amount  
 629 of clinopyroxene in the mantle, which does not change at this transition (Figure 7e,f).  
 630 The incompatibility of Ti in the mantle source sharply increases at depths shallower than  
 631 the spinel-garnet-transition zone. Gd and Sm also become more incompatible in the spinel  
 632 stability field, but incompatibility slowly increases as pressure decreases, rather than be-  
 633 ing strongly focussed at the spinel-garnet transition itself, as is the case with Ti (Fig-  
 634 ure 3). As a result, a deep spinel-garnet-transition zone generates deep melts with low  
 635  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  ratios (Figure 7f). These melts are down-weighted in the calculation  
 636 of the average  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  ratio at the ridge-axis, since deeper melts have lower melt-  
 637 ing rates. Consequently,  $10^4(\text{Gd}+\text{Sm})/\text{Ti}$  decreases from 9.26 to 8.81 as the spinel-garnet-  
 638 transition-zone depth increases from 59–60 to 79–80 km. Although the elements and ra-  
 639 tios investigated here are not greatly affected by spinel-garnet-transition-zone depth, we  
 640 note that this does have a significant effect on the concentrations of elements that are  
 641 more compatible in garnet than other upper-mantle minerals (Kay & Gast, 1973).

642 We note that in our simulations we do not consider the relative densities of fertile  
 643 and depleted mantle. Fertile mantle is denser than depleted mantle (Jordan, 1978). A  
 644 mid-oceanic ridge underlain by more fertile mantle may therefore sit deeper beneath the  
 645 sea surface. Deeper ridges will cease melting at higher pressures than shallow ridges, which  
 646 may effect the composition of erupted melts (Niu & O'Hara, 2008). However, mantle den-  
 647 sity and, therefore, ridge depth, may be more sensitive to temperature variations than  
 648 compositional variations, as highlighted by the negative correlation between sub-solidus  
 649 mantle shear-wave velocities and ridge depth (Dalton et al., 2014). For a list of all mid-  
 650 oceanic ridge simulation results see Supplementary Table S3.

## 651 5 Application to Lithospheric Steps

652 In intra-plate magmatic settings adjacent to lithospheric steps, melting is typically  
 653 deeper, less productive and shorter lived than at mid-oceanic ridges (Duvernay et al., 2021).  
 654 To ensure that our approach is also valid in these low-melt-productivity environments,  
 655 we construct and analyse a series of coupled geochemical-geodynamical simulations that  
 656 are applicable for such intra-plate settings.

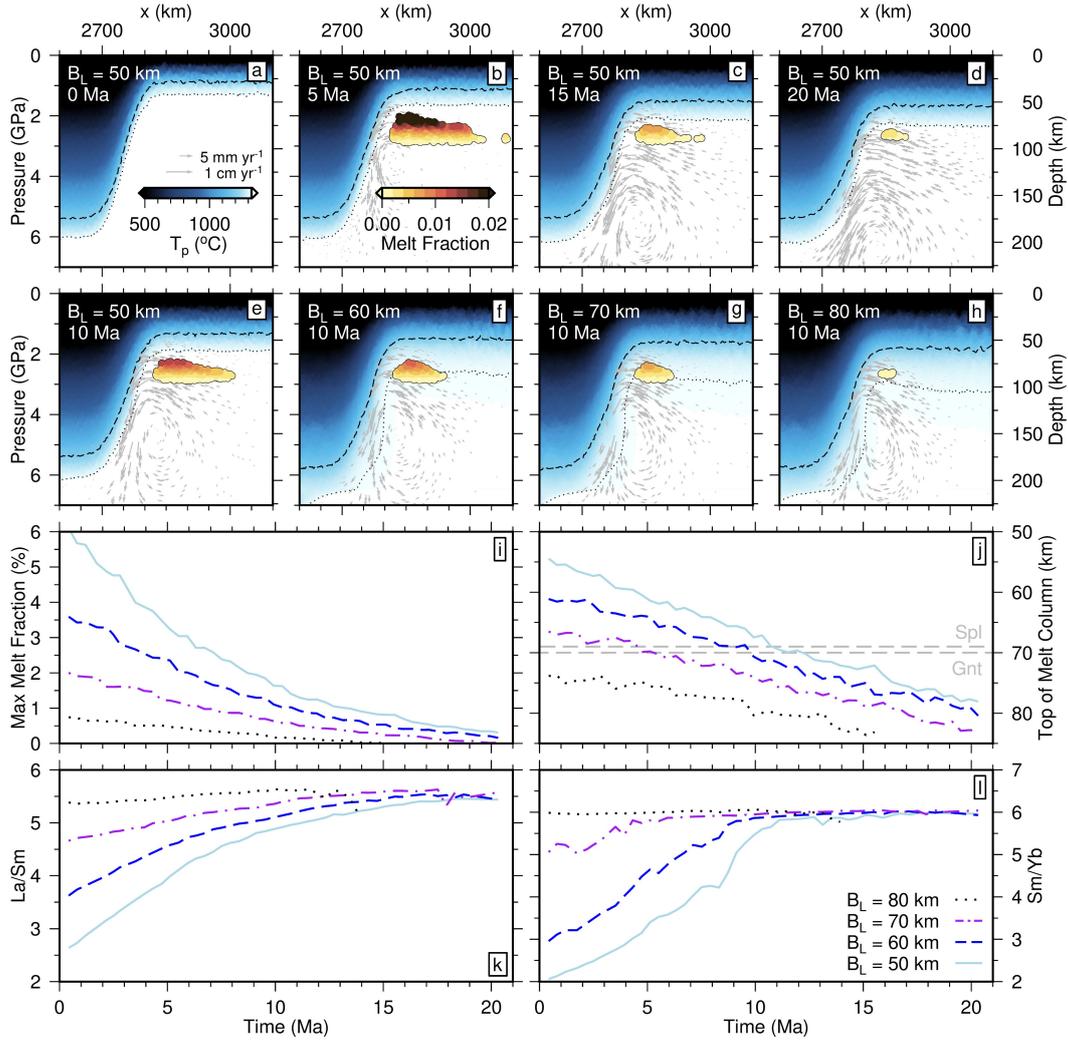
657 Edge-driven convective cells form along steps in lithospheric thickness, for exam-  
 658 ple where thick continental interiors abut thinner continental or oceanic margins (Figure  
 659 1b; King & Anderson, 1998). Locations where these cells have been suggested as causal  
 660 mechanisms for present-day volcanism include eastern Australia, Cameroon and the Col-

