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Sources, Fluxes and Residence Times of Trace Elements measured during the U.S. GEOTRACES East Pacific Zonal Transect.

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Abstract:	<p>Trace element (TE) fluxes and their residence times (Fe, Mn, Cu, Pb, Cd, and V) within the surface ocean were determined along the GEOTRACES East Pacific Zonal Transect (GP16/EPZT) and found to reflect the diverse physical and geochemical conditions encountered across the track. The TE flux from atmospheric deposition, vertical mixing, and upwelling into the mixed layer and into the particle production zone (PPZ) along the GEOTRACES EPZT transect were evaluated with ^{7}Be-based methods developed in earlier works. A horizontal input flux is driven from east to west by the South Equatorial Current (SEC), and estimated advection velocities were applied to horizontal gradients in the distributions of several TEs to approximate this term. There is a minimum in atmospheric deposition in the central gyre, with higher fluxes to the east due to large near-shore aerosol TE loadings, and higher to the west due to greater precipitation-driven deposition velocities (V_b). The ^{7}Be-derived vertical diffusion (K_z) values range from 2.5 to 39 m^2/d (0.29×10^{-4} to $4.5 \times 10^{-4} \text{ m}^2/\text{s}$) with higher values generally within the nearshore upwelling region and the lowest values within the stratified central gyre. Manganese displayed a well-defined gradient extending from the nearshore stations into the central gyre such that the advective term is a major component of the total input flux, particularly within the central gyre. Relative to other inputs the atmospheric input of soluble Mn is only of minor importance. Unlike Mn, there is no discernable horizontal gradient in the dissolved Fe data and therefore, there is no horizontal component of flux. Nearshore removal processes are more intense for dissolved Fe than for dissolved Mn and as a result, dissolved Mn remains elevated much farther offshore than does dissolved Fe. For the stratified mid-ocean gyre stations, the total input of Fe from all sources is relatively small compared to the inshore stations, and atmospheric deposition becomes the dominant mode of input. Aerosol Fe solubility determined by a 25% acetic acid leach with hydroxylamine hydrochloride was much greater than that derived from a leach using ultra-pure deionized water. This led to significant differences in the residence time of Fe calculated for the mid-ocean gyre using these different solubilities. Generally, each element displays relatively short (days-weeks) residence times within the nearshore region of robust upwelling, reflecting large input terms and rapid removal. Moving offshore, total input fluxes decrease and the residence times of the</p>

Dr. T.S. Bianchi
Editor in Chief
Marine Chemistry

September 8, 2019

Dear Tom

I am submitting a research paper for consideration in Marine Chemistry: “Sources, Fluxes and Residence Times of Trace Elements measured during the U.S. GEOTRACES East Pacific Zonal Transect”. This involves quite a few authors and is a synthesis of many data sets from the expedition.

Suggested Editors: Adina Paytan did a good job on my last submission

Thank you for your consideration.

Sincerely,

David Kadko

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- Trace element fluxes to the surface ocean were determined along the GEOTRACES EPZT
- Trace element residence times in the surface ocean were determined along the transect
- Trace element properties reflected diverse physical and geochemical conditions

1 **Sources, Fluxes and Residence Times of Trace Elements measured during the U.S.**
2 **GEOTRACES East Pacific Zonal Transect.**

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23 **Keywords:** *Trace elements; residence times; East Pacific Ocean; GEOTRACES*

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28 **ABSTRACT**

29 Trace element (TE) fluxes and their residence times (Fe, Mn, Cu, Pb, Cd, and V) within the surface ocean
30 were determined along the GEOTRACES East Pacific Zonal Transect (GP16/EPZT) and found to reflect
31 the diverse physical and geochemical conditions encountered across the track. The TE flux from
32 atmospheric deposition, vertical mixing, and upwelling into the mixed layer and into the particle
33 production zone (PPZ) along the GEOTRACES EPZT transect were evaluated with ^7Be -based methods
34 developed in earlier works. A horizontal input flux is driven from east to west by the South Equatorial
35 Current (SEC), and estimated advection velocities were applied to horizontal gradients in the distributions
36 of several TEs to approximate this term. There is a minimum in atmospheric deposition in the central
37 gyre, with higher fluxes to the east due to large near-shore aerosol TE loadings, and higher to the west
38 due to greater precipitation-driven deposition velocities (V_b). The ^7Be -derived vertical diffusion (K_z)
39 values range from 2.5 to 39 m^2/d (0.29×10^{-4} to $4.5 \times 10^{-4} \text{ m}^2/\text{s}$) with higher values generally within the
40 nearshore upwelling region and the lowest values within the stratified central gyre. Manganese displayed
41 a well-defined gradient extending from the nearshore stations into the central gyre such that the advective
42 term is a major component of the total input flux, particularly within the central gyre. Relative to other
43 inputs the atmospheric input of soluble Mn is only of minor importance. Unlike Mn, there is no
44 discernable horizontal gradient in the dissolved Fe data and therefore, there is no horizontal component of
45 flux. Nearshore removal processes are more intense for dissolved Fe than for dissolved Mn and as a
46 result, dissolved Mn remains elevated much farther offshore than does dissolved Fe. For the stratified
47 mid-ocean gyre stations, the total input of Fe from all sources is relatively small compared to the inshore
48 stations, and atmospheric deposition becomes the dominant mode of input. Aerosol Fe solubility
49 determined by a 25% acetic acid leach with hydroxylamine hydrochloride was much greater than that
50 derived from a leach using ultra-pure deionized water. This led to significant differences in the residence
51 time of Fe calculated for the mid-ocean gyre using these different solubilities. Generally, each element
52 displays relatively short (days-weeks) residence times within the nearshore region of robust upwelling,
53 reflecting large input terms and rapid removal. Moving offshore, total input fluxes decrease and the
54 residence times of the TEs increase markedly until the western edge of the transect. There, relaxation of
55 ocean stratification permits greater upward turbulent flux and greater rainfall leads to greater atmospheric
56 input of TEs.

57 **1.0 Introduction**

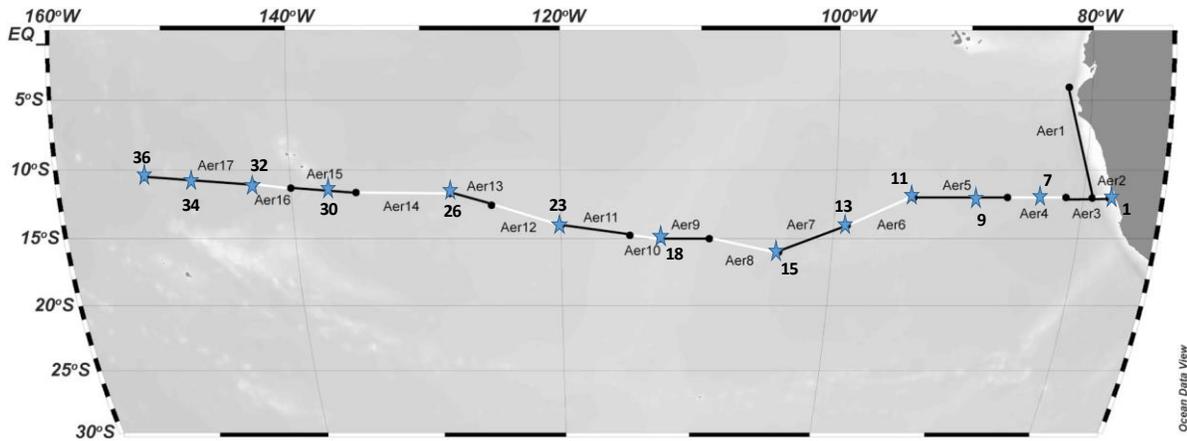
58 The International GEOTRACES Program was established to identify processes and quantify fluxes
59 that control the distributions of key trace elements (TEs) in the ocean as these chemical species play
60 important roles as nutrients, as tracers of current and past oceanographic processes, and as contaminants
61 derived from human activity (GEOTRACES Planning Group, 2006). Their biogeochemical cycling has
62 relevance to the carbon cycle, climate change, and ocean ecosystems. To accomplish these goals, the

63 Program established a global series of basin-scale transects to undertake comprehensive TE
64 measurements in the water column and air. One of these was the 2013 U.S. GEOTRACES East Pacific
65 Zonal Transect (EPZT, US GEOTRACES cruise GP16) (Moffett and German, 2018) which originated
66 within the highly productive upwelling region of the Peru Margin and extended westward across the
67 Central South Pacific Ocean to Tahiti, crossing some of the most remote and oligotrophic open-ocean
68 waters on Earth (Figure 1).

69 The Southeast Pacific Ocean along the EPZT transect (10-18°S) is an especially interesting
70 region to study trace metal fluxes to the surface ocean because low dust input supports only a very low
71 supply of trace metals to the open ocean (Wagener et al. 2008), and it has been suggested that mixing
72 processes might therefore dominate metal fluxes to the surface ocean (Fitzsimmons et al. 2016; Buck et
73 al. 2019). In the upwelling region near South America horizontal mixing and advection are especially
74 important and upwelling drives the ETSP (Eastern Tropical South Pacific) OMZ (Oxygen Minimum
75 Zone) (Karstensen et al. 2008) which results in unique metal signatures (Scholz et al. 2011, 2014) and
76 biological communities (Ohnemus et al. 2017).

77 Metal fluxes to the surface ocean are critical drivers of primary production, and the EPZT transect
78 is situated on the edge of several biological gradients: 1) the lowest productivity waters in the entire
79 global ocean in the South Pacific subtropical gyre (centered at 25-30°S; Claustre et al. 2008), 2) some of
80 the highest productivity waters in the ocean along the South American continent (e.g. Carr 2001), and 3)
81 the Fe-limited Equatorial Pacific Region (Martin et al. 1994). Sampling across such diverse
82 oceanographic settings would be expected to manifest different sources, sinks, distributions and residence
83 times of trace elements, which must be well constrained to predict or estimate biological responses.

84 The GEOTRACES program affords the opportunity to extensively sample TEs in the water column
85 and aerosols, including the determination of aerosol solubilities and tracer measurements such as the
86 isotope ^7Be , a cosmi-ray produced isotope (half-life = 53.3. days) that is deposited into the surface ocean
87 and used to study the atmospheric fluxes of TEs and their transport in the upper ocean (e.g. Young and
88 Silker, 1980; Kadko and Olson, 1996; Kadko and Johns; 2011; Haskell et al., 2015; Kadko et al., 2015;
89 Shelley et al., 2016). In this work we utilize the extensive dissolved TE, ^7Be tracer, and aerosol
90 concentration and solubility datasets from EPZT to calculate TE fluxes into the ocean mixed layer and the
91 Particle Production Zone (PPZ; depth where fluorescence reaches 10% of maximum fluorescence) across
92 the transect; from this, we evaluate the residence time of TEs in the upper ocean of this region.



93
 94 **Figure 1. The 2013 US GEOTRACES EPZT cruise track with aerosol sample deployment locations marked as**
 95 **alternating black and white lines; recovery location of each sample coincides with the deployment location of**
 96 **the next sample (details in Buck et al., 2019). The water sampling stations are indicated by a blue star and the**
 97 **station number; these did not always coincide with aerosol collection deployments.**

98
 99 **2.0 Methods**

100 **2.1 Sample Collection-** Seawater and aerosol samples were collected during the U.S. GEOTRACES
 101 EPZT (Eastern Pacific Zonal Transect -GEOTRACES cruise GP16) expedition on the R/V Thomas G.
 102 Thompson (TGT303, 25 October 2013 - 20 December 2013, Manta, Ecuador - Papeete, Tahiti) along the
 103 cruise track shown in Fig. 1.

104 **2.1.1. Aerosols:** Details of the aerosol collection methods have been presented in prior work (Buck et al.,
 105 2019). Briefly, bulk aerosol samples were collected on 12-replicate acid-washed 47 mm Whatman 41
 106 (W41) ash-less filter discs mounted in Advantec-MFS polypropylene inline filter holders (PP47). When
 107 the wind was directed from the bow, air was pumped through the filters using a high-volume aerosol
 108 sampler (model 5170V-BL, Tisch Environmental) at approximately 100 L min⁻¹ through each filter. The
 109 sampler was mounted on the starboard rail of the 03-deck approximately 16 m above sea level and
 110 forward of both the ship's superstructure and exhaust stacks. Each collection period lasted approximately
 111 three days.

112 **2.1.2. Water samples for trace elements:** Seawater was collected following established GEOTRACES
 113 sample collection protocols (Cutter and Bruland 2012) using a trace metal-clean CTD mounted on an
 114 epoxy-coated aluminum rosette housing 24 GO-Flo bottles (12 L each) on a Vectran conducting cable.
 115 Seawater was filtered in a clean sampling van under ~0.5 atm of filtered air through 0.2 μm Acropak-200
 116 polyethersulfone capsule filters (Pall) into pre-cleaned bottles following three 10% volume rinses.
 117 Samples were acidified to pH 2 with ultrapure hydrochloric acid (Optima grade, Fisher Scientific) or pH
 118 ~1.7 with quartz-distilled hydrochloric acid (Bruland lab) and stored at least 3 months prior to analysis.

119 **2.1.3. Water column ⁷Be:** Details of sample collection for this expedition have been presented in Kadko,
120 (2017). Briefly, samples were collected at selected depths by pumping 400–700 L of seawater via a ~4cm
121 hose into large plastic barrels on deck. From these barrels, the seawater was then pumped through iron
122 impregnated acrylic fibers at ~10 L/min (Lal et al., 1988; Krishnaswami et al., 1972; Lee et al., 1991) to
123 extract the ⁷Be from seawater.

124 **2.2 Analytical methods**

125 **2.2.1. Aerosol samples:** Total trace metal concentrations in the samples were determined at the University
126 of Alaska Fairbanks by inductively couple plasma mass spectrometry (ICPMS) using a Thermo Element-
127 2 ICP mass spectrometer following digestion with concentrated nitric and hydrofluoric acids. Method
128 fitness was tested by digesting three reference materials and quantifying elemental recoveries. Details can
129 be found in Buck et al. (2019). For ⁷Be, the Whatman-41 aerosol filters were stacked three-high in a
130 plastic Petri dish and counted by gamma spectroscopy. This configuration was calibrated with a
131 commercially prepared mixed solution of known gamma activities.

132 Soluble TE concentrations were measured in both ultrapure deionized water (DI water; 18.2 MΩ·cm)
133 after Buck et al. (2006), and in ammonium acetate solution with a reducing agent after modifications to
134 one of the methods (“HAc_{T*+R},”) as described in Berger et al. (2008). Three filters from each deployment
135 were extracted by each treatment producing triplicate DI water soluble aerosol samples and triplicate
136 acetic acid soluble aerosol samples which will be called “HAc” samples from this point forward.

137 Then DI water method produces a lower-limit of elemental solubility and the HAc method an upper
138 limit such that a “solubility window” for each element is provided (Shelley et al., 2018). The HAc leach
139 treatment is designed to extract TEIs associated with biogenic material, carbonates, most forms of Fe and
140 Mn (oxyhydr)oxides and those adsorbed to clay minerals. The reducing conditions may simulate the
141 conditions particles experience during passage through a zooplankton gut or inside a fecal pellet or
142 organic aggregate while the low pH (2.1) serves to prevent any reprecipitation or adsorption artifacts.

143 All sample handling was performed under Class-100 laminar flow conditions and all filters were
144 stored frozen prior to processing. Extractions with DI water took place while at sea, while extractions
145 with HAc took place at the University of Alaska Fairbanks. The W41 filter discs were extracted with 100
146 mL of ultrapure DI water (>18 MΩ cm) under a vacuum. The extract flowed through a Teflon-PFA filter
147 holder (Savillex) loaded with an acid-washed 0.4 μm Nuclepore backing filter and directly into an acid-
148 washed 100 mL 100% LDPE receiving bottle. Samples were immediately acidified with Teflon-distilled
149 6M hydrochloric acid to a final concentration was 0.024M HCl. DI water soluble aerosol TE
150 concentrations were determined by inductively coupled plasma mass spectrometry (Perkin Elemer Nexion
151 300D) at the Skidaway Institute of Oceanography. This quadrupole ICP-MS utilizes NH₃ as a reaction gas
152 to eliminate polyatomic interferences. Additionally, the analytes were concentrated on an in-line chelating

153 resin (Nobias PA-1) prior to introduction to the plasma (ESI seaFAST S3 system). Instrument detection
154 limits were determined by measuring the reagent blank of the system multiple times during each run and
155 calculating three times the standard deviation of those analyses. Deployment filter blank extracts were
156 prepared at sea with W41 discs that had been deployed in the aerosol samplers for 1 h while not in
157 operation. Three replicate blanks were extracted each time a group of samples were extracted thereby
158 accounting for any variability in the quality of the ship-provided DI water.

