

1 **Low biodegradability of dissolved organic matter from Southeast Asian**
2 **peat-draining rivers**

3 **Robert S. Nichols^{1*}, Patrick Martin¹**

4 ¹Asian School of the Environment, Nanyang Technological University Singapore,

5 50 Nanyang Dr, Singapore 637459

6 Corresponding Author: Robert S. Nichols (robertsc001@e.ntu.edu.sg)

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8 **Key points**

- 9 • Low biodegradability of peatland DOC during 56-day incubations, including when
10 diluted with seawater or amended with nutrients
- 11 • No measurable phenol oxidase activity in the peat-draining rivers and coastal waters
12 of Sarawak
- 13 • Auto-oxidation of the phenol oxidase assay substrate L-DOPA occurs at $\text{pH} \geq 7$; we
14 recommend ultrafiltration to generate enzyme-free controls
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19 **Abstract**

20 Southeast Asia's extensive tropical peatlands account for a significant proportion of the global
21 riverine dissolved organic carbon (DOC) flux to the ocean. Peat-derived DOC is rich in
22 polyphenolic compounds, the microbial degradation of which is thought to rely on extracellular
23 phenol oxidases. Despite substantial interest in the biogeochemical fate of terrigenous DOC
24 (tDOC), few studies have quantified phenol oxidase activity in aquatic environments, and
25 microbial remineralization rates of tDOC have never been measured in Southeast Asia. Here,
26 we assess the potential for using phenol oxidase assays as a proxy of tDOC biodegradation
27 across peat-draining rivers and coastal waters of Sarawak, Borneo, and report experimental
28 measurements of microbial tDOC remineralization rates from this region. We show first that
29 phenol oxidase assays in aquatic samples are problematic because of the rapid, pH-dependent
30 auto-oxidation of the assay substrate. Our field measurements of phenol oxidase activity
31 detected only substrate auto-oxidation, suggesting that real phenol oxidase activity was low or
32 absent. Second, we report that peatland tDOC, collected from one of the few remaining intact
33 peatlands on Borneo, showed at most very limited biodegradation (0–6% loss of DOC, and 0–
34 12% loss of coloured dissolved organic matter) during several 56-day incubation experiments
35 at in-situ temperature of ~30°C, even when diluted with seawater or amended with nutrients.
36 Our results suggest that direct microbial respiration is perhaps not a major pathway for peatland
37 tDOC remineralization in Southeast Asia, and that photo-oxidation is more likely to control the
38 fate of this carbon.

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40

41 **1 Introduction**

42 Peatlands in Southeast Asia contain approximately 69 Gt of soil organic carbon, mainly in
43 Sumatra and Borneo (Dommain et al., 2014; Page et al., 2011). Rivers draining Southeast Asian
44 peatlands have amongst the highest dissolved organic carbon (DOC) concentrations reported
45 globally (Alkhatib et al., 2007; Baum et al., 2007; Martin et al., 2018; Moore et al., 2011;
46 Müller et al., 2015), and are thought to account for around 10% of the global land-to-ocean
47 flux of terrigenous DOC (tDOC) (Baum et al., 2007; Moore et al., 2011). tDOC derived from
48 soils has traditionally been thought of as largely refractory to microbial decomposition, owing
49 to its inherent chemical properties (Alexander, 1965; Sollins et al., 1996). Peat-derived tDOC
50 is rich in phenolic lignin degradation products that are generally considered resistant to
51 degradation (Gandois et al., 2014; Moore et al., 2013), and which inhibit extracellular enzymes
52 (Freeman et al., 2001; Mann et al., 2014). Tropical peat may in fact be particularly phenol-rich
53 compared to peat at high latitudes (Hodgkins et al., 2018, Yule et al., 2018).

54 However, it is usually thought that the DOC pool in the deep ocean only contains a relatively
55 minor contribution from tDOC (Dittmar & Stubbins, 2014; Opsahl & Benner, 1997), even
56 though the flux of tDOC to the ocean is sufficient to account for the entire oceanic DOC
57 turnover (Williams & Druffel, 1987). This implies that tDOC is biogeochemically labile over
58 relatively short time-scales. Microbial degradation is recognized to be a key pathway for tDOC
59 remineralization in aquatic environments (Bianchi, 2011; Cai, 2011; Ward et al., 2013) and
60 was shown to quantitatively dominate over photodegradation in the Arctic Ocean and on the
61 Louisiana Shelf (Fichot & Benner, 2014; Kaiser et al., 2017). Aarnos et al. (2018) recently
62 estimated that 71% of the global riverine tDOC flux is remineralised within 1 year of entering
63 the ocean by a combination of bio- and photodegradation. However, some studies report tDOC
64 to be resistant to bio- and/or photodegradation (Stubbins et al., 2017; Shirokova et al., 2017).
65 Moreover, both Follett et al. (2014) and Zigah et al. (2017) argued, based on carbon isotopic

66 data, that a greater proportion of DOC in the ocean is tDOC than previously believed.
67 Resolving whether tDOC in different regions of the world is labile to biodegradation and/or to
68 photodegradation is therefore important for improving our understanding of the global carbon
69 cycle. Moreover, it is now increasingly thought that the biodegradability of organic matter is
70 not purely determined by its chemical properties, but depends also on the microbial community
71 composition, physical protection of the organic matter, environmental drivers such as
72 temperature and pH, and enzyme kinetics (Bianchi, 2011; Kleber, 2010). Consequently, it is
73 also necessary to determine whether the biodegradability of tDOC changes across the land–
74 ocean aquatic continuum.

75 In Southeast Asia, several studies have concluded from measurements of air–water CO₂ fluxes
76 that the majority of the peatland tDOC is remineralized within rivers, estuaries, and the coastal
77 ocean (Müller et al., 2015, 2016; Wit et al., 2015, 2018). Low O₂ concentrations have also been
78 taken as evidence for remineralization of tDOC within the peat-draining rivers (Müller et al.
79 2015; Rixen et al., 2008). However, tDOC often appears to mix conservatively across peat-
80 draining estuaries in Southeast Asia (Alkhatib et al., 2007; Martin et al., 2018; Rixen et al.,
81 2008; Zhou et al., 2019). Such conservative mixing suggests that tDOC actually experiences
82 limited biogeochemical processing at least until it reaches the coastal sea. Moreover, Southeast
83 Asian shelf seas have been classed both as sources (Borges et al., 2005; Kartadikaria et al.,
84 2015) and as sinks (Laruelle et al., 2014) for atmospheric CO₂, highlighting the need for a
85 better understanding of carbon biogeochemistry in this region.

86 Microbial processing of organic matter relies on enzymes, and enzyme activity rates are
87 therefore widely measured as indices of microbial processes in terrestrial and aquatic
88 environments (Arnosti et al., 2014; Hoppe, 1991; Sinsabaugh et al., 2008). Phenol oxidases
89 (POx) are a key class of enzymes that oxidize phenolic groups, and are released by a broad
90 range of microbes to detoxify metal ions and degrade phenolic and humic compounds

91 (Sinsabaugh, 2010). POx activity is often measured in soil biogeochemical studies because
92 POx are thought to exert a key control over soil organic matter degradation rates (Allison &
93 Vitousek, 2004; Carreiro et al., 2000; Freeman et al., 2001; Freeman et al., 2004; Mazzon et
94 al., 2018; Prescott, 2010; Sinsabaugh, 2010; Sinsabaugh & Shah, 2011; Stursova &
95 Sinsabaugh, 2008; Wang et al., 2015). POx activity in soils is strongly inhibited by lack of
96 oxygen (Freeman et al., 2001), low pH (Stursova & Sinsabaugh, 2008; Tahvanainen &
97 Haraguchi, 2013), and high concentrations of phenolic compounds (Williams et al., 2000). The
98 environmental conditions in peatlands have therefore been hypothesized to act as a positive
99 feedback that promotes further organic matter accumulation by inhibiting POx activity
100 (Freeman et al., 2001; Freeman et al., 2004). This “enzymatic-latch hypothesis” (Freeman et
101 al., 2001) would predict that POx rates, and thus the potential for biodegradation of tDOC,
102 should increase as phenol-rich tDOC is diluted, and as oxygen and pH increase during transit
103 through estuaries and into coastal seas.

104 Yet despite the interest in the biogeochemical fate of tDOC in aquatic environments (Bauer et
105 al., 2013; Cai, 2011; Ciais et al., 2013), to our knowledge only two studies have attempted to
106 measure POx in aquatic samples. Mann et al. (2014) reported that POx and other extracellular
107 enzyme activities increased after experimental removal of phenolic DOC from samples taken
108 in the permafrost-draining Kolyma River (Mann et al., 2014). This result is similar to the
109 reported effect of reducing the phenol concentration in soil samples (Freeman et al., 2004).
110 Moreover, POx activity in the Kolyma River was correlated with the microbial respiration rate
111 (Mann et al., 2014). These results are consistent with a link between POx activity and tDOC
112 biodegradation rate in rivers. A second study by Sieczko & Peduzzi (2014) found that flooding
113 of the Danube river initially promoted the activity of sugar- and protein-degrading enzymes
114 (glucosidases and protease) due to the introduction of fresh DOM, but this was later followed
115 by increasing POx activity after the water table started to drop, suggesting that the microbial

116 community switched to lignin degradation. Sieczko & Peduzzi (2014) therefore proposed that
117 changes in the ratio of hydrolytic enzyme activity to POx activity could be used as a proxy to
118 track the utilization of allochthonous versus autochthonous DOM. These two studies (Mann et
119 al., 2014; Sieczko & Peduzzi, 2014) thus suggest that POx activity measurements hold promise
120 as a proxy of tDOC biodegradation in aquatic environments.

121 Here, we test whether POx activity can be measured as a proxy of tDOC biodegradation across
122 salinity gradients from peatland-draining rivers to coastal waters of Sarawak, Borneo.
123 Moreover, we report experimental measurements of tDOC biodegradability from Southeast
124 Asian peatland-draining rivers.