661 orado Plateau (Davies & Rawlinson, 2014; Adams et al., 2015; Ballmer et al., 2015; Afonso  
 662 et al., 2016; Rawlinson et al., 2017). In these regions, intricate details of 3-D lithospheric  
 663 architecture, the magnitude and direction of plate motion and/or asthenospheric flow,  
 664 and potential complications arising from complex flow induced by adjacent subducting  
 665 slabs and/or mantle plumes can influence the distribution and extent of melting (Ballmer  
 666 et al., 2007; Conrad et al., 2011; Davies & Rawlinson, 2014; Kennett & Davies, 2020; Du-  
 667 vernay et al., 2021). As such, replicating specific melting observations at one of these lo-  
 668 cations is beyond the scope of this paper. Instead, our aim is to reproduce expected first-  
 669 order differences between melting at mid-oceanic ridges and at lithospheric steps, and  
 670 to confirm that these simulations generate similar melt chemistry to lavas observed ad-  
 671 jacent to lithospheric steps. Furthermore, we test the sensitivity of our 2-D lithospheric  
 672 step simulations to the thickness of the overlying oceanic thermal-boundary-layer, high-  
 673 lighting the role of pressure and the overlying lid in dictating melt compositions.

674 In the case presented in Figure 8a, we impose an initial oceanic lithospheric thick-  
 675 ness ( $B_L$ ) of 50 km, the spinel-garnet-transition zone is placed at 69–70 km depth, and  
 676 an equal mix of primitive and depleted mantle with a water content of 200 ppm is as-  
 677 sumed. Under these conditions the solidus is situated at  $\sim 90$  km depth and, hence, melt  
 678 is generated in locations where asthenospheric mantle upwells at depths  $\lesssim 90$  km. The  
 679 temperature contrast between cold continental lithosphere and horizontally adjacent as-  
 680 thenosphere causes the asthenosphere close to the continent to cool. As this asthenosphere  
 681 cools, its density increases and it begins to sink, displacing deeper asthenosphere which  
 682 upwells as a passive return flow (see grey arrows in Figure 8b-h). This convection cell  
 683 intensifies as the instability grows and continuously supplies fresh mantle to the the base  
 684 of the plate, thus maintaining a melting zone close to the lithospheric step (Figures 8b-  
 685 e). Over time, the oceanic lithosphere thickens through thermal diffusion. This thick-  
 686 ening decreases the distance between the solidus and the base of the thermal boundary  
 687 layer and, as a result, the melting region becomes increasingly restricted. Consequently,  
 688 cases with thicker initial thermal boundary layers cease melting sooner and have lower  
 689 maximum melt fractions (Figure 8e-j). The edge-driven cell next to the lithospheric step  
 690 continuously supplies hot asthenosphere to the base of the plate and retards the rate of  
 691 conductive cooling. Nevertheless, by 25 Myr, melting has ceased in all simulations.

692 Variations in the extent and depth of melting are reflected in the compositions of  
 693 the melts generated. When calculating erupted-melt compositions from these simulations,  
 694 we weight melts according to their melting rates and assume that all melt is instantane-  
 695 ously extracted (i.e.,  $A$  encompasses the whole melting region; Equation 18). La is more  
 696 incompatible within the mantle than Sm and hence higher melt fractions yield lower La/Sm  
 697 ratios within the integrated melt (Figure 3). Accordingly, melts generated towards the  
 698 beginning of each simulation have lower La/Sm ratios (Figure 8k). Yb is much more com-  
 699 patible in garnet than spinel, and so Sm/Yb values are sensitive to the proportion of melt-  
 700 ing in the spinel and garnet stability fields. As the depth to the top of the melt column  
 701 increases, melting ceases within the spinel stability field (Figure 8j). As each simulation  
 702 progresses, Sm/Yb values therefore increase as a higher proportion of melting occurs in  
 703 the garnet stability field.

