

1 Spatio-temporal variability of the carbonate system and air-sea CO₂ fluxes in the
2 South Yellow Sea and East China Sea during the warm seasons

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17 **Key Points:**

- 18 • Strong CO₂ sinks turned into weak sources in the entire study area from spring
19 to summer

- 20 • The study area was divided into three subregions to separately examine the
21 driver mechanisms of the $p\text{CO}_2$ variations
- 22 • The $p\text{CO}_2$ variability was controlled by the combined influences of
23 temperature, biological activity and physical mixing

24

25 **Abstract**

26 Due to the complex physical and biogeochemical conditions, the adjacent South
27 Yellow Sea (SYS) and East China Sea (ECS) are ideal sites for studying different
28 carbonate characteristics in marginal seas. The distributions of carbonate system
29 parameters were investigated in this region in early spring and summer. Overall,
30 dissolved inorganic carbon (DIC) and alkalinity concentrations in the SYS were
31 higher than those in the ECS due to the Yellow River runoff which was featured with
32 intensive carbonate weathering and erosion. Low DIC, alkalinity and high pH values
33 were observed in the Zhe-Min Coastal Current with intensive primary production in
34 spring caused by the Changjiang River and Taiwan Warm Current. Temperature and
35 biological activities were the primary drivers in controlling the partial pressure of CO_2
36 ($p\text{CO}_2$) variability in the SYS, whereas temperature was the only dominant factor in
37 the outer shelf of the ECS, which was heavily impacted by the Kuroshio Current. The
38 $p\text{CO}_2$ dynamics was controlled by primary production and physical mixing in the
39 Changjiang River plume and the inner and middle shelves of the ECS, due to the
40 influence of the Changjiang River with high nutrient supply. Overall, strong CO_2
41 sinks ($-4.11 \pm 5.28 \text{ mmol m}^{-2}\text{d}^{-1}$) turned into weak sources ($0.88 \pm 5.09 \text{ mmol m}^{-2}\text{d}^{-1}$)

42 in the entire study area from spring to summer. Specifically, the SYS and ECS
43 offshore waters changed from CO₂ sinks in spring to sources in summer, while the
44 Changjiang River plume always served as a CO₂ sink.

45

46 **Keywords:** Carbon cycle, air-sea CO₂ flux, the South Yellow Sea and East China Sea,
47 seasonal variations

48

49 **Plain Language Summary**

50 Although the South Yellow Sea (SYS) and East China Sea (ECS) are adjacent to
51 each other, the biogeochemical characteristics of them are very different. Therefore, a
52 study of the SYS and ECS together will allow us a better understanding of the
53 processes determining the spatial and temporal variations of the carbonate system in
54 continental seas, which is an important step toward estimating global air-sea CO₂
55 fluxes. In this study, two cruises were conducted in early spring and summer to
56 investigate the distributions of the carbonate system in the SYS and ECS. High
57 dissolved inorganic carbon (DIC) and alkalinity were observed in the SYS under the
58 influence of Yellow River with high DIC and alkalinity discharges. However, low
59 DIC, alkalinity and high pH values occurred in the Zhe-Min Coastal Current. The
60 distribution and variability of the partial pressure of CO₂ were associated with
61 temperature, biological activity and physical mixing. Under the combined impacts of
62 the above factors, the SYS and ECS offshore waters changed from CO₂ sinks in
63 spring to sources in summer, while the Changjiang River plume always served as a

64 CO₂ sink. Overall, Strong CO₂ sinks turned into weak sources in this region from
65 spring to summer.

66

67 **1 Introduction**

68 In the last two decades, continental shelves have drawn increasing attention in
69 the global ocean carbon cycle research due to the fact that they provide 15–30% of the
70 oceanic primary production and represent 50% of ocean organic carbon burial in
71 sediments, although shelves just comprising approximately 7% of the world's ocean
72 surface area (Bauer et al., 2013; Borges et al., 2005; Cai, 2011; Chen, 2003). The
73 continental shelves are distinguished by the intense physical, chemical and biological
74 processes, and considered to be significant contributors to the global carbon cycle
75 (Mackenzie et al., 1991). It is therefore necessary to understand and accurately
76 account for the variability of air-sea CO₂ flux and carbonate chemistry of continental
77 shelves although it remains a tremendous challenge to make satisfactory progresses
78 (Bauer et al., 2013; Zhai et al., 2014).

79 There are large spatial and temporal variations in surface seawater
80 biogeochemistry and air-sea CO₂ exchanges across heterogeneous continental shelves
81 (Takahashi et al., 2009). An ocean margin province-based synthesis by Cai et al.
82 (2006) suggested that continental marginal seas at low latitudes were the major CO₂
83 sources (Cai et al., 2003, 2004; Goyet et al., 1998), but those in mid-high latitudes
84 were the sinks of atmospheric CO₂ (DeGrandpre et al., 2002; Thomas et al., 2005;
85 Tsunogai et al., 1999). Recently, Dai et al. (2013) suggested that margins receiving

86 river inputs (RiOMar) were largely a CO₂ sink while those receiving ocean inputs
87 (OceMar) were largely sources of CO₂ to the atmosphere. However, these broad
88 classifications may oversimplify the field observations and the mechanisms leading to
89 these synthesis statements require further supports from many field studies in
90 dissimilar margins. For example, the North Sea, a European continental shelf sea, is a
91 CO₂ sink in the northern part but a CO₂ source in the southern part under the influence
92 of temperature, terrestrial inputs and biological activity from north to south (Bozec et
93 al., 2005; Kempe & Pegler, 1991; Omar et al., 2010; Thomas et al., 2005). Similarly,
94 the partial pressure of CO₂ ($p\text{CO}_2$) distribution also shows a different spatial trend in
95 the adjacent South Yellow Sea (SYS) and East China Sea (ECS), resulting from
96 different topography, ocean circulation and biogeochemical conditions. In the north,
97 the SYS, as a semi-enclosed marginal sea, is a net annual CO₂ source because of its
98 long water residence time and limited water exchange with open ocean (Qu et al.,
99 2014; Xue et al., 2011; Zhang et al., 2010). In contrast, in the south, the ECS is an
100 eminent continental shelf pump for efficient transferring atmospheric CO₂ to the deep
101 sea and serves as an annual net sink of atmospheric CO₂ (Song et al., 2018; Tsunogai
102 et al., 1999; Wang et al., 2000; Zhai & Dai, 2009). Therefore, the carbonate
103 characteristics of these two seas are different. However, the previous studies tend to
104 study their air-sea flux separately, even if the SYS and ECS are adjacent to each other.
105 To our best knowledge, the carbonate system of the entire SYS and ECS (together
106 called the South Yellow-East China Sea region) has only been investigated in the
107 spring and summer of 2011 and the summer of 2013 (Qu et al., 2015, 2017) to date.

108 With those limited studies, the spatial resolution of $p\text{CO}_2$ distributions is still lacking
109 because the $p\text{CO}_2$ in Qu et al. (2015, 2017) was calculated by CO2SYS based on the
110 discrete pH and total alkalinity (TA) instead of continuous underway measurement,
111 and the air-sea CO_2 flux based on discrete estimations also had great uncertainties.

112 A better understanding of spatial and temporal variations of carbonate system in
113 the South Yellow-East China Sea region is important to merge its CO_2 flux into the
114 global carbon cycle. Therefore, in this study, we investigated high spatial resolutions
115 of discrete pH, dissolved inorganic carbon (DIC) and TA samples as well as
116 underway $p\text{CO}_2$ in the South Yellow-East China Sea region with intensive biological
117 activity based on two cruises from 27 March to 11 April, 2017 and 27 June to 17 July,
118 2018. We also calculated the air-sea CO_2 fluxes and examined the driver mechanisms
119 to impact the $p\text{CO}_2$ variations and air-sea CO_2 fluxes. This study fills in the
120 knowledge gap by not only providing an updated dataset on the air-sea CO_2 fluxes in
121 the South Yellow-East China Sea region, but also improving our understanding of
122 carbon cycles in continental seas.

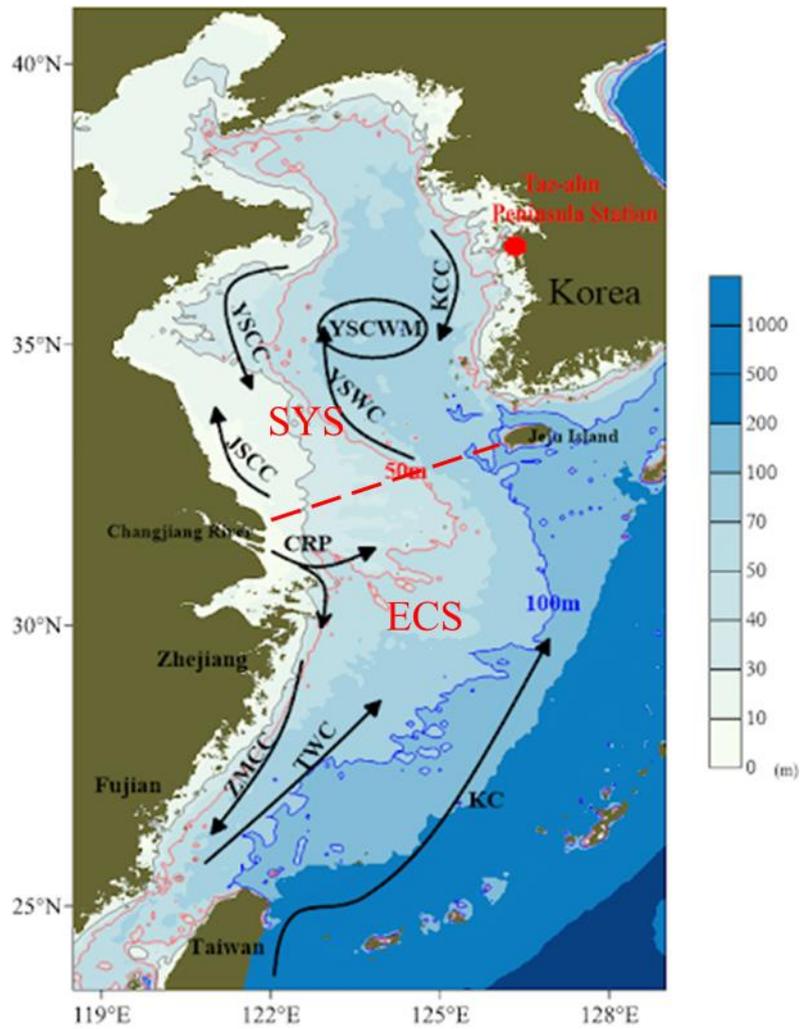
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124 **2 Materials and Methods**