125

126 **2 Materials and Methods**

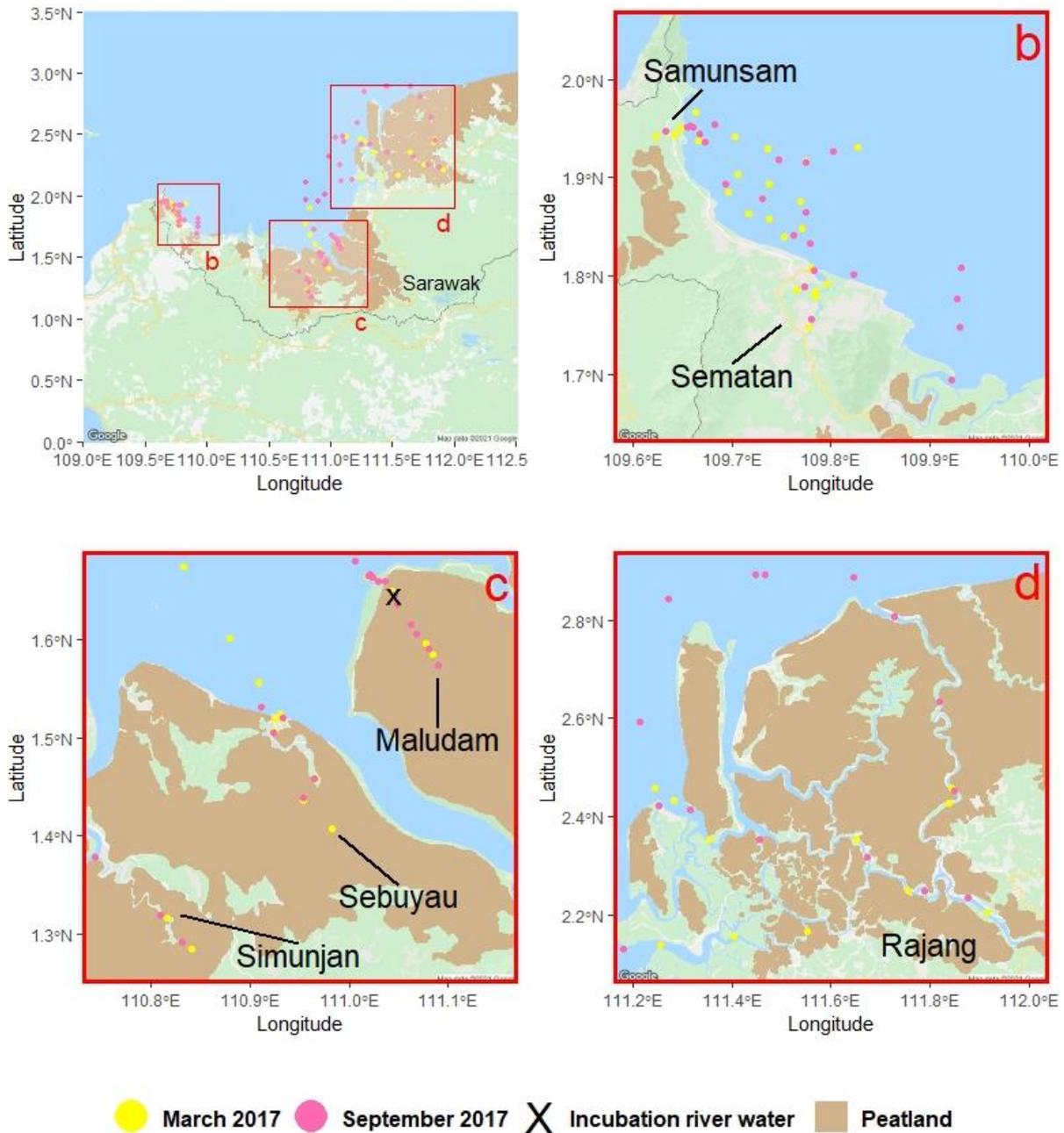
127 **2.1 Sites and sample collection**

128 Six rivers, their estuaries, and surrounding coastal seawater were sampled in Sarawak (Borneo)
129 during March and September 2017 (Figure 1). The rivers Maludam, Simunjan, Sebuyau, and
130 Samunsam are blackwater rivers that drain catchments containing large areas of peatland, and
131 have DOC concentrations of 1,000–4,400 $\mu\text{mol l}^{-1}$. In contrast, the rivers Rajang and Sematan
132 drain catchments that consist to a greater proportion of mineral soils, and have DOC
133 concentrations below 500 $\mu\text{mol l}^{-1}$ (Martin et al., 2018). Data for DOC concentration, coloured
134 dissolved organic matter, and fluorescent dissolved organic matter for these sampling
135 campaigns were already reported by Martin et al. (2018) and Zhou et al. (2019). March
136 corresponds to the end of the wettest season of the year (northeast monsoon), while September
137 marks the end of the drier southwest monsoon. Rainfall in Sarawak is relatively high at all
138 times of the year though, averaging >100 mm per month year-round (Sa'adi et al., 2017). Water
139 temperature averaged 28.5–29.5°C in both seasons.

140 Water samples for enzyme activity assays were collected from the upper 1m of water using a
141 bucket or hand-held jug and immediately frozen unfiltered in dry shippers (-190°C) for
142 transport back to the laboratory, and then stored at -80°C until analysis up to 3 months later.
143 Salinity and temperature were measured at all stations using either a YSI CastAway or a
144 Valeport FastCTD. Water pH was measured at most stations using a YSI Aquaread AP-2000.

145 To further test the POx assay method, water samples were collected from two sites around
146 Singapore (Singapore Strait: 1.226°N 103.746°E, 5 m depth, December 2018; Johor Strait:
147 1.403°N 104.002°E 1 m depth, March 2019). Depth profiles of water were also collected at
148 two oceanic sites north of Hawaii (22.76°N -158.077°E and 22.77°N -158.056°W) using a
149 Niskin rosette between 10 m and 500 m depth. All samples were frozen immediately after
150 collection in a dry shipper (-190 °C) and stored at -80°C until analysis within 3 months.

151 Additional synthetic solutions to measure pure auto-oxidation of the POx assay substrate were
152 prepared from ultrapure Elga water (here and below, 18.2 MΩ cm⁻¹) either alone or with added
153 NaHCO₃ (Sigma-Aldrich S6014) or synthetic sea salt (Sigma-Aldrich S9883) to manipulate
154 the pH and salinity.



155

156 **Figure 1.** Locations of March (yellow circles) and September (pink circles) sampling locations for phenol oxidase
 157 and Leucine-aminopeptidase activity measurements in Sarawak, 2017. The sampling location of river water
 158 collected for all biodegradation incubations is shown (X). Labelled boxes correspond to the maps in panels (b-d)
 159 and annotated with rivers sampled. Brown shading delineates peatland as reported by Dommain et al., 2014.

160

161 2.2 Enzyme assays and data analysis

162 LAP activity was measured to provide an index of total heterotrophic microbial activity. Note

163 that the presence of measurable LAP does not imply that tDOC biodegradation is taking place,

164 as LAP activity might simply reflect the processing of autochthonous organic matter. All
165 enzyme assays were conducted with unfiltered water (200 μ l) in triplicate with a Spark Tecan
166 10m microwell plate reader. Proteinase activity was measured as leucine aminopeptidase
167 activity (LAP) and was assayed using the substrate L-leucine-7-amido-4-methylcoumarin
168 (Sigma-Aldrich, A9891) at a final concentration of 40 μ mol l⁻¹ by measuring fluorescence
169 (excitation 365 nm, emission 450 nm) at 10-minute intervals for 5 hours (Hoppe 1993).
170 Standard curves were created for each assay in Elga water using the fluorescent standard 7-
171 Amino-4-methylcoumarin (Sigma-Aldrich, A9891), and differences in quenching of sample
172 fluorescence were corrected for by adding a known concentration of standard to each sample
173 and calculating the quench factor compared to the standard curve.

174 POx activity was measured with the intention of providing a relative measure of tDOC
175 degradation activity. POx was assayed using the substrate L-DOPA (L-3,4-
176 dihydroxyphenylalanine, Sigma-Aldrich D9628) at a final concentration of 1 mM by
177 measuring the change in absorbance at 460 nm. POx assays consistently showed sigmoidal
178 reaction kinetics (Figure S1), which is a known feature of L-DOPA-based POx assays
179 (Sinsabaugh 2010). To account for this, absorbance measurements were taken at 10-minute
180 intervals for 6 hours (Figure S1), and the substrate oxidation rates were calculated from the
181 maximum linear slope of absorbance or fluorescence over at least 5 consecutive time points.
182 The steepest slope for POx assays was typically found between 1 and 2 hours after starting the
183 assay. Because the oxidation product of L-DOPA (Dopachrome) is not commercially available,
184 we determined the extinction coefficient empirically during each assay by reacting a known
185 quantity of L-DOPA with a solution of mushroom tyrosinase (0.2 mg/ml final concentration,
186 Sigma-Aldrich CAS 9002-10-2). We then applied the average extinction coefficient of all
187 assays (0.1832 μ mol l⁻¹ m⁻¹ at 460 nm) to convert absorbance to moles of L-DOPA oxidized.

188 Enzyme activity assays require matrix-matched controls to correct for possible non-enzymatic
189 changes in sample absorbance or fluorescence. Duplicate samples for POx and LAP samples
190 were collected in September 2017 to provide individual autoclaved controls for each sample,
191 but in March 2017, controls for autoclaving were only collected from four stations covering
192 the main gradients in water chemistry across our study region (i.e., peatland-draining
193 blackwater rivers, mid-salinity estuaries, and coastal seawater). Controls for LAP assays were
194 autoclaved for 20 minutes at 121°C, and never showed any significant change in fluorescence.
195 Controls for POx were autoclaved for 1 hour at 121°C. As discussed below, autoclaved controls
196 for POx activity all showed similar L-DOPA oxidation rates to samples, prompting additional
197 experiments to distinguish between enzymatic and chemical oxidation of L-DOPA.
198 Ultrafiltration to create enzyme-free samples was performed by pre-filtering samples with 0.2
199 µm Acrodisc syringe filters, followed by 3 kDa Amicon centrifugal filtration.

200 For the LAP assays, all data are presented as the autoclaved control-corrected LAP activities,
201 but the autoclaved controls all showed uniformly minimal change in fluorescence so that this
202 correction was negligible. Because the autoclaved controls for POx assays had similarly high
203 substrate oxidation rates as the samples, we present the POx data simply as the measured L-
204 DOPA oxidation rates in the samples and in the controls (Substrate oxidation rate).

205 2.3 Biodegradation incubations

206 We conducted two separate experiments to test whether tDOC from an intact Southeast Asian
207 peatland is labile to microbial remineralization. Water for experimental incubations was
208 collected from the upper 1 m in the Maludam River (at salinity 0, upstream of any human
209 infrastructure in and around the village of Maludam, 1.645°N, 111.046°E) in July 2019 and
210 December 2019. Coastal seawater was collected from the Singapore Strait in December 2019.
211 The Maludam River was selected because it is a fully peat-draining river that originates within
212 an intact peat dome and drains a catchment that is a designated national park (Müller et al.,

213 2015). The Maludam therefore represents the closest to a pristine example of a tropical peat-
214 draining river within our sample area, and is one of the few peat swamp forests in Southeast
215 Asia that is still intact (i.e. has not been subjected to large-scale drainage or deforestation),
216 unlike most peatlands in the other river catchments we studied.

217 All glassware and containers used for biodegradation and DOC analysis were either pre-baked
218 at 450°C for 4 hours or acid-washed and dried before use. All filters were pre-washed with
219 ~300 ml of Elga water before use. 250-ml Duran bottles with polypropylene screw caps were
220 used for all incubations. Incubation bottles were kept in a dark box in a covered location
221 outdoors at ambient temperature (ranging from 26°C at night to 31°C during the day) and
222 swirled gently every 2-3 days. All incubation bottles were kept tightly sealed to avoid
223 evaporation.

