

Supporting Online Material

Abating HCl and HNO₃ is more effective than NH₃ for mitigating high aerosol loading over the Indo-Gangetic Plain

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Supplementary text

Table S1

Figure S1

Supporting Text

1. Data and measurements

1.1 Measurement site

The gas and particle-phase species measurements were conducted at the Indira Gandhi International Airport (IGIA), Delhi (28.56° N, 77.09°E, 237 m asl) under the Winter Fog Experiment (WiFEX) field campaign (Ghude et al., 2017). The measurement site fairly represents the general characteristics of highly polluted and densely populated urban locations in the IGP (Sinha et al., 2019; Ali et al., 2019). The measurements were conducted from 19 December 2017 to 10 February 2018 and a detailed description of the observation site and the prominent weather features associated with the winter period in Delhi has been well described in our earlier publications Ali et al., 2019; Acharja et al., 2021.

1.2 Instrumentation

1.2.1 Measurements of trace gases and inorganic constituents of PM₁, PM_{2.5}

In this study, the measurement of trace gases (HCl, HONO, HNO₃, SO₂, and NH₃) and the inorganic constituents (Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, K⁺, Ca²⁺, and Mg²⁺) of fine particulates (PM₁ and PM_{2.5}) were simultaneously conducted at hourly resolution using Monitor for Aerosol and Gases in ambient Air (MARGA-2S, Make: Metrohm Applikon B.V, Netherlands) instrument (Rumsey et al., 2014).

The instrument is an assembly of equipment consisting of sampling unit, denuder for separating trace gases from the particles, aerosol collector, and analytical unit (Makkonen et al., 2012). MARGA-2S can provide a physicochemical characterization of particle matter along with trace gases on hourly time resolution (Schaap et al., 2011). In the measurement system, ambient air is first drawn through the PM₁, PM_{2.5} inlets that pass through horizontal wet rotating denuder (WRD), and gases get diffused into an aqueous film. The aerosol gets collected in the steam jet aerosol collector (SJAC), where water-soluble aerosols grow in a supersaturated environment (Stieger et al., 2018). The water-soluble inorganic ions are quantified with two ion chromatography systems (anion + cation) using the principle presented by Oms et al., 1996. The MARGA-2S instrument was installed on the first floor of the building, whereas PM₁ and PM_{2.5} impactors were installed on the terrace of the building nearly 8m above the ground and 2m above the roof surface. The instrument was continuously calibrated during its operation to ensure the quality of observation. A detailed description of

the MARGA-2S instrument, its operating principle, calibration method, and study location is discussed in our previous publication (Acharja et al., 2021). Meteorological parameters like temperature, relative humidity, wind speed, and wind direction were monitored with co-located automatic weather station (AWS).

2. Methodology

2.1 Thermodynamic equilibrium model ISORROPIA-II

The thermodynamic state of aerosols depends on ionic strength, dissociation of electrolyte, temperature of the solution, pH of aerosol, water activity, surface tension, density, and hygroscopicity (Kim et al., 1993; Tilgner et al., 2021). In the present study, ISORROPIA-II (Cl^- - NO_3^- - SO_4^{2-} - NH_4^+ - Na^+ - K^+ - Mg^{2+} - Ca^{2+} - H_2O system) model was used (Fountoukis and Nenes, 2007; Weber et al., 2016; Liu et al., 2017). In the model, the equilibrium is characterized by simultaneously considering a set of nonlinear algebraic equations coupled with appropriate electro-neutrality and mass conservation equations using the Newton-Raphson and bisectional methods (Ansari and Pandis., 1999). It is also assumed that the system is closed with respect to the total constituents, and the Gibbs free energy is at the global minimum. At this state, the variation in entropy concerning neighboring states is zero, and the system may remain at this energy level for arbitrarily long times.

The observational dataset of concentration of total ammonium ($\text{TNH}_4 = \text{NH}_3 + \text{NH}_4^+$), total nitrate ($\text{TNO}_3 = \text{HNO}_3 + \text{NO}_3^-$), and total chloride ($\text{TCl} = \text{HCl} + \text{Cl}^-$) along with Na^+ , K^+ , Ca^{2+} , and Mg^{2+} were used as inputs to run the model. The model can run in the "backward" mode based on the particle-phase data and in the "forward" mode based on the total (particle + gas) phase concentration data (Xu et al., 2015; Yao et al., 2007; Guo et al., 2017; Song et al., 2007). Studies have shown the "forward" mode to be more accurate and less sensitive to the measurement uncertainties compared to the "backward" mode (Hennigan et al., 2015; Bian et al., 2014). In the "forward" mode, the aerosols are assumed to exist in a metastable or aqueous state. These aqueous particles being larger than dry crystalline solids, generally scatter more light, degrade visibility, and influence mass transfer kinetics of trace gases (Rood et al., 1989).