704 The geochemical results from our simulations resemble compositions observed along  
 705 present-day lithospheric steps. Using a recent global compilation, average La/Sm val-  
 706 ues for high-MgO basalts from Cameroon, Western USA and the Newer Volcanic Province,  
 707 Australia, are  $5.53 \pm 0.77$ ,  $5.17 \pm 1.09$  and  $5.94 \pm 2.42$ , respectively (Ball et al., 2021).  
 708 Both geochemical modeling and mantle temperature estimates derived from seismic to-  
 709 mographic models suggest that the lithosphere beneath these volcanic provinces is  $\sim 50$ -  
 710 65 km thick (Hoggard et al., 2020; Ball et al., 2021). Under these conditions, our sim-  
 711 ulations predict lower La/Sm values than those recorded at the aforementioned locations  
 712 (La/Sm  $\approx 4$ ; Figure 8k). However, complexities that we do not include, such as the pres-  
 713 ence of trace-element-enriched fusible mantle phases, crystal fractionation, and the in-



**Figure 8.** Melting geometry along lithospheric steps as a function of initial thermal-boundary-layer thickness ( $B_L$ ) and time. a–h) Lithospheric-step centred melting simulation with background coloured by temperature; dotted/dashed contours = isothermal surfaces with temperature of 1325/1200 °C; coloured circles = melt fraction on each particle;  $B_L$  and duration the simulation has run for indicated top-left corner. i) Maximum melt fraction as a function of simulation time for four simulations with  $B_L$  of 50–80 km. j) Top of melt column as a function of simulation duration; dashed gray lines = top and bottom of spinel-garnet transition zone. k) Cumulative La/Sm ratio in the melt as a function of simulation duration. l) Cumulative Sm/Yb ratio in the melt as a function of simulation duration.

714 corporation of metasomatised lithosphere can act to increase La/Sm values in the final  
715 melt.

716 Given the strong sensitivity of our modelling predictions to thermo-chemical structure,  
717 it is clear that accurate replication of incompatible element concentrations and ratios  
718 of lavas at lithospheric steps requires full integration of all available geochemical and  
719 geophysical constraints. Such efforts have begun, particularly those exploiting geophysical  
720 observations of lithospheric structure and thickness (e.g., Rawlinson et al., 2017).

721 However, the use of geochemical constraints remains in its infancy and will doubtless be-  
 722 come a fruitful avenue for future research. It is therefore not surprising that our predic-  
 723 tions do not perfectly match the composition of igneous rocks observed at some of these  
 724 locations. Nevertheless, the results presented in this study are encouraging. Our find-  
 725 ings demonstrate that BDD21 provides a framework within which to test different geo-  
 726 dynamical scenarios, and to discount those that cannot be reconciled with geochemical  
 727 observations from igneous rocks.

## 728 6 Discussion

729 In this study, we have presented BDD21, a new peridotite melting parameteriza-  
 730 tion that predicts melt productivity and chemistry through space and time. We coupled  
 731 BDD21 to incompressible, single-phase-flow simulations generated using Fluidity, to repli-  
 732 cate mantle melting at two contrasting geologic settings: mid-oceanic ridges and litho-  
 733 spheric steps.

734 Our mid-oceanic-ridge simulations were analysed over a wide range of spreading  
 735 rates and mantle temperatures. They reproduce key melt-related diagnostics that are  
 736 consistent with those observed at present-day ridge segments: (i) crustal thickness and  
 737 incompatible-element concentrations (e.g.,  $\text{Na}_2\text{O}$ ) increase and decrease as a function of  
 738 mantle temperature, respectively; (ii) crustal thickness and incompatible-element con-  
 739 centrations are less sensitive to spreading rate than to mantle temperature at plate ve-  
 740 locities  $> 1.5 \text{ cm yr}^{-1}$ ; and (iii)  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  is approximately constant across the  
 741 parameter space examined, as is observed throughout the mid-oceanic-ridge system (Figures  
 742 4 and 6; Klein & Langmuir, 1987; White et al., 1992; O'Neill & Jenner, 2016). To our  
 743 knowledge, our study is the first to reconcile observed mid-oceanic ridge patterns in  $\text{Na}_2\text{O}$   
 744 and  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  within the same modelling framework.

745 Our lithospheric step simulations illustrate that melt productivity and composi-  
 746 tion are strongly sensitive to the thickness of the overlying lithospheric lid. As these sim-  
 747 ulations progress, melt volumes decrease, incompatible-element concentrations increase,  
 748 and the proportion of melting in the garnet stability field increases as the melting re-  
 749 gion is suppressed by the thickening thermal-boundary layer above. Our mid-oceanic ridge  
 750 and lithospheric step simulations demonstrate that BDD21 can replicate first-order dif-  
 751 ferences in melt distribution and composition between these different geologic settings.  
 752 This result implies that BDD21 can be used to investigate the geodynamical mantle pro-  
 753 cesses responsible for the emplacement of igneous provinces, both at the present day and  
 754 into the geological past.

755 A number of previous studies have calculated melt compositions at mid-oceanic ridges  
 756 by coupling geochemical melting parameterizations to mantle flow simulations (e.g., McKen-  
 757 zie & O'Nions, 1991; Langmuir et al., 1992; Plank & Langmuir, 1992; Asimow et al., 2004;  
 758 Gregg et al., 2009; Behn & Grove, 2015; Brown & Leshner, 2016; Brown et al., 2020; Krein  
 759 et al., 2020). We next consider the implications of our modeling results at mid-oceanic  
 760 ridges and lithospheric steps, compare BDD21 to other melting parameterizations, dis-  
 761 cuss uncertainties that are inherent to parameterizations of this nature, and briefly out-  
 762 line potential future research avenues utilizing our approach.