224 To quantify tDOC biodegradation rates in undiluted river water, and test whether tDOC
225 biodegradation might be limited by nutrients, Maludam River water was collected in July 2019.
226 Half of the water was immediately filtered (0.2 µm Whatman Polycap TC 75 capsule filter)
227 upon collection and half left unfiltered. The samples were stored in separate 10-L HDPE jerry
228 cans and shipped to Singapore. Incubations were started 7 days after collection. Unfiltered
229 water was then further split into two biodegradation treatments, one of which was amended
230 with nutrients (5 µmol l⁻¹ KNO₃ (Sigma-Aldrich, product number 221295) and 1 µmol l⁻¹
231 KH₂PO₄ (Fisher Scientific, catalog number P285)). The filtered water was then re-filtered (0.22
232 µm Supor membrane, Merck Millipore) as a microbe-free control. Triplicate initial samples
233 were taken from each homogenized treatment to measure DOC and coloured dissolved organic
234 matter (CDOM). The remaining water was then equally split into 250-ml Duran bottles with
235 PTFE-lined caps each with ~150 mL water and 100 mL headspace. At each sampling point
236 (after 7, 14, 28, and 56 days), 3 sacrificial replicates were taken per treatment to measure DOC
237 & CDOM.

238 To test the hypothesis that biodegradation of peatland tDOC might occur after the tDOC has
239 been substantially diluted with coastal seawater and exposed to a coastal marine microbial
240 community, water was again collected from the Maludam river in December 2019. The water
241 was immediately filtered as in July 2019, shipped to Singapore, and then re-filtered through a
242 0.22 μm Supor membrane. Coastal surface seawater was then freshly collected in the Singapore
243 Strait (1.228°N, 103.750°E), part of which was kept unfiltered as an inoculum and the rest
244 filtered through a 0.22 μm Supor filter. Incubations were started 21 days after the Maludam
245 water was collected, and one day after the seawater was collected, by creating four treatments:
246 filtered seawater only (as sterile control), filtered seawater + unfiltered seawater inoculum (to
247 measure background DOC remineralization rate), filtered seawater + filtered Maludam water
248 (as sterile control), and filtered seawater + filtered Maludam water + unfiltered seawater
249 inoculum (to measure remineralization rate of tDOC). The two treatments with added tDOC
250 received 1.25% of final volume of Maludam water, which raised the DOC concentration by
251 $\sim 45 \mu\text{mol l}^{-1}$. The two treatments to which the seawater inoculum was added received 5% of
252 final volume of unfiltered seawater. After taking initial DOC and CDOM samples from each
253 treatment, the remaining water was equally split into 250-mL Duran bottles per treatment to
254 give 3 sacrificial replicates of ~ 150 mL for each sampling point (after 8, 15, 29, and 56 days)
255 with single DOC and CDOM samples taken from each replicate. Mixing ratios for all
256 incubations are shown in table S1.

257 Oxygen measurements were not taken during incubations. However, a headspace of air of at
258 least 40% of the total bottle volume was left in all incubation flasks. We estimate that this
259 provided an O_2 :DOC molar ratio of at least 1.9 even in the treatment with highest DOC
260 concentration, which contained $\sim 470 \mu\text{mol DOC}$ in 150 ml sample volume, while the
261 headspace of 100 ml would have contained $\sim 840 \mu\text{mol O}_2$). While the rate of O_2 diffusion from
262 headspace into the sample could have slowed the rate of biodegradation, we can confidently

263 rule out that O₂ limitation prevented carbon decomposition over the duration of our
264 experiments. This is also demonstrated by the fact that river water samples from Sarawak lost
265 up to 400 μmol l⁻¹ DOC during five-day photodegradation experiments using lower
266 headspace:sample volume ratios in sealed bottles (Martin et al., 2018).

267 2.4 Chemical analyses

268 2.4.1 Dissolved organic carbon (DOC) analysis

269 DOC samples (30 mL) for all experiments were syringe-filtered (0.22 μm Acrodisc, pre-rinsed
270 with 180 ml Elga water and flushed with sample before collection) into EPA vials, acidified
271 with 100 μL 50% H₂SO₄ and stored at 4°C. Samples were analysed within one month on a
272 Shimadzu TOC-L system with high-salt kit, with an injection volume of 150 μl and a sparge
273 time of 5 min, and 5-7 replicate injections to ensure a coefficient of variance ≤2%. Calibration
274 and drift monitoring were performed using potassium hydrogen phthalate standards. The
275 analytical accuracy was determined using deep-sea Certified Reference Material from the
276 University of Miami, USA (42–45 μmol l⁻¹ DOC), which returned a long-term mean and
277 standard deviation of 48 ± 3.9 μmol l⁻¹.

278 2.4.2 Coloured dissolved organic matter (CDOM) analysis

279 Samples for CDOM were syringe-filtered (0.22 μm Acrodisc, pre rinsed with 180 ml Elga
280 water and flushed with sample before collection) into EPA vials, which were then either run
281 immediately, or stored at 4°C and then warmed to room temperature before analysis. The
282 absorbance was measured at 230-900 nm at 1-nm resolution against an Elga water reference
283 on a Thermo Evolution 300 dual-beam spectrophotometer, using either a 10-cm or 0.2-cm
284 quartz cuvette. Instrument performance was checked prior to analysis according to Mitchell et
285 al., (2000).

286 Absorbance spectra were converted to Napierian absorbance coefficients as:

$$a_{\lambda} = \frac{2.303 \times A_{\lambda}}{l}$$

Where a_{λ} is the absorption coefficient at wavelength λ , A_{λ} is the absorbance at wavelength λ , and l is the pathlength of the cuvette in meters. Spectra were baseline-corrected by subtracting the mean absorbance from 700-800 nm (Green and Blough, 1994). The absorption coefficient at 350 nm (a_{350}) was determined, and CDOM spectral slopes calculated for the intervals 275-295 nm ($S_{275-295}$) and 350-450 nm ($S_{350-400}$) as the absolute value of the linear regression of log-transformed Napierian absorption coefficients against wavelength (Helms et al., 2008). The spectral slope ratio S_R was calculated as the ratio of $S_{275-295}$ to $S_{350-400}$.

2.5 Data analysis

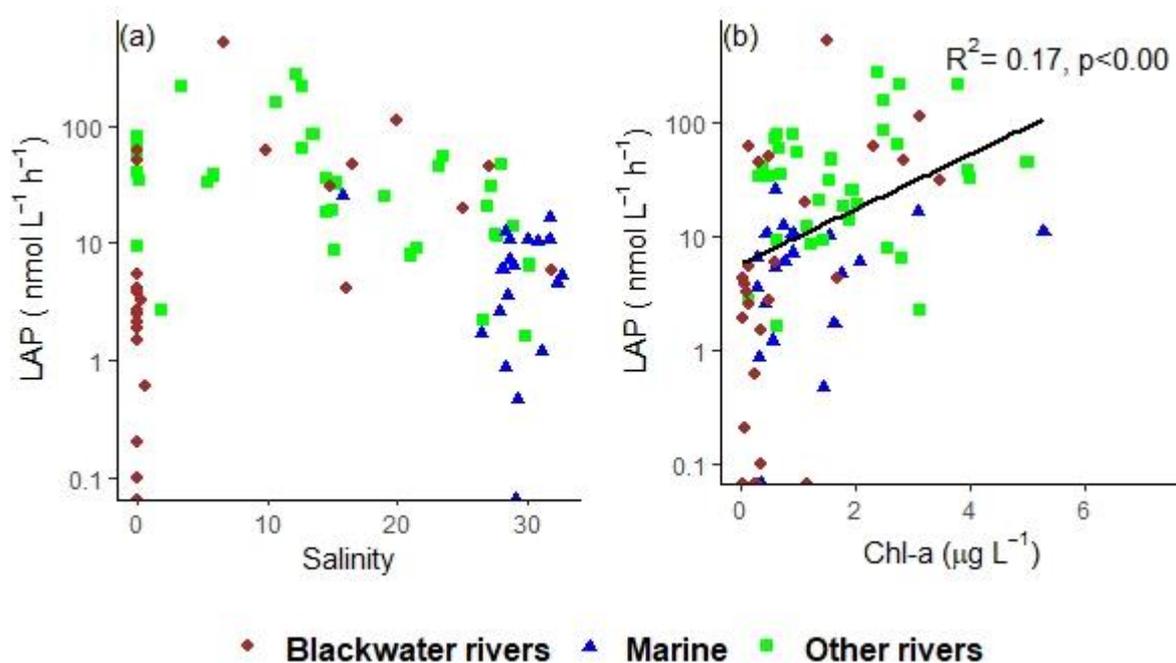
All statistical analyses and CDOM spectral calculations were conducted using R (R core team 2020) and the R packages ‘tidyverse’ (Wickham et al., 2019), and ‘hyperspec’ (Beleites & Sergio, 2020). All concentrations and activity rates are quoted as mean \pm 1 standard deviation unless otherwise stated.

3 Results

3.1 Leucine aminopeptidase (LAP) activity

Extracellular LAP activity spanned four orders of magnitude (Figure 2a). Freshwater stations (i.e. salinity = 0) showed higher LAP activity in the Rajang and Sematan (57 ± 26.7 nmol l⁻¹ h⁻¹, n=8, DOC concentration 100–400 μ mol l⁻¹) than in the blackwater rivers (2.5 ± 5.4 nmol l⁻¹ h⁻¹, n=20, DOC concentration 1100-4400 μ mol l⁻¹), except for two outliers (59.9 and 49.9 nmol l⁻¹ h⁻¹), which were collected next to villages in the Maludam and Sebuyau blackwater rivers. LAP activities reached highest values in the estuaries at salinities between 3 and 12 (159 ± 158 nmol l⁻¹ h⁻¹, n = 10) and then decreased with increasing salinity, and only averaged 11.1 ± 11.5 nmol l⁻¹ h⁻¹(n=30) at stations with salinity>26. Across the entire dataset, there was a statistically

310 significant relationship between LAP activity and chlorophyll-*a* concentration ($R^2 = 0.17$,
 311 $p < 0.001$, Figure 2b; chlorophyll-*a* data taken from Martin et al., 2018).



312

313 **Figure 2.** Leucine-aminopeptidase (LAP) of blackwater rivers (Maludam, Sebuyau, Samunsam and Simunjan, all
 314 $> 500 \mu\text{M}$ DOC), other rivers (Rajang, Sematan and Lundu, all $< 500 \mu\text{M}$ DOC) and marine waters from Sarawak,
 315 Borneo during surveys in March and September 2017. (b) Linear regression of LAP (this study) and chlorophyll-*a*
 316 concentrations as reported by Martin et al. (2018).