The particle acidity (pH) is estimated using the following equation:

$$\text{pH} = -\log_{10} \frac{1000 \times [\text{H}_{\text{air}}^+]}{\text{ALWC}} \quad (1)$$

where $[\text{H}_{\text{air}}^+]$ concentration and aerosol liquid water content (ALWC) are the ISORROPIA-II model outputs (Liu et al., 2014; Xue et al., 2011). In the model, the ALWC was estimated by

considering the Raoult effect, curvature effect, and water uptake property (Bian et al., 2014; Hennigan et al., 2008). The water uptake of aerosols is calculated with the Zdanovskii–Stokes–Robinson (ZSR) hypothesis (Stokes and Robinson, 1966)

$$A = \sum_i \frac{M_i}{m_{oi}(a_w)} \quad (2)$$

where M_i is the mole concentration of the electrolyte i (mol m^{-3}), $m_{oi}(a_w)$ is the corresponding molality of the binary solution of the electrolyte i under the same a_w (Seinfeld and Pandis, 2006). At the phase equilibrium a_w is equal to the ambient RH:

$$a_w = RH \quad (3)$$

This approach ignores the contribution of organic species (e.g., oxalate, formate, acetate) to ALWC, but studies have shown that organics do not significantly affect the aerosol liquid water content (Ziemba et al., 2007; Malm and Day, 2001).

In this study, the data of RH in the 30–95% range were considered for the model simulation and data of $RH > 95\%$ were excluded since the exponential growth in ALWC with RH introduces large pH uncertainties (Guo et al., 2015). During the study period, the ambient humidity was high ($\geq 50\%$), much higher than the efflorescence RH (10–30 %), so the "forward" mode consideration and the assumption that aerosols were in a completely deliquesced phase and remained at the "metastable" side of the hysteresis is reasonable. The model-derived pH of $\text{PM}_{2.5}$ was compared to make a comprehensive analysis of the model simulation.

2.2 Gas-particle partitioning ratio

The gas-to-particle partitioning and formation of secondary aerosols are significantly impacted by pH, ALWC, and T. High ALWC results in uptake of a gaseous species onto liquid and phase partitioning depending on the multiphase and heterogeneous reaction rates and the measurement or module of the model used (Nemitz et al., 2004; Hanson et al., 1994; Murphy et al., 2017). The partitioning ratio of ammonium [$\varepsilon(\text{NH}_4^+)$], nitrate [$\varepsilon(\text{NO}_3^-)$], and chloride [$\varepsilon(\text{Cl}^-)$] can be estimated using the following equations:

$$\varepsilon(\text{NH}_4^+) = \frac{[\text{NH}_4^+]}{[\text{NH}_3] + [\text{NH}_4^+]} \quad (4)$$

$$\varepsilon(\text{NO}_3^-) = \frac{[\text{NO}_3^-]}{[\text{HNO}_3] + [\text{NO}_3^-]} \quad (5)$$

$$\varepsilon(Cl^-) = \frac{[Cl^-]}{[HCl] + [Cl^-]} \quad (6)$$

where $[NH_4^+]$, $[NO_3^-]$, $[Cl^-]$, $[NH_3]$, $[HNO_3]$, $[HCl]$ are the molar concentrations ($\mu\text{mol m}^{-3}$) of particulate and gaseous species respectively. The agreement between the predicted and measured partitioning ratios has been evaluated by estimating the mean bias (MB) and normalized mean bias (NMB) using the following equations:

$$MB = \frac{1}{N} \times \sum_{i=1}^N (M_i - O_i) \quad (7)$$

$$NMB = \frac{\sum_{i=1}^N (M_i - O_i)}{\sum_{i=1}^N O_i} \times 100\% \quad (8)$$

where, M_i and O_i are the model predicted and observed data, respectively, and N is the number of data points (Sudheer et al., 2015). These metrics are commonly used to evaluate the model predictions, and negative values for MB and NMB imply lower prediction, whereas positive values show the over-prediction.