### 763 6.1 Observational Uncertainties at Mid-Oceanic Ridges

764 We use BDD21 to predict primary magmatic compositions. It is important to re-  
 765 member that when we compare these predictions to observations at mid-oceanic ridges,  
 766 we are comparing to basaltic glasses that represent the vestigial melts of complex mag-  
 767 matic plumbing systems (O'Hara, 1977). Observed  $\text{Na}_2\text{O}$  concentrations are corrected  
 768 for fractional crystallisation so that the melt is in equilibrium with the upper mantle (Na<sub>90</sub>;  
 769 Gale et al., 2014). However, this correction assumes that fractionation occurs in a closed

770 system, and that plagioclase begins crystallising when MgO wt% < 8.5. Modeling frac-  
 771 tional crystallisation in a magmatic system that replenishes over time, or varying the point  
 772 at which plagioclase begins crystallising, can significantly affect calculated  $\text{Na}_{90}$  values  
 773 (e.g., O'Neill & Jenner, 2012; Till et al., 2012). Although corrected compositions should  
 774 be treated with caution, it is encouraging that our mid-oceanic ridge simulations repli-  
 775 cate the broad trends found in two very different geochemical systems -  $\text{Na}_2\text{O}$  and  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$ .  
 776 In future, given that each incompatible element behaves differently during fractional crys-  
 777 tallisation, it may be informative to use our predictive primary melt compositions as a  
 778 starting point to investigate these fractional crystallisation processes.

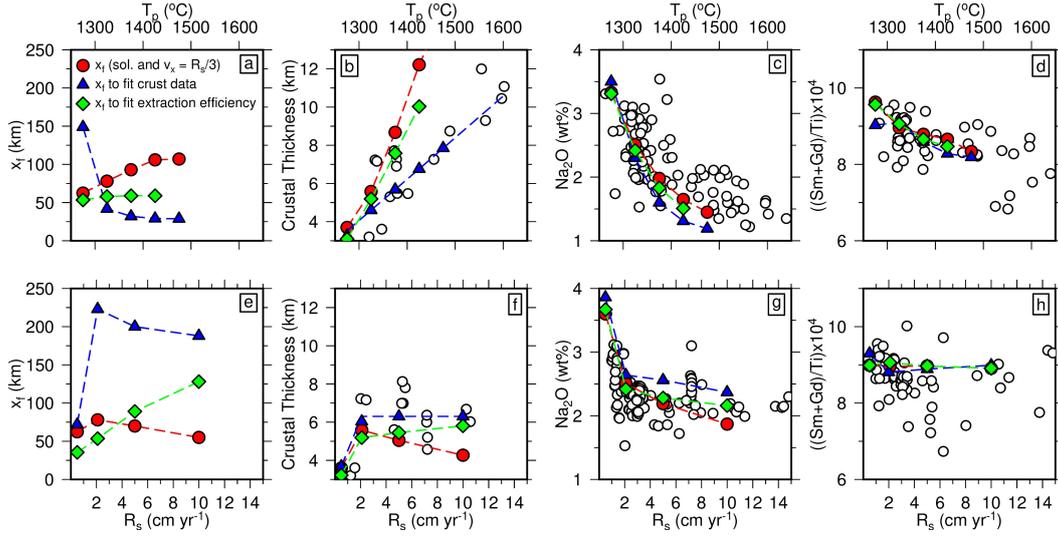
779 We compare our simulations to mid-oceanic ridges that have potential tempera-  
 780 tures that are estimated to be between 1250–1650 °C using techniques that convert from  
 781 seismic to physical structure (Richards et al., 2020). The hottest segments in the mid-  
 782 oceanic ridge system, such as those near Iceland and the Galápagos Islands, are under-  
 783 lain by mantle plumes. All our simulations assume passive upwelling in response to plate  
 784 spreading. However, near plumes, upwelling is likely to be thermally driven by buoyancy  
 785 contrasts between the hot plume and the surrounding mantle (e.g., Ito & Mahoney, 2005).  
 786 If upwelling speeds beneath the ridge are increased by an order of magnitude, crustal thick-  
 787 nesses are predicted to increase by a factor of  $\sim 4$ –5 and incompatible-element concen-  
 788 trations are also predicted to increase (MacLennan et al., 2001). These aspects are not  
 789 captured by our simulations. Moreover, we do not account for possible compositional dif-  
 790 ferences between these plumes and depleted mantle (Fitton et al., 1997; T. Jones et al.,  
 791 2016). A more enriched mantle plume source may be responsible for the slight increase  
 792 in observed  $\text{Na}_{90}$  concentrations at temperatures  $\gtrsim 1425$  °C (Figure 4h). Moreover, an  
 793 increase in enriched mantle material may explain observed low  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values  
 794 (i.e., < 8) at some ridges with  $T_p \gtrsim 1500$  °C (Figure 4i). Taken together, these obser-  
 795 vations suggest that incorporating enriched mantle plumes in future simulations may re-  
 796 duce discrepancies between predicted and observed compositions at higher potential tem-  
 797 peratures.