125 **2.1 Study area**

126 The South Yellow and East China Sea, which is located in the northwestern
127 Pacific Ocean between $26\text{--}37^\circ\text{N}$ and $119\text{--}125^\circ\text{E}$ (Figure 1). The SYS is a
128 semi-enclosed marginal sea, which is surrounded by the Shandong and Korean
129 Peninsulas and bordered on the north by the North Yellow Sea and on the south by the
130 ECS. It is distinguished and dominated by several major water masses in different

131 seasons, including the Yellow Sea Warm Current, Yellow Sea Cold Water Mass,
132 coastal water currents along both the Chinese and Korean coasts and Changjiang
133 River (Figure 1). Moreover, the Subei Shoal waters in the southwestern SYS, being
134 one of the most turbid coasts in China, load numerous sediments into the SYS
135 annually (Wang et al., 2011a). Hence, the SYS is strongly influenced by nearshore
136 biogeochemical processes, intense anthropogenic activities to a great extent and is
137 isolated from open sea (Choi et al., 2019). However, the ECS is located to the south of
138 the SYS and the western three quarters of the ECS is occupied by the continental shelf,
139 while the eastern part is deep and opens to the Pacific Ocean (Chen, 2009). The
140 materials from the Pacific Ocean is easily exchanged with the ECS by the Kuroshio
141 Current (KC) and Taiwan Warm Current (TWC), while the Changjiang River brings
142 large amounts of terrigenous nutrients into the ECS (Qu et al., 2017). Thus, the
143 carbonate system in the ECS is strongly subjected to the continental shelf pump
144 between the nearshore and open sea (Tsunagai et al., 1999). Given these points, the
145 South Yellow-East China Sea region becomes a biogeochemical hotspot all the time
146 due to the large differences in the physical and biogeochemical conditions between
147 the SYS and ECS.



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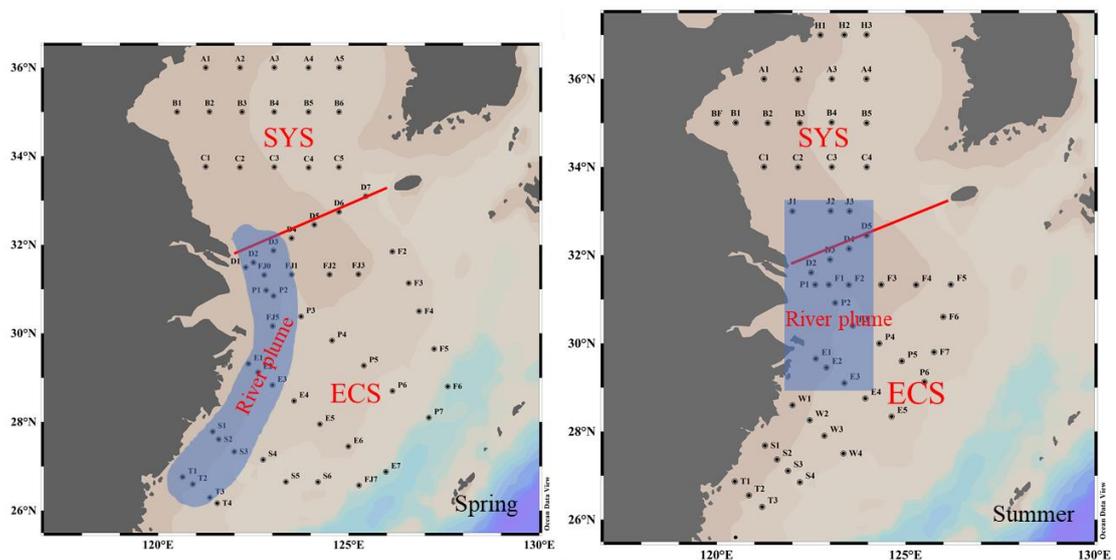
149 **Figure 1.** Topography and schematic map of the current system in the study area. The
 150 boundary of the South Yellow Sea (SYS) and East China Sea (ECS) is indicated by
 151 the red dashed-line. The currents described as arrows include the Yellow Sea Coastal
 152 Current (YSCC), Yellow Sea Warm Current (YSWC), Yellow Sea Cold Water Mass
 153 (YSCWM), Korea Coastal Current (KCC), Jiangsu Coastal Current (JSCC),
 154 Changjiang River plume (CRP), Zhe-Min Coastal Current (ZMCC), Taiwan Warm
 155 Current (TWC) and Kuroshio Current (KC) (Chen, 2009). The red and blue solid lines
 156 on the continental shelf are the depth contour of 50 and 100 m, respectively. The red
 157 dot (126.133°E, 36.738°N) in the west part of South Korea is the Tae-ahn Peninsula
 158 Station (<https://www.esrl.noaa.gov>).

159

160 2.2 Sampling

161 Two cruises were carried out aboard the R/V “Dongfanghong II” from 27 March
162 to 11 April, 2017 and 27 June to 17 July, 2018 to represent the early spring and
163 summer, respectively. The study area and sampling stations are shown in Figure 2.

164



165

166 **Figure 2.** Sampling stations of 2017 spring (left) and 2018 summer (right) cruises.

167 The blue shadows represent the Changjiang River plume; the South Yellow Sea (SYS)
168 and East China Sea (ECS) are divided by the red solid line.

169

170 Water samples were collected using 12 L Niskin bottles mounted onto a Seabird
171 911-plus Conductivity-Temperature-Depth system (CTD, SeaBird Inc. Bellevue, WA,
172 USA), which was used to measure the temperature and salinity of water column. DO
173 samples were collected, fixed and analyzed on board as described by the classic
174 Winkler procedure (Dickson, 1994). pH was collected into a 100 mL narrow-mouth

175 glass bottle and kept in a thermal bath (25 ± 0.1 °C) for 30–60 min before
176 determination. DIC and TA samples were stored in 250 mL borosilicate glass bottles
177 and overflowed at least twice their volume to minimize contact with air, then
178 poisoned with 100 μ L of saturated HgCl₂ immediately, sealed and preserved at room
179 temperature until determination (Huang et al., 2012). 300 mL water samples were
180 filtered through a 0.7 μ m-pore Whatman glass fiber filter (GF/F, pre-combusted at
181 400 °C for 4 h), and materials collected on the membrane were preserved at -20 °C
182 and used for analyzing Chlorophyll *a* (Chl *a*).

183 To monitor levels of $p\text{CO}_2$, surface water was continuously pumped from 1–2 m
184 below the sea level through an underway $p\text{CO}_2$ analyzer (AS-P2, Apollo SciTech Inc.,
185 USA) with a Picarro detector (G2301, Picarro Inc., USA) installed in the shipboard
186 laboratory. Briefly, surface water was pumped to the main shower-head equilibrator at
187 a rate of 2.5 L min⁻¹ to get rapid gas exchange. After the sample was equilibrated, the
188 well-mixed gas first passed through the water condenser and desiccant, which
189 removed most of the water vapor, and then the equilibrated gas was delivered to the
190 detector (Picarro G2301). The detector was calibrated every 9 h against three CO₂ gas
191 standards (0, 198 and 403 ppm), which were provided by Beijing Certified Reference
192 Material Center. Some values of $x\text{CO}_2$ outside of the concentration range of the
193 standard gases were also used in this study because the biases caused by the
194 out-of-range values are generally subtle (Pierrot et al., 2009). The uncertainty of $x\text{CO}_2$
195 measurements was less than 50 ppbv over 5 min internals (Li et al., 2017).

196

197 **2.3 Analytical methods**

198 pH was measured onboard at 25 ± 0.1 °C by Fisher pH meter (Star A211)
199 combining with a Ross Orion combination electrode (Ross-8102) on a National
200 Bureau of Standards (NBS) scale with a precision of ± 0.01 units. pH values were
201 corrected to the in-situ temperature using CO2SYS. 1 mL of each DIC sample was
202 acidified by 10% phosphoric acid, then the extracted CO₂ gas with carrier gas was
203 quantified via an infrared gas analyzer (AS-C3, Apollo SciTech Inc., USA) as
204 described in Cai et al. (2004). TA samples were determined by Gran titration on a 25
205 mL sample volume using the open-cell method with a semi-automatic potentiometric
206 titration system (AS-ALK2, Apollo SciTech Inc., USA) (Cai et al., 2010). Both DIC
207 and TA measurements were calibrated against certified reference materials from A. G.
208 Dickson's lab at Scripps Institute of Oceanography (Batches #162 and #171) at a
209 precision of ± 2 $\mu\text{mol kg}^{-1}$ (Huang et al., 2012).

210 Chl *a* on the membrane was extracted in 90% acetone and then analyzed by a
211 fluorescence spectrophotometer (F-4500, Hitachi, Japan) based on the procedure
212 described by Parsons et al. (1984).

213

214 **2.4 Air-sea CO₂ flux estimation**

215 In this study, the $p\text{CO}_2$ at the temperature of equilibration ($p\text{CO}_2(eq)$, unit: μatm)
216 in equation (1) is calculated as following and then the $p\text{CO}_2$ at the in-situ
217 temperature ($p\text{CO}_2(water)$, unit: μatm) in equation (2) is calculated by the expression
218 of Takahashi et al. (1993):

219 $p\text{CO}_2(\text{eq}) = x\text{CO}_2 \times [P_b(\text{eq}) - P_w(\text{eq})]$ (1)

220 $p\text{CO}_2(\text{water}) = p\text{CO}_2(\text{eq}) \times \exp [0.0423 \times (\text{SST} - T_{\text{eq}})]$ (2)

221 where $x\text{CO}_2$ is the CO_2 mole fraction concentration of the seawater CO_2 in the dried
222 sample gas flow (ppm), $P_b(\text{eq})$ is the barometric pressure of equilibration and $P_w(\text{eq})$
223 is the water vapor pressure at 100% humidity calculated by the equilibrated
224 temperature (T_{eq} , °C) and in-situ salinity (Weiss & Price, 1980). SST (°C) is the in-situ
225 temperature of surface water, the temperature difference between the in-situ water and
226 the equilibration was less than 0.5 °C.