159 For the determination of HAc soluble aerosol trace element loadings, the thawed W41 filter discs
160 were folded twice into a wedge and transferred to acid cleaned centrifuge tubes. Extractions were carried
161 out following the “HAc_{T*+R},” procedure described in Berger et al., 2008, which includes a reducing agent
162 and a heating step. We modified the method by adding a centrifugation step to prevent transfer of
163 unextracted particulate trace elements. Briefly, 1mL of 25% acetic acid with 0.02M hydroxylamine
164 hydrochloride was added to the folded filters, which were immediately placed in a 90 °C hot water bath
165 for 10 minutes, then left to cool for 2 hours. Samples were centrifuged, and the solution was carefully
166 transferred into Teflon beakers. Filters were then rinsed 3 times with 500µl of ultra-pure DI water, with a
167 centrifugation step after each rinse and careful transfer into the corresponding Teflon beaker. The
168 solutions on the beakers were set to dry on a hot plate down to a pearl. The residue was then reconstituted
169 with 1mL of Optima grade concentrated nitric acid and decanted into an acid-washed, wide-mouth, 30 ml
170 polyethylene bottle (Nalgene). Each beaker was rinsed 3 times with 5ml of 0.1 % v/v Optima grade nitric
171 acid, with rinses decanted into the corresponding 30 ml bottle. All sample handling was performed under
172 Class 100 laminar flow conditions. Deployment blanks were processed in the same manner as samples.
173 HAc soluble aerosol trace metal concentrations were determined by inductively coupled plasma mass
174 spectrometry (Thermo Element 2) using external standard curves. Indium was added to standards and
175 samples to track instrument fluctuations. Prior to direct injection, sample solutions were diluted as needed
176 by a factor of 2 to 20 with 1 M Optima grade nitric acid, and deployment blank solution were diluted by a
177 factor of 2. Analysis took place at the University of Alaska Fairbanks.

178 **2.2.2. Dissolved trace metals**

179 The dissolved metal data used in this paper have been culled from the GEOTRACES
180 Intermediate Data Product (IDP) 2017 (Schlitzer et al. 2018) and from several published papers (Resing et
181 al. 2015 for Mn; Fitzsimmons et al. 2017 and John et al. 2018 for Fe; Roshan and Wu 2018 for Cu; Ho et
182 al., 2019 for V); the details of these methods can be found in these publications. Dissolved Fe, Mn, Cu,
183 Pb, and Cd were also measured using established methods (Biller and Bruland 2012; Parker et al. 2016)
184 after UV irradiation and preconcentration onto Nobias PA1 chelating resin, followed by ICP-MS analysis
185 and comparison to matrix-matched standard curves. Dissolved Pb was also analyzed across the transect
186 using isotope dilution ICP-MS with ²⁰⁶Pb on a quadrupole ICP-MS (Fisons PQ+; Lee et al. 2011). These

187 datasets were all rigorously intercalibrated by the originating labs, and the intercalibration procedures
188 were reviewed and approved by the GEOTRACES Standards & Intercalibration Committee before being
189 published in the GEOTRACES IDP 2017.

190

191 **2.2.3. Water column ^7Be** – Details of the ^7Be analysis for this expedition are presented in Kadko (2017).
192 On land, the fibers were dried and then ashed. The ash was subsequently pressed into a pellet (5.8 cm
193 diameter) and placed on a low background germanium gamma detector. The isotope ^7Be has a readily
194 identifiable gamma peak at 478 keV. The detector was calibrated for the pellet geometry by adding a
195 commercially prepared mixed solution of known gamma activities to an ashed fiber, pressing the ash into
196 a pellet, and counting the activities to derive a calibration curve. The uncertainty of the extraction
197 efficiency (4%) and the detector efficiency (2%) was in all cases smaller than the statistical counting error
198 and the uncertainty in the blank.

199

200 2.3 Flux calculations.

201 The flux of TEs into the mixed layer and the PPZ along the GEOTRACES EPZT transect are composed
202 of four components: i) atmospheric deposition ii) vertical mixing from below, iii) upwelling if present,
203 and iv) horizontal advection from east to west driven by the South Equatorial Current (SEC) if horizontal
204 gradients in the TE distribution exist. In the Northern Equatorial Pacific, Landing and Bruland (1987)
205 reported rapid off- shore advective transport from the eastern boundary by the North Pacific Equatorial
206 Current, indicating the importance of this transport term in any TE mass budget calculation.

207 i) *Atmospheric deposition*. The bulk atmospheric flux for aerosol trace elements (F_{atm}) is estimated from
208 the concentration of trace elements in the aerosols (C_{TE}) and the bulk deposition velocity (V_b) which
209 includes dry + wet deposition, such that

$$210 F_{\text{atm}} = C_{\text{TE}} \times V_b \quad \text{Eqn 1}$$

211 It has been shown that V_b can be derived from the upper ocean inventory of ^7Be and the aerosol ^7Be
212 activity (Young and Silker, 1980):

$$213 V_b = (\text{Inventory}^7\text{Be} \times \lambda) / (^7\text{Be})_{\text{aerosol}} \quad \text{Eqn 2}$$

214 where λ is the ^7Be decay constant (0.013 d^{-1}).

215

216 This method has been used to derive the TE atmospheric flux for sites in the Atlantic (Kadko et al., 2015;
217 Anderson et al., 2016; Shelley et al., 2017), the Pacific (Buck et al., 2019), and the Arctic (Kadko et al.,
218 2016; 2019; Marsay et al., 2018) oceans.

219 The most biogeochemically relevant fraction of atmospheric flux is that which is operationally

220 defined as soluble. We adjust Eq. (1) to account for the elemental aerosol fractional solubility, β such that
 221 the atmospheric flux of soluble TE equals the product of the bulk atmospheric flux and the measured
 222 fractional solubility determined for each element:

$$223 \quad F_{atm} = C_{TE} \times V_b \times \beta \quad \text{Eqn. 3}$$

224 In this paper we will consider the atmospheric flux of soluble TEs based on the solubilities determined by
 225 both the DI and HAc leach approaches.

226

227 ii) *Vertical diffusion*. Diffusive transport below the mixed layer is described by $F_D = K_z dC/dz$, where K_z
 228 is the turbulent diffusion coefficient and dC/dz is the vertical gradient in TE concentration. The shape of
 229 the ^7Be profile below the ocean mixed layer can be used to derive vertical diffusion rates (Kadko and
 230 Olsen, 1996; Kadko and Johns, 2011; Haskell et al., 2015; Kadko, 2017):

$$231 \quad C(z) = C_0 e^{\alpha(z-H)} \quad \text{Eqn 4a}$$

232 where $C(z)$ is the ^7Be activity at depth z and H is the depth of the mixed layer and

$$233 \quad \alpha = \frac{w}{2K_z} + \frac{1}{2} \left\{ \left(\frac{w}{K_z} \right)^2 + \frac{4\lambda}{K_z} \right\}^{\frac{1}{2}} \quad \text{Eqn 4b}$$

234 where w is the upwelling rate (m/d). In the absence of upwelling ($w=0$),

$$235 \quad C(z) = \exp[-(\lambda/K_z)^{1/2} \cdot Z] \quad \text{Eqn 5}$$

236 where Z is the depth below the mixed layer H .

237 iii). *Upwelling*. Within regions of upwelling the dilution of the water column ^7Be inventory by deep, ^7Be -
 238 free upwelled water provides a means to infer upwelling rates quantitatively (Kadko and Johns, 2011;
 239 Haskell et al., 2015; Kadko 2017). The upwelling flux is calculated using Eqn. 6, where w is the
 240 upwelling rate either at the base of the mixed layer or the base of the PPZ, and C is the trace element
 241 concentration at that horizon:

$$242 \quad \text{Upwelling flux } F_U = w [C] \quad \text{Eqn 6.}$$

243 The results and discussion for the upwelling stations of the EPZT transect (stations 1-15) have been
 244 presented elsewhere (Kadko, 2017).

245 iv). *Horizontal advection*. Near-surface east-to-west currents, calculated from satellite-tracked drogued
 246 drifter velocities, are in the range 0.1-0.5m/s for the EPZT region (Lumpkin and Johnson, 2013). The
 247 timescale of transport from nearshore to the central gyre (~ 5000 km) would then be in the range 580 –
 248 116 days respectively, with the upper end of this range more likely as the velocity for the entire mixed
 249 layer would be lower than that determined for the near-surface. Surface ^{228}Ra data (Sanial et al., 2018)

250 provide additional insight into the rate of offshore transport from the continental margin. The continental
 251 shelf is a source of ^{228}Ra which subsequently decays with a 5.75 y radioactive half-life upon transport
 252 offshore. The open water activities of the central gyre are \sim half the near-shore values. Based on
 253 radioactive decay, the transport timescale would be \sim 2000 days, equivalent to a horizontal velocity of
 254 0.03 m/s. This is not consistent with the Lumpkin and Johnson (2013) observations and likely is an
 255 underestimate, as ^{228}Ra dilution and loss due to surface uptake and particle settling were ignored in this
 256 simple calculation. In the discussions that follow, the effect of horizontal transport upon the TE
 257 distributions will be considered in detail, using a range of velocities (sections 3.1 and 3.2).

258 The net horizontal flux into the mixed layer is given by:

$$259 \quad F_{HZ} = H \cdot u \cdot \frac{dC}{dX} \quad \text{Eqn 7}$$

260 Where H is the height of the mixed layer, u is the horizontal velocity and dC/dX is the horizontal
 261 concentration gradient between profiles. Meridional transport is assumed to be small compared to the
 262 zonal transport in this section. For water below the mixed layer and within the PPZ, an estimate of
 263 0.05m/s, based on a high-resolution ocean general circulation model, was used (Masumoto et al., 2004).
 264 Including the mixed layer as part of the PPZ, the net horizontal flux into the PPZ is:

$$265 \quad F_{PPZ} = F_{PPZ} = H \cdot u \cdot \frac{dC}{dX} + (H_{PPZ} - H)u' \cdot \frac{dC}{dX} \quad \text{Eqn 8}$$

266 Where H_{ppz} is the base of the PPZ and u' is the horizontal velocity between the mixed layer and
 267 base of the PPZ.

268 These calculations will depend on the depths of the MLD and the PPZ, and these parameters may
 269 naturally vary in ways that cannot always be predicted or observed. However, as shown below this does
 270 not have a great effect on the calculated residence times.

271 2.4 Residence time calculations

272 Residence time is a useful concept that informs us of several aspects of trace element biogeochemical
 273 behavior in the upper ocean, including dissolution and scavenging processes, biological utilization, and
 274 the seasonal relationship of surface concentrations to atmospheric input (e.g. Jickells, 1999; Croot et al.,
 275 2004; Hayes et al., 2015; Bridgestock et al., 2016; Kadko et al., 2019). For the mixed layer it is defined
 276 by the dissolved TE inventory divided by the sum of the input fluxes.

$$277 \quad \tau_{ML} = [\text{mixed layer TE inventory}] / [F_{atm} + F_D + F_U + F_{HZ}] \quad \text{Eqn 9}$$

278 The residence time can also be calculated using the sum of the export fluxes (e.g. Black et al, 2019), since
 279 at steady state, the sum of the input fluxes should equal the sum of the output fluxes.

280 While the effect of variable mixed layer depth (MLD) would affect the horizontal flux term (F_{HZ}),

281 the effect on the residence time calculation is mitigated because the mixed layer TE inventory changes in
 282 the opposite sense of the flux term. For example, an increase in MLD might produce a higher mixed layer
 283 TE inventory (in the numerator) but also a higher F_{HZ} (eqn. 7) in the denominator. In this way, the effects
 284 of variable MLD on residence times tends to cancel out.

285 While residence times are often reported for the mixed layer, it is also useful to consider the
 286 geochemical behavior of TEs within the slightly deeper particle production zone (PPZ), defined from the
 287 surface to its base level where the fluorescence signal approaches 10% of the fluorescence maximum.
 288 This has recently been adopted as defining the euphotic zone (Owens et al, 2015; Ohnemus et al. 2016;
 289 Kadko, 2017). Then

$$290 \tau_{PPZ} = [\text{PPZ TE inventory}] / [F_{\text{atm}} + F_D + F_U + F_{PPZ}] \quad \text{Eqn. 10}$$

291 Both calculations will be considered here.

292

293 3.0 Results and Discussion

Sta no.	Aerosol no.	⁷ Be (dpm/m ³)	V (pmol/m ³)	Fe (nmol/m ³)	Mn (nmol/m ³)	Cu (nmol/m ³)	Cd (pmol/m ³)	Pb (pmol/m ³)	Al (nmol/m ³)	Ti (nmol/m ³)
1-15	1-7	0.211 ±0.092	6.780 ±7.734	0.684 ±0.824	0.0136 ±0.0161	0.0096 ±0.0113	0.372 ±0.448	1.392 ±1.667	2.140 ±2.695	0.071 ±0.081
18-36	8-17	0.133±0.039	0.288 ±0.139	0.058 ±0.026	0.0008 ±0.0004	0.00118 ±0.0004	0.028 ±0.024	0.377 ±0.160	0.147 ±0.040	0.015 ±0.005

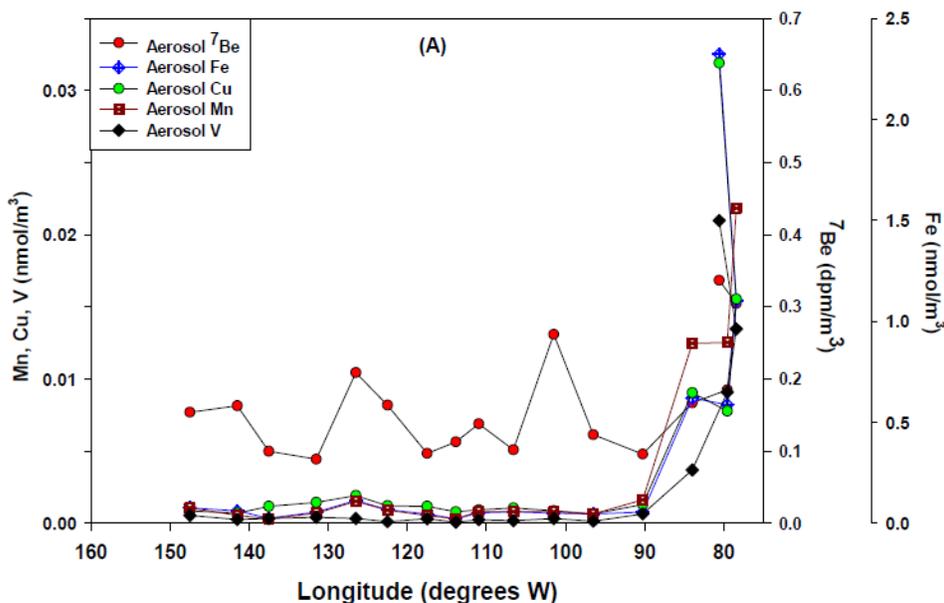
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295 3.1. Flux calculations

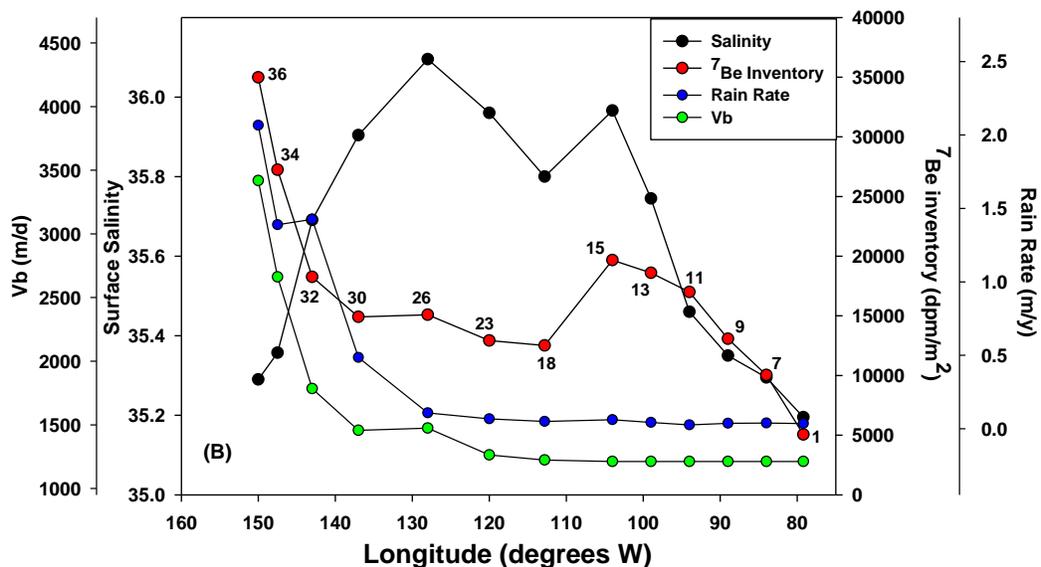
296 i) *Atmospheric flux.* The GEOTRACES transect crossed a significant gradient in atmospheric
 297 deposition driven by dust input into the eastern transect near the continental shelf (stations 1-15),
 298 coincident with the area of upwelling (Table 1, Buck et al., 2019). Moving westward over the open ocean
 299 (stations 18-36), the aerosol and TE concentrations were substantially lower, although this is not as
 300 apparent for ⁷Be. Here the region is characterized by the strong stratification and oligotrophic conditions
 301 of the subtropical gyre (Fig 2a).