## 798 6.2 Mid-Oceanic Ridge Simulation Uncertainties

799 A significant source of uncertainty within our single-phase flow modeling approach  
 800 is that we must assume a melt-focusing region (i.e., the region beneath the ridge from  
 801 which melt is extracted) and/or a melt-extraction efficiency (i.e., the proportion of melt  
 802 that is extracted at the ridge axis). In Sections 4.1 and 4.2, we assume that the lateral  
 803 melt-focusing distance ( $x_f$ ) coincides with the intersection between the anhydrous solidus  
 804 and  $v_z = R_s/3$ , and that within the melt-focusing region extraction efficiency is 100%  
 805 (red circles in Figure 9a,e; Keller et al., 2017). An alternative approach would be to choose  
 806 values of  $x_f$  that best fit crustal thickness observations (blue circles in Figure 9a,e). Here,  
 807 we investigate the effect of varying  $x_f$  on melt diagnostics.

808 The simulations shown in Section 4.1 yield crustal thickness estimates that exceed  
 809 those observed, and the difference between calculated and observed crustal thicknesses  
 810 increases as a function of  $T_p$  (Figure 9b). To fit crustal thickness observations,  $x_f$  must  
 811 be reduced to  $\sim 30$  km for simulations with  $T_p > 1275$  °C. However, narrowing the  
 812 melt-focusing region in this way reduces the relative contribution of deeper melts and  
 813 decreases calculated  $\text{Na}_2\text{O}$  concentrations such that they no longer lie within the observed  
 814 range (Figure 9c). Calculated  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values are largely unaffected by chang-  
 815 ing  $x_f$  (Figure 9d).

816 The increasing disparity between predicted and observed crustal thicknesses as po-  
 817 tential temperatures increase may instead result from our assumption that all material  
 818 within the melting region is fusible. The presence of refractory material (i.e., mantle ma-  
 819 terial that does not melt) will reduce calculated crustal thicknesses. If the proportion  
 820 of fusible material is set to a value < 100%, crustal thickness reduces as a function of



**Figure 9.** Varying  $x_f$  to fit mid-oceanic ridge crustal thickness observations. a)  $x_f$  as a function of  $T_p$ . Red circles =  $x_f$  values determined using intersection of anhydrous solidus and  $v_z = R_s/3$ . Blue triangles =  $x_f$  required to generate average observed crustal thickness. Green diamonds =  $x_f$  required to fit extraction efficiency data from Keller et al. (2017). b) Crustal thickness as a function of  $T_p$ . Colored symbols and dashed lines = calculated crustal thicknesses within melt focusing regions defined in panel a; white circles = observed crustal thicknesses at mid-oceanic ridge segments with  $R_s = 2.1 \pm 1 \text{ cm yr}^{-1}$  (Gale et al., 2013; Hoggard et al., 2017; Richards et al., 2020). c) Same as panel b with predicted  $\text{Na}_2\text{O}$  concentrations compared to  $\text{Na}_{90}$  values at each ridge segment (Gale et al., 2013, 2014; Richards et al., 2020). d) Same as panel b with simulation outputs compared to average  $10^4(\text{Sm}+\text{Gd})/\text{Ti}$  values at each ridge calculated using samples with  $\text{MgO} > 8.5 \text{ wt\%}$ . e-h) Same as panels a-d but showing melt diagnostics as function of  $R_s$ . Global databases filtered to include ridges above mantle with  $T_p = 1325 \pm 25 \text{ }^\circ\text{C}$ .

821 the absolute reduction in melt volume (e.g., Keller et al., 2017). An analogous case can  
 822 be made for decreasing crustal thickness by reducing extraction efficiency within the melt-  
 823 focusing region. We note that varying the concentration of refractory material within  
 824 the melt region or the melt-extraction efficiency will change calculated crustal thicknesses  
 825 without affecting melt composition, unless the amount of refractory material in the source  
 826 and/or extraction efficiency varies in space.

827 Away from slow spreading centers and mantle plumes, oceanic crustal thicknesses  
 828 show little variation ( $6.3 \pm 0.8 \text{ km}$ ; White et al., 2001). Like many previous studies, our  
 829 simulations suggest that this range of crustal thicknesses can be generated by ambient  
 830 mantle potential temperatures of  $\sim 1350 \text{ }^\circ\text{C}$  (e.g., Langmuir et al., 1992; Bown & White,  
 831 1994; Brown et al., 2020). To maintain a crustal thickness of 6.3 km within our sim-  
 832 ulations at a constant  $T_p$  of  $1325 \text{ }^\circ\text{C}$  and at a range of spreading rates requires  $x_f$  to in-  
 833 crease by  $\sim 150 \text{ km}$  (Figure 9e,f). Such a substantial widening of the melt-focusing re-  
 834 gion arises due to the sharp reduction of melt productivity in distal regions (i.e., smaller  
 835 melt volume contributions far from the ridge axis), and it leads to a rise of calculated  
 836  $\text{Na}_2\text{O}$  concentrations (Figure 9g). However, such an increase in  $x_f$  is likely unrealistic.  
 837 Instead, raising the ambient potential temperature in our simulations from  $1325 \text{ }^\circ\text{C}$  to  
 838  $\sim 1350 \text{ }^\circ\text{C}$  would increase and decrease calculated crustal thicknesses and  $\text{Na}_2\text{O}$  con-

839 concentrations, respectively, and, thus, improve our fit to observational data without requir-  
 840 ing any widening of the melt-focusing region.

