

From mass to structure: modified van Krevelen diagram and adjusted indexes for high-resolution mass data of organic matter

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Abstract

The traditional van Krevelen diagram, the nominal oxidation of organic carbon, and several aromaticity indexes generally considered C, H, O, N, P, and S factors, or just C, H, and O. However, these evaluated methods may provide incorrect interpretation for high-resolution mass data of organics containing halogen elements. Therefore, we adapted the van Krevelen diagram and the other indexes with halogen elements and verified with 2235 model organics. The results demonstrated that the modified van Krevelen diagram and the adjusted indexes were more accurate for the analysis of mass data, either in the compound classification or in the evaluation of oxidation state of organic matter.

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Dear Editor:

The development of high-resolution mass spectrometry (HRMS) has led to deeper insight into the molecular characterization of organic matter. However, a large amount of data obtained by HRMS cannot be immediately converted into specific substances, especially for Fourier transform ion cyclotron resonance mass spectrometry. Some methods have been developed to further explore the information obtained by HRMS by interpreting the data and predicting structures based on exact molecular weight and element constituent patterns.¹ Of these methods, the van Krevelen diagram is widely used to classify of organic matter by plotting atomic ratios of H/C and O/C in two-dimensional ordinations.² In addition, some index are employed to evaluate molecular characterizations, such as the nominal oxidation state of carbon (NOSC)³ and the aromaticity index (AI, AI_{mod} and X_c),^{4,5}. While the traditional van Krevelen diagram and the indexes were mainly targeted at natural organic matter (NOM), the calculation equations normally considered C, H, O, N, P, and S factors, or just C, H, and O. However, for organic matter, particularly environmental contaminants and artificial chemicals, halogen elements are important elements and their chemical and toxicological properties cannot be overlooked.⁶⁻⁸ Moreover, the application of HRMS in environmental pollutants and anthropogenic organics is increasing,^{8,9} and the ability to more accurately interpret large amounts of data is essential. Therefore, we adapted the van Krevelen diagram and the other indexes with halogen elements for the analysis of organic matter with complex constituents.

To verify the modified van Krevelen diagram and the adjusted indexes, chemical formulas of 2235 organics were evaluated, which originated from a web database of disinfection byproducts (<http://dbps.com.cn/#/main>). The results were compared and depicted as follows:

The van Krevelen diagram

The van Krevelen diagram is normally depicted as a two-dimensional plot using the molar ratio of hydrogen to carbon (H/C) as the ordinate and the molar oxygen-to-carbon ratio (O/C) as the abscissa.² The H/C

ratio separates compounds according to the degree of saturation, whereas O/C ratios separate compounds according to O classes.¹⁰ Halogens are generally bonded to organic matter by replacing hydrogen, so their contribution to valence bond saturation should be equivalent to that of hydrogen. For halogen-containing organics, if only the H/C ratio is considered, the material structure will be misjudged. For example, the H/C ratio of trichloromethane is 1, and it is classified as an unsaturated compound by the traditional van Krevelen diagram; however, trichloromethane is a typical saturated organic matter. When halogen is introduced to the calculation of the H/C ratio, H(X)/C (H(X) is the sum of H and halogen elements) of trichloromethane is 4, which is an apparent characteristic of saturated compounds. Fig. 1(a) and (b) show the traditional van Krevelen diagram with H/C versus O/C and the modified van Krevelen diagram with H(X)/C versus O/C, respectively. The distribution of organics clearly shifts upward in the diagram. In particular, compounds with H/C ratios lower than 0.5 are classified as condensed aromatic hydrocarbons,^{11,12} while almost no condensed aromatic compounds are found in the 141 compounds with H/C ratios lower than 0.5 in the database. These compounds are mostly multihalogen substituted organics with H(X)/C ranging from 0.67 to 4. The above results indicate that using the modified van Krevelen diagram plotted with H(X)/C versus O/C could provide a more accurate classification for the organic matter detected by HRMS.