302 Continuing west of 140°W, the rainfall is considerably higher than over the mid-gyre stations
 303 which is manifested by decreasing surface salinity and increasing ⁷Be inventories (Fig 2b); it has been
 304 observed that ⁷Be deposition rates correlate with the rate of precipitation (e.g. Young and Silker, 1980;
 305 Olsen et al., 1985; Uematsu et al., 1994; Kim et al., 1999; Kadko and Prospero, 2011; Peng et al., 2019).
 306 Accordingly, the higher ⁷Be inventories result in higher deposition velocities towards the western stations
 307 compared to stations to the east (Eqn. 2). This results in a minimum in atmospheric TE deposition in the
 308 central gyre, with higher fluxes to the east due to higher aerosol TE loads (Figure 2a), and higher to the

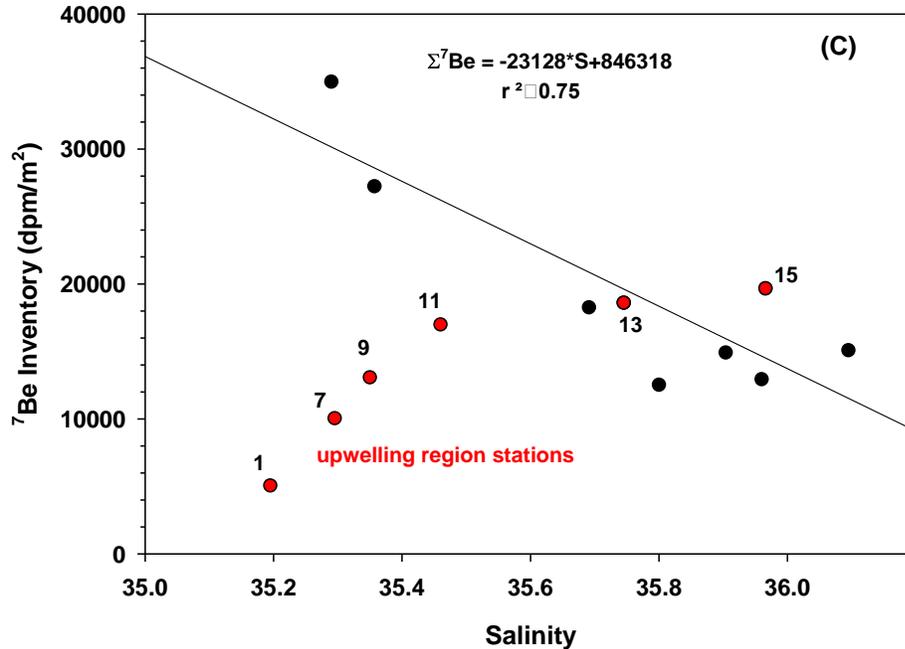
309 west due to higher V_b values resulting from increased precipitation (Figure 2b, Table 2). Our bulk
 310 atmospheric flux estimates of Mn (based on the ^7Be method) across the transect range between 1 and 16.5
 311 $\text{nmol}/\text{m}^2/\text{d}$ which is comparable to the range (0.1-16 $\text{nmol}/\text{m}^2/\text{d}$) of atmospheric particulate flux presented
 312 in Black et al (2019) who used dust-model estimates combined with upper crust ratios for each element.
 313 Buck et al. (2019) estimated bulk aerosol Mn deposition to range between 0.45 – 68 $\text{nmol}/\text{m}^2/\text{d}$ by
 314 assuming a constant deposition velocity of 1500 m/d applied to observed aerosol TE concentrations but
 315 those estimates are susceptible to the short-term variability of aerosol concentrations and the episodic
 316 nature of dust transport.



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 320 **Figure 2. A) Aerosol TE and ⁷Be concentrations along the EPZT transect. Note the high TE**
 321 **concentrations towards the eastern portion of the transect. This is not as apparent for ⁷Be. B)**
 322 **Rainfall (from Global Precipitation Climatology Project, GPCP) for the cruise period across the**
 323 **transect. Note the coincident increase in precipitation, decrease in salinity and increases in ⁷Be**
 324 **inventory and bulk deposition velocity (V_b) at the westernmost portion of the transect. C) ⁷Be**
 325 **inventory plotted against salinity. Stations within the upwelling region are plotted separately in red.**
 326 **The linear regression applies to non-upwelling region stations (black). Despite low rainfall, salinity**
 327 **is low at the eastern portion of the transect because of upwelling of low salinity water. Note that the**
 328 **⁷Be inventory is also lower because of upwelled low-⁷Be water (Kadko, 2017).**

329
 330 Note that the low salinity in the eastern stations (Fig 2a) is not the result of high precipitation but of
 331 upwelling of low salinity water (Fig 2c). Because of upwelling, Eqn. 2 does not hold for the easternmost
 332 stations 1-13, as the observed ⁷Be inventory does not reflect atmospheric input, but rather upwelling of
 333 deep, ⁷Be deficient water. For this reason, the inventories of these stations (for use in eqn. 2) are assigned
 334 that of station 15, the nearest station not affected by the upwelling process (Kadko, 2017; Buck et al.,
 335 2019). Thus, the nearshore stations display a constant deposition velocity and flux (Fig 2b, Table 2).

336 The fractional solubility for several TEs, determined by the DI and HAc leaches, are presented in
 337 Tables 3a and 3b. These were then applied to Eqn. 3 to derive the atmospheric fluxes of soluble (aerosol
 338 dissolvable) TEs.

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Sta	⁷ Be inventory (dpm/m ²)	⁷ Be Flux	Vb (m/d)	Fe Flux (nmol/m ² /d)	Mn Flux (nmol/m ² /d)	Cu Flux (nmol/m ² /d)	Cd Flux pmol/m ² /d	Pb Flux pmol/m ² /d	Al Flux (nmol/m ² /d)	V Flux (nmol/m ² /d)
1	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
7	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
9	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
11	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
13	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
15	19666	256	1212	828.8	16.48	11.63	450.7	1687	2593	8.2
18	12516	163	1223	71.0	0.98	1.35	34.25	461.2	179.8	0.35
23	12930	168	1264	73.3	1.01	1.39	35.39	476.5	185.8	0.36
26	15085	196	1474	85.5	1.18	1.62	41.29	555.9	216.2	0.42
30	14906	194	1457	84.5	1.17	1.60	40.80	549.3	214.18	0.42
32	18258	237	1785	103.5	1.43	1.96	49.97	672.8	262.3	0.51
34	27230	354	2662	154.4	2.13	2.93	74.52	1003	391.25	0.77
36	34981	455	3419	198.3	2.74	3.76	95.74	1289	502.6	0.98

a. Fluxes calculated using average aerosol concentrations reported in Table 1.

b. Soluble atmospheric flux= bulk atmospheric flux x solubility (see table 3).

Deployment	Start Date	Start Latitude	Start Longitude	Fe%	% Error	Mn%	% Error	Cu%	% Error
		°S	°W						
Aer1	26-Oct	4.07	81.99	2.42	0.12	43.7	2.2	42.8	4.0
Aer2	29-Oct	12.01	79.20	1.99	0.59	37.0	2.2	37.7	2.0
Aer3	1-Nov	12.05	77.66	1.27	0.54	32.1	2.0	38.2	2.6
Aer4	4-Nov	12.00	81.50	1.94	0.50	34.2	2.7	36.6	3.9
Aer5	7-Nov	12.00	86.50	4.75	0.50	61.7	32.4	46.0	20.8
Aer6	10-Nov	12.00	94.00	1.55	0.61	44.4	14.2	29.9	4.1
Aer7	13-Nov	14.00	99.00	0.94	0.52	35.5	4.4	24.7	6.4
Aer8	16-Nov	16.00	104.00	0.69	0.13	25.1	4.7	18.8	8.5
Aer9	19-Nov	15.00	109.19	0.56	0.11	26.0	8.6	10.3	4.1
Aer10	22-Nov	14.99	112.75	1.47	0.13	48.8	8.4	17.8	7.3
Aer11	25-Nov	14.77	115.00	0.83	0.15	30.2	4.6	12.8	5.7
Aer12	28-Nov	14.00	120.00	0.49	0.21	20.5	5.5	56.8	13.8
Aer13	1-Dec	12.54	125.00	0.35	0.27	18.2	7.5	15.7	6.2
Aer14	4-Dec	11.67	128.00	1.27	0.32	26.5	8.1	16.9	3.3
Aer15	7-Dec	11.60	135.00	1.23	0.25	51.8	18.0	17.3	1.7
Aer16	10-Dec	11.31	140.00	1.68	0.84	NA	NA	24.9	10.8
Aer17	13-Dec	11.03	142.95	0.69	0.34	38.8	3.5	24.7	5.4
			average	1.42 ± 1.04		35.9 ± 11.9		27.7 ± 13.2	

Deployment	Start Date	Start Latitude	Start Longitude	Fe%	% Error	Mn%	% Error	Cu%	% Error
		°S	°W						
Aer1	26-Oct	4.07	81.99	12.52	0.88	46.59	2.22	55.35	4.06
Aer2	29-Oct	12.01	79.20	8.45	0.48	40.25	2.05	53.44	2.33
Aer3	1-Nov	12.05	77.66	8.32	0.85	43.42	5.18	56.95	6.63
Aer4	4-Nov	12.00	81.50	10.45	2.40	39.27	6.01	65.57	4.82
Aer5	7-Nov	12.00	86.50	31.63	2.63	93.65	2.17	85.71	8.16
Aer6	10-Nov	12.00	94.00	16.54	3.14	52.78	19.16	60.98	6.43
Aer7	13-Nov	14.00	99.00	22.86	2.90	37.78	4.75	87.50	23.63
Aer8	16-Nov	16.00	104.00	6.88	0.97	12.89	2.55	22.06	8.93
Aer9	19-Nov	15.00	109.19	19.45	0.00	31.11	0.00	45.21	0.00
Aer10	22-Nov	14.99	112.75	60.66	12.92	55.56	10.80	72.55	36.92
Aer11	25-Nov	14.77	115.00	26.72	6.62	53.33	9.75	68.00	25.25
Aer12	28-Nov	14.00	120.00	13.03	5.41	28.00	9.01	65.38	14.37
Aer13	1-Dec	12.54	125.00	22.96	3.96	33.33	14.92	80.00	25.62
Aer14	4-Dec	11.67	128.00	16.08	2.29	24.19	18.95	36.96	6.80
Aer15	7-Dec	11.60	135.00	-	-	-	-	-	-
Aer16	10-Dec	11.31	140.00	68.21	33.06	123.91	57.45	126.67	53.00
Aer17	13-Dec	11.03	142.95	12.88	3.83	42.37	7.67	69.09	18.58
			average	22.35 ± 17.89		47.40 ± 27.01		65.71 ± 23.62	

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Table 4. Vertical transport parameters			
Station	upwelling rate		Diffusion
	W (ML) ^a	W (PPZ) ^b	K _z ^c
	(m/d)	(m/d)	(m ² /d)
1	4.4	3.3	38
7	1.3	0.75	39
9	1.5	0.68	30.65
11	0.04	0.005	14.7
13	-0.12	-0.007	24.1
15	0.15	0.01	26
18	0	0	28
23	0	0	24
26	0	0	6
30	0	0	2.5
32	0	0	2.5
34	0	0	20
36	0	0	7

352 a. Upwelling at the base of the mixed layer (Kadko, 2017)

353 b. Upwelling at the base of the PPZ (Kadko, 2017)

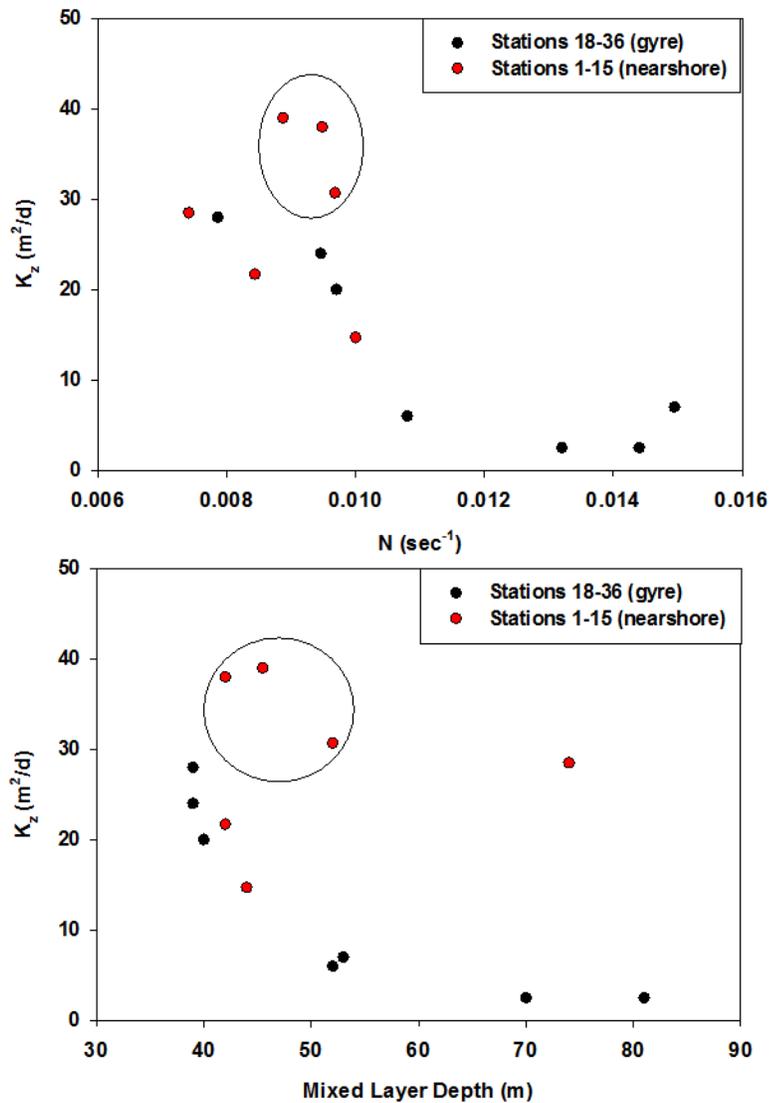
354 c. Turbulent diffusion coefficient. Values for stations 1-15 from Kadko, (2017); values for stations 18-36 from this
355 work.