841 In Section 4.2, we predict that crustal thicknesses decrease as a function of spread-  
 842 ing rate. To obtain these values, we make use of the observation of Keller et al. (2017)  
 843 that the  $x_f$  can be approximated through a relationship between upwelling rate, spread-  
 844 ing rate and depth of the anhydrous solidus. Thereby, our predicted decrease occurs be-  
 845 cause, using the approximation of Keller et al. (2017),  $x_f$  in our simulations narrows as  
 846  $R_s$  increases (when  $R_s > 2.1$  cm yr<sup>-1</sup>; Figure 9e). However, within the Keller et al. (2017)  
 847 simulations,  $x_f$  increases as a function of  $R_s$  and calculated crustal thicknesses remain  
 848 approximately constant. Such a change in behavior likely results from differences between  
 849 the thermal and rheological structure of our simulations and those of Keller et al. (2017).  
 850 Alternatively, we can define  $x_f$  according to the extraction efficiency observed in the sim-  
 851 ulations of Keller et al. (2017). In this approach, the total volume of melt generated within  
 852 a distance  $x_f$  from the ridge axis is equal to the extraction efficiency of the whole melt-  
 853 ing region (green line in Figure 9; see Supplementary Materials for method description).  
 854 This method yields increasing values of  $x_f$  as a function of  $R_s$ , in agreement with Keller  
 855 et al. (2017) (Figure 9e). Redefining  $x_f$  to fit extraction efficiency yields in a better fit  
 856 to crustal thickness data and little change to calculated geochemical diagnostics. In fu-  
 857 ture, documenting trade-offs between crustal thickness, mantle composition, melt focus-  
 858 ing, extraction efficiency,  $T_p$ , and  $R_s$  on the outputs of our coupled geochemical-geodynamical  
 859 simulations may play an important role in refining our understanding of mid-oceanic ridge  
 860 systems.

### 861 6.3 Lithospheric Step Simulation Uncertainties

862 The uncertainties that exist when predicting melt compositions at mid-oceanic ridges  
 863 (e.g., lithospheric processing, melt focusing and source heterogeneity) are amplified at  
 864 intraplate settings, since melt fractions are lower and the thermo-chemical structure and  
 865 underlying flow-regime are less well constrained. However, igneous rock compositions rep-  
 866 resent one of the few viable constraints for investigating how the upper mantle’s thermo-  
 867 chemical structure changes through space and time, and accordingly, the results gener-  
 868 ated by BDD21 deserve careful consideration. Our edge-driven convection simulations  
 869 show that, over time, the thermal-boundary layer at lithospheric steps should cool and  
 870 thicken. This thickening restricts the melting region and generates melts that are increas-  
 871 ingly enriched in incompatible elements and with more-pronounced garnet signatures (e.g.,  
 872 high Sm/Yb; Figure 8). This pattern is a robust prediction of our models and, as such,  
 873 may be observed in locations where edge-driven convection has been postulated as the  
 874 driving mechanism of volcanism. Future work should therefore analyse the spatio-temporal  
 875 evolution of lavas at volcanic provinces adjacent to lithospheric steps to determine whether  
 876 or not this trend is observed.

### 877 6.4 Comparison to Other Mantle Melting Parameterizations

878 Our approach initialises the upper mantle as homogeneous peridotite. However,  
 879 multiple lithologies, including harzburgitic and pyroxenitic variants, exist within the man-  
 880 tle (e.g., Allègre & Turcotte, 1986; Stracke, 2012). Each lithology has a different min-  
 881 eralogy, solidus, and composition and, accordingly, the distribution and composition of  
 882 melts will depend strongly on the upper mantle’s lithological make-up. Most modern geo-  
 883 chemical parameterizations, like BDD21, make use of melting experiments conducted on  
 884 MM3 and KR4003 lherzolites to constrain melt stoichiometry (Baker & Stolper, 1994;  
 885 Walter, 1998; Falloon et al., 1999). However, some parameterizations incorporate addi-  
 886 tional constraints to broaden their applicability to different compositions. Two such pa-  
 887 rameterizations, REEBOX PRO and Petrogen, have been used to study melt chemistry  
 888 at mid-oceanic ridges (Brown & Leshner, 2016; Krein et al., 2020).

889 REEBOX PRO uses the thermodynamic solver *Perple X* to parameterize melt sto-  
 890 ichiometry as a function of  $P$  and  $X$  for depleted peridotite, harzburgitic and pyroxenitic  
 891 lithologies (Connolly, 2005; Brown & Leshner, 2016). Using a purely peridotitic param-  
 892 eterization such as BDD21 to predict melt compositions in a region that contains pyrox-  
 893 enite or harzburgite could under- or over-predict melt productivity (Katz et al., 2003;  
 894 Lambart et al., 2016). Pyroxenites can vary greatly in composition and melts will react  
 895 with each wall-rock lithology differently on their way to the surface (e.g., Yaxley & Green,  
 896 1998; Lambart et al., 2012; Katz & Weatherley, 2012). Furthermore, the distribution of  
 897 lithologies with different densities within the upper-mantle will affect how mantle flow  
 898 and, therefore, melt geometry, develops through time (e.g., Ito & Mahoney, 2005). Al-  
 899 though Bayesian inversion schemes can be used to constrain the lithological make-up of  
 900 a heterogeneous mantle beneath a mid-oceanic ridge using a simple melt geometry and  
 901 a perfectly mixed mantle (Brown et al., 2020), it is currently computationally intractable  
 902 to implement such a scheme in combination with a geodynamical simulation.