227 The air-sea CO_2 fluxes (FCO_2 , $\text{mmol m}^{-2} \text{d}^{-1}$) were estimated as follows:

228 $FCO_2 = 0.24 \times k \times K_H \times [p\text{CO}_2(\text{water}) - p\text{CO}_2(\text{air})]$ (3)

229 where k (cm h^{-1}) is the gas transfer velocity of CO_2 , K_H ($\text{mol L}^{-1} \text{atm}^{-1}$) is the
230 solubility constant of CO_2 , calculated from in-situ temperature and salinity (Weiss,
231 1974), and $p\text{CO}_2(\text{water})$ and $p\text{CO}_2(\text{air})$ are the $p\text{CO}_2$ in the surface seawater and the
232 atmosphere, respectively. The atmospheric $p\text{CO}_2$ values (417 μatm in April 2017 and
233 398 μatm in July 2018) were estimated from the monthly atmospheric $x\text{CO}_2$ (418.49
234 ppm in April 2017 and 410.14 ppm in July 2018) at Tae-ahn Peninsula (126.133°E,
235 36.738°N, Figure1) (<https://www.esrl.noaa.gov>), after correction for water vapor
236 pressure at 100% humidity with in-situ temperature and salinity data (Weiss and
237 Price, 1980). A positive flux value represents a release of CO_2 from the water body to
238 the atmosphere, while a negative value indicates CO_2 transfer from the atmosphere to
239 the water body. The gas transfer coefficient was calculated from wind speed based on
240 the Wanninkhof (2014) empirical function:

241 $k \text{ (cm h}^{-1}\text{)} = 0.251 \times U^2 \times (Sc/660)^{-0.5}$ (4)

242 where U (m s^{-1}) is the wind speed at 10 m above the water surface, the reanalyzed
243 monthly averaged wind speed (5.95 and 6.00 m s^{-1} in spring and summer, respectively)
244 provided by the European Center for Medium-Range Weather Forecasts (ECMWF)
245 was employed to calculate the air-sea CO_2 fluxes; Sc is the Schmidt number of CO_2 in
246 seawater; 660 is the Sc value in seawater at $20 \text{ }^\circ\text{C}$.

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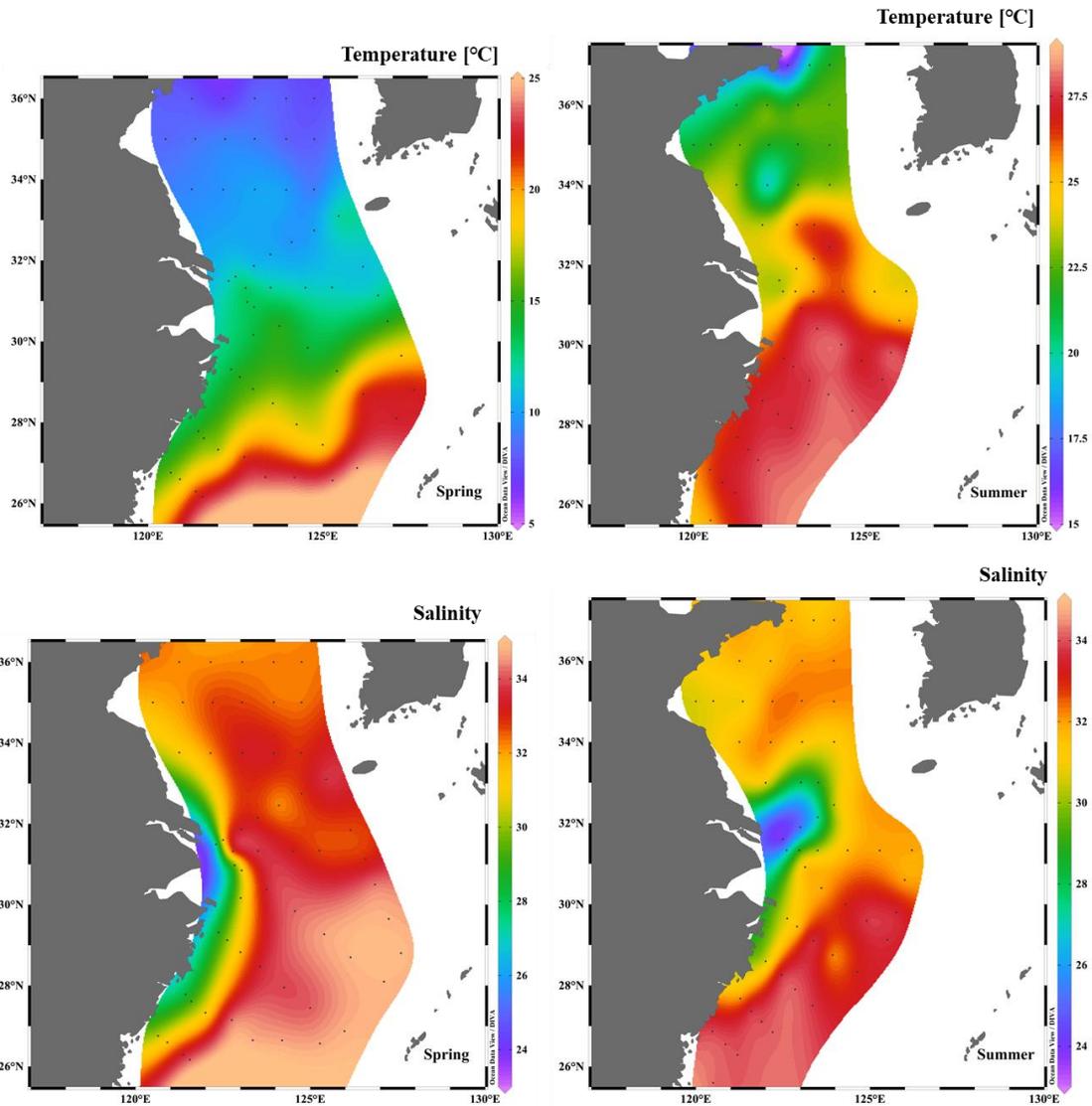
248 **3 Results**

249 **3.1 Hydrographic conditions**

250 Surface seawater temperature (SST) presented significant temporal and spatial
251 variations (Figure 3). SST was in the range of $6.91\text{--}24.47$ and $15.82\text{--}28.17 \text{ }^\circ\text{C}$, with
252 average (\pm SD) values of 14.00 ± 5.00 and $24.94 \pm 2.80 \text{ }^\circ\text{C}$, respectively, in the spring
253 and summer throughout the South Yellow-East China Sea region. clearly SST
254 increased homogeneously with the decreasing latitudes in spring and summer. On
255 average, SST in the SYS was 7.31 and $4.94 \text{ }^\circ\text{C}$ colder than that in the ECS in early
256 April and July, respectively. The temperature difference between the SYS and ECS in
257 April was due to the increase of atmospheric temperature from north to south in the
258 northern hemisphere, while in July, it could be mainly attributed to the intrusion of the
259 TWC or KC with the influence of the summer monsoon (Chen, 2009). The ranges of
260 salinity in spring (range: $25.81\text{--}34.94$, average: 32.59 ± 1.89) and summer
261 ($23.63\text{--}34.14$, average: 31.58 ± 2.26) were not greatly different. The main feature was
262 that a water tongue with relatively low salinity rushed out of Changjiang estuary and

263 formed a plume, indicating the influence of the Changjiang River on the junction of
264 the SYS and ECS. This plume feature was stronger in summer than spring.

265



266

267 **Figure 3.** Horizontal distributions of temperature (°C) and salinity in the surface
268 water of spring and summer.

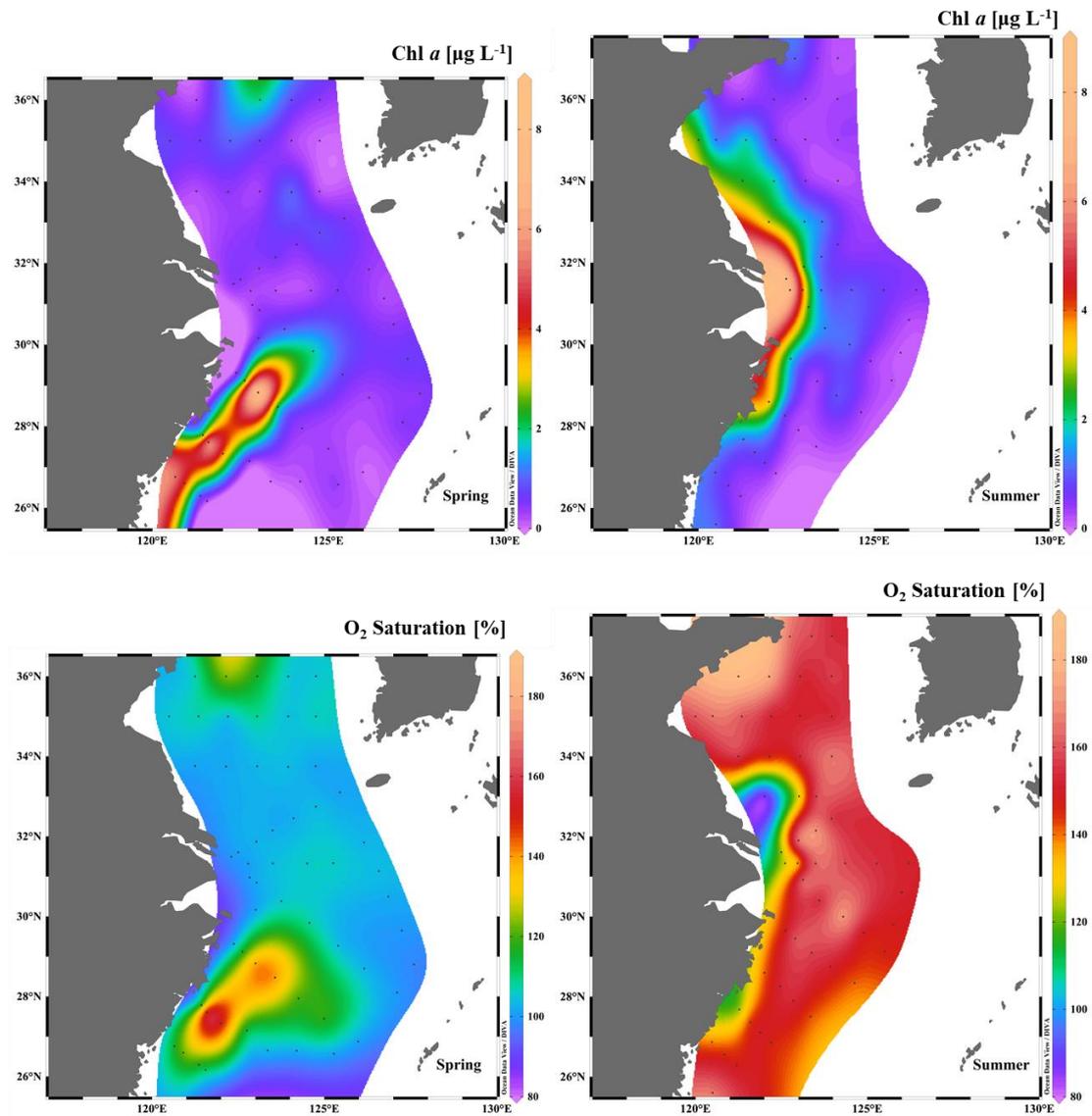
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270 From early spring to summer, the surface Chl *a* exhibited considerable variation
271 (Figure 4). Overall, Chl *a* concentration of the SYS in early spring ($0.59 \pm 0.51 \mu\text{g L}^{-1}$)