2.3 Volume growth factor (VGF)

Factors like pH, ALWC, RH, aerosol composition, aerosol size, and aging of aerosol play an important role in visibility degradation by impacting the growth of hygroscopic aerosols (Stainer et al., 2003; Salcedo et al., 2006). The volume growth factor (VGF) is defined as the ratio of the volume of the wet particle to the corresponding volume of the dry particle (Zheng et al., 2015; Sun et al., 2013). Li et al., (2019) have discussed the size-independent VGF estimated using the following equation:

$$VGF = \frac{\frac{\sum m_{i, MARGA}}{\rho_i} + \left(\frac{ALWC_{inorg} + ALWC_{org}}{\rho_{water}} \right)}{\frac{\sum m_{i, MARGA}}{\rho_i}} \quad (9)$$

where m_{iMARGA} is the mass concentration of species 'i' measured by MARGA-2S. In this calculation, we have also neglected the impact of organic fraction as studies have shown their growth is much less than water-soluble inorganic ions (Dusek et al., 2010; Petters and Kreidenweis, 2007; Guo et al., 2015). The major inorganic species are considered in the calculation, and the densities were assumed to be 1.52, 1.75, 1.75, 1.75, and 1.0 g cm^{-3} for chloride, nitrate, sulfate, ammonium, and water respectively (Salcedo et al., 2006).

3. Results and Discussions

3.1 Temporal evolution of pH and ALWC of PM_{10} and $PM_{2.5}$

We explored the diurnal variation of pH and ALWC of PM₁ and PM_{2.5} in Figure S1. ALWC increases from just after sunset to early morning, then decreases sharply due to daytime heating, increasing temperature, and decreasing RH. It reaches a maximum during morning traffic hours (08:00-10:00 hrs) due to high loadings of PM₁, PM_{2.5} and high humidity in the shallow boundary layer. It was lowest in the afternoon, due to the combined effect of temperature and increased boundary layer height, which dilutes the particle concentration at the ground level, consistent with the variations in other polluted regions experiencing fog (Guo et al., 2018). The pH variation generally followed the variation of liquid water content and it showed a double peak, one in the morning (10:00 hrs), and another in the evening (20:00 hrs), as shown in Figure S1. It was minimum in the afternoon, similar to ALWC, and studies have shown the diurnal variation of ALWC to be the primary driving factor in the diurnal variation of pH.

3.2 Influence of RH and ALWC on volume growth factor (VGF)

In addition to controlling the pH of aerosols, ALWC also impacts the loading of aerosols where higher aerosol concentration increases ALWC content, further enhancing the multiphase formation of aerosol, especially when RH exceeded $\geq 80\%$ resulting in the growth of aerosol during pollution events like fog (Su et al., 2020; Engelhart et al., 2011; Liu et al., 2011; Zhai et al., 2021). Limited research has been performed on this topic over IGP though the highest RH value was beyond 90% during the winter period.

Figures 3c, 3d give a clear view of VGF variability with RH. The figure shows that VGF of both PM₁ and PM_{2.5} increased sharply when ambient RH was $\geq 70\%$, which is due to the water uptake by hygroscopic constituents. Below 70% RH, the mean PM₁ and PM_{2.5} VGF is 2.02 ± 0.77 , 2.08 ± 0.97 , whereas at RH $\geq 70\%$ the mean PM₁ and PM_{2.5} VGF is 7.87 ± 4.58 , 7.38 ± 4.46 respectively. This indicates the increment of VGF by 70% at higher RH, which can be attributed to the increased phase partitioning and formation of more SIA under fog (Sjogren et al., 2007; Wang et al., 2019).

During the whole measurement period, the average VGF of PM₁ and PM_{2.5} was 4.83 ± 4.21 , 5.05 ± 4.44 showing nearly 5 times increase in the volume of foggy particles compared to the particle volume at non-foggy conditions, which is consistent with previous studies (Bian et al., 2014). The increase in VGF of PM_{2.5} aerosols was more compared to PM₁ as it has a larger volume fraction. This increased volume of particles at high RH would augment the particle mass loading, resulting in scattering more sunlight and lowering the surface temperature, and increasing fog sustenance. This results in a positive feedback

mechanism between RH and VGF that would reach equilibrium at some ALWC and pH (Wang et al., 2014; Guo et al. 2017).

Table S1. The measured total (gas + aerosol) concentration of chloride (TCl), nitrate (TNO₃), and ammonia (TNH₄) and estimated pH, ALWC, gas-to-particle partitioning ratio (ϵ) of ϵ (Cl⁻), ϵ (NO₃⁻), and ϵ (NH₄⁺) of PM₁ and PM_{2.5} during the WiFEX campaign period.