903 Rather than relying on a thermodynamic solver, *Petrogen* includes a wider variety  
 904 of peridotite melting experiments to constrain modal mineralogy (Krein et al., 2020). Since  
 905 the peridotites used in these experiments have different compositions, the calculated solidus  
 906 and melt stoichiometry are also functions of major element chemistry (Till et al., 2012).  
 907 This framework allows both the major and trace element compositions of melts to be  
 908 calculated, and for melts to re-equilibrate with their surroundings during ascent (Krein  
 909 et al., 2020). *Petrogen* reproduces the mineral modes from the majority of peridotite melt-  
 910 ing experiments, including those used to constrain BDD21 (e.g., Walter, 1998). However,  
 911 the agreement between observed and calculated clinopyroxene proportions for some start-  
 912 ing compositions is poor (correlation coefficient = 0.4–0.45; Krein et al., 2020). Since  
 913 many incompatible elements, including Na, La, Sm and Gd, are most compatible in clinopy-  
 914 roxene, the concentrations of these elements are particularly sensitive to the proportion  
 915 of clinopyroxene in the mantle source. Therefore, trace element concentrations calculated  
 916 using some peridotite compositions should be treated with caution. Nevertheless, the in-  
 917 corporation of major elements means that *Petrogen* offers functionality that cannot cur-  
 918 rently be replicated in BDD21 or REEBOX PRO. Both REEBOX PRO and *Petrogen* pre-  
 919 dominantly rely on fixed partition coefficients (Brown & Leshner, 2016; Krein et al., 2020).  
 920 Using the parameterization of Blundy et al. (1995), the partition coefficient for Na in  
 921 clinopyroxene varies from 7.45 at the solidus to 0.15 beneath the ridge axis in our mid-  
 922 oceanic ridge simulation (where  $R_s = 2.1 \text{ cm yr}^{-1}$  and  $T_p = 1325 \text{ }^\circ\text{C}$ ). We believe that  
 923 including pressure, temperature and composition dependent partition coefficients can sig-  
 924 nificantly affect predicted element concentrations, and so their inclusion in BDD21 can  
 925 be considered an improvement on existing parameterizations.

926 The applicability of each melting parameterization to other geologic settings and  
 927 to geodynamical simulations is defined by its geochemical framework. REEBOX PRO cal-  
 928 culates melt composition at mid-oceanic ridges using the “residual mantle column” method  
 929 pioneered by Langmuir et al. (1992) and Plank and Langmuir (1992). This method as-  
 930 sumes that when a mid-oceanic ridge is in steady state, flow is horizontal away from spread-  
 931 ing centres so that the distribution of melt depletion within a column of mantle outside  
 932 of the melt region can be used to calculate the composition of the crust above it (Langmuir  
 933 et al., 1992). This elegant solution removes the necessity to accurately constrain the ge-  
 934 ometry of the melting region and can be augmented with additional complexity such as  
 935 mantle temperature variations and buoyancy-driven flow (e.g., Plank & Langmuir, 1992;  
 936 Brown & Leshner, 2014). However, using this framework limits REEBOX PRO to calcu-  
 937 lating melting at spreading centers and it cannot be used to test how physical param-  
 938 eters, such as including a temperature and strain-rate dependent viscosity, impact melt  
 939 composition. In contrast, *Petrogen* can be used to calculate melt compositions based on  
 940 outputs from geodynamical simulations (Krein et al., 2020). One key advantage of the  
 941 *Petrogen* approach is that, unlike the other schemes that rely on the melting parameter-  
 942 ization of Katz et al. (2003), the solidus in *Petrogen* is compositionally dependent. How-

943 ever, since the solidus is calculated in the posterior within the geochemical calculation,  
 944 temperature and viscosity variations associated with melting are not fed back into the  
 945 geodynamical simulation (Behn & Grove, 2015). Moreover, the polybaric melting func-  
 946 tions used in *Petrogen* are simplistic compared to those used by *REEBOX PRO* and *BDD21*  
 947 ( $\frac{dF}{dP}$  is currently either set as a constant, e.g., 10% GPa<sup>-1</sup>, or set to linearly increase be-  
 948 tween the solidus and the surface; Krein et al., 2020). At present, *BDD21* can only es-  
 949 timate trace element compositions of melts generated within a homogeneous peridotite  
 950 mantle, and so it does not currently offer the geochemical flexibility of parameterizations  
 951 like *REEBOX PRO* and *Petrogen*. However, unlike these parameterizations, *BDD21* links  
 952 seamlessly to a geodynamical modelling framework on the fly, to calculate melting through  
 953 space and time. As such, it provides an opportunity to understand the how melting and  
 954 melt composition evolve in different geologic settings, and in response to changing man-  
 955 tle conditions.