272 was lower than that ($1.00 \pm 0.98 \mu\text{g L}^{-1}$) in summer because of the weak biological
273 activity at lower temperature. In summer, high Chl *a* value appeared in the
274 southwestern part of the SYS, which was enhanced by the abundant nutrient inputs
275 from the Changjiang River. In the ECS, high concentrations of surface Chl *a*
276 ($3.46\text{--}8.80 \mu\text{g L}^{-1}$) in spring were found in the southwest part of the shelf near the
277 coast. Also noted was high Chl *a* concentration in the northwest part of the ECS in
278 summer, which was also due to the large amount of nutrients loading from the
279 Changjiang River. In contrast, low surface Chl *a* ($< 0.50 \mu\text{g L}^{-1}$) was observed in the
280 middle and outer shelves of the ECS in both spring and summer.

281 O_2 saturation in surface water were oversaturated conditions albeit with a few
282 exceptions (Figure 4). The average O_2 saturations were 109 ± 13 and $152 \pm 17\%$ in
283 the surface layer of the South Yellow-East China Sea region during spring and
284 summer cruises, respectively. Thus, O_2 saturation in the surface layer of summer was
285 higher than that in spring, due to the growth of phytoplankton and water stratification,
286 which limited the exchanges between high DO in the surface layer and low DO in the
287 bottom layer.

288



289

290 **Figure 4.** Horizontal distributions of Chl *a* ($\mu\text{g L}^{-1}$) and O₂ saturation (%) in the
 291 surface water of spring and summer.

292

293 **3.2 Distributions of carbonate system parameters**

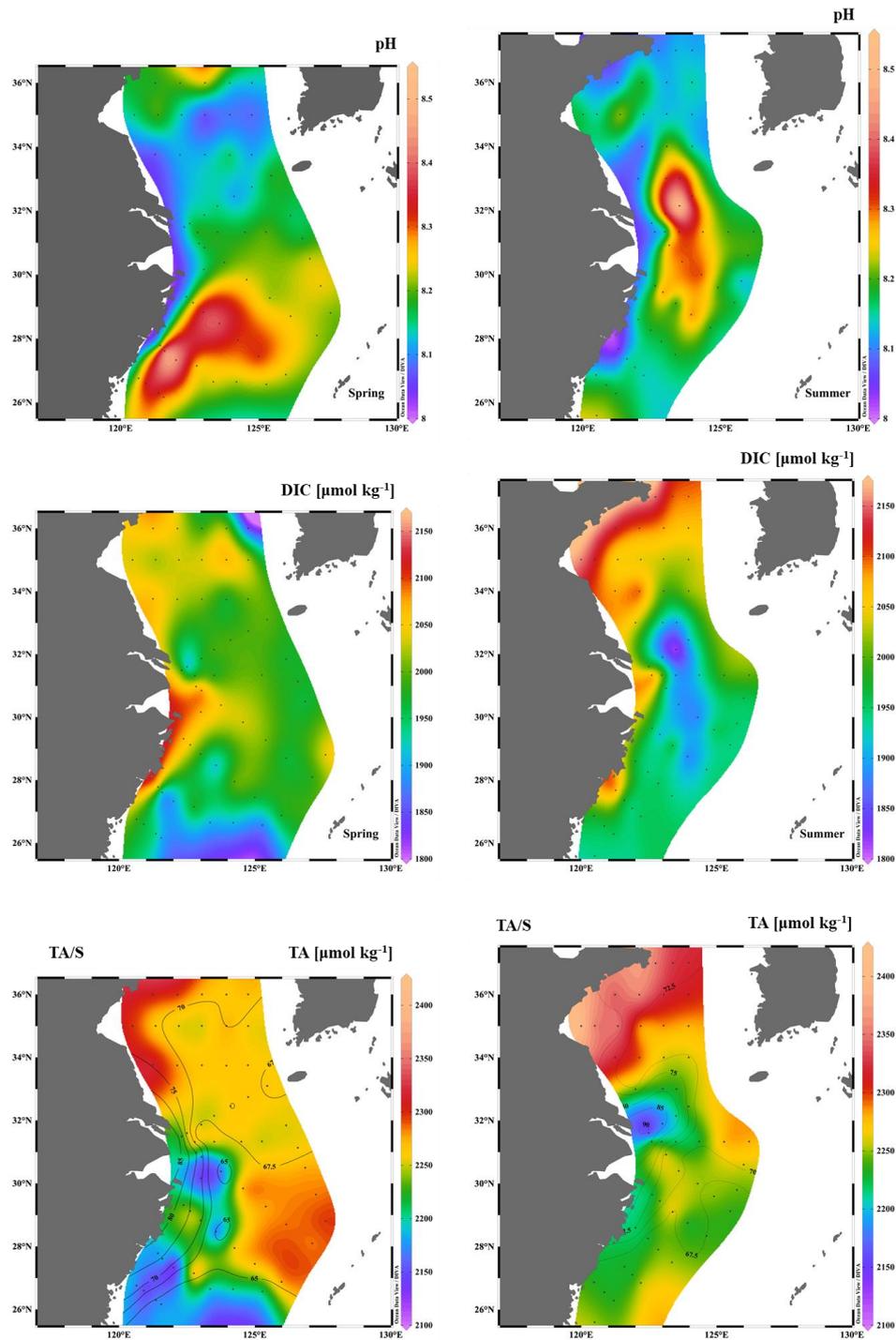
294 Surface pH of the South Yellow-East China Sea region from 8.06 to 8.50, and
 295 8.02 to 8.54 in spring and summer, with the averages of 8.20 ± 0.09 and 8.19 ± 0.09 ,
 296 respectively. Overall, pH increased from north to south, and the highest pH values
 297 (8.31–8.50) were mainly located in the southwestern part of the ECS. In summer, the

298 isoline of pH was basically perpendicular to the shoreline and the low pH appeared in
299 the coastal waters, whereas the high values of pH (8.27–8.54) were mainly observed
300 in the river plume region (Figure 5).

301 Surface DIC values were in the range of 1868 to 2089 and 1826 to 2159 $\mu\text{mol kg}^{-1}$
302 kg^{-1} , averaging 1997 ± 54 and $1982 \pm 82 \mu\text{mol kg}^{-1}$ in spring and summer, respectively.

303 The surface DIC concentration exhibited both spatial variations and clear seasonal
304 patterns in different subregions: a slight decreasing in the whole ECS region and an
305 increasing trend in the SYS region from spring to summer. surface TA values were
306 observed in the range of 2113–2318 and 2146–2410 $\mu\text{mol kg}^{-1}$ in spring and summer,
307 respectively, averaging 2244 ± 47 and $2264 \pm 50 \mu\text{mol kg}^{-1}$. In early spring, TA was
308 homogenous in the central SYS, the middle and outer shelves of the ECS. A few
309 higher values ($2306 \pm 9 \mu\text{mol kg}^{-1}$) existed in the coastal waters of the SYS while
310 lower values ($2154 \pm 29 \mu\text{mol kg}^{-1}$) appeared in the southwestern ECS. On the other
311 hand, TA in summer decreased from north to south in the SYS and fluctuated with a
312 narrow range in the ECS ($2247 \pm 17 \mu\text{mol kg}^{-1}$). However, the distribution of
313 normalized TA (TA/S) decreased with the increasing salinity, which was greatly
314 different from TA distribution (Figure 5).

315



316

317 **Figure 5.** Horizontal distributions of pH, DIC ($\mu\text{mol kg}^{-1}$) and TA ($\mu\text{mol kg}^{-1}$) in the
 318 surface water during spring and summer. The contour in the lower panel represent
 319 the distribution of TA/S.