356

357

358 ii) *Vertical diffusion*. The turbulent diffusion coefficients for the upwelling stations of the EPZT transect
359 (stations 1-15) based on ⁷Be profiles have been derived previously (Kadko, 2017) and are summarized in
360 Table 4. The ⁷Be profiles for stations 18-36 are shown in supplemental figure S1, and the derived
361 diffusion coefficients presented in Table 4. The ⁷Be-derived K_z values range from 2.5 to 39 m²/d (0.29 x
362 10⁻⁴ to 4.5 x 10⁻⁴ m²/s) with higher values generally within the nearshore upwelling region and the lowest
363 values within the central gyre. This pattern is consistent with the greater surface ocean stratification of the
364 gyre which is reflected in a plot of K_z vs. Brunt- Väisälä frequency (N) derived for the water column
365 between the base of the mixed layer and the depth of the PPZ (Figure 3A). In addition, the ⁷Be-derived K_z
366 generally decreases as the mixed layer depth increases (Figure 3B) suggesting that the deeper mixed
367 layers here penetrate into more stratified water. In these figures the stations with the most robust
368 upwelling (stations 1, 7, and 9) manifest K_z values greater than the trend for the other stations, which is
369 consistent with the energetic environment of the upwelling regime.

370



371

372 **Figure 3. A). The ⁷Be-derived K_z plotted against Brunt- Väisälä frequency. B). The ⁷Be-derived K_z**
 373 **plotted against mixed layer depth. Nearshore stations 1-15 (red) and offshore stations 18-36 (black)**
 374 **are indicated separately. Nearshore stations 1,7, and 9 with the greatest rates of upwelling are**
 375 **circled and fall off the trend set by the rest of the stations.**

376

377 *iii) Upwelling flux.* The upwelling rates for the eastern stations of the EPZT transect (stations 1-15) have
 378 been derived elsewhere (Kadko, 2017) and are summarized in Table 4. The upwelling rate is modeled to
 379 linearly decrease from the base of the mixed layer to zero at 100 m below the mixed layer. The upwelling
 380 TE fluxes are significant for these stations and in some cases are the dominant source of a TE into the
 381 mixed layer and the PPZ. A summary of all the input fluxes for the mixed layer and the PPZ are shown in
 382 Table 5.

383

384 iv). *Horizontal advection*. Horizontal velocities discussed above were combined with horizontal gradients
 385 along the EPZT transect to derive horizontal fluxes according to eqns. 7 and 8. These are summarized in
 386 Table 5.

387 *Manganese*- In both the mixed layer and at the depth of the PPZ, Mn shows a clear east-to-west
 388 gradient (Figure 4). For this calculation, we choose a current speed of 0.1m/s which is at the lower end of
 389 the range of Lumpkin and Johnson (2013) and consistent with the ^{228}Ra data of Sanial et al. (2018). The
 390 data in Table 5 indicate that despite a large atmospheric dust input to the near shore stations, the greatest
 391 flux of Mn is upwelled, suggesting a shelf-sediment porewater origin (e.g. Sanial et al., 2018). A similar
 392 observation was made off NW Africa, where high Mn occurs in a region associated with both upwelling
 393 and dust input, but the dominant source was upwelling (Shiller, 1997). We also note that some dissolved
 394 Mn at the easternmost stations is derived from upwelling of waters where *in situ* reduction of particulate
 395 Mn(IV) to dissolved Mn(II) is favored due to the oxygen deficient conditions (Cutter et al. 2018; Lee et
 396 al. 2018), but this is likely less than the porewater-diffused Mn flux.

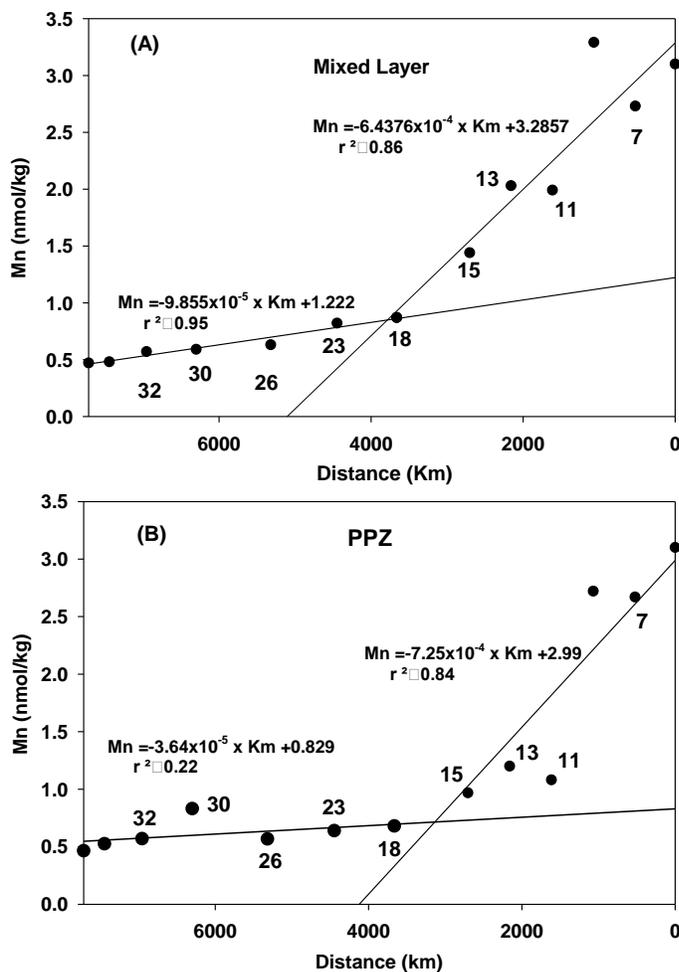


Figure 4.

A) Dissolved Mn plotted against distance from station 1 for the mixed layer across the EPZT.

B) Dissolved Mn plotted against distance from station 1 for the PPZ across the EPZT.

397

398 For the offshore gyre stations, horizontal transport is the dominant flux term. Dissolved Mn
 399 profiles of most stations display a maximum below the mixed layer suggestive of this advective input
 400 (supplemental Figure S2). The horizontal Mn concentration gradient between stations is steepest for the
 401 near-shore stations, indicating removal as the Mn, diffused from sediments, is transported offshore and
 402 begins to be oxidized (Figure 4). Towards the open ocean (west of station 15 at 104°W) the gradient is
 403 diminished suggesting a slower removal rate. This will be apparent in the residence time calculations
 404 discussed below. Sanial et al. (2018) found a similar discontinuity in the upper 200m ²²⁸Ra distribution
 405 across the transect but noted that even in the gyre stations ²²⁸Ra was still above the detection limit which
 406 suggested these surface waters had recently (years) been in contact with the continental margin. They
 407 suggested that the correlation between dissolved Mn and ²²⁸Ra over the transect indicates that Mn was
 408 transported over considerable length scales into the open ocean. They also used ²²⁸Ra as a “flux gauge” to
 409 derive a Mn flux from the shelf into the open water of 9000 nmol/m²/d from the upper 200 m, but they
 410 noted that estimates of Mn benthic flux (the ultimate source of Mn for offshore transport) have been as
 411 low as 1100 nmol/m²/d from this area (Scholz et al., 2011). Black et al (2019) derived an offshore flux of
 412 1,960 nmol/m²/d. Normalizing our PPZ water column to 200m, we derive an average offshore Mn flux of
 413 ~750 nmol/m²/d.

414 *Iron*- Unlike dissolved Mn, there is no discernable horizontal gradient in the dissolved Fe

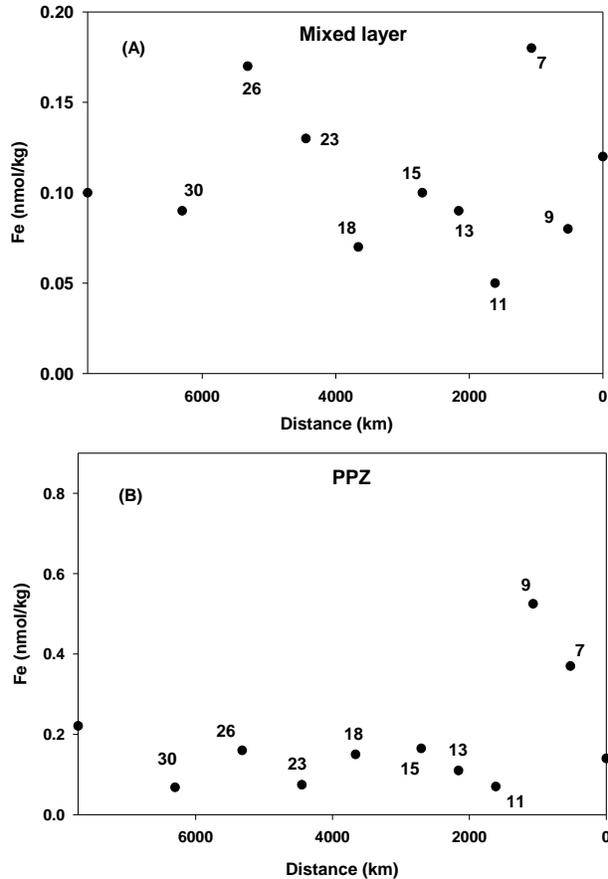


Figure 5.

A) Dissolved Fe plotted against distance from station 1 for the mixed layer across the EPZT.

B) Dissolved Fe plotted against distance from station 1 for the PPZ across the EPZT.

416 concentration field. (Figure 5). Therefore, there is no horizontal component of flux. Nearshore removal
 417 processes are more intense for dissolved Fe than for dissolved Mn (Landing and Bruland, 1987). Fe(II),
 418 supplied by porewater diagenesis, is rapidly oxidized via abiotic reactions in the presence of oxygen
 419 (Millero et al. 1987) and is thus rapidly attenuated nearshore to the South American continent, while the
 420 kinetically inhibited dissolved Mn must await microbially-catalyzed oxidation and thus persists farther
 421 offshore.

422 *Copper*- For the mixed layer, as with Mn, there is a relatively steep gradient in concentrations leaving
 423 the coastal zone, with a diminished gradient further offshore into the open ocean (Figure 6). At the depth
 424 of the PPZ there is also a steep decline in concentration nearshore, although the gradient is not as well-
 425 defined as in the mixed layer. Further offshore, there appears to be a small increase in Cu concentration

426

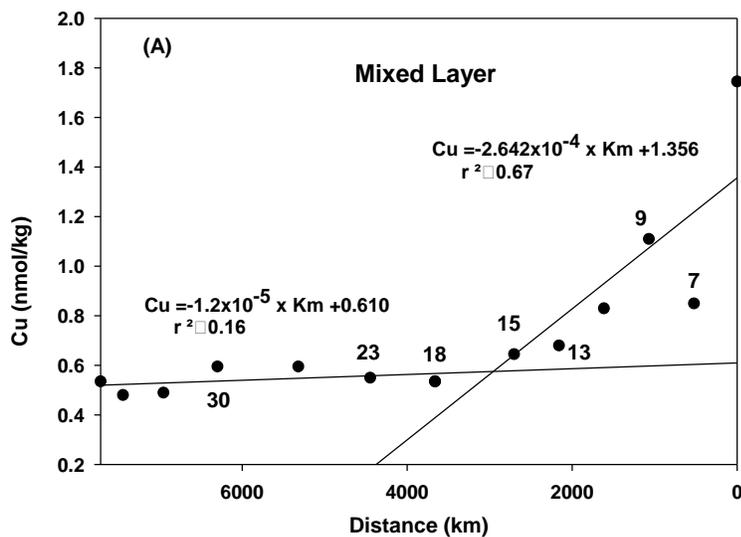
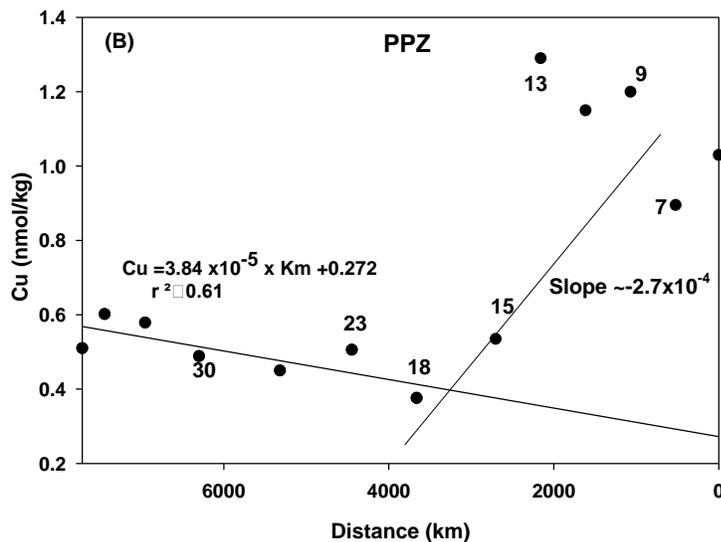


Figure 6.
 A) Dissolved Cu plotted against distance from station 1 for the mixed layer across the EPZT.
 B) Dissolved Cu plotted against distance from station 1 for the PPZ across the EPZT.



427

428 going westward, which may be related to the fact that the seaward transect is bearing somewhat
429 northwest, back above more Cu-rich subsurface waters.

430 *Lead* – As with Mn and Cu, within the mixed layer there is a relatively steep gradient in dissolved Pb
431 concentrations leaving the coastal zone, with a diminished gradient further offshore into the open ocean
432 (Figure 7). At the depth of the PPZ there is also a steep near-shore decline in Pb concentration although
433 the gradient is less well-defined.

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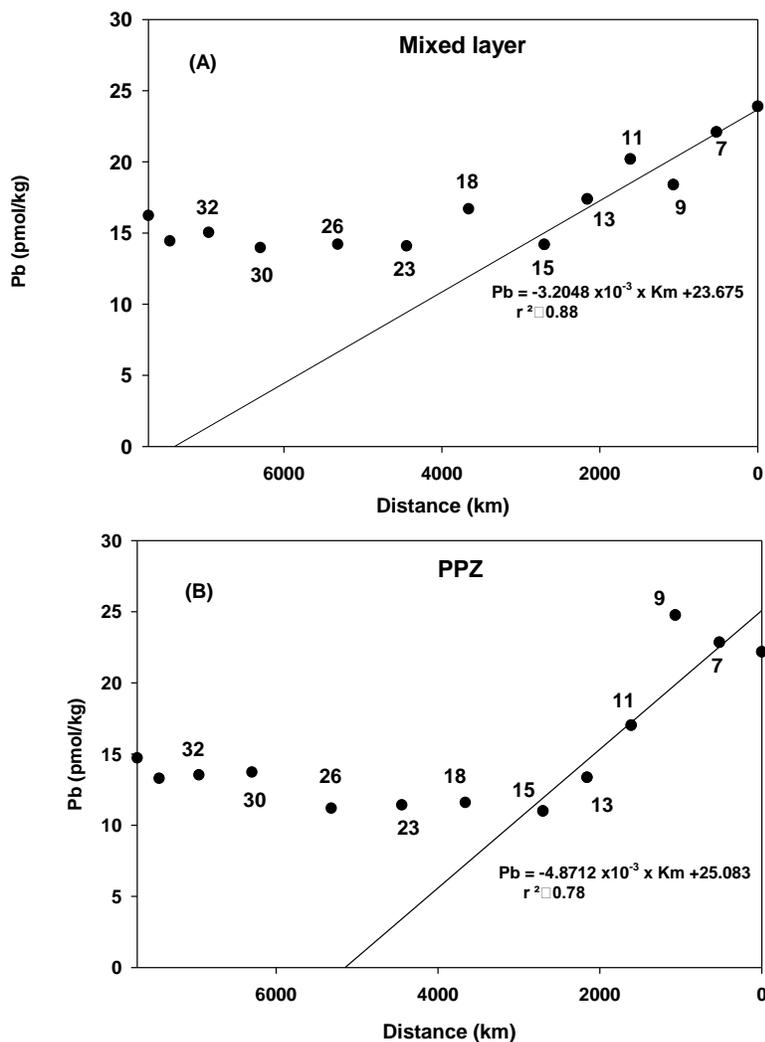


Figure 7.

A) Dissolved Pb plotted against distance from station 1 for the mixed layer across the EPZT.

B) Dissolved Pb plotted against distance from station 1 for the PPZ across the EPZT.