Variables	PM ₁	PM _{2.5}
TNO ₃ (μg m ⁻³)	19.74 ± 9.81	31.67 ± 14.57
TCl (μg m ⁻³)	22.28 ± 20.57	39.71 ± 35.41
TNH ₄ (μg m ⁻³)	43.41 ± 19.81	58.04 ± 24.06
Gas-to-particle partitioning ratio (ϵ)		
ϵ (Cl ⁻)	93 ± 09 %	96 ± 07 %
ϵ (NO ₃ ⁻)	83 ± 11 %	89 ± 08 %
ϵ (NH ₄ ⁺)	42 ± 17 %	55 ± 15 %
Predicted from ISORROPIA-II		
pH	4.49 ± 0.53	4.58 ± 0.48
ALWC (μg m ⁻³)	169 ± 205	324 ± 396

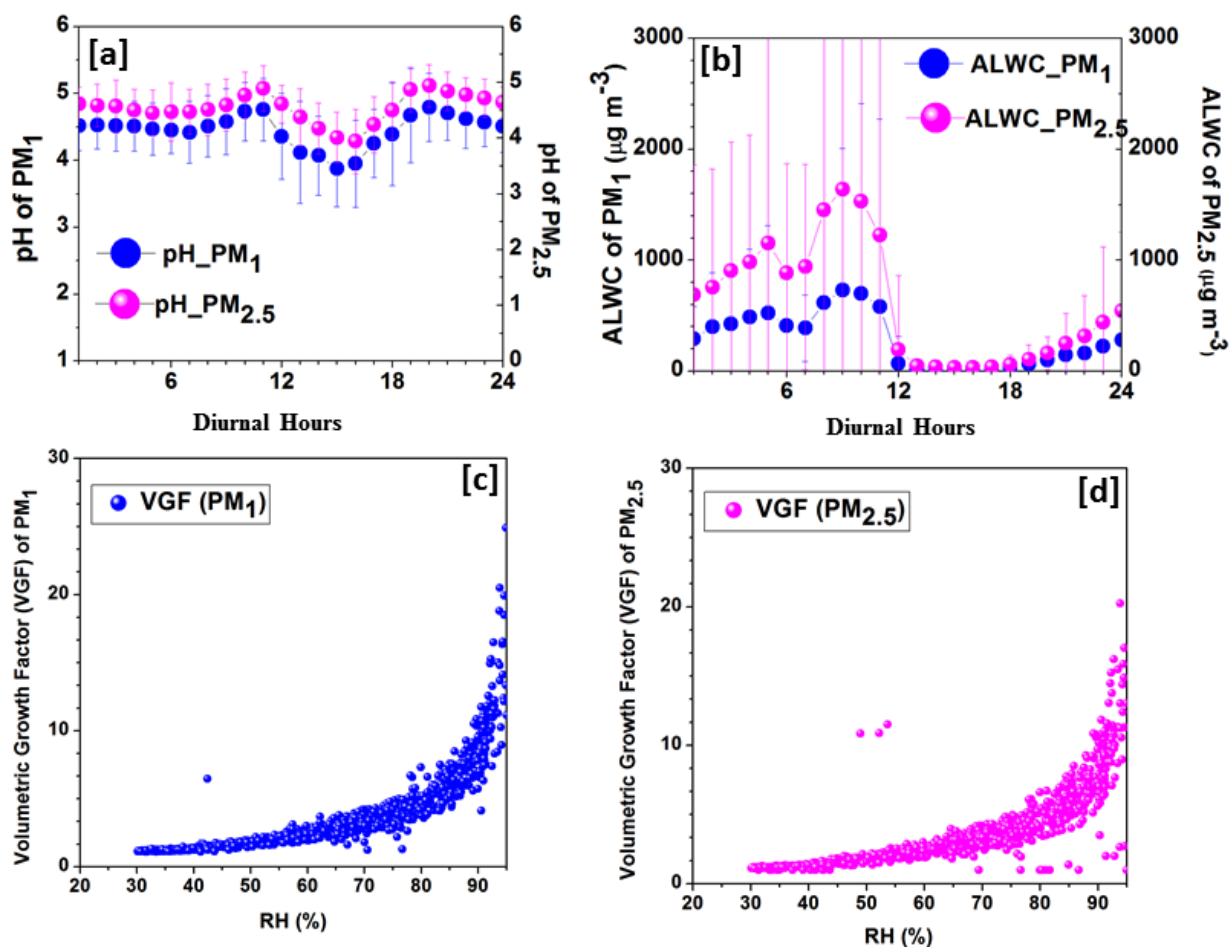


Figure S1: Diurnal variation of (a) pH and (b) ALWC of PM_1 and $PM_{2.5}$ predicted from the ISORROPIA-II model simulation. Variation of volume growth factor (VGF) of (c) PM_1 and (d) $PM_{2.5}$ as a function of RH during the WiFEX campaign period.

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