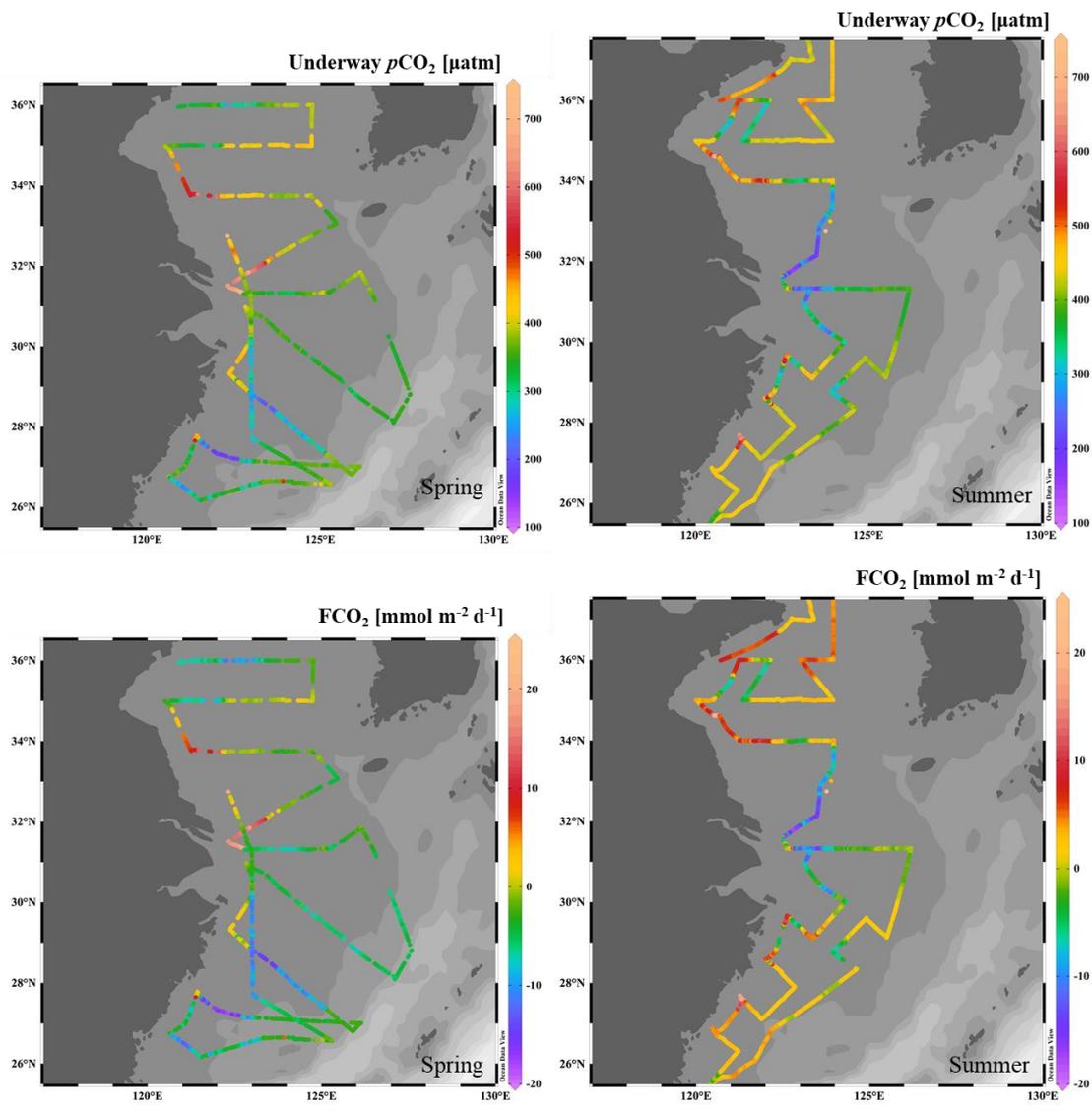
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321 The underway $p\text{CO}_2$ data during the two cruises allow us to examine the

322 high-resolution distribution of $p\text{CO}_2$ in the South Yellow-East China Sea region
323 (Figure 6). During our sampling period, even though $p\text{CO}_2$ values showed similar
324 ranges between early spring (151–689 μatm) and summer (149–709 μatm), the
325 averaged $p\text{CO}_2$ values were significantly higher in summer ($409 \pm 71 \mu\text{atm}$) than
326 those in spring ($360 \pm 75 \mu\text{atm}$). In spring, supersaturated $p\text{CO}_2$ values were observed
327 near the Subei shoal waters (447–611 μatm), whereas $p\text{CO}_2$ values in the rest of SYS
328 were generally low (258–407 μatm), except that $p\text{CO}_2$ in the central SYS (420–444
329 μatm), which was a little higher than the atmospheric $p\text{CO}_2$. However, in summer,
330 $p\text{CO}_2$ in the SYS was generally high ($>420 \mu\text{atm}$) except that in the southern SYS,
331 resulting from Changjiang diluted water input. In addition, few sporadic low $p\text{CO}_2$
332 values were found near the south of the Shandong Peninsula. In the river plume, $p\text{CO}_2$
333 in spring kept in a low range ($<355 \mu\text{atm}$) except in the area near the Changjiang
334 mouth and the northwestern corner (513–689 μatm), as well as in Hangzhou Bay
335 (421–455 μatm). To the contrary, Changjiang estuary mouth and the northwestern
336 corner showed relatively low $p\text{CO}_2$ (149–386 μatm) in summer, however, high $p\text{CO}_2$
337 in the range of 420–470 μatm occurred in the coastal water near the Hangzhou Bay.
338 The seasonal patterns in the northern and southern of ECS offshore water were
339 different: $p\text{CO}_2$ in the northern ECS always showed relatively low $p\text{CO}_2$ in spring and
340 summer; while southern ECS had low $p\text{CO}_2$ ($<380 \mu\text{atm}$) in spring and high $p\text{CO}_2$ in
341 summer ($>430 \mu\text{atm}$).

342 Similar to the seasonality of $p\text{CO}_2$ in the South Yellow-East China Sea region,
343 the air-sea CO_2 fluxes also had strong seasonal variations in the range of -18.63–19.46

344 and -18.03 – 22.61 $\text{mmol m}^{-2} \text{d}^{-1}$, with the average values of -4.11 ± 5.28 and $0.88 \pm$
 345 5.09 $\text{mmol m}^{-2} \text{d}^{-1}$ in spring and summer, respectively (Figure 6). The SYS and ECS
 346 were significant atmospheric CO_2 sinks in spring, with average values of -2.11 ± 4.57
 347 and -5.56 ± 3.12 $\text{mmol m}^{-2} \text{d}^{-1}$, but shifted into CO_2 sources (2.35 ± 4.30 and $1.73 \pm$
 348 3.05 $\text{mmol m}^{-2} \text{d}^{-1}$) in summer. However, the river plume always acted as a strong
 349 CO_2 sink in both spring and summer (-3.78 ± 7.44 and -5.17 ± 6.65 $\text{mmol m}^{-2} \text{d}^{-1}$). In
 350 addition, the Subei shoal waters was always a CO_2 source during these two surveys.
 351



352

353 **Figure 6.** The trajectory of underway $p\text{CO}_2$ (μatm) and air-sea CO_2 flux (mmol m^{-2}
354 d^{-1}) in the surface water during spring and summer.

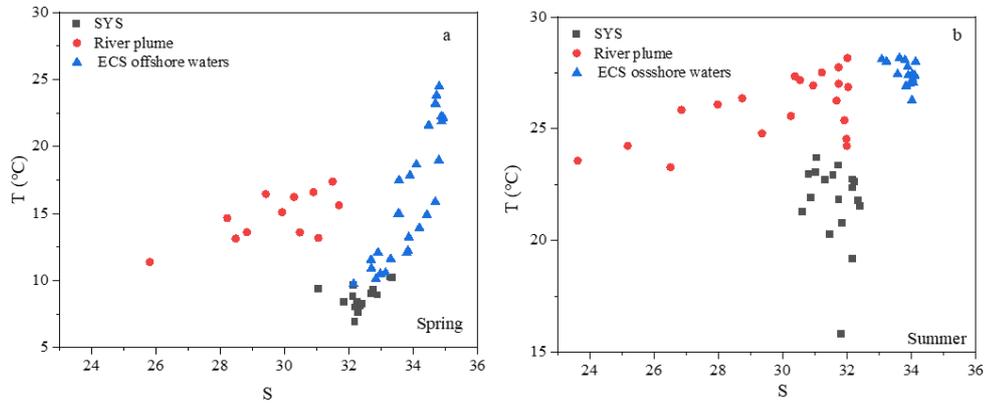
355

356 **4 Discussion**

357 **4.1 Classification of the water mass types**

358 In order to understand the biogeochemistry of the South Yellow-East China Sea
359 region and explore the influencing mechanism of carbonate system, it is necessary to
360 describe typical water mass distributions in this study. On the basis of the
361 temperature-salinity characteristics (Figure 7), surface waters were divided into three
362 subregions: (1) SYS, (2) River plume and (3) ECS offshore waters. The ranges of the
363 hydrological and carbonate system parameters in these three subregions are
364 summarized in Table 1, and the areal distributions of these water masses are presented
365 in Figure 2. The SYS has low temperature and moderate salinity; the ECS offshore
366 waters are mainly impacted by the KC and TWC (Chen, 2009). The KC flows
367 northeast along the shelf break and TWC enters the ECS through Taiwan Strait and
368 traverses in the middle shelf (Lee & Chao, 2003). Both KC and TWC are
369 characterized by high salinity and temperature (Chou et al., 2009). It is worth noting
370 that the location of river plume varied from April to July under the influence of
371 seasonal monsoons (Qi et al., 2014), which was further illustrated by the distribution
372 of salinity (Figure 3). For example, the low salinity in the river plume was roughly
373 confined to the nearshore from 26 to 32 °N in spring because the influence of the
374 prevailing northeast wind (Wu et al., 2014). When it comes to summer, the

375 Changjiang River, carrying the nutrients, rushed out towards its northeast and formed
 376 river plume, triggering the high primary productivity there (Isobe & Matsuno, 2008).
 377



378
 379 **Figure 7.** Temperature (T) vs. salinity (S) in surface water of the South Yellow-East
 380 China Sea region during spring (a) and summer (b) for three water masses: SYS, river
 381 plume and ECS offshore waters.

382
 383 **Table 1.** Hydrological and carbonate parameters of the different water masses in the
 384 surface layer defined in this study.