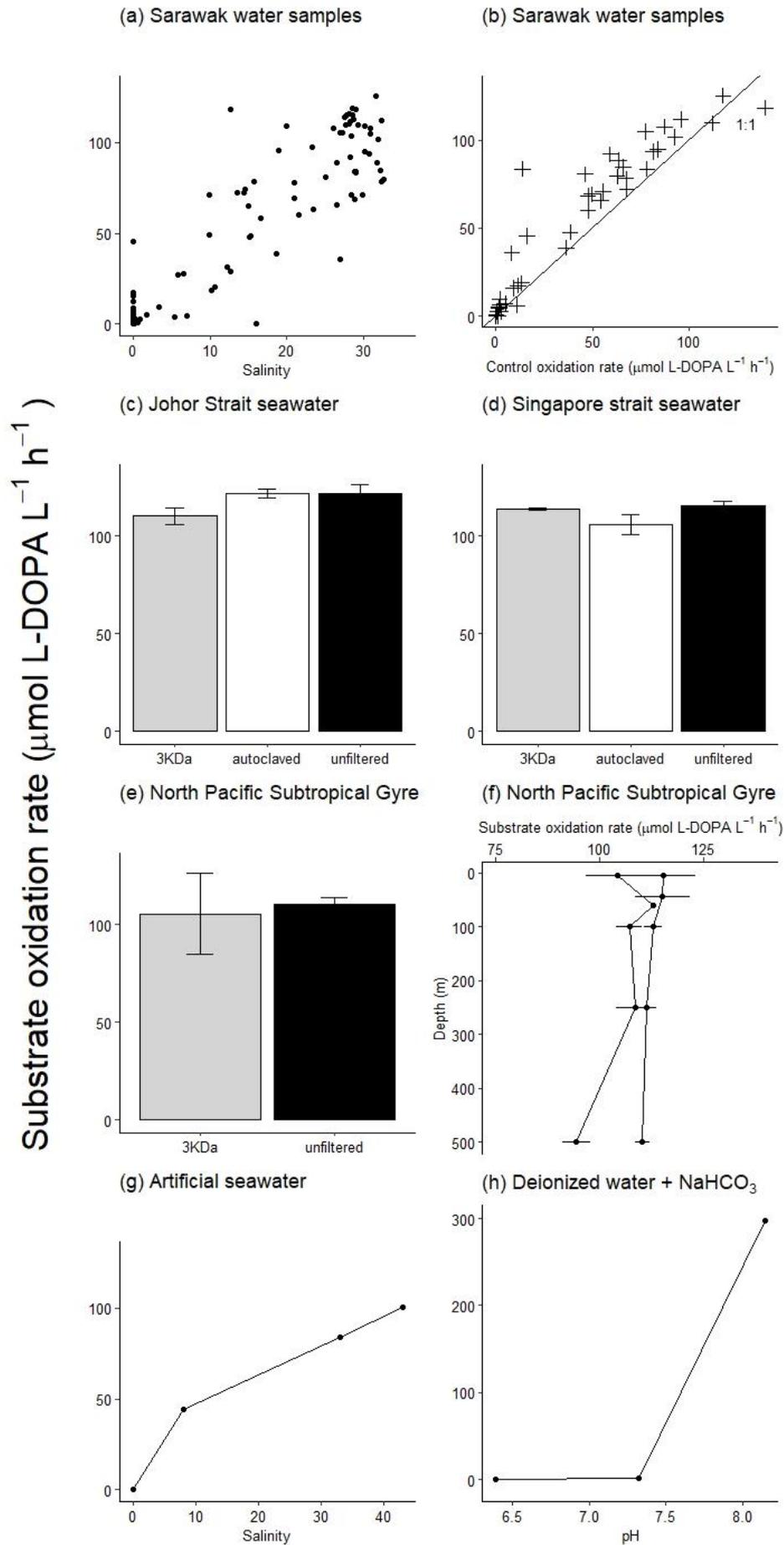
317

318 3.2 Phenol oxidase (POx) activity and auto-oxidation of L-DOPA

319 Strikingly, we observed a strong increase in the oxidation rate of the POx assay substrate, L-
 320 DOPA, with increasing salinity, ranging from 0 to about $140 \mu\text{mol l}^{-1} \text{h}^{-1}$ (Figure 3a). However,
 321 our autoclaved controls showed essentially the same substrate oxidation rates as the un-
 322 autoclaved samples, which means that the L-DOPA oxidation rate in our assays was due to
 323 non-enzymatic oxidation of the substrate (Figure 3b).

324 To verify that L-DOPA is oxidized non-enzymatically in seawater, we used ultrafiltration to
 325 generate enzyme-free seawater samples. We found that samples collected in the Johor and
 326 Singapore Straits (two coastal sites around Singapore with salinity ≥ 31), as well as from two
 327 depth profiles north of Station ALOHA in the North Pacific Subtropical Gyre, all showed

328 similarly high L-DOPA oxidation rates in untreated water as in 3-kDa ultrafiltered and
329 autoclaved controls (Figure 3c-f). The small decrease in oxidation rate observed in the 3 kDa
330 ultrafiltered controls from the Johor Strait (Figure 3c) was most likely due to slight dilution
331 with residual Elga water used to rinse the filters. Notably, the L-DOPA oxidation rates were
332 very similar for all Sarawak samples with salinity >30, all samples from Singapore, and all
333 samples from Station ALOHA from the surface down to 500 m (all between 100–140 $\mu\text{mol l}^{-1}$
334 h^{-1}). Additional assays with solutions of artificial sea salts and sodium bicarbonate in Elga
335 water showed that the L-DOPA oxidation rate increases approximately linearly with salinity
336 up to $\sim 100 \mu\text{mol L-DOPA l}^{-1} \text{ h}^{-1}$ (salinity 44), and that substrate oxidation is observed above
337 pH ~ 7 (Figure 3g,h), similar to the trend observed in our samples from Sarawak.



339 **Figure 3.** Substrate oxidation rate (phenol oxidase assay) in: (a,b) Water samples (substrate oxidation rate),
 340 autoclaved controls (control oxidation rate) collected from Sarawak, Borneo during surveys in March and
 341 September 2017. Comparisons of unfiltered water, autoclaved samples and 3KDa ultrafiltration of water from the
 342 (c) Johor Strait, (Singapore, salinity 31), (d) Singapore Strait (Singapore, salinity 31), (e,f) two depth profiles (0-
 343 500m) from the North Pacific Subtropical Gyre. (g) Artificial seawater. (h) Deionized water with pH manipulation
 344 (NaHCO_3 addition). Data show means \pm 1 standard deviation of (c,d,e) 3 replicate measurements of one water
 345 sample; (f) 8 individual water samples from differing depths.

346

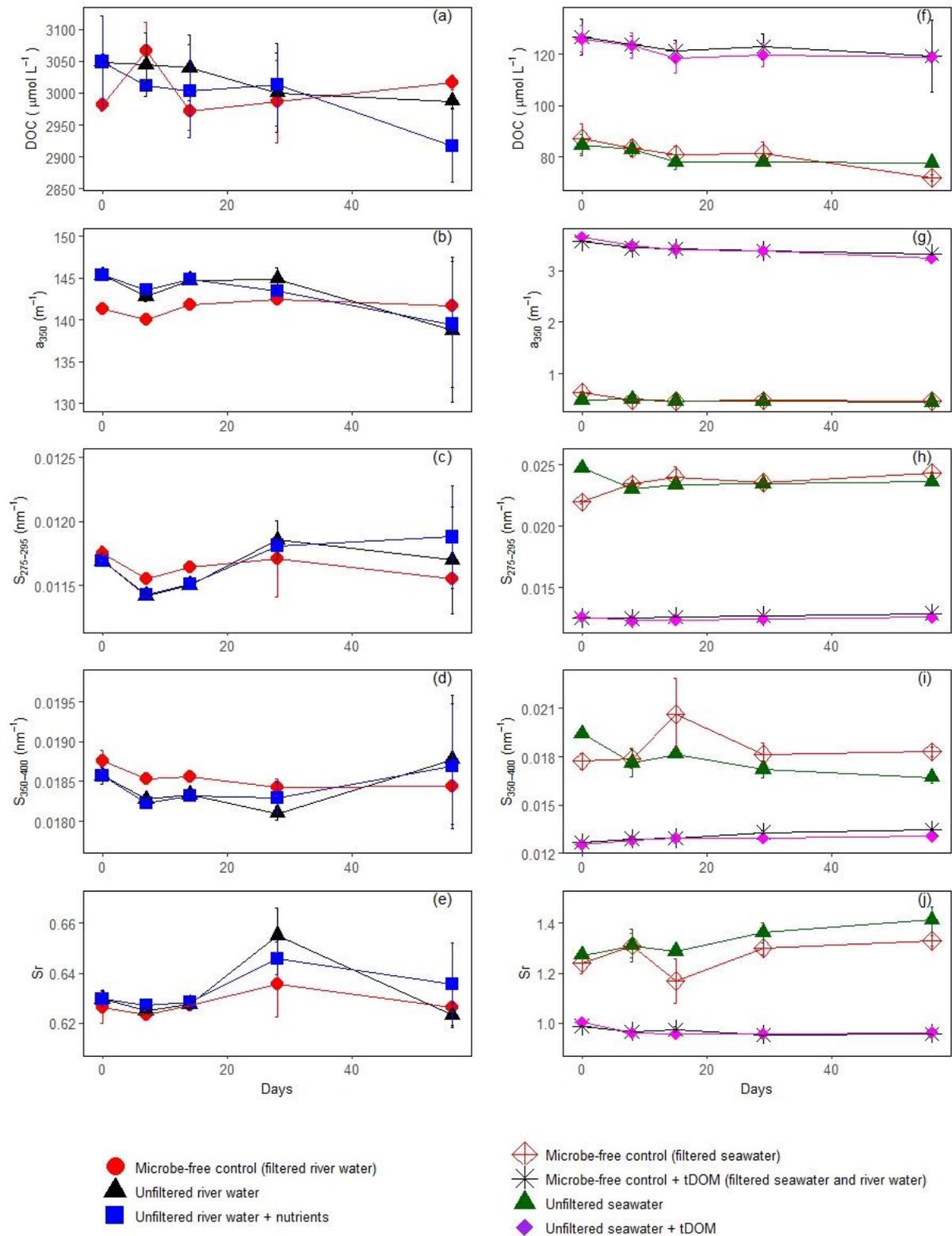
347 3.3 Biodegradation of tDOC

348 Both incubation experiments showed few significant changes in DOC and CDOM parameters,
 349 and where these changes were significant, they were small. Statistical parameters for those
 350 treatments and measurements that did show significant changes are given in Table S2.
 351 Incubations of Maludam river water with and without nutrient addition showed minor DOC
 352 variation in treatment means by $<90 \mu\text{mol l}^{-1}$, corresponding to $<3.5\%$ (Figure 4a). The data
 353 could not be fit with exponential decay curves. Linear regressions of mean DOC over time
 354 were non-significant in the filtered controls, but showed small significant decreases in
 355 unfiltered and unfiltered + nutrients river water treatments (Figure 4a). Corresponding to total
 356 tDOC decreases over 56 days by 0–4.4% the untreated river water ($61 \pm 73 \mu\text{mol l}^{-1}$ DOC), and
 357 of 2.5–6% in the river water with added nutrients ($131 \pm 53 \mu\text{mol l}^{-1}$ DOC) (Figure 4a);
 358 calculated as the mean change in DOC after 56 days \pm the combined sd of the three initial and
 359 three final replicates for each treatment.

360 No significant changes in CDOM parameters were observed in the unfiltered or filtered river
 361 water. The nutrient-amended treatment showed a significant decrease in a_{350} over 56 days (by
 362 5.9 m^{-1} , or 4%), but CDOM spectral slopes and slope ratios did not show any significant change
 363 (Figure 4b–e).