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438 *Cadmium* – Dissolved Cd concentrations within the mixed layer and the PPZ show a relatively steep
439 gradient in dissolved Cd concentrations leaving the coastal zone, with no discernible gradient offshore
440 into the open ocean (Supplemental Figure 3).

441 Vanadium - The dissolved V concentration field has no discernable horizontal gradient (Supplemental
 442 Figure 4).

Table 5a TE fluxes into the mixed layer																
sta	ML flux Fe nmol/m2/d				ML Flux Mn nmol/m2/d				ML Flux Cu nmol/m2/d				ML Flux Pb nmol/m2/d			
	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal
1	198	968	829	0	257	13024	16.48	0	367	3872	11.63	0	-3.2	105	1.69	0
7	0	117	829	0	-119	3627	16.48	253	109	1131	11.63	104	-0.29	29	1.69	1258
9	0.0	195	829	0	392	5070	16.48	289	-318	1545	11.63	119	-2.4	27.6	1.69	1438
11	8.4	2	829	0	122	82	16.48	245	-112	40.6	11.63	100	-17.6	0.8	1.69	1217
13	12.7	-12	829	0	38.1	-243.6	16.48	234	622	-79.2	11.63	96	1.3	-2	1.69	1161
15	34.8	15	829	0	0.0	237	16.48	406	4.4	87.8	11.63	167	-0.8	2	1.69	2018
18	-12.0	0	70.96	0	178	0	0.98	217	-6.0	0	1.35	89	-1.12	0	0.46	0
23	-40.1	0	73.3	0	64.1	0	1.01	33	0.0	0	1.39	4	-0.6	0	0.48	0
26	0.0	0	85.52	0	8.6	0	1.18	44	-4.1	0	1.62	5	0.07	0	0.56	0
30	-9.8	0	84.5	0	17.9	0	1.17	60	-4.2	0	1.6	7	-0.04	0	0.55	0
32	4.6	0	103.51	0	13.0	0	1.43	69	-1.1	0	1.96	8	-0.05	0	0.67	0
34	25.6	0	154.37	0	6.4	0	2.13	34	0.0	0	2.93	4	-0.44	0	1.00	0
36	42.3	0	198.31	0	43.0	0	2.74	45	-19.3	0	3.76	5	-0.1	0	1.29	0

Table 5b TE fluxes into the PPZ																
sta	PPZ Flux Fe nmol/m2/d				PPZ Flux Mn nmol/m2/d				PPZ Flux Cu nmol/m2/d				PPZ Flux Pb nmol/m2/d			
	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal	Diffusive	Upwelling	atmosphere (BULK)	Horizontal
1	144	462	829	0	-558	10230	16.48	0	252	3399	11.63	0	120	73.2	1.69	0
7	702	277.5	829	0	-2049	2002.5	16.48	405	72.5	671.25	11.63	158.0	2.80	17.1	1.69	2278
9	444	357	829	0	913	1849.6	16.48	471	383	816	11.63	119	-0.31	16.8	1.69	2658
11	50.0	0.35	829	0	71.9	5.4	16.48	514	-55.3	5.75	11.63	100	0.26	0.1	1.69	3026
13	84.1	-0.77	829	0	337	-8	16.48	559	-13.3	-9	11.63	96	3.28	-0.1	1.69	3349
15	109	2	829	0	-413	10	16.48	436	-22	5.35	11.63	277	-1.22	0.1	1.69	4017
18	-33.3	0	71.0	0	-229	0	0.98	259	2.8	0	1.35	246	0.76	0	0.46	0
23	12.0	0	73.3	0	-267	0	1.01	53.7	-2.0	0	1.39	4	-0.41	0	0.48	0
26	0.0	0	85.5	0	-21.5	0	1.18	62.7	-7.2	0	1.62	5.4	-0.09	0	0.56	0
30	-1.6	0	84.5	0	-4.8	0	1.17	92.1	-4.8	0	1.6	7.3	0.03	0	0.55	0
32	0.935	0	103.5	0	-6.1	0	1.43	80.3	-1.1	0	1.96	8.4	-0.05	0	0.67	0
34	5.12	0	154	0	-59.4	0	2.13	55.3	-10.6	0	2.93	4.1	-0.26	0	1.00	0
36	13.0	0	198	0	-32.3	0	2.74	65.0	-7.2	0	3.76	5.5	-0.17	0	1.29	0

443
 444 a. Horizontal fluxes based on an east-to-west mixed layer velocity of 0.1m/s.

445
 446 3.2. Residence Time Calculations.

447 The residence time of the TEs were calculated using eqns. 9 and 10, the flux data from Table 5,
 448 the aerosol solubility data of Table 3, and the TE inventories in Table 6. The results are presented in
 449 Table 7. Note that only the positive flux values in Table 5 are used to calculate the residence times with
 450 respect to input fluxes. We consider the residence time of each element within the mixed layer and PPZ,
 451 using solubilities based on both the DI and HAc leaching methods.

452 *Manganese-* There is not a significant difference between the solubility of aerosol Mn determined
 453 by the DI and HAc leach methods, and as discussed below, the atmospheric input of soluble Mn along
 454 this transect is minor compared to the other inputs. Consequently, the residence times based on
 455 solubilities determined by either method are nearly identical (Table 7).

456 For nearshore stations 1,7, and 9 the dissolved Mn residence times within the mixed layer range
 457 from 10 to 29 days and average 22 ± 10 days, requiring intense removal processes in the nearshore
 458 upwelling zone to keep the dissolved Mn distribution at steady-state. The Mn residence times within the

459 mixed layer for stations 11-18 range from 89 to 294 days, with an average of 188 ± 84 days. This is
460 comparable to the 0.4 y nearshore scavenging residence time derived by Landing and Bruland (1987) for
461 the Central California coast using a one-dimensional advection-diffusion approach and to the 0.22-1.8 y
462 residence time for the upper 150 m of the central North Pacific calculated by comparing surface
463 inventories to sediment trap fluxes (Martin & Knauer, 1980).

464 However, residence times based on the ^{234}Th -derived particulate export flux of Mn at 100m depth
465 along the EPZT transect ranged from 1.5 - 3 y, considerably longer than this (Black et al. 2019). This is
466 likely because the Mn flux determined by the ^{234}Th method cannot account for lateral removal of Mn
467 from upwelling zones which would lead to a lower apparent total Mn export flux. It has been suggested
468 for example that the export flux of POC determined by the ^{234}Th method underestimates primary
469 production (Kadko, 2017), as export production can become spatially decoupled from new production.
470 This is because a fraction of the newly produced organic material can be transported laterally before
471 leaving the euphotic zone, a process known to be particularly important in dynamic coastal upwelling
472 systems (Plattner et al., 2005). Here we calculate the input flux (upwelling, atmospheric deposition, and
473 diffusion) of Mn at station 1 to be $\sim 13300 \text{ nmol/m}^2/\text{d}$ (Table 5a) while Black et al., (2019) report an
474 export flux of only $1000 \text{ nmol/m}^2/\text{d}$. Clearly, there is a decoupling between surface production and export
475 flux at these eastern sites. These authors acknowledge that if net dissolved TE removal via offshore
476 transport exceeds the particulate TE export flux, then their residence times would be an upper limit. Their
477 residence time applies only to the specific removal process arising from the vertical export of particles.

478 For the offshore stations 23-36, dissolved Mn residence times increase to the range of 337 to 572
479 days with an average of 490 ± 150 days. This is comparable to the 1.9 y reported by Black et al. (2019)
480 for the gyre region of this study where decoupling of new production and export flux does not occur.
481 This estimate is shorter than the 19 y residence time presented in Landing and Bruland (1987) where the
482 lack of strong westward advection off the Central California coast into the North Pacific results in much
483 longer Mn residence times. Our results are consistent with Figure 4a, where the steeper gradient for the
484 nearshore stations suggests more rapid Mn removal within the upwelling environment. The longer Mn
485 residence time at the open ocean stations is comparable to the advective timescale of 580 days for the
486 transport 5000 km offshore to the central gyre (with a velocity of 0.1m/s) and is consistent with the
487 diminishment of the ^{228}Ra signal (Sanial et al., 2018).

488 Over the PPZ, nearshore (stations 1,7, and 9) residence times range from 20 to 111 days, with an
489 average of 79 ± 51 days. For the stations 11-18, residence times range from 297 to 757 days averaging
490 506 ± 197 days. For the gyre stations 23-36, residence times of the PPZ range from 1180 to 2970 days
491 with an average of 1780 ± 690 days. The average residence times of Mn for the open ocean mixed layer
492 (1.3 y) and the PPZ (5 y) are comparable to the lower end of the 5-20 year Mn residence times discussed
493 in Shiller (1997) and the 5-6 year range modeled by Wu et al (2014) for a depth of 200m (~depth of the

494 PPZ in this study) in the N. Pacific and the N. Atlantic based on Mn oxidation rates. The maximum PPZ
 495 residence times are more comparable to the 9.6-11 y residence times estimated by comparing dissolved
 496 Mn seawater inventories with aerosol fluxes (Jickells 1999).

497 As described in section 3.1, a value of 0.1 m/s has been chosen for the ocean horizontal velocity
 498 (hence the horizontal flux) calculations. The sensitivity of the residence times to this parameter is
 499 illustrated in Figure 8. For current speeds between 0.05 and 0.5 m/s (range of a factor of 10), the Mn
 500 residence times for the mixed layer (Figure 8a) of upwelling stations 1,7, and 9 range within a factor of
 501 only ≤ 1.2 as vertical transport terms dominate the input flux. For stations to the west (11-36), the
 502 residence time range is between a factor of 2.8-7.5 since the horizontal flux term has greater relative
 503 importance. As current speed increases the residence time decreases for all stations. For the PPZ (Figure
 504 8b), the Mn residence times for upwelling stations 1,7, and 9 range over a factor of ≤ 1.5 . For stations to
 505 the west (11-36) the range is between a factor of 2.3-8.4.

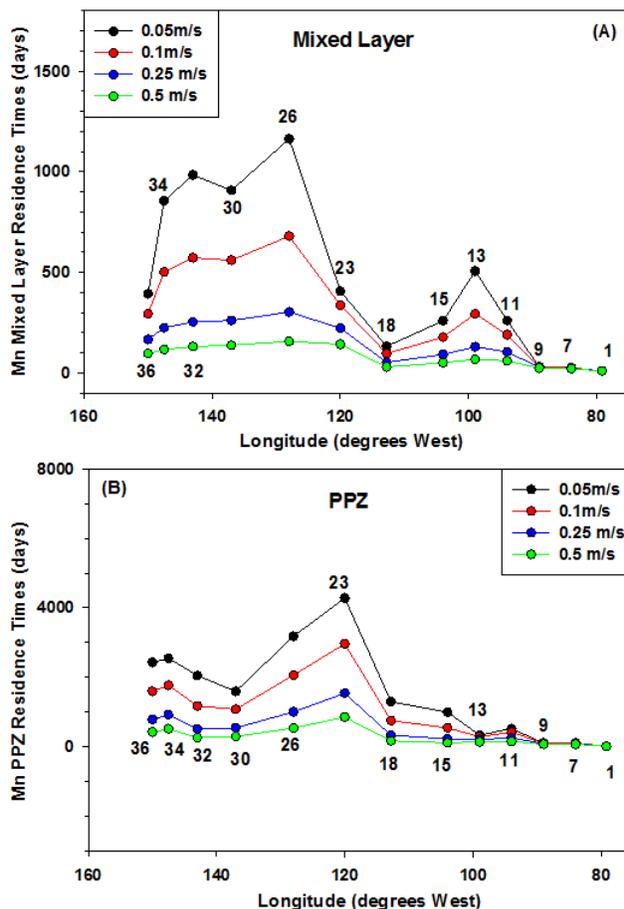


Figure 8.
 The relationship of the residence time of dissolved Mn to surface horizontal current velocity for A) the mixed layer and B) the PPZ.

506
 507
 508 Relative to other inputs the atmospheric input of soluble Mn is only of minor consequence in these
 509 calculations. Nearshore, where the atmospheric input of TEs are greatest, the upwelling Mn source is
 510 dominant. Within the gyre, the total input flux decreases and horizontal flux dominates Mn input into

511 these surface waters (Sanial et al., 2018; Black et al., 2019). Zheng et al. (2019) suggest sources other
512 than aerosols are more significant contributors to Mn in the N. Pacific as well, and Kadko et al (2019)
513 show that atmospheric deposition of Mn is insignificant relative to other inputs in the Arctic Ocean

514 *Iron-* The observed Fe characteristics differ from those of Mn in two distinct ways. First, the DI
515 and HAc aerosol Fe solubilities are appreciably different. Second, atmospheric Fe input is significant
516 relative to other input terms within the gyre. For nearshore stations 1, 7, and 9 the Fe residence times
517 within the mixed layer using the two leaching methods are similar because the upwelling input of Fe is
518 more significant than the atmospheric input. Using the DI water solubility, residence times range from 5
519 to 40 days, averaging 25 ± 18 days, and using the HAc solubility the range is from 4.6 to 22 days,
520 averaging 13.0 ± 8.6 days. These short residence times indicate intense removal processes in the
521 nearshore upwelling zone. Despite the relatively large dust input for these stations, the DI soluble
522 atmospheric Fe input is $< 4\%$, and the HAc soluble input is $< 60\%$ of the total input from all sources. A
523 similar conclusion was drawn by Buck et al. (2019) for these same stations and by Fitzsimmons et al.
524 (2016) for stations farther south but equally close to the South American continent.

525 Moving offshore, but still in the zone of upwelling (stations 11,13, and 15), the average residence
526 time increases to 140 ± 100 and 23 ± 8.5 days with the DI and HAc solubilities, respectively. Upwelling
527 is diminished in importance, and the atmospheric soluble input is $\sim 6\%$ of the total Fe input flux using the
528 DI solubility and as high as 94% with the HAc solubility. For the stratified mid-gyre stations 18-32
529 upward turbulent diffusion is limited (Fig 3) and the total Fe input is relatively small compared to the
530 inshore stations. The choice of solubility in the residence time calculation is therefore critical as the
531 atmospheric delivery of Fe becomes the dominant input term. The resulting average residence time using
532 the DI solubility ~ 11 y which is much longer than the 290 ± 110 days based on the HAc solubility. The
533 shorter HAc-based residence time is more consistent with most surface dissolved Fe residence times
534 reported in the literature (e.g. Jickells et al. 1999; Croot et al., 2004; Boyle et al., 2005; Bergquist and
535 Boyle 2006; Hayes et al. 2015) except for that reported in the Arctic (Kadko et al., 2019) where low
536 particle scavenging likely occurs.

537 As discussed elsewhere (e.g., Aguilar-Islas et al., 2010; Sholkovitz et al. 2012), provenance, more
538 than leaching methodology, determines the fractional solubility of aerosol Fe, with anthropogenic Fe
539 being more readily soluble relative to mineral Fe. In the gyre area of this region characterized by minimal
540 mineral dust deposition, the soluble Fe is likely dominated by highly soluble anthropogenic or pyrogenic
541 sources (Conway et al., 2019; Ito et al., 2019). Higher solubilities obtained from the HAc method might
542 be more representative of the actual solubility of aerosol Fe in seawater, as the heating step, reducing
543 conditions, and lower pH of the HAc method may artificially account for time-dependent Fe dissolution
544 pathways, such as those supported by naturally occurring organic Fe-binding ligands in seawater, and the

545 conditions particles experience during passage through a zooplankton gut or inside a fecal pellet or
546 organic aggregate (Shelley et al., 2018).

547 In the westernmost stations, moving out of the gyre, atmospheric input increases, and stratification
548 is relaxed resulting in enhanced turbulent diffusion. These factors contribute to shorter dissolved Fe
549 residence times for stations 34 and 36. Using the DI solubility, these are 113 and 71 days, respectively,
550 and with the HAc solubility, 52 and 37 days.