385

Subregion	Longitude (°E)	Latitude (°N)	T (°C)	S	Chl <i>a</i> (µg L ⁻¹)	DO Saturation (%)
<i>Spring</i>						
SYS	120.0–125.5	33.0–36.0	6.91–10.25 (8.71 ± 0.91)	31.05–33.36 (32.39 ± 0.55)	0.07–1.93 (0.59 ± 0.51)	101–125 (106 ± 6)
River plume	122.0–123.5	28.0–32.0	11.37–17.38	25.81–32.26	0.23–8.80	100–167
	120.0–122.0	26.0–28.0	(14.55 ± 1.85)	(29.92 ± 1.76)	(2.66 ± 2.97)	(117 ± 22)

ECS offshore	123.5–127.6	28.0–33.0	9.75–24.47	32.15–34.94	0.03–1.88	98–141
	122.0–126.2	25.5–28.8	(16.47 ± 5.01)	(33.90 ± 0.82)	(0.49 ± 0.41)	(106 ± 9)
<i>Summer</i> SYS	120.0–125.5	33.0–37.5	15.82–23.70	30.61–32.40	0.16–3.03	82–188
			(21.71 ± 1.87)	(31.64 ± 0.57)	(1.00 ± 0.98)	(156 ± 21)
River plume	122.0–124.3	29.0–33.0	23.28–27.76	23.63–31.74	0.57–8.22	122–173
			(25.98 ± 1.43)	(29.12 ± 2.56)	(2.74 ± 2.41)	(150 ± 18)
ECS offshore	124.3–126.2	29.0–33.0	24.23–28.17	31.93–34.14	0.13–1.21	120–174
	120.0–125.5	25.5–29.0	(27.08 ± 1.15)	(33.37 ± 0.86)	(0.52 ± 0.36)	(149 ± 10)

386

387

(Table 1 continued)

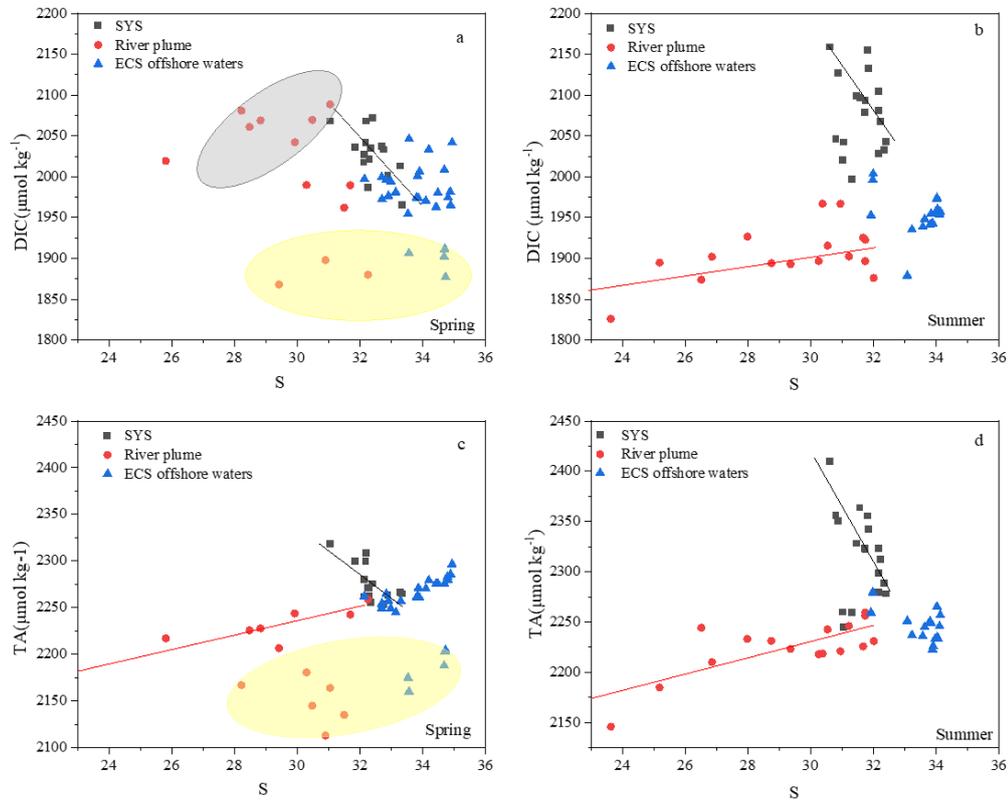
Subregion	pH	DIC ($\mu\text{mol kg}^{-1}$)	TA ($\mu\text{mol kg}^{-1}$)	Underway $p\text{CO}_2$ (μatm)	$F\text{CO}_2$ ($\text{mmol m}^{-2} \text{d}^{-1}$)
<i>Spring</i> SYS	8.06–8.24	1965–2071	2251–2318	258–611	-12.44–14.08
	(8.14 ± 0.05)	(2028 ± 30)	(2275 ± 20)	(393 ± 61)	(-2.11 ± 4.57)
River plume	8.11–8.50	1868–2089	2113–2259	170–689	-17.48–19.46
	(8.24 ± 0.13)	(2001 ± 78)	(2195 ± 47)	(365 ± 104)	(-3.78 ± 7.44)
ECS offshore	8.11–8.39	1877–2046	2160–2297	151–411	-18.62–-0.68
	(8.22 ± 0.06)	(1976 ± 41)	(2250 ± 39)	(336 ± 44)	(-5.56 ± 3.12)
<i>Summer</i> SYS	8.08–8.36	1997–2159	2245–2410	261–709	-9.79–22.61
	(8.16 ± 0.07)	(2078 ± 47)	(2316 ± 43)	(432 ± 60)	(2.35 ± 4.30)
River plume	8.10–8.54	1826–1967	2146–2259	149–564	-18.03–11.99
	(8.26 ± 0.12)	(1905 ± 34)	(2224 ± 28)	(326 ± 92)	(-5.17 ± 6.65)
ECS offshore	8.02–8.31	1879–2004	2223–2280	296–664	-7.11–19.02
	(8.18 ± 0.06)	(1954 ± 26)	(2247 ± 17)	(420 ± 44)	(1.73 ± 3.05)

388

389 **4.2 The carbonate parameters variability in subregions**

390 Two opposite trends can be obtained from the TA-S and DIC-S scatter plots
391 (Figure 8): one with a negative relationship encompassing the data from the SYS and
392 the other with a positive relationship including the river plume samples. Both
393 regressions converge towards the ECS offshore waters and represent the mixing of
394 these three water masses. The corresponding water mass with high TA and DIC
395 originated from the SYS which could be attributed to the very high TA and DIC
396 discharges of the Yellow River, which has intensive carbonate weathering and erosion
397 in the drainage basin (Liu et al., 2014; Zhang et al., 1990). The positive relationship
398 was characterized by the mixing between the Changjiang River and ECS offshore
399 waters. However, DIC near the Hangzhou Bay was little high in spring ($>2000 \mu\text{mol}$
400 kg^{-1} , shaded grey in Figure 8a), because the DIC-enrich TWC bottom water flowed
401 northward and extended to around 30°N (Li et al., 2012), then mixed well with the
402 surface water in the water column, leading to the high DIC in spring.

403 DIC and TA concentrations in spring were low near the Zhe-Min Coastal Current
404 (Yellow shaded circles in Figures 8a and 8c). It could be related to the river inputs
405 with low DIC and TA values, such as Minjiang River with low DIC concentration of
406 about $500 \mu\text{mol L}^{-1}$ in April (Qian et al., 2019). Moreover, the high TA/S values
407 (Figure 5) with the low salinity (Figure 3) in the coastal waters also indicated the
408 influences of terrestrial inputs on DIC and TA values (Jiang et al., 2014). On the other
409 hand, the nutrient enrichment phenomenon was observed in the Zhe-Min Coastal
410 Current, which was caused by river runoff and coastal upwelling, enhancing the
411 primary production and lowering DIC values in this area (Wang & Wang, 2007).



413

414 **Figure 8.** The relationships between surface DIC vs salinity (upper panel) and TA vs
 415 salinity (lower panel) in spring and summer in three water masses. The grey shaded
 416 circle represents the stations near the Hangzhou Bay and the yellow shaded circles
 417 represent the stations in the Zhe-Min Coastal Current.

418

419 4.3 Temperature and non-temperature effects on $p\text{CO}_2$ variability

420 Field investigations conducted in the global oceans indicate that temperature is
 421 one of the most important factors controlling seawater CO_2 (Millero, 1995; Takahashi
 422 et al., 2002) as a consequence of the dependence of the solubility of CO_2 with the
 423 temperature (Weiss, 1974). Meanwhile, field observations in the productive
 424 continental shelf seas have demonstrated that biological activity also plays a key role

425 in carbonate characteristics of ocean systems (Takahashi et al., 2002; Thomas et al.,
426 2005; Zhai et al., 2014). In order to quantify the relative importance of temperature
427 and biological effects on the seasonal changes of $p\text{CO}_2$ from spring to summer, we
428 calculated the temperature effect in all three subregions with the method proposed by
429 Takahashi et al. (2002). Briefly, the temperature effect can be removed by normalizing
430 $p\text{CO}_2$ data at each station to an average temperature (19.31°C) for spring and summer
431 seasons:

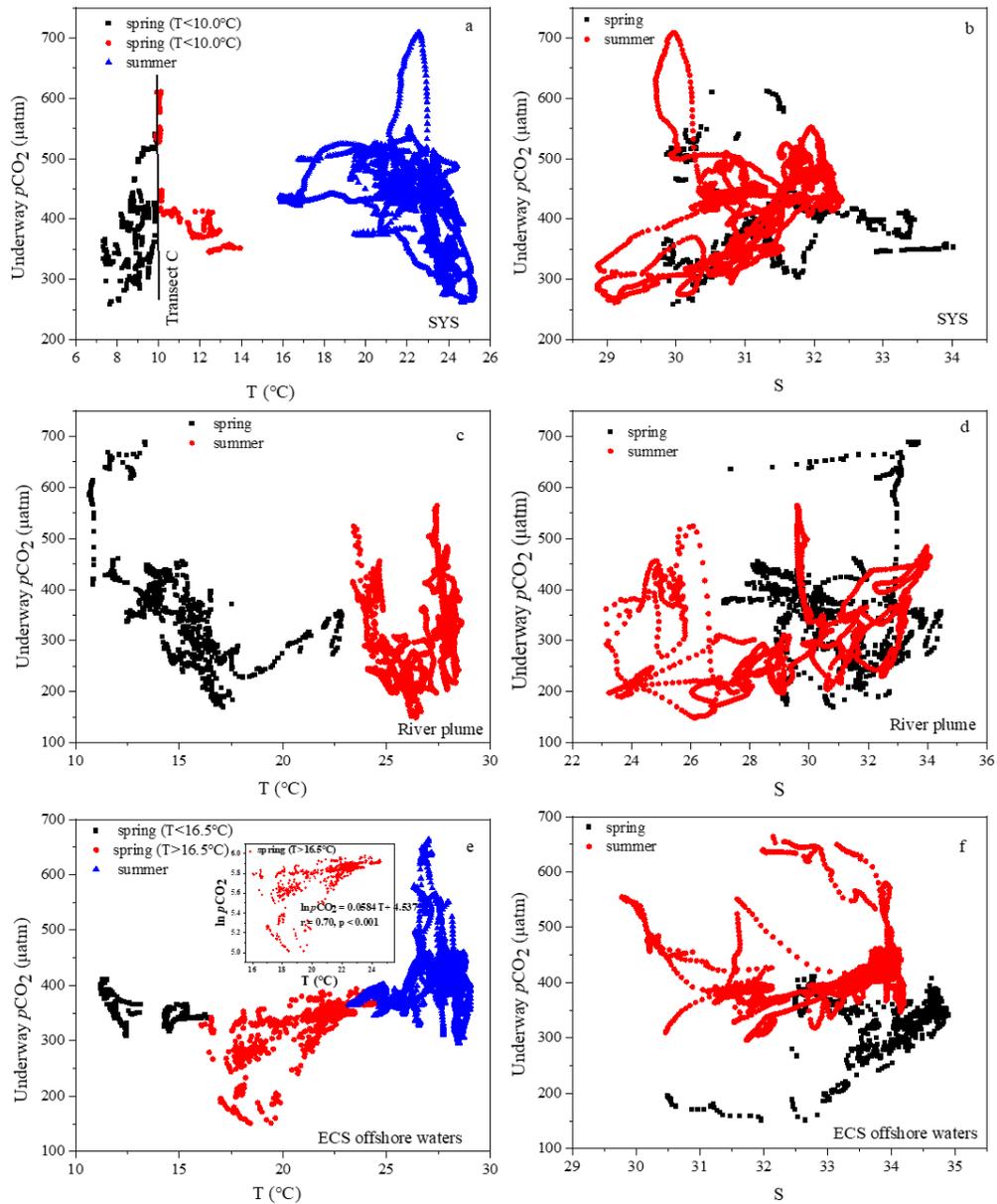
$$432 \quad p\text{CO}_2 \text{ at } T_{\text{mean}} = p\text{CO}_{2, \text{obs}} * \exp [0.0423 (T_{\text{mean}} - T_{\text{obs}})], \quad (5)$$