The oxidation of organic carbon

Three indexes are used to evaluate the oxidation state of organic carbon, including Cox,³ NOSC,¹³ and CHO.¹¹ The equations of these three indexes are calculated as follows:

$$\text{Cox} = (2\text{O} + 2\text{S} - \text{H} + 3\text{N} - 5\text{P})/\text{C} \quad (1) \quad \text{NOSC} = -((-Z + 4\text{C} + \text{H} - 3\text{N} - 2\text{O} + 5\text{P} - 2\text{S})/\text{C}) + 4 \quad (2) \quad \text{CHO} = (2\text{O} - \text{H})/\text{C} \quad (3)$$

In these three equations, C, O, H, N, P, and S represent the stoichiometric numbers of the elements C, O, H, N, P and S, respectively. Z in Eqn. (2) is the net charge of organic matter. Cox and CHO are special cases of NOSC. When Z=0, Cox = NOSC, and when the stoichiometric numbers of the heteroatoms are zero, CHO = NOSC. However, even NOSC ignores the contribution of halogen to the oxidation state of organic carbon. According to the definition of NOSC and Cox,^{3,13} we recommend the equation used to adjust NOSC, which is depicted as NOSC': $\text{NOSC}' = -((-Z + 4\text{C} + \text{H} - 3\text{N} - 2\text{O} + 5\text{P} - 2\text{S} - \text{X})/\text{C}) + 4$. In the equation of NOSC', X is the sum of stoichiometric numbers of halogen elements (F, Cl, Br, I).

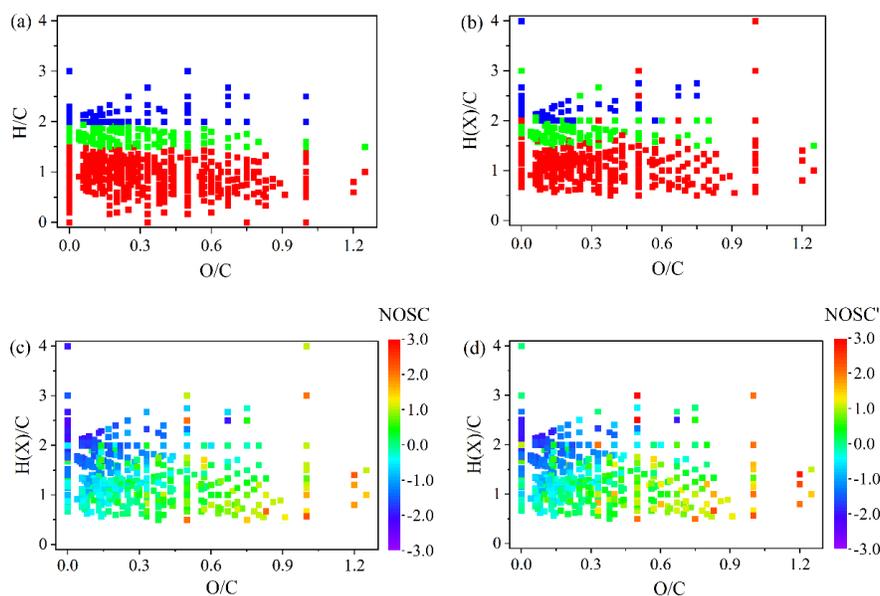


Fig. 1 (a) The traditional van Krevelen diagram; (b) The modified van Krevelen diagram; (c) The modified van Krevelen diagram associated with NOSC; (d) The modified van Krevelen diagram associated with NOSC'.

(In (a) and (b), the red dots symbol the compounds with $H/C < 1.5$, the green dots are the compounds with $1.5 \leq H/C < 2.0$, and the blue dots are the compounds with $H/C \geq 2.0$; NOSC: the nominal oxidation state of carbon; NOSC': the adjusted nominal oxidation state of carbon)

Fig. 1(c) and (d) demonstrate the modified van Krevelen diagrams associated with NOSC and NOSC', respectively. It can be seen that the compounds with $NOSC' > 0$ are slightly higher than those with $NOSC > 0$. Among the 2235 organics in the database, approximately 21.3% with $NOSC < 0$ in Fig. 1(c) change into that with $NOSC' \geq 0$ in Fig. 1(d). Generally, compounds with $NOSC > 0$ are more oxidized and less degradable, while those with $NOSC < 0$ are more likely degraded. When the impact of halogen is not considered, the oxidation state of organic carbon is underestimated, and the degradation potential of organic matter is overestimated.