364 Incubations in which tDOC-rich Maludam river water was mixed with coastal seawater showed
 365 a small decrease in DOC in both treatments, by $6.8 \pm 2.5 \mu\text{mol l}^{-1}$ (8%) in the unamended
 366 seawater and by $7 \pm 3.0 \mu\text{mol l}^{-1}$ (5.5%) in the seawater treatment with added Maludam river
 367 water, Figure 4f). However, neither linear nor exponential regressions of DOC over time were

368 significant. This indicates that the added Maludam tDOC was not being remineralized. A small,
369 significant linear decrease in a_{350} was observed in both the unfiltered + tDOM and filtered
370 control + tDOM treatments, amounting to $0.413 \pm 0.02 \text{ m}^{-1}$ (11% decrease) and 0.253 ± 0.018
371 m^{-1} (7% decrease), respectively (Figure 4g). A much smaller (but significant) decrease in a_{350}
372 was observed in the unfiltered seawater without added Maludam river water ($0.04 \pm 0.01 \text{ m}^{-1}$),
373 with no significant change in the filtered seawater control. Little overall change was observed
374 in slope parameters during incubations. No significant changes were observed in $S_{275-295}$ except
375 for a small significant increase in the filtered seawater controls (Figure 4h). In the unamended
376 seawater $S_{350-400}$ decreased slightly but not significantly, and increased very slightly (but
377 significantly) in the filtered and unfiltered treatments with added Maludam river water (Figure
378 4i). A significant increase in slope ratio was observed in the unfiltered seawater treatment,
379 which was not observed in the other treatments and controls (Figure 4j).



380

381 **Figure 4.** Biodegradation incubation experiments of tDOC rich river water collected from the peat draining
 382 Maludam river, Sarawak, Borneo. Maludam river water, with and without the addition of nutrients (Left column,
 383 a-e), and seawater with and without the addition of Maludam river water (right column, f-j). Results shown are
 384 dissolved organic carbon, absorption coefficient at 350 nm (a_{350}), spectral slopes $S_{275-295\text{nm}}$ and $S_{350-400\text{nm}}$, and
 385 the slope ratio of $S_{275-295\text{nm}}:S_{350-400\text{nm}}$ (Sr). In all incubations, filtered controls represent microbe free controls whereas
 386 unfiltered water contains the natural microbial community. Nutrient additions consisted of $5 \mu\text{mol L}^{-1}$ nitrate and

387 1 $\mu\text{mol L}^{-1}$ phosphate. All data are mean \pm 1 standard deviation of three replicate incubations. Significant
388 regressions are detailed in Table S2.

389

390 **4 Discussion**

391 4.1 Leucine aminopeptidase (LAP) activity

392 Leucine aminopeptidase is typically associated with the activity of heterotrophic bacteria
393 degrading organic matter for protein synthesis (Kirchman et al., 1985). Our data therefore
394 suggest that heterotrophic microbial activity was lower in blackwater rivers than the less DOC-
395 rich Rajang River, and highest in the estuaries. Estuaries are generally recognized as areas of
396 high biogeochemical activity (Cai, 2011), so this result was not unexpected. The higher LAP
397 activity in the Rajang River compared to the blackwater rivers is likely a consequence of the
398 differences in river chemistry promoting higher microbial activity in the Rajang, with higher
399 pH and lower DOC concentrations than blackwater rivers (Martin et al., 2018) and high
400 dissolved inorganic nitrogen levels likely due to anthropogenic inputs (Jiang et al., 2019).
401 Conversely, chlorophyll concentrations were relatively low at all marine stations (on average
402 around $1 \mu\text{g l}^{-1}$ and mostly $<2 \mu\text{g l}^{-1}$; Martin et al., 2018), and LAP activities were consistently
403 low at high salinities. This indicates that LAP is likely an accurate index of relative variation
404 in heterotrophic microbial activity across the study region. Crucially, however, this enzyme
405 plays no role in the degradation of lignin or other phenolic molecules and therefore these results
406 do not indicate biodegradation of peatland derived tDOC. The fact that we observed a
407 significant relationship between LAP and chlorophyll-*a* across the entire dataset indicates
408 instead that LAP activity was most likely related to the degradation of autochthonous organic
409 matter, and therefore does not imply that microbial processing of tDOC was taking place. This
410 is consistent with the interpretation by Sieczko & Peduzzi, (2014) that LAP provides an index
411 of the processing of labile DOM but not of more refractory, terrigenous DOM.

412 4.2 Phenol oxidase assay methodological limitations

413 Our POx results clearly suggest that this assay did not return data that reflect microbial tDOC
414 remineralization across the region. It could be tempting to interpret the trend in substrate
415 oxidation rate with salinity as evidence in support of the hypothesis that dilution of phenol
416 concentration and increasing pH over the salinity gradient promote POx activity and tDOC
417 remineralization (Sinsabaugh, 2010; Williams et al., 2000). However, because the oxidation
418 rate in our autoclaved controls accounted for most of the substrate oxidation rate, these data
419 cannot be interpreted as representing enzymatic activity. Moreover, we observed very similar
420 rates of L-DOPA oxidation in the marine samples from Sarawak, from Singapore, and from
421 the surface down to 500 m depth in the North Pacific Subtropical Gyre, which are
422 biogeochemically very different environments, and, in the case of the depth profiles, span a
423 large gradient in expected microbial activity. This further indicates that our POx assay results
424 cannot be interpreted as representing enzymatic activity, since we would expect far more
425 variable oxidation rates across such different sampling locations.

426 The additional experiments we conducted clearly demonstrated that the L-DOPA oxidation
427 rate in POx assays is highly sensitive to pH and ionic concentration. Significant auto-oxidation
428 of L-DOPA at alkaline pH was also reported by two previous studies (Tahvanainen &
429 Haraguchi, 2013; Zhou et al., 2012). Autoclaving causes significant changes to solution
430 chemistry: although autoclaved seawater usually becomes more alkaline because CO₂ is
431 released (Harrison & Berges, 2005), autoclaving of soil extracts can release organic acids and
432 result in a more acidic pH (Skipper & Westermann, 1973). In our coastal water samples, the
433 presence of varying quantities of organic and inorganic particulate matter of terrestrial and
434 aquatic origin may either have led to small decreases in pH or otherwise changed the solution
435 chemistry in a way that led to small reductions in L-DOPA oxidation rate in most controls.
436 Consequently, we conclude that any POx activity rate we could calculate from our data is

437 unlikely to reflect real enzymatic activity, and is more likely the result of small changes in auto-
438 oxidation rate between samples and controls.

439 Although POx assays can also be conducted with the alternative substrates pyrogallol and
440 ABTS, L-DOPA is much more commonly used because it has a more suitable redox potential
441 and can be used over a much wider pH range (Bach et al., 2013). Thus, there are no alternative
442 substrates known at present that could overcome the limitations of L-DOPA and be used over
443 the range of chemical gradients that are found across the land–ocean aquatic continuum.

444 Our data do not preclude the possibility that genuine microbial POx activity might be
445 measurable in aquatic samples from other regions using L-DOPA. However, our results clearly
446 show that great care must be taken to ensure that such assays are not confounded by the auto-
447 oxidation of L-DOPA. Moreover, because autoclaving alters solution chemistry in a way that
448 is likely to influence the L-DOPA auto-oxidation rate, we would strongly recommend that
449 ultrafiltered controls are used. Because POx are larger than ~40 kDa (Dean & Eriksson, 1994;
450 Van Gelder et al., 1997; Goulart et al., 2003; Thurston 1994; Weemaes et al., 1998),
451 ultrafiltration through a suitably small pore size can generate enzyme-free controls without
452 altering the L-DOPA auto-oxidation rate.

453 We would therefore recommend that POx activities should also be measured in regions where
454 rapid rates of tDOC biodegradation have been observed. It is possible that in such
455 environments, real microbial POx activity could significantly overwhelm the auto-oxidation
456 rate, in which case L-DOPA-based POx assays could prove to be valuable after all for tracing
457 tDOC biodegradation in aquatic environments. However, we recommend that ultrafiltration
458 should be used to generate enzyme-free controls without altering the water chemistry in a way
459 that could affect the L-DOPA auto-oxidation rate.

460 4.3 Lack of peatland tDOC biodegradation in Sarawak

461 The fact that we only observed 0–4.4% DOC loss in river water incubations over 56 days at
462 around 30°C indicates very low biodegradability of tDOM from an intact tropical peatland. In
463 comparison, around 5–20% of DOC in temperate peat-draining rivers is typically biodegradable
464 over time-scales of 5–55 days at incubation temperatures of 10–22°C (Asmala et al., 2014;
465 Fovet et al., 2020; Hulatt et al., 2014; Stutter et al., 2013). Biodegradability of tDOC is
466 determined not only by inherent chemical characteristics but also by environmental constraints
467 (Guggenberger et al., 2011; Kleber, 2010). In Southeast Asian peat-draining rivers, these
468 constraints might include low concentrations of nutrients (Alkhatib et al., 2007; Bange et al.,
469 2019; Gandois et al., 2020; Wickland et al., 2012) and low pH due to organic acids (Borges et
470 al., 2015; Müller et al., 2015). However, our nutrient-amended treatment showed a maximum
471 of 6% DOC loss, indicating that microbial degradation of peatland tDOM in our region is not
472 simply limited by nutrients. Moreover, addition of tDOM-rich Maludam River water to coastal
473 seawater did not result in any excess DOC loss compared to the unamended seawater
474 treatments, with only $\sim 7 \mu\text{mol l}^{-1}$ lost in both cases, indicating that Maludam river tDOM does
475 not become more biodegradable when mixed with coastal seawater and microbes. A linear
476 decrease in CDOM concentration (a_{350}) was observed in both treatments and filtered controls
477 where Maludam River tDOM was added to coastal seawater, and $S_{350-400}$ showed a very small
478 increase typically associated with a shift towards lower molecular weight and/or decreasing
479 aromaticity (Hansen, et al., 2016). Because these changes occurred in both unfiltered and
480 filtered treatments, this possibly reflects slow, abiotic transformations of CDOM after mixing
481 with seawater, although this was clearly not associated with a decrease in the added tDOC.

482 Porewater DOC in Southeast Asian peatlands likely undergoes significant processing by the
483 soil microbial community, resulting in DOC that has a very young radiocarbon age (~ 10 years)
484 but already shows chemical and optical characteristics of being highly degraded by the time it
485 enters rivers (Gandois et al., 2014; Müller et al., 2015; Zhou et al., 2019). Such pre-degradation

486 would likely reduce the biodegradability of the DOC that is ultimately exported to rivers and
487 coastal seas. A lack of biodegradability was also recently reported for DOC in surface waters
488 from peat bogs in the permafrost zone of Siberia (Shirokova et al., 2019).