433 where T is the temperature in $^\circ\text{C}$, and the subscripts “mean” and “obs” indicate the
434 average and observed values, respectively. Thus, the biological activity effect on
435 $p\text{CO}_2$ in different subregions described above are discussed based on the relationship
436 between normalized $p\text{CO}_2$ ($np\text{CO}_2$, normalized to average temperature in this study
437 based on the equation (5)) and Chl a in each station.

438 In the SYS, the different trends between underway $p\text{CO}_2$ and SST in spring and
439 summer were found (Figure 9a): a positive correlation ($r = 0.32$, $p < 0.001$) in the
440 middle and north parts (north of the transect C) in April, and negative correlations in
441 the southern part (south of the transect C) in April ($r = -0.72$, $p < 0.001$) and in
442 summer ($r = -0.46$, $p < 0.001$). The positive correlation in the north and middle parts
443 in spring indicated that temperature was the leading role in the $p\text{CO}_2$ distribution,
444 which was also found by Liu et al. (2008) in late March and May. Underway $p\text{CO}_2$
445 decreased from the west to east along the transect C (Figure 6), which was consistent
446 with the results of high $p\text{CO}_2$ in the southwestern SYS (Qu et al., 2017) and low $p\text{CO}_2$

447 in the southeastern part (Choi et al., 2019). It is due to the fact that the southwest SYS
448 was occupied by the Subei shoal waters with extremely high concentration of
449 suspended sediment (Wang et al., 2011a) while high biological activity in the
450 southeast SYS (Choi et al., 2019). The negative correlation in July suggested that
451 temperature was no longer the primary factor in controlling the $p\text{CO}_2$ variability, it
452 tended to be ascribed to the biological activity in this study area (Qu et al., 2014;
453 Zhang et al., 2010). The low $p\text{CO}_2$ in the southern part was mainly affected by the
454 Changjiang diluted water, which could be seen from the positive correlation between
455 $p\text{CO}_2$ and salinity in summer (Figure 9b). Similar negative relationships of $p\text{CO}_2$ and
456 temperature were also reported in the SYS in summer by Qu et al. (2014) and Zhang
457 et al. (2010) and the inner and middle shelves of the ECS in winter (Chou et al., 2011).
458 Therefore, the controlling mechanisms on $p\text{CO}_2$ variation in the SYS varied with
459 different seasons.

460



461

462 **Figure 9.** Relationships among underway $p\text{CO}_2$, temperature and salinity in three

463 subregions in spring and summer. The insert graph in panel (e) shows the relationship

464 between $\ln p\text{CO}_2$ and temperature when temperature is higher than 16.5°C in the ECS

465 offshore waters ($\partial p\text{CO}_2 / \partial T = 0.0584 \text{ }^\circ\text{C}^{-1}$).

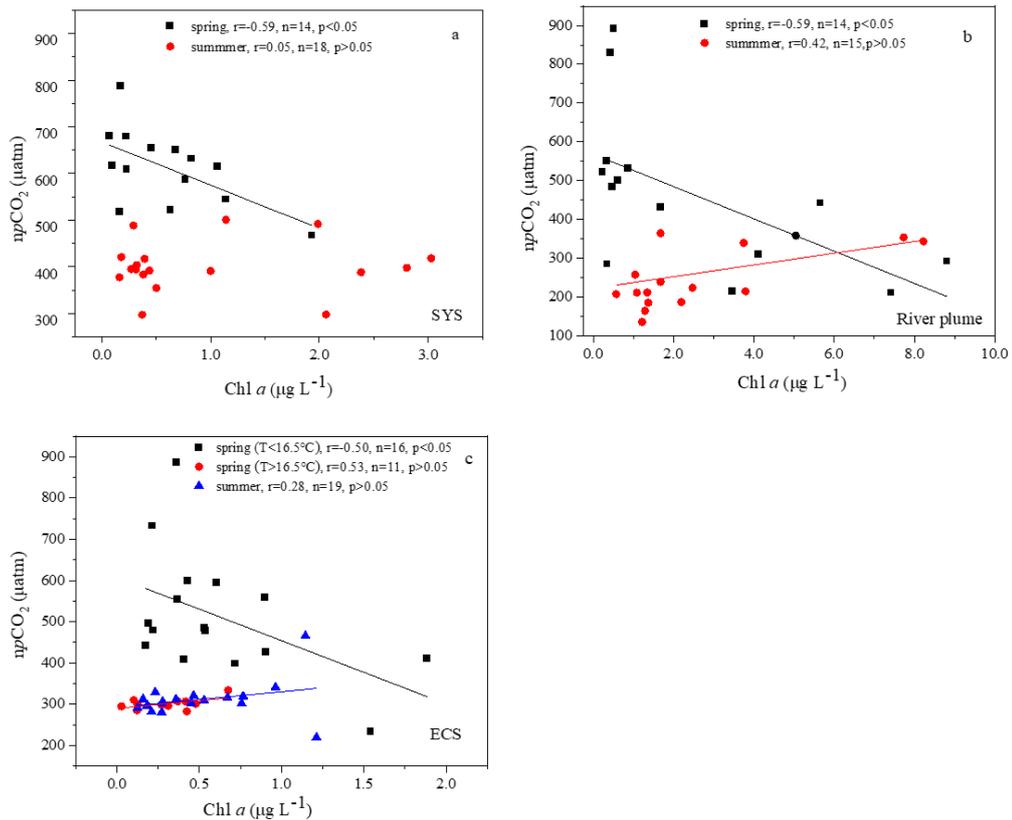
466

467 The negative relationship between $\ln p\text{CO}_2$ and Chl a in spring ($r = -0.59$, $n = 14$,

468 $p < 0.05$) and no conspicuous relationship between them in the SYS during summer

469 (Figure 10a) demonstrated that the effect of biological production on CO₂
470 sequestration processes varied with seasons, which was also found by the observation
471 of Qu et al. (2017) in spring and summer of the SYS, 2011. Noted that a weak CO₂
472 source replaced a significant CO₂ sink in the central SYS in spring (Figure 6), which
473 was opposed to the results of Qu et al. (2014). The reason could be attributed to the
474 fact that a strong vertical mixing still existed in the late-March, and a little bit high
475 *p*CO₂ in the central SYS thus was caused by the upwelling of CO₂-riched Yellow Sea
476 Warm Current, which was characterized by high-temperature saline water (Figure S1,
477 see the supplement). Moreover, the onset of the spring phytoplankton bloom in the
478 central SYS usually started from April (Jin et al., 2013; Liu et al., 2015). Therefore,
479 low biological uptake in the central SYS was not strong enough to offset the enhanced
480 *p*CO₂ from the upwelling of the CO₂-enriched bottom water. The net result was
481 elevated *p*CO₂ value in the central SYS in early spring.

482



483

484 **Figure 10.** Relationship $npCO_2$ and Chl a in spring and summer in three water masses:
 485 SYS (a), River plume (b) and ECS offshore waters (c).

486

487 In the plume area, the relationship of underway pCO_2 and SST in spring is
 488 negative but random in summer, while the relationship of pCO_2 and salinity in spring
 489 is random but positive in summer (Figures 9c and 9d). In summer, the positive
 490 relationship between pCO_2 and salinity was similar to the result of the cruise in July
 491 2007, which was found by Zhai and Dai (2009). Relatively high pCO_2 (Figure 9d)
 492 near the Changjiang River mouth ($S= 24\text{--}26$) was mainly subjected to the physical
 493 mixing effect. Moreover, photosynthesis was at a low level given the light limitation
 494 and strong vertical mixing in the turbid area in spite of high concentration nutrients

495 supply, therefore resulting in high $p\text{CO}_2$ distribution (He et al., 2013; Zhai et al.,
496 2007). Besides, this area was also more affected by the SYS through the Yellow Sea
497 Coastal Current in spring that carried higher CO_2 water southward, thereby resulting
498 in high $p\text{CO}_2$ in the northwest corner of the Changjiang estuary with moderate salinity
499 (Su, 1998). In the outer estuary, low $p\text{CO}_2$ values were distributed in the northeast of
500 the Changjiang estuary. It demonstrated in addition to the physical processes but
501 reinforcement of biological action also greatly affected $p\text{CO}_2$ dynamics. In the south
502 branch of the Changjiang estuary, on one hand, high primary production in this region
503 was fueled by abundant nutrients with the Changjiang River supply, thus leading to
504 biological CO_2 uptake and low $p\text{CO}_2$; On the other hand, the upwelling carrying
505 CO_2 -rich water, which from the flow-northward TWC bottom water, mixed with the
506 surface water and thereby elevated $p\text{CO}_2$ (Wang et al., 2011b). Therefore, the
507 upwelling effect canceled out the intensive biological activity to some extent, then
508 resulting in $p\text{CO}_2$ of the south branch slightly higher than the northwest branch
509 (Figure 6).