551 For the PPZ, using the DI solubility, the Fe residence times for the nearshore stations 1,7, and 9
552 range from 11-29 days, and average 19 ± 9 days, and using the HAc solubility range from 9 to 24 days,
553 and average 16 ± 8 days. These residence times are similar because of the dominance of the upwelling
554 and diffusive terms compared to the atmospheric input. As with the mixed layer calculation, the residence
555 time increases for stations 11, 13, and 15, averaging 153 ± 6 days using the DI solubility and 53 ± 13 days
556 using the HAc solubility. For the stratified mid-gyre stations 18-32 the average residence time increases
557 to 30 ± 34 y using the DI solubility and to 2.3 ± 1.2 y using the HAc solubility. Moving out of the gyre,
558 the residence times for stations 34 and 36 are 6.6 and 4.4 y, respectively using the DI solubility. Both
559 stations have a residence times of 1.2 y with the HAc leach. As discussed above, the HAc solubilities
560 appear to be more appropriate for Fe in these calculations.

561 *Copper*-The HAc solubility is ~twice that based on the DI leach but the soluble atmospheric flux
562 across the transect is relatively small compared to other inputs into the mixed layer, contributing < 1% to
563 16% of the total Cu input into the mixed layer using the DI solubility, and < 1% to 31% with the HAc
564 solubility, increasing in both cases towards the western edge of the transect. Thus the choice of aerosol
565 leach solubility is not critical in these calculations.

566 The average mixed layer residence time of Cu for nearshore stations 1,7, and 9 is 25 ± 11 days
567 using both the DI and HAc solubilities. Moving offshore, but still in the zone of upwelling (stations 11,
568 13, and 15), the average residence time increases to 0.50 ± 0.37 y for both solubilities. For the stratified
569 mid-gyre stations 18-32 the average residence time increases to 11.5 ± 6.2 y using the DI solubility which
570 is not significantly greater than the 10.4 ± 5.6 y using the HAc solubility. The increase of residence time
571 within the gyre reflects diminished input from upward diffusion and a smaller advective input term
572 (smaller gradient, Figure 6a). For stations 34 and 36, just outside of the gyre to the west, the residence
573 time based on the DI leach is 11.4 ± 1.35 y which again is not significantly different than the 9.3 ± 1.1 y
574 based on the HAc solubility. For comparison, Bruland (1980) derived a dissolved Cu surface residence
575 time with respect to atmospheric deposition of 10 years, and Boyle et al. (1977) estimated 2.1-50 years
576 with respect to scavenging, both for surface waters of the North Pacific.

577 For the PPZ, there is generally a similar pattern, with the nearshore stations 1, 7, and 9 manifesting
578 an average Cu residence time of 66 ± 40 days, using both solubilities. The residence times increase to an
579 average of 2.5 ± 1.3 y for offshore stations 11, 13, and 15, again using both solubilities. For stations 18-

580 36, using the DI solubility, the residence times range from 0.9-62 y, averaging 38 ± 22 y. For the HAc
 581 solubility, the residence times range from 0.9-56 y, averaging 33 ± 20 y.

582 The upwelling of Cu at stations 1, 7 and 9 is the largest component of Cu flux anywhere along the
 583 EPZT transect. It has been reported that the stability constant for Cu-binding ligands at these nearshore
 584 stations is an order of magnitude greater than those of stations further offshore (Boiteau et al., 2016)
 585 suggesting the possibility that organisms have upregulated genes to produce these high-K ligands to
 586 protect against Cu toxicity associated with the high rate of Cu supply by upwelling.

587 *Lead-* The concentrations of aerosol Pb across the EPZT were too low to allow reliable determination
 588 of solubilities. We use instead the aerosol solubilities of marine background samples presented in Shelley
 589 et al. (2018). Average DI and HAc solubilities were respectively, 33% and 63%. The average mixed layer
 590 residence time of Pb for nearshore stations 1-15 was 2 ± 3.5 days using both the DI and HAc solubilities.
 591 These very short residence times are indicative of intense scavenging removal processes in the nearshore
 592 upwelling zone as observed for other TEs discussed here. Upwelling and horizontal advection were the
 593 dominant input components for these stations. Moving out to the open ocean gyre stations 18-36, the total
 594 input of Pb to the upper ocean was very small compared to the inshore stations (Table 5) with atmospheric
 595 input dominant. With the HAc solubility, average residence times were 5.2 ± 2.1 y which is consistent with
 596 estimates of 2-3 y presented elsewhere (Bruland, 1980; Nozaki et al., 1976). With the DI solubility the
 597 average residence time was 9.8 ± 5.1 y.

598

Table 6. Mixed layer and PPZ TE inventories (nmol/m²)								599
station	Fe ML	Fe PPZ	Mn ML	Mn PPZ	Cu ML	Cu PPZ	Pb ML	Pb PPZ
1	6145	10863	131375	202680	52137	73867	972	1518
7	3796	10608	124786	269037	41167	84228	950	2023
9	8290	23901	169883	341952	53471	112356	990	2225
11	3155	9091	86439	250776	43451	132174	740	2180
13	3785	15264	81531	267501	27792	122892	650	2325
15	7646	18666	115687	248070	55085	107899	1090	2280
18	2635	17193	37990	196395	85932	21609	645	2380
23	4806	12776	32916	160501	24723	100809	549	2207
26	8585	26786	36231	130590	32650	95026	740	2246
30	3970	8327	43677	100201	34758	68028	986	1913
32	8575	18017	47165	95115	44667	85736	1195	2224
34	3145	17436	20706	99552	18917	92410	590	2523
36	3218	25217	26207	105635	29501	98935	817	2677

612

613 For the PPZ, there is generally a similar pattern, with the residence time of Pb for nearshore stations
 614 1-15 manifesting an average of 1.9 ± 2.9 days, using both solubilities. The residence times increase to an

Table 7. Trace Element Residence Times (days)																
Station	Mixed Layer ^b								PPZ ^b							
	Fe		Mn ^a		Cu ^a		Pb ^a		Fe		Mn ^a		Cu ^a		Pb ^a	
	DI	HAc	DI	HAc	DI	HAc	DI	HAc	DI	HAc	DI	HAc	DI	HAc	DI	HAc
1	5.2	4.6	9.9	9.9	12.3	12.3	9.2	9.2	17.6	13.7	19.8	19.8	20.2	20.2	9.2	9.2
7	29.5	13	32.1	32.1	30.6	30.5	0.8	0.8	10.7	9.1	111	111	93	93	0.8	0.7
9	40.1	22	29.4	29.5	32.1	32.0	0.7	0.7	29.4	24.2	106	105	85	85	0.7	0.7
11	143	16	190	189	301	292	0.6	0.6	147	39	420	419	1209	1162	0.6	0.6
13	156	20	294	292	38.5	38.3	0.6	0.6	159	57	297	296	1241	1188	0.6	0.6
15	125	33	178	178	210	207	0.6	0.6	153	63	550	547	377	372	0.6	0.5
18	2652	166	89	89	242	241	4249	2226	17307	1084	757	756	344	344	4249	2226
23	4683	293	337	337	5583	4988	3466	1815	981	450	2968	2962	22763	20337	3466	1815
26	7170	449	680	678	5591	5057	2882	1742	22370	1400	2068	2064	16270	20228	2882	1742
30	3356	210	560	559	4490	4162	5433	2846	7037	441	1083	1082	8852	8187	5433	2846
32	1418	309	572	572	4996	4611	5405	2831	7670	750	1177	1174	9589	8852	5405	2831
34	113	52	502	499	3815	3115	1788	937	2400	440	1774	1767	18636	15218	1788	937
36	71.4	37	294	293	4513	3704	1919	1007	1600	440	1600	1593	15136	12420	1919	1005

615 a. Based on horizontal mixed layer velocity of 0.1 m/s

616 b. In each column residence times are shown based on the DI and HAc solubility.

617

618 average of 13.2 ± 4.9 y for offshore stations 18-36 using the HAc solubility and 24.3 ± 10.5 y using the DI
619 solubility.

620 *Cadmium and Vanadium*- The relevant information for these elements are shown in Supplemental
621 Tables 1-3.

622

623 4.0 Conclusions

624 The TE fluxes from atmospheric deposition, vertical mixing, and upwelling into the mixed layer
625 and the PPZ along the GEOTRACES EPZT were evaluated with ⁷Be-based methods developed in earlier
626 works. Horizontal advection velocity from east to west, driven by the South Equatorial Current, was
627 estimated from literature values and when horizontal gradients in the TE distributions were observed,
628 fluxes due to horizontal advection were determined. Atmospheric deposition is lowest in the central gyre,
629 with higher fluxes to the east due to large near-shore aerosol TE loads, and higher to the west due to
630 higher deposition velocities (V_b) driven by greater precipitation. The ⁷Be-derived K_z values range from
631 2.5 to 39 m²/d (0.29×10^{-4} to 4.5×10^{-4} m²/s) with higher values generally within the nearshore upwelling
632 region and the lowest values within the stratified central gyre. Mn displayed a well-defined gradient
633 extending from the nearshore stations into the central gyre. Even with a modest choice of horizontal
634 velocity the advective term was a major component of the total input flux, particularly within the central
635 gyre. Relative to other inputs the atmospheric input of soluble Mn was minor. Unlike manganese, there

636 was no discernable horizontal gradient in the Fe concentration field and therefore there is no horizontal
637 component of flux. Nearshore removal processes were more intense for dissolved Fe than for dissolved
638 Mn and as a result, dissolved Mn remained elevated much farther offshore than dissolved Fe. For the
639 stratified mid-ocean gyre stations upward turbulent diffusion of Fe was limited and atmospheric
640 deposition became dominant, although the total input from all sources was relatively small compared to
641 the inshore stations. Aerosol Fe solubility determined by a 25% acetic acid leach with hydroxylamine
642 hydrochloride was much greater than that derived from a leach using ultrapure deionized water. This led
643 to significant differences in the residence time of Fe calculated for the mid-ocean gyre depending on the
644 aerosol fractional solubility chosen for the calculation. For Cu, Cd, and Pb, as with Mn, within the
645 mixed layer there is a relatively steep gradient in concentrations leaving the coastal zone.

646 The residence times of the TEs calculated with the measured water column inventories and the
647 input fluxes discussed above reflected the significantly different physical and geochemical conditions
648 encountered across the zonal transect. Generally, each element displayed relatively short (days-weeks)
649 residence times within the nearshore region of intense upwelling, reflecting large input terms and rapid
650 removal. Moving offshore, total input fluxes decreased and the residence times of the TEs increased
651 markedly until the western edge of the transect. There, relaxation of ocean stratification permitted greater
652 upward turbulent flux and greater rainfall led to greater atmospheric input of the trace elements. These
653 results suggest that one size does not fit all when considering ocean residence times as there is a range of
654 scavenging removal pathways for different elements as well as natural gradients in oceanic fluxes.
655 Regional difference should be expected.

656 As discussed in Twining and Baines (2013), trace metals can influence the growth and structure
657 of natural phytoplankton communities and, conversely, the composition and structure of phytoplankton
658 communities influence the distribution of metals in the ocean. The metal contents of phytoplankton reflect
659 biochemical demands and environmental availability. The ambient trace metal distributions reflect the
660 complex interplay between external sources, microbial uptake and recycling, and physical mixing.
661 Studies of the sources and residence times of these elements across diverse oceanic settings, as discussed
662 here, will provide further insight into the relationship between trace element abundance and
663 phytoplankton community structure.

664

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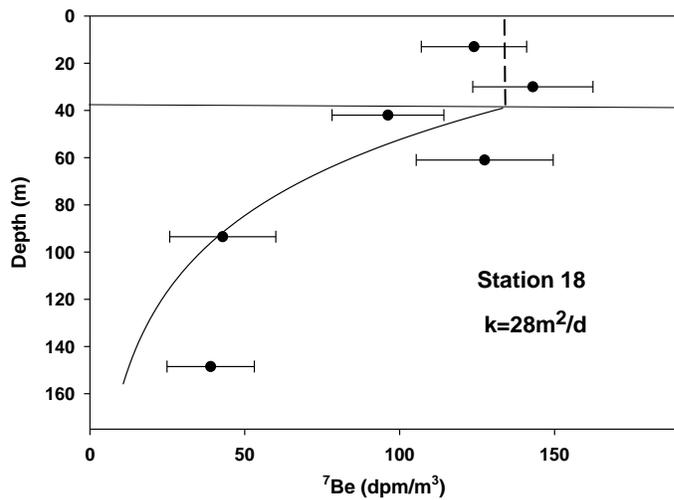
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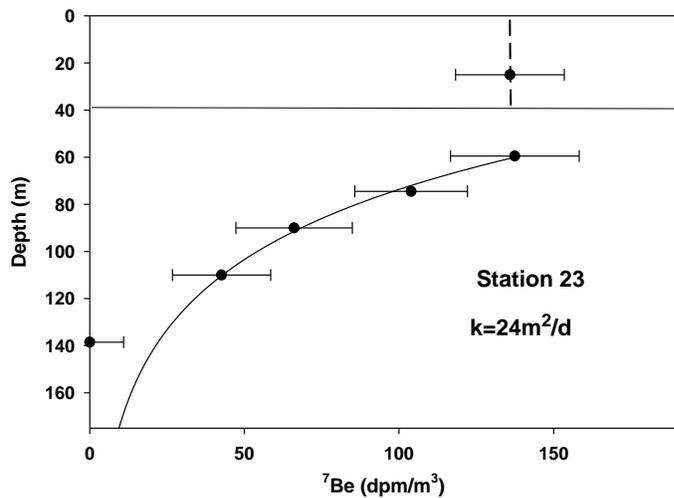
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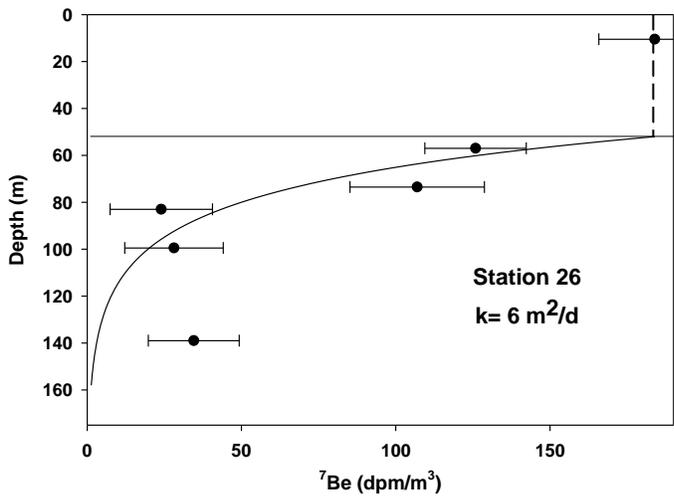
986 Supplemental Figure 1. ^7Be profiles and plot of eqn. 5 for stations 26-36. The derived K_z is indicated. The
987 horizontal line indicates base of the mixed layer. The vertical dashed line indicates mixed layer ^7Be
988 activity.