489 Our data are consistent with the predominantly conservative mixing pattern of tDOC that has
490 been reported for peat-draining rivers in Southeast Asia (Alkhatib et al., 2007; Baum et al.,
491 2007; Martin et al., 2018; Zhou et al., 2019), as well as with the absence of measurable POx
492 enzyme activity across our study region in Sarawak. However, these results contrast with
493 reports of consistent outgassing of CO₂ from peat-draining rivers in Southeast Asia, including
494 the Maludam (Müller et al., 2015; Müller-Dum et al., 2019; Wit et al., 2015). A recent analysis
495 of the riverine and coastal carbon budget off Sumatra also concluded that around 60% of the
496 total carbon exported by rivers is emitted as CO₂ in the estuaries and coastal ocean, which was
497 attributed to respiration (Wit et al., 2018). This apparent discrepancy between our results and
498 CO₂ outgassing measurements could potentially be explained by lateral transport of CO₂ and
499 CH₄ from peat pore water into rivers (Clymo & Pearce, 1995; Johnson et al., 2008; Jones &
500 Mulholland, 1998), and subsequent methanotrophy consuming the appreciable methane
501 concentrations in these rivers (Bange et al., 2019). Moreover, tDOC from Borneo has been
502 shown to be highly photolabile: filtered blackwater samples exposed for five days to natural
503 sunlight lost up to 26% of DOC (Martin et al., 2018), and fluorescence spectra from a
504 blackwater river system in Borneo were also interpreted recently as showing evidence of
505 photodegradation (Gandois et al., 2020). To our knowledge, the only other tDOC degradation
506 experiment in Southeast Asia exposed unfiltered water from a Sumatran blackwater river to
507 natural sunlight and reported that ~27% of DOC was labile to combined photo- and
508 biodegradation, mostly within 8 days (Rixen et al., 2008). However, the relative importance of
509 photo- *versus* biodegradation was not determined, and it is possible that these results primarily
510 reflect photo-chemical remineralization.

511 It is possible that biodegradation of tDOC in Southeast Asia takes place after partial
512 photodegradation: it is now widely recognized that photodegradation can render DOC more
513 labile to biodegradation (Cory & Kling, 2018; Ward et al., 2017). Overall, our data suggest that
514 photodegradation may be more likely to act as the main control over peatland tDOC
515 remineralization in coastal waters of Southeast Asia, as has been shown for carbon processing
516 in freshwater systems in the Arctic (Cory et al., 2014; Bowen et al., 2020).

517 **5 Conclusions**

518 We found that the commonly used phenol oxidase enzyme assay substrate L-DOPA undergoes
519 significant auto-oxidation at typical environmental pH values in estuaries and coastal seas,
520 which could significantly confound results unless adequate care is taken to prepare appropriate
521 enzyme-free controls. Our data suggest that autoclaved samples are not appropriate controls
522 because autoclaving can change sample chemistry; we therefore recommend that ultrafiltration
523 is necessary. We conclude that there was no measurable POx activity across our study region
524 in Sarawak, but recommend that this assay should also be tested (with ultrafiltered controls) in
525 aquatic environments where substantial tDOC biodegradation is anticipated. LAP data
526 indicated that microbial heterotrophic activity follows the expected trends with low activity in
527 blackwater rivers and highest activities in estuaries. However, this most likely reflects
528 heterotrophic processing of autochthonous organic matter, not tDOC biodegradation. Our
529 incubation experiments further demonstrated that tDOC from one of the few pristine peat-
530 draining rivers in Southeast Asia was largely refractory to biodegradation over eight weeks at
531 in-situ temperatures, even after adding nutrients or diluting the DOC with coastal seawater. We
532 surmise that the remineralization of Southeast Asian peatland DOC in the aquatic environment
533 is likely dependent on photodegradation.

534

535

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558 **References**

- 559 Aarnos, H., Gélinas, Y., Kasurinen, V., Gu, Y., Puupponen, V-M., & Vähätalo, A. V. (2018). Photochemical
 560 Mineralization of Terrigenous DOC to Dissolved Inorganic Carbon in Ocean. *Global Biogeochemical Cycles*,
 561 32(2), 250–266. <https://doi.org/10.1002/2017GB005698>
- 562 Alexander, M., (1965). Biodegradation: problems of molecular recalcitrance and microbial fallibility. *Advances*
 563 *in Applied Microbiology* (Vol. 7, pp. 35–80). Elsevier.
- 564 Alkhatib, M., Jennerjahn, T. C. & Samiaji, J. (2007). Biogeochemistry of the Dumai River estuary, Sumatra,
 565 Indonesia, a tropical black-water river. *Limnology and Oceanography*, 52(6), 2410–2417.
 566 <https://doi.org/10.4319/lo.2007.52.6.2410>
- 567 Allison, S. D. & Vitousek, P. M. (2004). Extracellular enzyme activities and carbon chemistry as drivers of
 568 tropical plant litter decomposition. *Biotropica*, 36(3), 285–296. <https://doi.org/10.1111/j.1744-7429.2004.tb00321.x>
- 570 Arnosti, C., Bell, C., Moorhead, D. L., Sinsabaugh, R. L., Steen, A. D., Stromberger, M., et al. (2014).
 571 Extracellular enzymes in terrestrial, freshwater, and marine environments: Perspectives on system variability and
 572 common research needs. *Biogeochemistry*, 117(1), 5–21. <https://doi.org/10.1007/s10533-013-9906-5>

- 573 Asmala, E., Autio, R., Kaartokallio, H., Stedmon, C. A. & Thomas D N (2014). Processing of humic-rich riverine
574 dissolved organic matter by estuarine bacteria: effects of predegradation and inorganic nutrients. *Aquatic Sciences*,
575 76(3), 451–463. <https://doi.org/10.1007/s00027-014-0346-7>
- 576 Bach, C. E., Warnock, D. D., Van Horn, D. J., Weintraub, M. N., Sinsabaugh, R. L., Allison, S. D. & German, D.
577 P. (2013). Measuring phenol oxidase and peroxidase activities with pyrogallol, 1-DOPA, and ABTS: Effect of
578 assay conditions and soil type. *Soil Biology and Biochemistry*, 67, 183–191.
579 <https://doi.org/10.1016/j.soilbio.2013.08.022>
- 580 Bange, H. W., Sim, C. H., Bastian, D., Kallert, J., Kock, A., Mujahid, A. & Müller, M. (2019). Nitrous oxide
581 (N₂O) and methane (CH₄) in rivers and estuaries of northwestern Borneo. *Biogeosciences*, 16(22), 4321–4335.
582 <https://doi.org/10.5194/bg-16-4321-2019>
- 583 Bauer, J. E., Cai, W.-J., Raymond, P. A., Bianchi, T. S., Hopkinson, C. S. & Regnier, P. A. G. (2013). The changing
584 carbon cycle of the coastal ocean. *Nature*, 504(7478), 61. <https://doi.org/10.1038/nature12857>
- 585 Baum, A. & Rixen, T. (2014). Dissolved inorganic nitrogen and phosphate in the human affected blackwater river
586 Siak, central Sumatra, Indonesia. *Asian Journal of Water, Environment and Pollution*, 11(1), 13–24.
- 587 Baum, A., Rixen, T. & Samiaji, J. (2007). Relevance of peat draining rivers in central Sumatra for the riverine
588 input of dissolved organic carbon into the ocean. *Estuarine, Coastal and Shelf Science*, 73(3–4), 563–570.
589 <https://doi.org/10.1016/j.ecss.2007.02.012>
- 590 Bianchi, T. S. (2011). The role of terrestrially derived organic carbon in the coastal ocean: A changing paradigm
591 and the priming effect. *Proceedings of the National Academy of Sciences*, 108(49), 19473–19481.
592 <https://doi.org/10.1073/pnas.1017982108>
- 593 Borges, A. V., Darchambeau, F., Teodoru, C. R., Marwick, T. R., Tamooh, F., Geeraert, N., et al. (2005). Do we
594 have enough pieces of the jigsaw to integrate CO₂ fluxes in the coastal ocean? *Estuaries*, 28(1), 3–27.
595 <https://doi.org/10.1007/BF02732750>
- 596 Borges, A. V., Darchambeau, F., Teodoru, C. R., Marwick, T. R., Tamooh, F., Geeraert, N., et al. (2015). Globally
597 significant greenhouse-gas emissions from African inland waters. *Nature Geoscience*, 8(8), 637.
598 <https://doi.org/10.1038/ngeo2486>
- 599 Bowen, J. C., Ward, C. P., Kling, G. W., & Cory, R. M. (2020). Arctic amplification of global warming
600 strengthened by sunlight oxidation of permafrost carbon to CO₂. *Geophysical Research Letters*, 47,
601 e2020GL087085. <https://doi.org/10.1029/2020GL087085>
- 602 Cai, W.-J. (2011). Estuarine and coastal ocean carbon paradox: CO₂ sinks or sites of terrestrial carbon incineration?
603 *Annual Review of Marine Science*, 3, 123–145. <https://doi.org/10.1146/annurev-marine-120709-142723>
- 604 Carreiro, M. M., Sinsabaugh, R. L., Repert, D. A. & Parkhurst, D. F. (2000). Microbial Enzyme Shifts Explain
605 Litter Decay Responses To Simulated Nitrogen Deposition. *Ecology*, 81(9), 2359–2365.
606 [https://doi.org/10.1890/0012-9658\(2000\)081\[2359:MESELD\]2.0.CO;2](https://doi.org/10.1890/0012-9658(2000)081[2359:MESELD]2.0.CO;2)
- 607 Ciais, P., Sabine, C., Bala, G., Bopp, L., Brovkin, V. & Canadell, J. (2013). Carbon and Other Biogeochemical
608 Cycles In: Stocker T F, Qin D, Plattner G K, Tignor M, Allen S K, Boschung J et al. *Climate Change 2013: The*
609 *Physical Science Basis Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental*
610 *panel on climate change*. Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA.
- 611 Clymo, R. S. & Pearce, D. M. E. (1995). Methane and carbon dioxide production in, transport through, and efflux
612 from a peatland. *Philosophical Transactions of the Royal Society of London. Series A: Physical and Engineering*
613 *Sciences*, 351(1696), 249–259. <https://doi.org/10.1098/rsta.1995.0032>
- 614 Cory, R. M. & Kling, G. W. (2018). Interactions between sunlight and microorganisms influence dissolved
615 organic matter degradation along the aquatic continuum. *Limnology and Oceanography Letters*, 3(3), 102–116.
616 <https://doi.org/10.1002/lol2.10060>
- 617 Cory, R. M., Ward, C. P., Crump, B. C. & Kling, G. W. (2014). Sunlight controls water column processing of
618 carbon in arctic fresh waters. *Science*, 345, 925–928. <https://doi.org/10.1126/science.1253119>