510 In spring, the overall random relationship between $p\text{CO}_2$ and salinity as well as
511 the relatively narrow salinity range revealed a strong mixing in water column.
512 Moreover, the negative correlation between $p\text{CO}_2$ and SST in spring also suggested
513 that the water mixing between high- $p\text{CO}_2$ /low-temperature river water and
514 low- $p\text{CO}_2$ /high-temperature ECS water. As for Zhe-Min Coastal Current, intensive
515 primary production occurred with the high nutrients loading from the Changjiang
516 River leading to low $p\text{CO}_2$ values (He et al., 2013; Guo et al., 2015). The negative

517 relationship between $npCO_2$ and Chl *a* in spring ($r = -0.59$, $n = 14$, $p < 0.05$, Figure
518 10b) demonstrated that biological activity was also the main driver on the pCO_2
519 distribution.

520 All summer stations in the ECS were confined to the shelf break of the ECS,
521 while some spring stations were expanded to the outer shelf of the ECS, which was
522 influenced by the KC and open ocean to some extent. The southern and outer shelf of
523 ECS was characterized by warm and saline water with the influences of TWC and KC.
524 In spring, considering that the isotherm of $16.5\text{ }^\circ\text{C}$ coincided with the depth contour
525 of 100m in the ECS, the relationship underway pCO_2 and SST in the ECS offshore
526 waters was divided into two parts to discuss (Figure 9e): one is the negative
527 correlation in the inner and middle shelves with $SST < 16.5\text{ }^\circ\text{C}$ ($r = -0.49$, $p < 0.001$);
528 another is the significant positive correlation in the outer shelf with $SST > 16.5\text{ }^\circ\text{C}$ (r
529 $= 0.74$, $p < 0.001$). This indicated that the pCO_2 variation in the outer shelf of the ECS
530 was mainly dominated by temperature and the apparent coefficient of temperature
531 effect on surface seawater pCO_2 was $0.0584\text{ }^\circ\text{C}^{-1}$, which was higher than that of the
532 $0.0423\text{ }^\circ\text{C}^{-1}$ determined for North Atlantic surface water by Takahashi et al. (1993).
533 This suggested that the temperature effect on seawater pCO_2 was enhanced by other
534 important processes (i.e. mixing with Kuroshio subsurface CO_2 -rich water). Moreover,
535 the apparent temperature coefficient was similar to that in South Atlantic Bight
536 ($0.058\text{ }^\circ\text{C}^{-1}$; Jiang et al., 2008), and higher than that in northern South China Sea
537 (0.024 , 0.049 and $0.03\text{ }^\circ\text{C}^{-1}$ in spring, autumn and winter, respectively; Tseng et al.,
538 2007), and northern Yellow Sea ($0.0205\text{ }^\circ\text{C}^{-1}$; Xu et al., 2016). In the inner and middle

539 shelves of the ECS, the combined influence of the physical mixing and the biological
540 activity masked the effects of temperature on the $p\text{CO}_2$ distribution. The negative
541 relationship between $\eta p\text{CO}_2$ and Chl a ($r = -0.50$, $n = 16$, $p < 0.05$) also showed that
542 the phytoplankton production mainly led to the $p\text{CO}_2$ variation in the inner and
543 middle shelves. Moreover, the positive correlation of the $p\text{CO}_2$ and salinity in spring
544 (Figure 9f) could be an evidence to affirm the intrusion of high salinity CO_2 -rich KC
545 water to the ECS.

546

547 **4.4 Air-sea CO_2 fluxes in subregions and implication**

548 With the three subregions categorized in the South Yellow-East China Sea region,
549 the seasonality variation of both $p\text{CO}_2$ and air-sea CO_2 fluxes were found in each
550 individual region. Theoretically, the variation of the CO_2 flux was mainly attributed to
551 the variability in $\Delta p\text{CO}_2$ and wind speed (Jiang et al., 2008). In the present study, we
552 used the same reanalyzed monthly averaged wind speed from ECMWF to calculate
553 the CO_2 flux inside each cruise. Therefore, there was no spatial pattern in the
554 intra-seasonal variation of the wind speed and the flux difference was solely ascribed
555 to the $\Delta p\text{CO}_2$. Therefore, all above factors that influencing $p\text{CO}_2$ variability also
556 affected the distributions of CO_2 flux.

557 In this study, the SYS served as a CO_2 sink in early spring and a CO_2 source in
558 summer with the average air-sea CO_2 flux of -2.11 ± 4.57 and 2.35 ± 4.30 mmol m^{-2}
559 d^{-1} , respectively. This was in good agreement with the findings in the surveys of Xue
560 et al. (2011) and Qu et al. (2017). To be more specific, it was unquestionable that the

561 SYS served as a CO₂ sink in spring because of low temperature and spring bloom.
562 The trajectory of underway pCO₂ data clearly verified that spring pCO₂ in the SYS
563 was all undersaturated except for the Subei shoal in the southwest SYS (Figure 6).
564 However, in summer, a strong CO₂ sink in the SYS from the June gradually reduced
565 to a weak sink or source in July and further converted to an obvious CO₂ source in
566 August, according to the compilation of our data and the results from Qu et al. (2014,
567 2017). In the SYS, the outbreak of *Ulva prolifera* bloom between the April and June
568 every year absorbs a large amount of CO₂, which promotes the SYS to become a
569 temporary CO₂ sink in early summer (Van Alstyne et al., 2015), then the degradation
570 of the *U. prolifera* in the mid-July to late August also accelerate it to act as a CO₂
571 source (Deng et al., 2018). Therefore, the temporal changes of pCO₂ and CO₂ flux are
572 combined results from biological activity, the Changjiang River plume and the
573 upwelling of the Yellow Sea Cold Water. An exception is that the southeastern SYS
574 along the Korean coast was a net strong CO₂ sink during spring, summer and fall,
575 whereas it was a net weak source during winter (Choi et al., 2019). The sharp contrast
576 in CO₂ fluxes could be attributed to regional variations in magnitude and the
577 terrestrial influences, especially the influences of Changjiang and Yellow Rivers on
578 the western SYS in summer. In short, the factors affecting the pCO₂ distribution and
579 CO₂ flux, especially in summer, are complicated, and more detailed mechanisms for
580 these changes in summer need further intense investigation.

581 As a whole, the river plume region always acted as a CO₂ sink with the fluxes of
582 -3.78 ± 7.44 and -5.17 ± 6.65 mmol m⁻² d⁻¹ from spring to summer. The estimated

583 CO₂ flux in summer were close to those of Zhai and Dai (2009) (-8.8 ± 5.8 and $-4.9 \pm$
584 $4.0 \text{ mmol m}^{-2} \text{ d}^{-1}$ in spring and summer) and multiple observations by Guo et al. (2015)
585 (-10.7 ± 8.2 and $-6.5 \pm 10.7 \text{ mmol m}^{-2} \text{ d}^{-1}$ in spring and summer), whereas the CO₂
586 flux we calculated in spring was lower than those. It could be ascribed to the fact that
587 spring is the transitional season from winter to summer and the plume area is also the
588 transitional zone from the inland to the ECS. Thus, the magnitude of CO₂ flux
589 changes associated with many factors, including wind speed, biological activity and
590 river discharge, etc. Therefore, the estimated CO₂ fluxes in this study are reasonable
591 in comparison with the CO₂ fluxes in spring from multiple observations, such as those
592 of Zhai and Dai (2009) and Guo et al. (2015). As for the ECS, the northern ECS
593 always was a sink in both spring and summer, due to the jointly impacts of primary
594 production and Changjiang River input (Kim et al., 2013), while the southern ECS
595 shifted the CO₂ sink in spring to source in summer with the increasing intrusion of
596 KC and TWC in summer. Overall, the ECS offshore water was also a strong sink in
597 spring with an average CO₂ flux of $-5.56 \pm 3.12 \text{ mmol m}^{-2} \text{ d}^{-1}$. When it comes to
598 summer, it turned into a CO₂ source with the seasonal average CO₂ flux of 1.73 ± 3.05
599 $\text{mmol m}^{-2} \text{ d}^{-1}$, which might be due to the fact that high temperature in the ECS
600 offshore waters, especially in the southern ECS, contributed to the high $p\text{CO}_2$ and a
601 CO₂ source in summer. This is also in good consistency with the flux reported by Guo
602 et al. (2015) in summer.

603

604 **5 Conclusion**

605 Based on the hydrological and carbonate parameters observations with high

606 spatial resolution in the South Yellow-East China Sea region from spring and summer
607 cruises, the results showed that DIC and TA values decreased generally from
608 nearshore to offshore, moreover, the low TA and DIC values as well as high pH,
609 occurred in the river plume, especially in the Zhe-Min Coastal Current in spring. The
610 DIC and TA values in the SYS were higher than those in the ECS due to the Yellow
611 River with intensive carbonate weathering and erosion in the drainage basin. The
612 South Yellow Sea and East China Sea, as a whole, turned from an important CO₂ sink
613 in spring to a weak CO₂ source in summer, Specifically, the air-sea CO₂ flux
614 displayed large spatial and temporal variations in three subregions: CO₂ sinks in the
615 SYS and ECS offshore water converted into sources from spring to summer, while the
616 river plume always served as a CO₂ sink both in spring and summer. The controlling
617 factors of *p*CO₂ differ in different regions and seasons. In general, temperature and
618 biological activity were the primary drivers in controlling the *p*CO₂ variability in the
619 SYS, and primary production was more important than temperature in summer. For
620 the river plume area, *p*CO₂ distribution was largely controlled by physical mixing and
621 biological activity, which could be attributed to the Changjiang runoff with high
622 nutrients supply, while temperature was the dominant factor in the outer shelf of the
623 ECS as the high-temperature KC saline water always occupied in the region.

624 This study improved the understand of sea surface carbonate chemistry dynamics
625 in the two adjacent margin seas with different physical conditions and may also help
626 to better understand the chemical dynamics in many other marine systems. To better
627 understand the role of the continental shelf seas in the global carbon cycle, more

628 comprehensive surveys and researches of direct $p\text{CO}_2$ measurements in the entire
629 South Yellow Sea and East China Sea, especially the seasons (autumn and winter)
630 with weak primary production, are needed.

631

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641

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