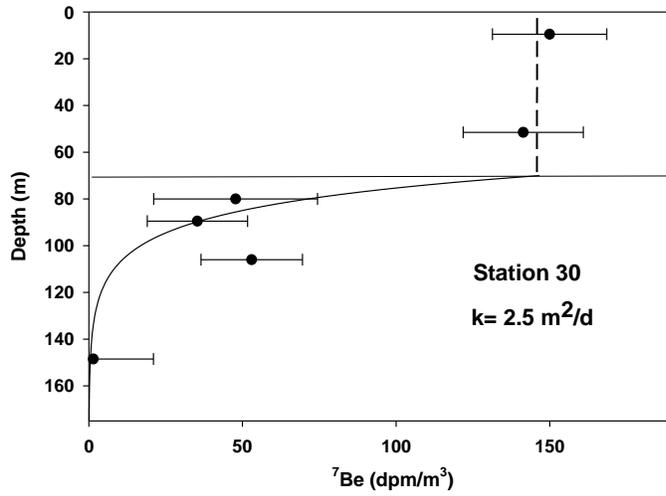
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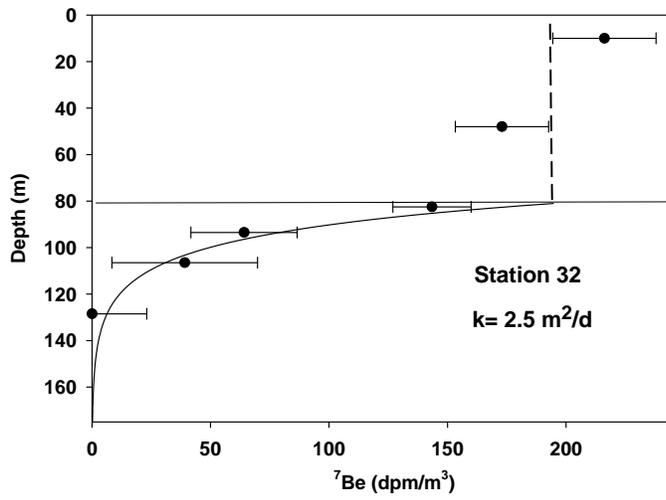
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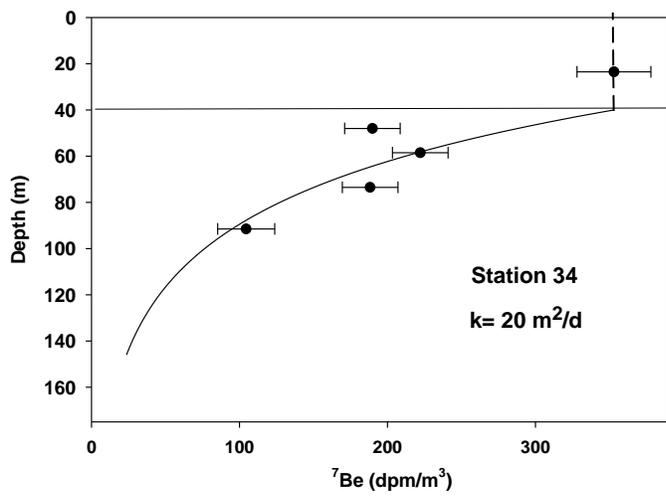


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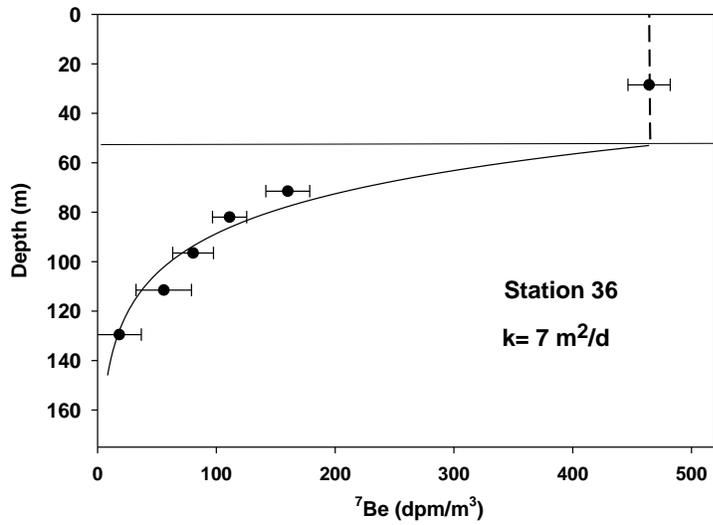


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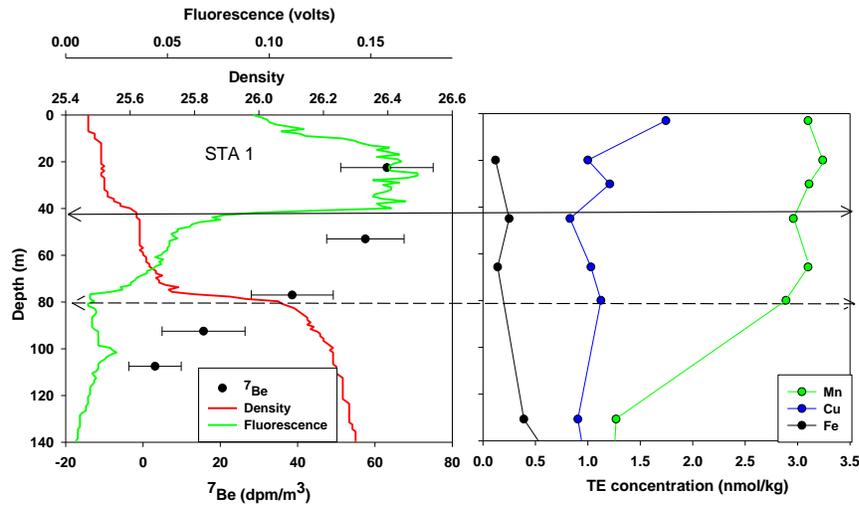
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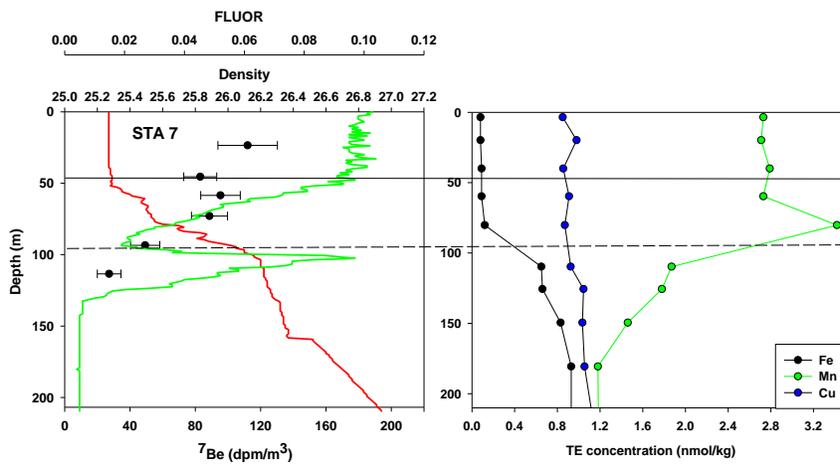
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999 Supplemental Figure 2

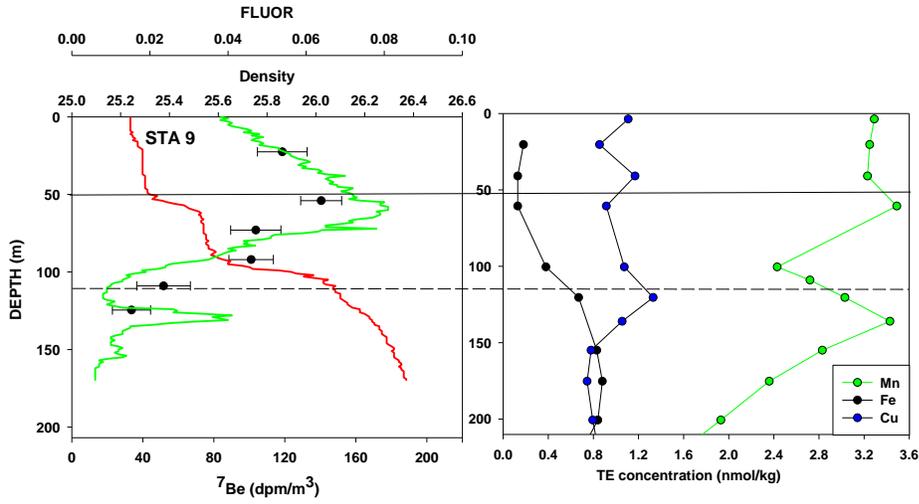
1000 Left: ^7Be , density, and fluorescence profiles. Right: Selected TE concentration profiles. The solid
 1001 horizontal line indicates the mixed layer depth. The dashed horizontal line indicates the depth of the PPZ.



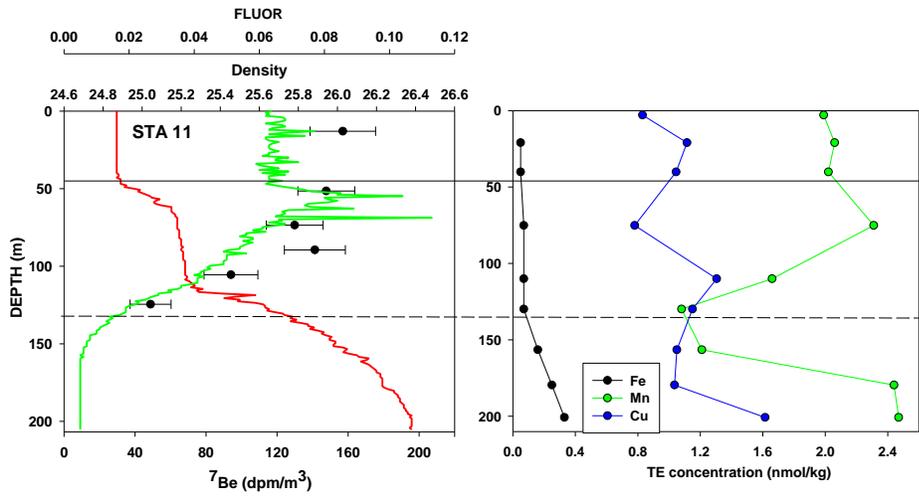
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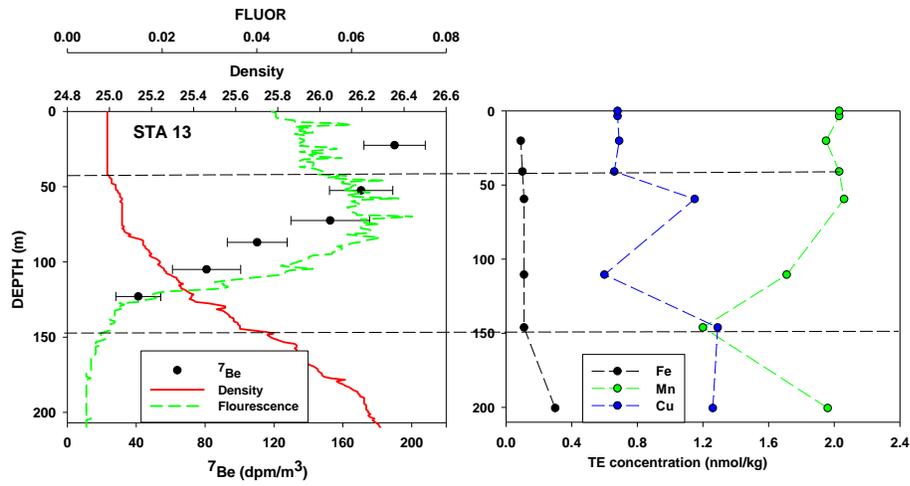
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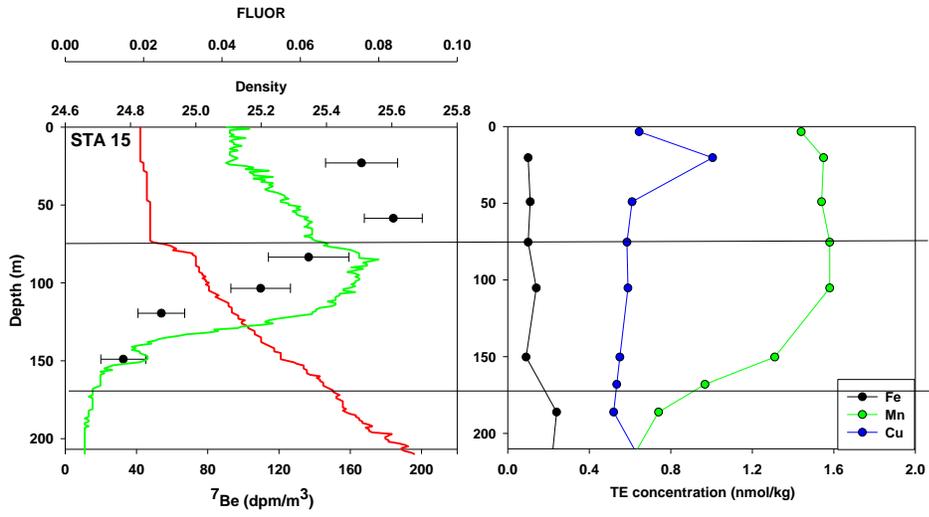
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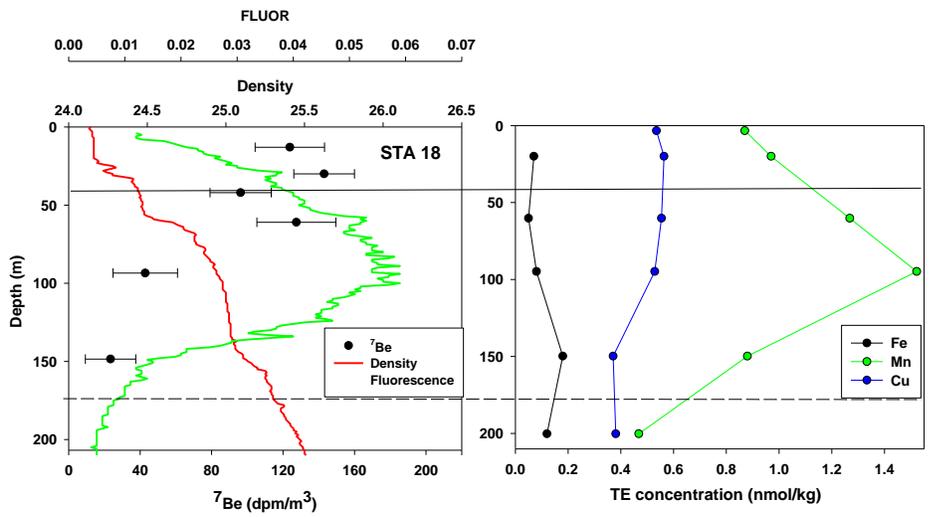
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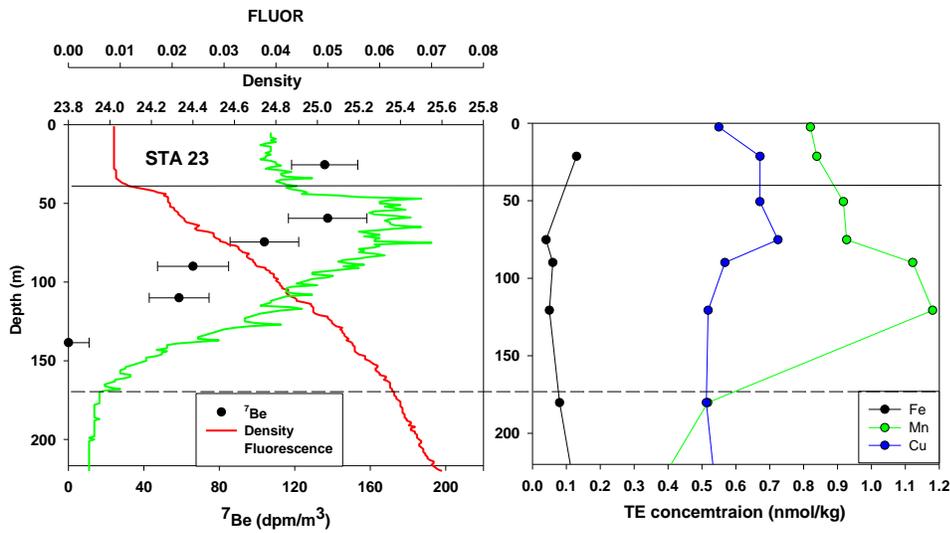
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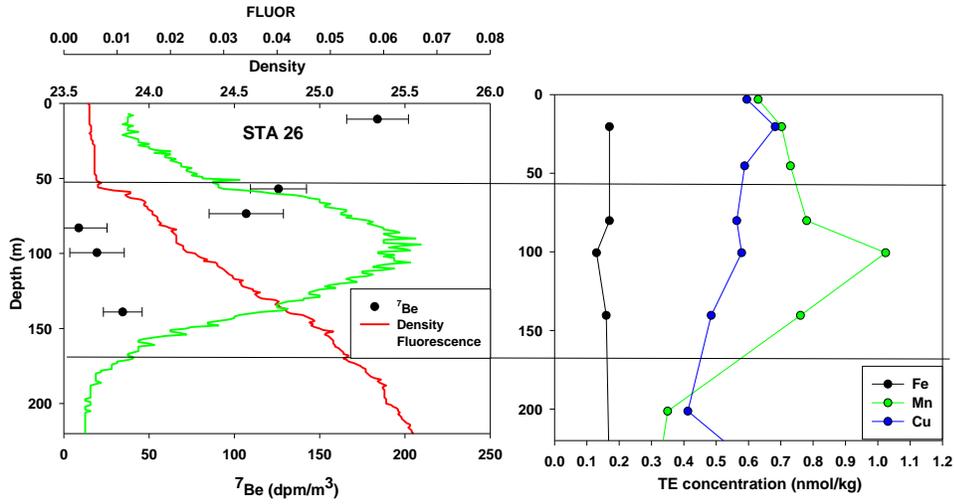
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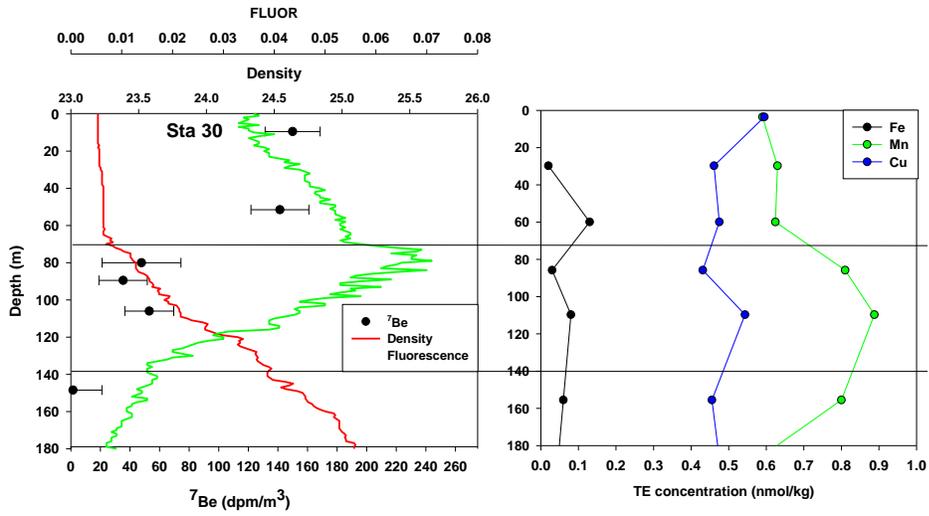
1008