- 619 Beleites, C. & Sergo, V. (2020). hyperSpec: a package to handle hyperspectral data sets in R. R package version
620 0.99-20200527. <https://github.com/cbeleites/hyperSpec>
- 621 Dean, J. F. D. & Eriksson, K-E. L. (1994). Laccase and the deposition of lignin in vascular plants. *Holzforschung-*
622 *International Journal of the Biology, Chemistry, Physics and Technology of Wood*, 48(s1), 21–33.
623 <https://doi.org/10.1515/hfsg.1994.48.s1.21>
- 624 Dittmar, T., & Stubbins, A. (2014). 12.6 - Dissolved Organic Matter in Aquatic Systems (H. D. Holland & K. K.
625 B. T.-T. on G. (Second E. Turekian (eds.); pp. 125–156). *Elsevier*. [https://doi.org/https://doi.org/10.1016/B978-](https://doi.org/https://doi.org/10.1016/B978-0-08-095975-7.01010-X)
626 [0-08-095975-7.01010-X](https://doi.org/10.1016/B978-0-08-095975-7.01010-X)
- 627 Dommain, R., Couwenberg, J., Glaser, P. H., Joosten, H. & Suryadiputra, I. N. N. (2014). Carbon storage and
628 release in Indonesian peatlands since the last deglaciation. *Quaternary Science Reviews*, 97, 1–32.
629 <https://doi.org/10.1016/j.quascirev.2014.05.002>
- 630 Fichot, C. G. & Benner, R. (2014). The fate of terrigenous dissolved organic carbon in a river-influenced ocean
631 margin. *Global Biogeochemical Cycles*, 28(3), 300–318. <https://doi.org/10.1002/2013GB004670>
- 632 Fovet, O., Cooper, D. M., Jones, D. L., Jones, T. G. & Evans, C. D. (2020). Dynamics of dissolved organic matter
633 in headwaters: comparison of headwater streams with contrasting DOM and nutrient composition. *Aquatic*
634 *Sciences*, 82(2), 29. <https://doi.org/10.1007/s00027-020-0704-6>
- 635 Freeman, C., Ostle, N. & Kang, H. (2001). An enzymic “latch” on a global carbon store. *Nature*, 409(6817), 149.
636 <https://doi.org/10.1038/35051650>
- 637 Freeman, C., Ostle, N. J., Fenner, N. & Kang, H. (2004). A regulatory role for phenol oxidase during
638 decomposition in peatlands. *Soil Biology and Biochemistry*, 36(10), 1663–1667.
639 <https://doi.org/10.1016/j.soilbio.2004.07.012>
- 640 Follett, C. L., Repeta, D. J., Rothman, D. H., Xu, L., & Santinelli, C. (2014). Hidden cycle of dissolved organic
641 carbon in the deep ocean. *Proceedings of the National Academy of Sciences*, 111(47), 16706 LP – 16711.
642 <https://doi.org/10.1073/pnas.1407445111>
- 643 Gandois, L., Teisserenc, R., Cobb, A. R., Chieng, H. I., Lim, L. B. L., Kamariah, A. S., et al (2014). Origin,
644 composition, and transformation of dissolved organic matter in tropical peatlands. *Geochimica et Cosmochimica*
645 *Acta*, 137, 35–47. <https://doi.org/10.1016/j.gca.2014.03.012>
- 646 Gandois, L., Hoyt, A. M., Mounier, S., Le Roux, G., Harvey, C. F., Claustres, A., et al (2020). From canals to the
647 coast: dissolved organic matter and trace metal composition in rivers draining degraded tropical peatlands in
648 Indonesia. *Biogeosciences*, 17(7), 1897–1909. <https://doi.org/10.5194/bg-17-1897-2020>
- 649 Goulart, P. de FP., Alves, J. D., Magalhaes, M. M., Lima, L. C. O. & Meyer, L. E. (2003). Purification of
650 polyphenoloxidase from coffee fruits. *Food Chemistry*, 83(1), 7–11. [https://doi.org/10.1016/S0308-](https://doi.org/10.1016/S0308-8146(03)00030-X)
651 [8146\(03\)00030-X](https://doi.org/10.1016/S0308-8146(03)00030-X)
- 652 Green, S. A., and Blough, N. V. (1994). Optical absorption and fluorescence properties of chromophoric dissolved
653 organic matter in natural waters. *Limnology and Oceanography*, 39, 1903–1916.
654 <https://doi.org/10.4319/lo.1994.39.8.1903>
- 655 Guggenberger, G., Janssens, I. A., Kleber, M., Kgel-Knabner, I., Lehmann, J., Manning, D. A. C., et al. (2011).
656 Persistence of soil organic matter as an ecosystem property. *Nature*, 478, 4956.
657 <https://doi.org/10.1038/nature10386>
- 658 Hansen, A. M., Kraus, T. E. C., Pellerin, B. A., Fleck, J. A., Downing, B. D., & Bergamaschi, B. A. (2016).
659 Optical properties of dissolved organic matter (DOM): Effects of biological and photolytic degradation.
660 *Limnology and Oceanography*, 61(3), 1015–1032. <https://doi.org/10.1002/lno.10270>
- 661 Harrison, P. J. & Berges, J. A. (2005). Marine culture media. *Algal Culturing Techniques*, 21–34.

- 662 Helms, J. R., Stubbins, A., Ritchie, J. D., Minor, E. C., Kieber, D. J., and Mopper, K. (2008). Absorption spectral
663 slopes and slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric dissolved
664 organic matter. *Limnology and Oceanography*, 53, 955-969. <https://doi.org/10.4319/lo.2008.53.3.0955>
- 665 Hodgkins, S. B., Richardson, C. J., Dommain, R., Wang, H., Glaser, P. H., Verbeke, B., et al. (2018). Tropical
666 peatland carbon storage linked to global latitudinal trends in peat recalcitrance. *Nature Communications*, 9(1),
667 3640. <https://doi.org/10.1038/s41467-018-06050-2>
- 668 Hoppe, H-G. (1991). Microbial extracellular enzyme activity: a new key parameter in aquatic ecology. *Microbial*
669 *enzymes in aquatic environments* (pp. 60–83). Springer.
- 670 Hoppe, H-G. (1993). Use of fluorogenic model substrates for extracellular enzyme activity (EEA) measurement
671 of bacteria. *Handbook of Methods in Aquatic Microbial Ecology*, 423–431.
- 672 Hulatt, C. J., Kaartokallio, H., Asmala, E., Autio, R., Stedmon, C. A., Sonninen, E., et al. (2014). Bioavailability
673 and radiocarbon age of fluvial dissolved organic matter (DOM) from a northern peatland-dominated catchment:
674 effect of land-use change. *Aquatic Sciences*, 76(3), 393–404. <https://doi.org/10.1007/s00027-014-0342-y>
- 675 Jiang, S., Müller, M., Jin, J., Wu, Y., Zhu, K., Zhang, G., et al. (2019). Dissolved inorganic nitrogen in a tropical
676 estuary in Malaysia: transport and transformation. *Biogeosciences*, 16(14), 2821–2836.
677 <https://doi.org/10.5194/bg-16-2821-2019>.
- 678 Johnson, M. S., Lehmann, J., Riha, S. J., Krusche, A. V., Richey, J. E., Ometto, J. P. H. B. & Couto, E. G. (2008).
679 CO₂ efflux from Amazonian headwater streams represents a significant fate for deep soil respiration. *Geophysical*
680 *Research Letters*, 35, L17401, doi:10.1029/2008GL034619.
- 681 Jones, J. B., Mulholland, P. J. (1998). Methane input and evasion in a hardwood forest stream: Effects of
682 subsurface flow from shallow and deep pathway. *Limnology and Oceanography*, 6, doi:
683 10.4319/lo.1998.43.6.1243.
- 684 Kaiser, K., Benner, R. & Amon, R. M. W. (2017). The fate of terrigenous dissolved organic carbon on the Eurasian
685 shelves and export to the North Atlantic. *Journal of Geophysical Research: Oceans*, 122(1), 4–22.
686 <https://doi.org/10.1002/2016JC012380>
- 687 Kartadikaria, A. R., Watanabe, A., Nadaoka, K., Adi, N. S., Prayitno, H. B., Soemorumekso, S., et al. (2015). CO₂
688 sink/source characteristics in the tropical Indonesian seas. *Journal of Geophysical Research: Oceans*, 120(12),
689 7842–7856. <https://doi.org/10.1002/2015JC010925>
- 690 Kirchman, D., K'nees, E. & Hodson, R. (1985). Leucine incorporation and its potential as a measure of protein
691 synthesis by bacteria in natural aquatic systems. *Applied and Environmental Microbiology*, 49(3), 599–607.
- 692 Kleber, M. (2010). What is recalcitrant soil organic matter? *Environmental Chemistry*, 7(4), 320–332.
693 <https://doi.org/10.1071/EN10006>
- 694 Laruelle, G. G., Lauerwald, R., Pfeil, B. & Regnier, P. (2014). Regionalized global budget of the CO₂ exchange
695 at the air-water interface in continental shelf seas. *Global Biogeochemical Cycles*, 28(11), 1199–1214.
696 <https://doi.org/10.1002/2014GB004832>
- 697 Mann, P. J., Sobczak, W. V., Larue, M. M., Bulygina, E., Davydova, A., Vonk, J. E., et al. (2014). Evidence for
698 key enzymatic controls on metabolism of Arctic river organic matter. *Global Change Biology*, 20(4), 1089–1100.
699 <https://doi.org/10.1111/gcb.12416>
- 700 Martin, P., Cherukuru, N., Tan, A. S. Y., Sanwlani, N., Mujahid, A. & Müller, M. (2018). Distribution and cycling
701 of terrigenous dissolved organic carbon in peatland-draining rivers and coastal waters of Sarawak, Borneo.
702 *Biogeosciences*, 15(22), 6847--6865. <https://doi.org/10.5194/bg-15-6847-2018>
- 703 Mazzon, M., Cavani, L., Margon, A., Sorrenti, G., Ciavatta, C., & Marzadori, C. (2018). Changes in soil phenol
704 oxidase activities due to long-term application of compost and mineral N in a walnut orchard. *Geoderma*, 316,
705 70–77. <https://doi.org/https://doi.org/10.1016/j.geoderma.2017.12.009>