Aromatic indexes

Three methods were used for calculating the aromaticity index, including AI, AI_{mod}⁴ and X_c.⁵ Of them, AI was suggested for compounds in which oxygen was mainly present as carbonyl functional groups; AI_{mod} was adjusted on the basis of AI when oxygen could also be bound with π -bonds in carboxyl groups;¹² and X_c was aimed at aromatic compounds with long alkylation.⁵ Considering the contribution of halogen elements to the saturation of organic matter, we adjusted these three indexes based on the original equations and put forward them as AI', AI_{mod}' and X_c', respectively. The equations used to calculate the tree adjusted indexes are as follows:

$$AI' = (1 + C - O - S - 0.5(N + P + H + X)) / (C - O - N - S - P)$$

$$AI_{mod}' = (1 + C - 0.5O - S - 0.5(N + P + H + X)) / (C - 0.5O - N - S - P)$$

$$Xc' = [3((1 + C - 0.5(H + X) + 0.5(N + P) - (O + S)) - 2)] / (1 + C - 0.5(H + X) + 0.5(N + P) - (O + S))$$

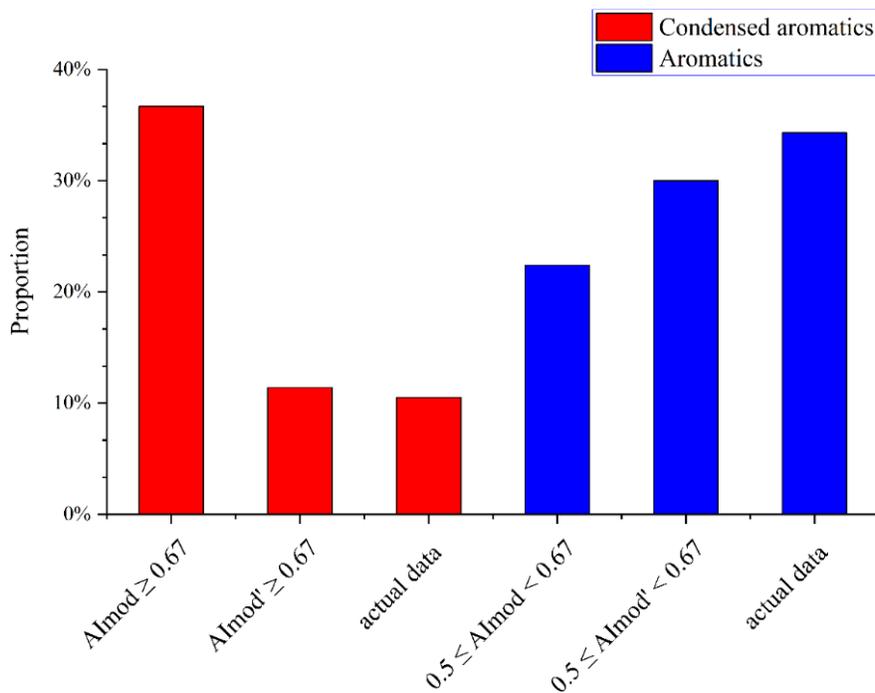


Fig. 2 Comparison of AI_{mod} and AI_{mod}' with actual data (organics with AI_{mod} or AI_{mod}' ≥ 0.67 are classified as condensed aromatics, and those with $0.5 \leq AI_{mod}$ or $0.5 \leq AI_{mod}' < 0.67$ are categorized as aromatics)

According to the structures of the 2235 organics, a comparison between AI_{mod} and AI_{mod}' was chosen to determine the validity of the adjusted aromatic indexes (as shown in Fig. 2). The proportions of condensed aromatics and aromatics predicted by AI_{mod}' are closer to the actual data than those predicted by AI_{mod}.

In summary, introducing halogen to the interpretation models of HRMS data for organic matter improves the accuracy. The modified van Krevelen diagram, NOSC and aromatic indexes are equivalent for multiple organic matter, including NOM, emerging environmental pollutants, and artificial products.

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