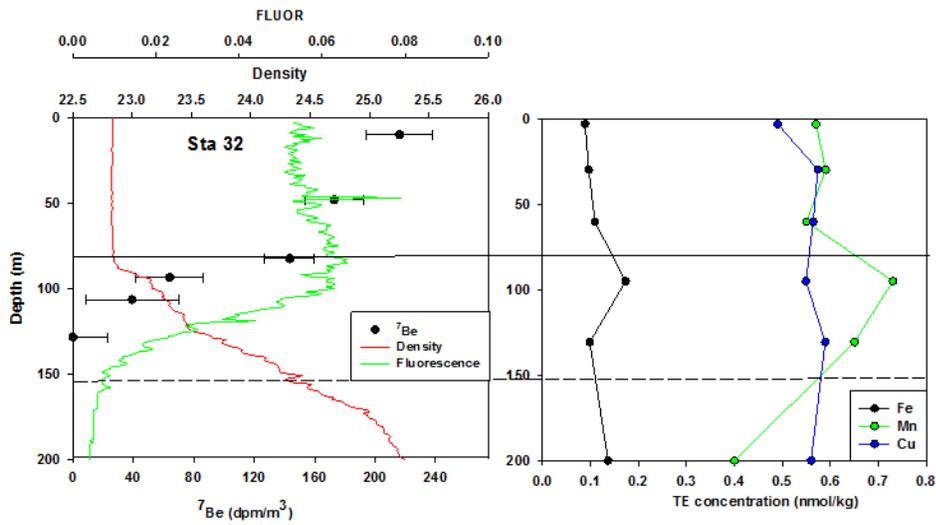
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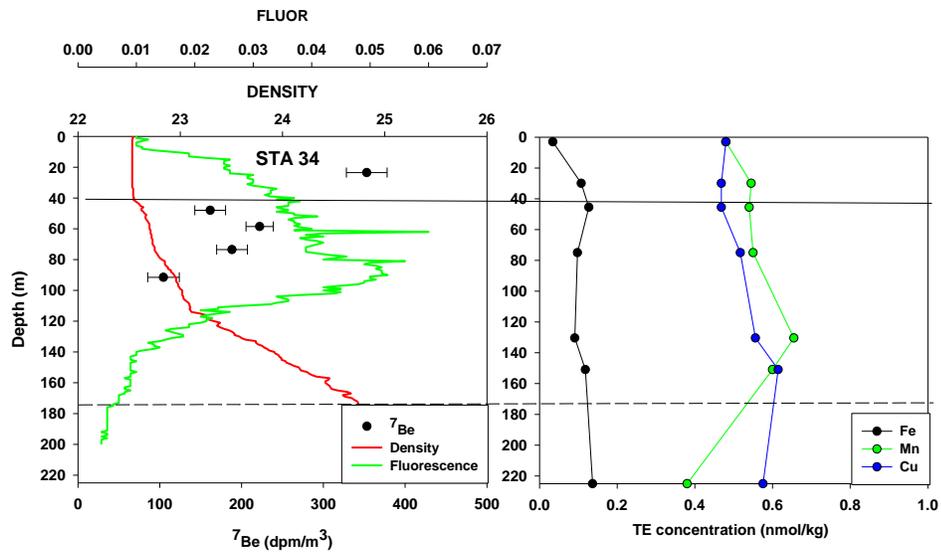
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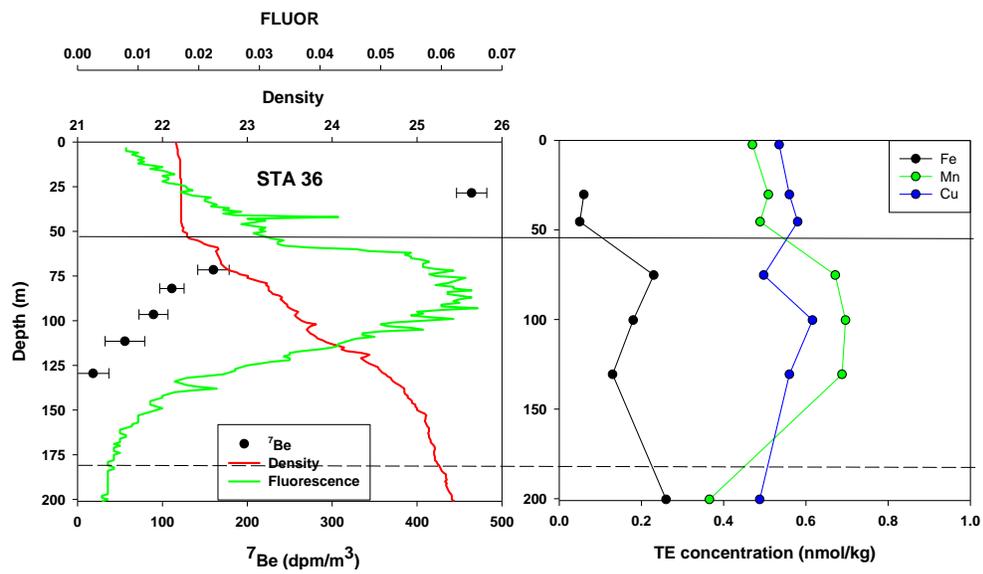
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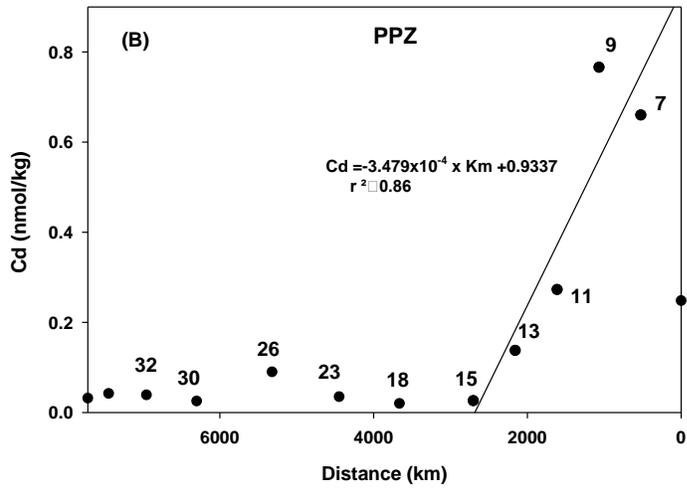
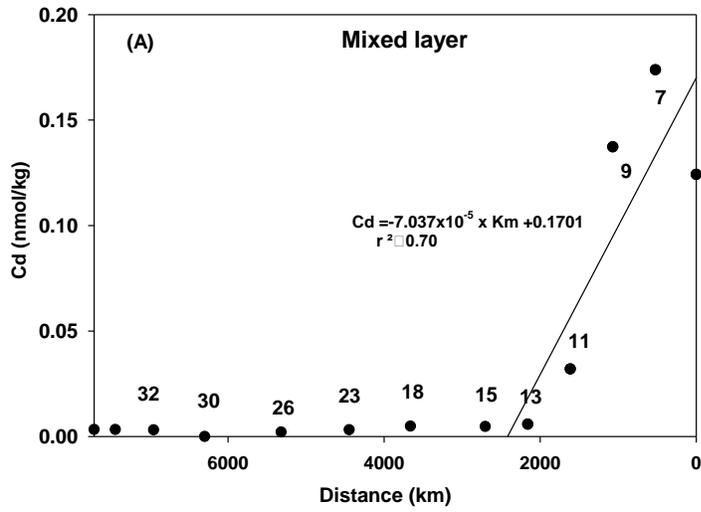


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Supplemental Figure 3.

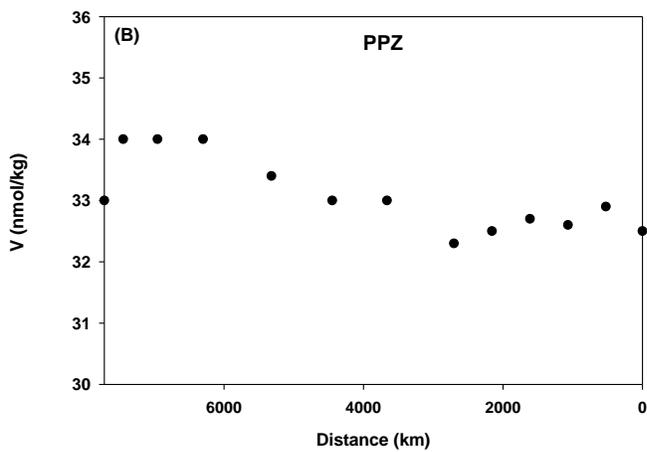
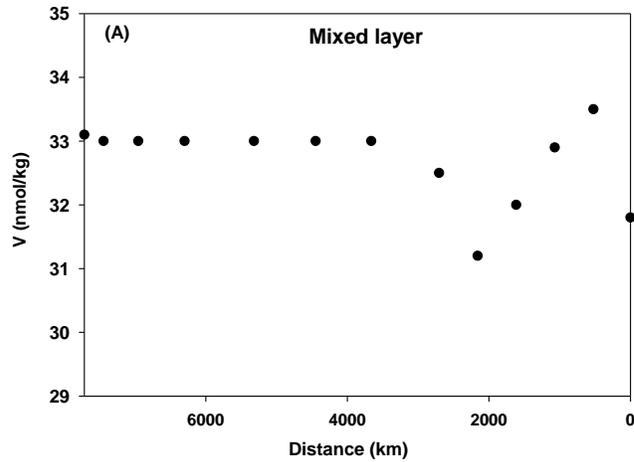
A) Dissolved Cd plotted against distance from station 1 for the mixed layer across the EPZT.

B) Dissolved Cd plotted against distance from station 1 for the PPZ across the EPZT.

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Supplemental Figure 4.

A) Dissolved V plotted against distance from station 1 for the mixed layer across the EPZT.

B) Dissolved V plotted against distance from station 1 for the PPZ across the EPZT.

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Supplemental Table 1a. Cd and V fluxes into the mixed layer									
STATION	Cd ML Flux (nmol/m ² /d)					V ML Flux (nmol/m ² /d)			
	Diffusive	Upwelling	Atmosphere (BULK)	Horizontal		Diffusive	Upwelling	Atmosphere (BULK)	Horizontal
1	154	682	0.45	0		-182	143440	8.21	0
7	-25.4	161.2	0.45	28		-593	43160	8.21	0
9	138	319.5	0.45	32		-828	48900	8.21	0
11	19.4	1.19	0.45	27		250	1280	8.21	0
13	-1.0	-0.81	0.45	26		1015	-3900	8.21	0
15	2.4	0.44	0.45	44		0.0	4950	8.21	0
18	-1.1	0	0.03	24		0.0	0	0.35	0
23	7.9	0	0.04	0		0.0	0	0.36	0
26	-3.4	0	0.04	0		170	0	0.42	0
30	0.2	0	0.04	0		0.0	0	0.42	0
32	-0.4	0	0.05	0		142.5	0	0.51	0
34	-2.2	0	0.07	0		0.0	0	0.77	0
36	1.0	0	0.10	0		378	0	0.98	0

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Supplemental Table 1b. Cd and V fluxes into the PPZ									
STATION	Cd PPZ Flux (nmol/m ² /d)					V PPZ Flux (nmol/m ² /d)			
	Diffusive	Upwelling	Atmosphere (BULK)	Horizontal		Diffusive	Upwelling	Atmosphere (BULK)	Horizontal
1	1971	818	0.45	0		1862	107250	8.21	0
7	308	495	0.45	101		0.0	24675	8.21	0
9	-85.8	521	0.45	119		-294.5	22170	8.21	0
11	88.2	1.365	0.45	157		-764	163.5	8.21	0
13	220.5	-1	0.45	184		44.1	-228	8.21	0
15	31.2	0	0.45	188		143.0	323	8.21	0
18	9.5	0	0.03	0.0		-1070	0	0.35	0
23	8.6	0	0.04	0.0		0.0	0	0.36	0
26	5.6	0	0.04	0.0		111.6	0	0.42	0
30	-0.1	0	0.04	0.0		0.0	0	0.42	0
32	1.1	0	0.05	0.0		-53.3	0	0.51	0
34	5.6	0	0.07	0.0		-400.0	0	0.77	0
36	2.7	0	0.10	0.0		-98.0	0	0.98	0

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Supplemental Table 2. Mixed layer and PPZ Cd and V inventories (nmol/m ²)				
STATION	Cd ML inv	Cd PPZ inv	V ML inv	V PPZ inv
1	5720	10450	1363000	2131000
7	6140	20760	1510000	3106000
9	8290	38780	1704000	3590000
11	1250	10630	1412000	4200000
13	241	4340	1306000	4762000
15	206	1500	2450000	5491000
18	158	1090	1287000	5742000
23	238	3120	1287000	5639300
26	523	5060	1717000	5641000
30	328	1610	1977000	4568100
32	871	2710	2647000	5048000
34	194	2770	1320000	5847300
36	187	1840	1731000	5891000

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Supplemental Table 3. Mixed layer and PPZ Cd and V Residence Times								
	Mixed Layer Residence Times (d)				PPZ Residence Times (d)			
	Cd	Cd	V	V	Cd	Cd	V	V
STATION	HAc ^a	DI ^b	HAc ^c	DI	HAc ^a	DI ^d	HAc ^c	DI ^d
1	7	7	9.5	9.5	4	4	19.5	19.5
7	32	31	35.0	35.0	23	23	12.6	12.6
9	17	17	34.8	34.8	61	61	162	162
11	26	26	920	920	43	43	24800	24990
13	9	9	1280	1282	11	11	96470	99115
15	4.4	4.4	494	495	6.8	6.8	11650	11680
18	7	7	5631150	7463250	115	115	25123605	33297570
23	30	30	5474730	7255940	361	362	23988855	31793630
26	13620	20115	10084	10090	897	899	50240	50270
30	1376	1451	7208490	9553770	41971	62000	16656090	22075150
32	18150	26800	18530	18540	2361	2393	15157790	20089380
34	2887	4264	262525	347940	489	491	11629240	15412820
36	171	176	4570	4574	657	664	9205550	12200580

- 1030 a. Cd HAc solubility is 96%
1031 b. Cd DI solubility is 65%
1032 c. V HAc solubility is 65.3%
1033 d. V DI solubility is 49.3%