## 956 7 Conclusions and Future Directions

957 We have presented *BDD21*, a new geochemical melting parameterization for cal-  
 958 culating melt incompatible element concentrations as a function of temperature, pres-  
 959 sure, melt fraction and mantle composition. This parameterization is designed in such  
 960 a way that it can be seamlessly coupled with geodynamical simulations, as demonstrated  
 961 herein through the *Fluidity* computational modelling framework. The versatility of our  
 962 combined geochemical-geodynamical approach is demonstrated in the context of melt-  
 963 ing at two geologic settings: mid-oceanic ridges and lithospheric steps. Our simulations  
 964 reproduce first-order differences in melt volumes and compositions between these set-  
 965 tings. Furthermore, we replicate geological and geochemical observations throughout the  
 966 present-day mid-oceanic ridge system. Taken together, our results imply that coupled  
 967 geochemical-geodynamical approaches, such as the one advocated herein, will reveal new  
 968 insights into the mechanisms and processes controlling magmatism, both at the present  
 969 day and into the geological past.

970 For simplicity, we have only analysed results for a handful of incompatible elements  
 971 (e.g., Na, Ti, La, Sm, Gd and Yb). However, *BDD21* can currently calculate melt con-  
 972 centrations for 34 elements, and calculating additional element concentrations incurs lim-  
 973 ited computational cost. Furthermore, the modular nature of *BDD21* allows for seam-  
 974 less refinement of the current framework as more experimental data become available,  
 975 as well as future incorporation of heterogeneous mantle lithologies and major-element  
 976 systematics. These potential improvements would increase the versatility of *BDD21* but  
 977 also highlight the number of unknowns present when conducting studies of this nature.  
 978 We believe that analysing how melt chemistry evolves through space and time in response  
 979 to varying individual parameters in isolation, such as mantle  $T_p$ , lithospheric geometry  
 980 and thickness, source lithology, mantle rheology, plate velocity, or the presence/absence  
 981 of mantle plumes, will be a fruitful avenue for improving our understanding of past and  
 982 present magmatic phenomena.

## 983 8 Data and Software Availability Statement

984 The *BDD21* melting parameterization can be accessed via [https://github.com/  
 985 Pato175/BDD21](https://github.com/Pato175/BDD21). The *Fluidity* computational modeling framework, including source code,  
 986 demonstrable examples and documentation, are available from [https://fluidityproject  
 987 .github.io/](https://fluidityproject.github.io/); the latest release, with tag 4.1.18 and which we used for the simulations  
 988 presented herein, is archived at <https://doi.org/10.5281/zenodo.4896641>.

## 989 Appendix A Physical Parameters of Geodynamical Simulations

**Table A1.** Physical parameters used in the mid-oceanic ridge simulations examined herein. UD = user-defined variable.

Quantity	Symbol	Units	Value
Gravity	$g$	$\text{m s}^{-2}$	9.81
Coefficient of thermal expansion	$\alpha$	$\text{K}^{-1}$	$3 \times 10^{-5}$
Thermal diffusivity	$\kappa$	$\text{m}^2 \text{s}^{-1}$	$10^{-6}$
Reference density	$\rho_s$	$\text{kg m}^{-3}$	3300
Surface temperature	$T_s$	K	273
Mantle temperature	$T_m$	K	UD
Gas constant	$R$	$\text{J K}^{-1} \text{mol}^{-1}$	8.3145
Max. viscosity	$\mu_{\text{max}}$	Pa s	$10^{25}$
Min. viscosity	$\mu_{\text{min}}$	Pa s	$10^{18}$
Half spreading rate	$R_s$	$\text{cm yr}^{-1}$	UD
<b><i>Diffusion creep</i></b>			
Activation energy	$E$	$\text{kJ mol}^{-1}$	300
Activation volume	$V$	$\text{cm}^3 \text{mol}^{-1}$	4
Prefactor	$A$	$\text{Pa}^{-n} \text{s}^{-n}$	$3 \times 10^{-11}$
Exponent	$n$	-	1
<b><i>Dislocation creep</i></b>			
Activation energy	$E$	$\text{kJ mol}^{-1}$	540
Activation volume	$V$	$\text{cm}^3 \text{mol}^{-1}$	16
Prefactor	$A$	$\text{Pa}^{-n} \text{s}^{-n}$	$5 \times 10^{-16}$
Exponent	$n$	-	3.5

**Table A2.** Physical parameters used in the EDC simulations examined herein. UD = user-defined variable.

Quantity	Symbol	Units	Value
Gravity	$g$	$\text{m s}^{-2}$	9.8
Coefficient of thermal expansion	$\alpha$	$\text{K}^{-1}$	$3 \times 10^{-5}$
Thermal diffusivity	$\kappa$	$\text{m}^2 \text{s}^{-1}$	$6 \times 10^{-7}$
Mantle density		$\text{kg m}^{-3}$	3370
Continental density		$\text{kg m}^{-3}$	3300
Crust density		$\text{kg m}^{-3}$	2900
Surface temperature	$T_s$	K	290
Mantle temperature	$T_m$	K	1598
Adiabatic gradient		$\text{K.km}^{-1}$	0.4
Oceanic lithosphere age		Myr	UD
Continent depth		km	200
Step width		km	200
Gas constant	$R$	$\text{J K}^{-1} \text{mol}^{-1}$	8.3145
Activation energy	$E$	$\text{kJ mol}^{-1}$	350
Activation volume	$V$	$\text{cm}^3 \text{mol}^{-1}$	6.8
Prefactor	$A$	$\text{Pa}^{-1}$	$8.6 \times 10^{-8}$
Max. viscosity	$\mu_{\text{max}}$	Pa s	$10^{24}$
Min. viscosity	$\mu_{\text{min}}$	Pa s	$10^{18}$
Water Content (Melting)		ppm	200

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