- 706 Mitchell, G., Bricaud, A., Carder, K., Cleveland, J., Ferrari, G., Gould, R., Kahru, M., Kishino, M., Maske, H.,
707 Moisan, T., Moore, L., Nelson, N., Phinney, D., Reynolds, R., Sosik, H., Stramski, D., Tassan, S., Trees, C. C.,
708 Weidemann, A., Wieland, J. and Vodacek, A. (2000). Determination of spectral absorption coefficients of
709 particles, dissolved material and phytoplankton for discrete water samples, in: *Ocean Optics Protocols for Satellite*
710 *Ocean Color Sensor Validation, Revision 2*, edited by: Fargion G S, and Mueller J L, National Aeronautical and
711 Space Administration, Greenbelt, Maryland, 125-153.
- 712 Mohammad Razi, M. A., Mokhtar, A., Mahamud, M., Rahmat, S. N., & Al-Gheethi, A. (2020). Monitoring of
713 river and marine water quality at Sarawak baseline. *Environmental Forensics*, 1–22.
714 <https://doi.org/10.1080/15275922.2020.1836076>
- 715 Moore, S., Gauci, V., Evans, C. D., Page, S. E., Hall, W., Keynes, M., et al. (2011). Fluvial organic carbon losses
716 from a Bornean blackwater river. *Biogeosciences*, 8(4), 901. <https://doi.org/10.5194/bg-8-901-2011>
- 717 Moore, S., Evans, C. D., Page, S. E., Garnett, M. H., Jones, T. G., Freeman, C., et al. (2013). Deep instability of
718 deforested tropical peatlands revealed by fluvial organic carbon fluxes. *Nature*, 493(7434), 660.
719 <https://doi.org/10.1038/nature11818>
- 720 Müller-Dum, D., Warneke, T., Rixen, T., Müller, M., Baum, A., Christodoulou, A., et al. (2019). Impact of
721 peatlands on carbon dioxide (CO₂) emissions from the Rajang River and Estuary, Malaysia. *Biogeosciences*,
722 16(1), 17–32. <https://doi.org/10.5194/bg-16-17-2019>
- 723 Müller, D., Warneke, T., Rixen, T., Müller, M., Jamahari, S., Denis, N., et al. (2015). Lateral carbon fluxes and
724 CO₂ outgassing from a tropical peat-draining river. *Biogeosciences*, 12(20), 5967–5979.
725 <https://doi.org/10.5194/bg-12-5967-2015>
- 726 Müller, D., Warneke, T., Rixen, T., Müller, M., Mujahid, A., Bange, H. W. & Notholt, J. (2016). Fate of terrestrial
727 organic carbon and associated CO₂ and CO emissions from two Southeast Asian estuaries. *Biogeosciences*, 13,
728 691–705. <https://doi.org/10.5194/bg-13-691-2016>
- 729 Opsahl, S. & Benner, R. (1997). Distribution and cycling of terrigenous dissolved organic matter in the ocean.
730 *Nature*, 386(6624), 480. <https://doi.org/10.1038/386480a0>
- 731 Page, S. E., Rieley, J. O. & Banks, C. J. (2011). Global and regional importance of the tropical peatland carbon
732 pool. *Global Change Biology*, 17(2), 798–818. <https://doi.org/10.1111/j.1365-2486.2010.02279.x>
- 733 Prescott, C. E. (2010). Litter decomposition: what controls it and how can we alter it to sequester more carbon in
734 forest soils? *Biogeochemistry*, 101(1–3), 133–149. <https://doi.org/10.1007/s10533-010-9439-0>
- 735 R Core Team (2020) R: A language and environment for statistical computing. R Foundation for Statistical
736 Computing, Vienna, Austria. <https://www.R-project.org/>
- 737 Rixen, T., Baum, A., Pohlmann, T., Balzer, W., Samiaji, J. & Jose, C. (2008). The Siak, a tropical black water
738 river in central Sumatra on the verge of anoxia. *Biogeochemistry*, 90(2), 129–140. <https://doi.org/10.1007/s10533-008-9239-y>
- 740 Sa'adi, Z., Shahid, S., Ismail, T., Chung, E-S. & Wang, X-J. (2017) Distributional changes in rainfall and river
741 flow in Sarawak, Malaysia. *Asia-Pacific Journal of Atmospheric Sciences*, 53(4), 489–500.
742 <https://doi.org/10.1007/s13143-017-0051-2>
- 743 Sieczko, A. & Peduzzi, P. (2014). Origin, enzymatic response and fate of dissolved organic matter during flood
744 and non-flood conditions in a river-floodplain system of the Danube (Austria). *Aquatic Sciences*, 76(1), 115–129.
745 <https://doi.org/10.1007/s00027-013-0318-3>
- 746 Sinsabaugh, R. L. (2010). Phenol oxidase, peroxidase and organic matter dynamics of soil. *Soil Biology and*
747 *Biochemistry*, 42(3), 391–404. <https://doi.org/10.1016/j.soilbio.2009.10.014>
- 748 Sinsabaugh, R. L. & Shah, J. J. F. (2011). Ecoenzymatic stoichiometry of recalcitrant organic matter
749 decomposition: The growth rate hypothesis in reverse. *Biogeochemistry*, 102(1), 31–43.
750 <https://doi.org/10.1007/s10533-010-9482-x>

- 751 Sinsabaugh, R. L., Lauber, C. L., Weintraub, M. N., Ahmed, B., Allison, S. D., Crenshaw, C., et al. (2008).
752 Stoichiometry of soil enzyme activity at global scale. *Ecology Letters*, 11(11), 1252–1264.
753 <https://doi.org/10.1111/j.1461-0248.2008.01245.x>
- 754 Skipper, H. D. & Westermann, D. T. (1973). Comparative effects of propylene oxide, sodium azide, and
755 autoclaving on selected soil properties. *Soil Biology and Biochemistry*, 5(4), 409–414.
756 [https://doi.org/10.1016/0038-0717\(73\)90067-9](https://doi.org/10.1016/0038-0717(73)90067-9)
- 757 Sollins, P., Homann, P. & Caldwell, B. A. (1996). Stabilization and destabilization of soil organic matter:
758 mechanisms and controls. *Geoderma*, 74(1–2), 65–105. [https://doi.org/10.1016/S0016-7061\(96\)00036-5](https://doi.org/10.1016/S0016-7061(96)00036-5)
- 759 Stubbins, A., Mann, P. J., Powers, L., Bittar, T., Dittmar, T., McIntyre, C. P., et al. (2017). Low photolability of
760 yedoma permafrost dissolved organic carbon. *Journal of Geophysical Research: Biogeosciences*, 122, 200–211,
761 <https://doi.org/10.1002/2016JG003688>
- 762 Stursova, M. & Sinsabaugh, R. L. (2008). Stabilization of oxidative enzymes in desert soil may limit organic
763 matter accumulation. *Soil Biology and Biochemistry*, 40(2), 550–553.
764 <https://doi.org/10.1016/j.soilbio.2007.09.002>
- 765 Stutter, M. I., Richards, S. & Dawson, J. J. C. (2013). Biodegradability of natural dissolved organic matter
766 collected from a UK moorland stream. *Water Research*, 47(3), 1169–1180.
767 <https://doi.org/10.1016/j.watres.2012.11.035>
- 768 Tahvanainen, T. & Haraguchi, A. (2013). Effect of pH on phenol oxidase activity on decaying Sphagnum mosses.
769 *European Journal of Soil Biology*, 54, 41–47. <https://doi.org/10.1016/j.ejsobi.2012.10.005>
- 770 Thurston, C. F. (1994). The structure and function of fungal laccases. *Microbiology*, 140(1), 19–26.
771 <https://doi.org/10.1099/13500872-140-1-19>
- 772 Van Gelder, C. W. G., Flurkey, W. H. & Wichers, H. J. (1997). Sequence and structural features of plant and
773 fungal tyrosinases. *Phytochemistry*, 45(7), 1309–1323. [https://doi.org/10.1016/S0031-9422\(97\)00186-6](https://doi.org/10.1016/S0031-9422(97)00186-6)
- 774 Wang, H., Richardson, C. J., & Ho, M. (2015). Dual controls on carbon loss during drought in peatlands. *Nature*
775 *Climate Change*, 5(6), 584–587. <https://doi.org/10.1038/nclimate2643>
- 776 Ward, N. D., Keil, R. G., Medeiros, P. M., Brito, D. C., Cunha, A. C., Dittmar, T., et al. (2013). Degradation of
777 terrestrially derived macromolecules in the Amazon River. *Nature Geoscience*, 6(7), 530–533.
778 <https://doi.org/10.1038/ngeo1817>
- 779 Ward, C.P., Nalven, S.G., Crump, B.C. *et al.* Photochemical alteration of organic carbon draining permafrost soils
780 shifts microbial metabolic pathways and stimulates respiration. *Nat Commun* **8**, 772 (2017).
781 <https://doi.org/10.1038/s41467-017-00759-2>
- 782 Weemaes, C., Ludikhuyze, L., Van den Broeck, I. & Hendrickx, M. (1998). High pressure inactivation of
783 polyphenoloxidases. *Journal of Food Science*, 63(5), 873–877. <https://doi.org/10.1111/j.1365-2621.1998.tb17917.x>
- 785 Wickland, K. P., Aiken, G. R., Butler, K., Dornblaser, M. M., Spencer, R. G. M. & Striegl, R. G. (2012).
786 Biodegradability of dissolved organic carbon in the Yukon River and its tributaries: Seasonality and importance
787 of inorganic nitrogen. *Global Biogeochemical Cycles*, 26(4). <https://doi.org/10.1029/2012GB004342>
- 788 Wickham, H., Averick, M., Bryan, J., Chang, W., McGowan, L. D., François, R., et al. (2019). Welcome to the
789 Tidyverse. *Journal of Open Source Software*, 4(43), 1686. <https://doi.org/https://doi.org/10.21105/joss.01686>
- 790 Williams, C. J., Shingara, E. A. & Yavitt, J. B. (2000). Phenol oxidase activity in peatlands in New York State:
791 response to summer drought and peat type. *Wetlands*, 20(2), 416–421. [https://doi.org/10.1672/0277-5212\(2000\)020\[0416:POAIP\]2.0.CO;2](https://doi.org/10.1672/0277-5212(2000)020[0416:POAIP]2.0.CO;2)
- 793 Williams, P. M. & Druffel, E. R. M. (1987). Radiocarbon in dissolved organic matter in the central North Pacific
794 Ocean. *Nature*, 330, 246. <http://dx.doi.org/10.1038/330246a0>

- 795 Wit, F., Müller, D., Baum, A., Warneke, T., Pranowo, W. S., Müller, M. & Rixen, T. (2015). The impact of
796 disturbed peatlands on river outgassing in Southeast Asia. *Nature Communications*, 6, 10155.
797 <https://doi.org/10.1038/ncomms10155>
- 798 Wit, F., Rixen, T., Baum, A., Pranowo, W. S. & Hutahaean, A. A. (2018). The Invisible Carbon Footprint as a
799 hidden impact of peatland degradation inducing marine carbonate dissolution in Sumatra, Indonesia. *Scientific*
800 *Reports*, 8(1), 17403. <https://doi.org/10.1038/s41598-018-35769-7>
- 801 Yule, C. M., Lim, Y. Y. & Lim, T. Y. (2018). Recycling of phenolic compounds in Borneo's tropical peat swamp
802 forests. *Carbon Balance and Management*, 13(1), 3. <https://doi.org/10.1186/s13021-018-0092-6>
- 803 Zhou, Y., Martin, P. & Müller, M. (2019). Composition and cycling of dissolved organic matter from tropical
804 peatlands of coastal Sarawak, Borneo, revealed by fluorescence spectroscopy and PARAFAC analysis.
805 *Biogeosciences*, 16(13), 2733–2749. <https://doi.org/10.5194/bg-16-2733-2019>
- 806 Zhou, Y. Z., Alany, R. G., Chuang, V. & Wen, J. (2012). Studies of the rate constant of L-DOPA oxidation and
807 decarboxylation by HPLC. *Chromatographia*, 75(11–12), 597–606. <https://doi.org/10.1007/s10337-012-2229-1>
- 808 Zigah, P. K., McNichol, A. P., Xu, L., Johnson, C., Santinelli, C., Karl, D. M. & Repeta, D. J. (2017).
809 Allochthonous sources and dynamic cycling of ocean dissolved organic carbon revealed by carbon isotopes.
810 *Geophysical Research Letters*, 44(5), 2407–2415. <https://doi.org/10.1002/2016GL071348>